

Carbazole-based Bis-Imidazole Ligand-Involved Synthesis of Inorganic-Organic Hybrid Polyoxometalates as Electrochemical Sensors for Detecting Bromate and Efficient Catalysts for Selective Oxidation of Thioether

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Table S1 Selected bond lengths (Å) and angles (°) of complexes **1-3**.

Complex 1			
Ni(1)-O(7)	2.119(4)	Ni(1)-O(5)	2.191(4)
Ni(1)-N(10)	2.058(5)	Ni(1)-N(9)	2.056(5)
Ni(1)-N(8)	2.123(5)	Ni(1)-N(4)	2.079(5)
O(7)-Ni(1)-O(5)	71.61(14)	N(10)-Ni(1)-O(7)	94.09(16)
N(10)-Ni(1)-O(5)	165.68(17)	N(9)-Ni(1)-O(7)	87.83(17)
N(9)-Ni(1)-O(5)	89.21(18)	N(9)-Ni(1)-N(10)	89.5(2)
Complex 2			
O(2W)-Cu(1)	2.087(4)	O(8)-Cu(1)	2.303(3)
N(4)-Cu(1)	1.951(4)	N(4)-Cu(1)-O(2W)	91.57(17)
O(2W)-Cu(1)-O(8)	89.84(13)	N(4)-Cu(1)-O(8)	90.57(15)
Complex 3			
N(5)-Ni(1)	2.051(3)	N(1)-Ni(1)	2.063(4)
Ni(1)-O(1W)	2.091(4)	Ni(1)-O(2W)	2.111(3)
Ni(1)-O(5)	2.087(3)	Ni(1)-O(6)	2.083(3)
N(5)-Ni(1)-O(2W)	172.82(14)	N(5)-Ni(1)-O(5)	91.01(14)
O(5)-Ni(1)-O(1W)	173.74(15)	O(5)-Ni(1)-O(2W)	85.46(14)
N(5)-Ni(1)-N(1)	96.21(14)	N(5)-Ni(1)-O(1W)	94.96(16)

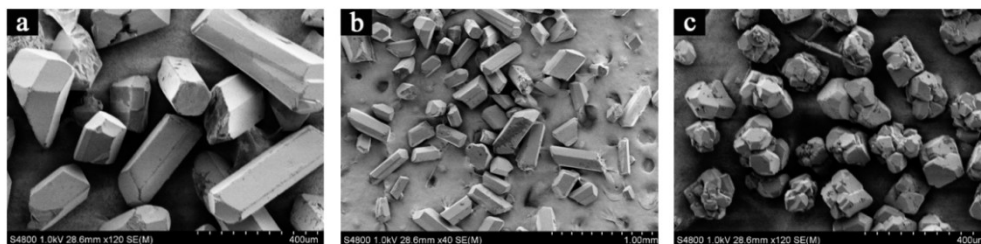


Fig. S1 The SEM images for complexes 1(a), 2(b) and 3(c).

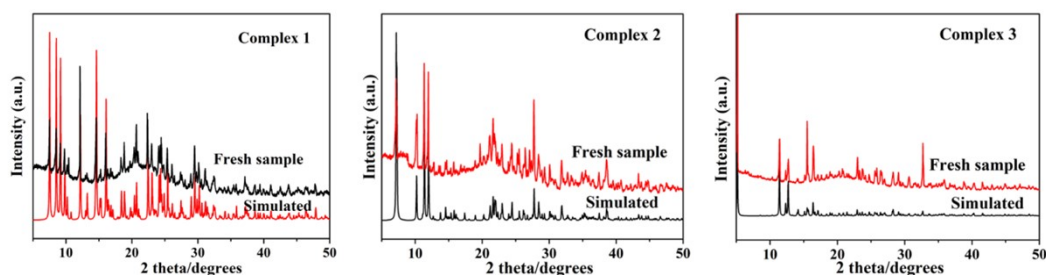


Fig. S2 The PXRD patterns for complexes 1-3.

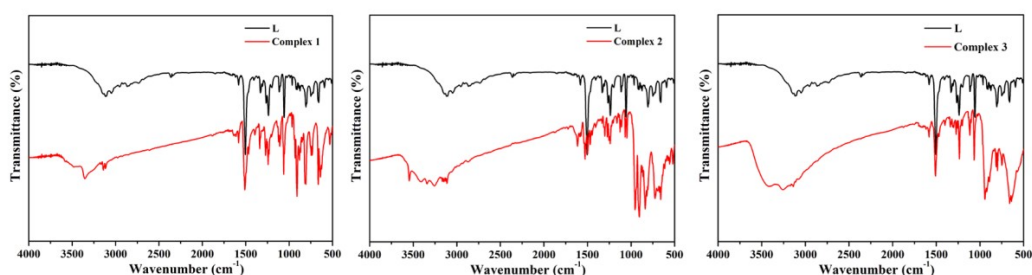


Fig. S3 IR spectra of complexes 1-3.

Table S2 The comparison of LODs of 1–3–CPEs and reported sensors.

Sensor	Concentration (M)	Coefficient of association	sensitivity ($\mu\text{A}\cdot\text{mM}^{-1}$)	LOD (μM)	ref
1–CPE	2×10^{-5} – 1×10^{-3}	0.998	8.58	0.315	this work
2–CPE	2×10^{-5} – 1×10^{-3}	0.999	27.61	0.098	this work
3–CPE	2×10^{-5} – 1×10^{-3}	0.999	0.54	0.551	this work
$\{[\text{Cu}_8(\text{H}_2\text{O})_6](\text{dpyh})_4(\alpha\text{-}\gamma\text{-Mo}_8\text{O}_{26})\cdot(\beta\text{-Mo}_8\text{O}_{26})\cdot 8.5\text{H}_2\text{O}\}$	1×10^{-5} – 1×10^{-4}	0.998	0.05	83.2	1
$[\text{CoL}^1(\beta\text{-Mo}_8\text{O}_{26})_{0.5}]$	1×10^{-5} – 4.2×10^{-3}	0.992	0.69	1503	2
$[\text{Co}(\text{H}_2\text{O})_2\text{L}^2(\beta\text{-Mo}_8\text{O}_{26})_{0.5}]$	1×10^{-5} – 9×10^{-4}	0.995	8.0	25.4	2
$[\text{Zn}(\text{HL}_3)_2(\beta\text{-Mo}_8\text{O}_{26})]$	1×10^{-5} – 5×10^{-3}	0.999	2.0	24	2

$[\text{Ag}_4\text{L}^a_5(\text{SiW}_{12}\text{O}_{40})]\cdot 28\text{H}_2\text{O}$	1.2×10^{-5} – 5.8×10^{-4}	0.991	12.7	5.61	3
$[\text{Ag}_3\text{L}^a_4(\text{PW}_{12}\text{O}_{40})]$	3.4×10^{-3} – 8×10^{-3}	0.997	101.7	16.9	3
$(\text{Ni}_4[(\text{P}_8\text{W}_{48}\text{O}_{184})(\text{WO}_2)]^{28-})$	1×10^{-4} – 1×10^{-3}	0.998	69.4	0.2	4
NENU-3/CC	5.0×10^{-5} – 5.6×10^{-3}	0.997	45.1	0.55	5
NENU-5/CC	1.5×10^{-5} – 3.8×10^{-5}	0.996	18.8	1.18	5
NENU-3/film	5.0×10^{-5} – 7.2×10^{-4}	0.998	0.02	12	6
Ag/PMO ₁₂ /PBz-modified electrode	1×10^{-3} – 7.5×10^{-3}	0.999	2.2	86.3	7
$[(\text{C}_3\text{H}_7)_2\text{-bim}]_2[\text{CdCl}_4]$	5×10^{-9} – 2×10^{-8}	0.998	0.10	0.01	8

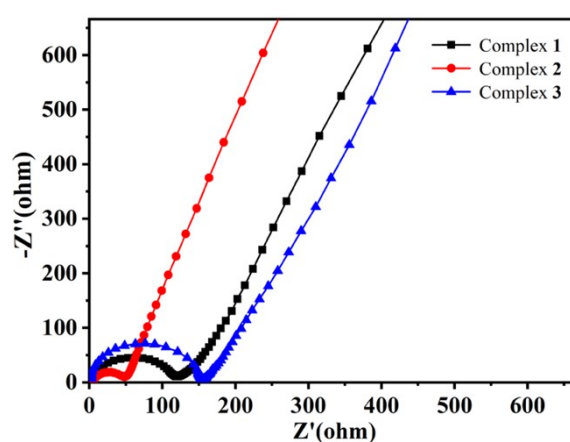


Fig. S4 The EIS spectra of 1–3–based electrodes

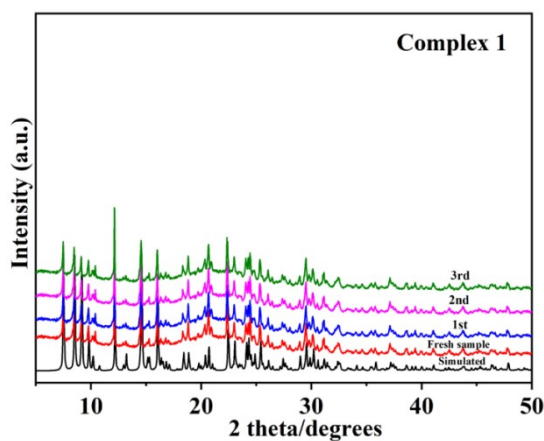


Fig. S5 The PXRD patterns of simulated and recycled after acting as catalyst of **1**.

References

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