

## Electronic Supplementary Information

for

### **A high-spin diradical dianion and its bridged chemically switchable single-molecule magnet**

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## Experimental Section

**General considerations.** All experiments were carried out under an atmosphere of dry nitrogen by using modified Schlenk line or glovebox techniques. Toluene, THF and n-hexane were freshly distilled over Na/benzophenone and degassed three times before using. Elemental analyses were performed on an Elementar Vario EL III instrument at Shanghai Institute of Organic Chemistry, the Chinese Academy of Sciences. EPR spectra were obtained on Bruker plus-6/1 X-band variable-temperature apparatus. Considering the  $^{57}\text{Fe}$  Mössbauer spectroscopic studied, the solid sample was the as-isolated complex, and it was placed into a Delrin Mössbauer sample cup for measurements and loaded under liquid nitrogen. The spectrum was recorded on a conventional spectrometer with alternating constant acceleration of the  $\gamma$ -source ( $^{57}\text{Co}/\text{Rh}$ , 1.8 GBq) at room temperature. Magnetic measurements were performed using a Quantum Design SQUID VMS magnetometer. For the single-crystal X-ray diffraction analyses, the data were collected on Bruker D8 CMOS detectors at 193 K. The structures were solved by direct methods and all refined on  $F^2$  with the SHELX-2018/3 software package. The positions of the H atoms were calculated and considered isotropically according to a riding model. Commercially available reagents were purchased from Energy Chemical and Alfa-Assar, and used as received.  $2,7\text{-}t\text{Bu}_2\text{-PTO}$ ,<sup>1</sup>  $\text{LMg-MgL}^2$  and  $\text{LFe}(\text{Tol})^3$  were synthesized according to the reported literatures.

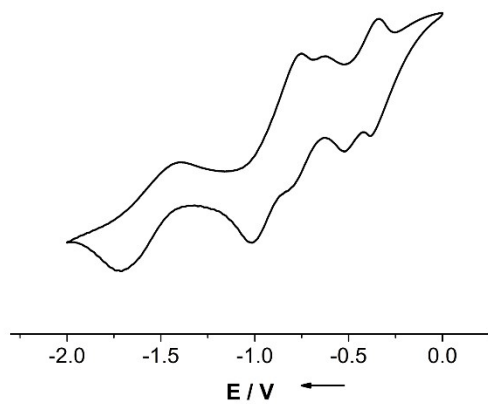
**Synthesis of 1.** A mixture of  $2,7\text{-}t\text{Bu}_2\text{-PTO}$  (136.4 mg, 0.36 mmol) and  $\text{LMg-MgL}$  (338.1 g, 0.38 mmol) in toluene (ca. 60 ml) was stirred at room temperature in an  $\text{N}_2$ -filled glovebox for 12 h to give a deep green solution. The solution was filtered, the filtrate was concentrated to ca. 10 mL and stored at  $-20\text{ }^\circ\text{C}$  for 24 h to afford deep green crystals of **1** (185.2 mg, 40.4%). M.p.  $> 300\text{ }^\circ\text{C}$ . Elemental analysis (%) Calcd for  $\text{C}_{82}\text{H}_{104}\text{Mg}_2\text{N}_4\text{O}_4 \cdot 2(\text{C}_7\text{H}_8)$ : C, 79.93; H, 8.38; N, 3.88; Found: C, 79.94; H, 8.41; N, 3.49.

**Synthesis of 2.** A mixture of  $2,7\text{-}t\text{Bu}_2\text{-PTO}$  (59.2 mg, 0.16 mmol) and  $\text{LFe}(\text{Tol})$  (188.33 mg, 0.33 mmol) in toluene (ca. 30 ml) was stirred at room temperature in an  $\text{N}_2$ -filled glovebox for 12 h to give a brownish-red solution. The solution was filtered, the filtrate was concentrated to ca. 10 mL and stored at  $-20\text{ }^\circ\text{C}$  for 24 h affording red crystals of **2** (90.1 mg, 43.1 %). Decomposed at  $289\text{ }^\circ\text{C}$ . Elemental analysis (%) Calcd for  $\text{C}_{82}\text{H}_{104}\text{Fe}_2\text{N}_4\text{O}_4 \cdot 2(\text{C}_7\text{H}_8)$ : C, 76.58; H, 8.03; N, 3.72. Found: C, 76.61; H, 8.04; N, 3.48.

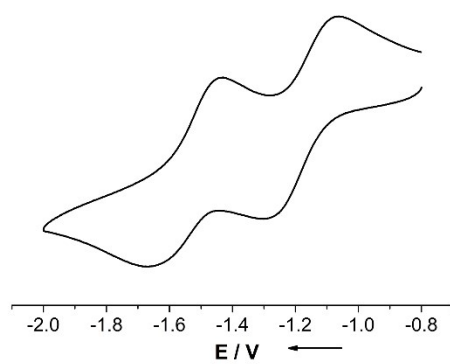
**Synthesis of  $1\text{K}_2$ .** A mixture of **1** (90.5 mg, 0.07 mmol) and potassium (6.3 mg, 0.16 mmol) in THF ( $\approx 40$  ml) was stirred at room temperature in an  $\text{N}_2$ -filled glovebox, whereupon the color of the solution was changed from deep green to light green. After stirring for 12 h, the solution was filtered, and the filtrate was concentrated to ca. 10 mL and stored at  $-20\text{ }^\circ\text{C}$  for 24 h yielding light green crystals of  $1\text{K}_2$ . Removing all the volatiles in vacuo, the light green crystals soon became a light green powder (32 mg, 33.2 %). Decomposed at  $170\text{-}173\text{ }^\circ\text{C}$ .  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ ):  $\delta = 8.42$  (s, 4H, Ar-H), 6.77 (d,  $J = 7.6$  Hz, 8H, Ar-H), 6.57 (t,  $J = 7.6$  Hz, 4H, Ar-H), 5.10 (s, 2H,  $\gamma\text{-H}$ ), 3.72 (m, 8H,  $\text{CH}(\text{CH}_3)_2$ ), 3.56 (br, 16H,  $\text{THF-OCH}_2$ ), 3.30 (m, 3H), 2.98 (m, 2H), 1.82 (s, 15H,  $\text{CH}_3$ ), 1.80 (s, 10H,  $\text{CH}_3$ ), 1.56 (m, 22H,  $\text{CH}_3$ ), 1.40 (br, 16H,  $\text{OCH}_2\text{CH}_2$ ), 1.22-1.15 (m, 38H,  $\text{CH}_3$ ) ppm.  $^{13}\text{C}$  NMR (101 MHz,  $\text{THF-d}_8$ ):  $\delta = 167.72$ , 161.77, 147.16, 146.77, 144.37, 144.14, 142.91, 129.46, 125.85, 124.48, 123.69, 123.54, 110.97, 108.66, 94.32, 94.02,

68.02, 32.65, 28.87, 28.61, 26.19, 24.80, 24.56, 23.56, 23.27, 20.71 ppm. Elemental analysis (%) Calcd for  $C_{82}H_{104}K_2Mg_2N_4O_4 \cdot 6(C_4H_8O)$ : C, 71.96; H, 8.66; N, 3.17; Found: C, 71.94; H, 8.01; N, 3.81.

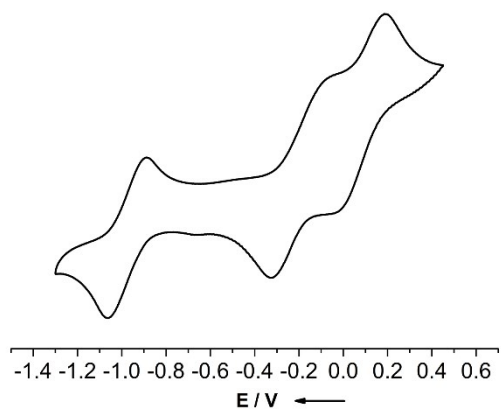
**Synthesis of  $2K_2$ .** A mixture of **2** (120 mg, 0.09 mmol) and potassium (8.3 mg, 0.21 mmol) in THF (ca. 30 ml) was stirred at room temperature in an  $N_2$ -filled glovebox for 24 h, whereupon the color of the solution was changed from brownish red to brown. The solution was filtered, and the filtrate was concentrated to ca. 12 mL and stored at  $-20\text{ }^\circ\text{C}$  for 24 h affording brown crystals of  $2K_2$ . Removing all the volatiles in vacuo, the crystals soon became a reddish brown powder (73 mg, 57.3 %). M.P.  $> 300\text{ }^\circ\text{C}$ . Elemental analysis (%) Calcd for  $C_{82}H_{104}Fe_2K_2N_4O_4 \cdot 2(C_4H_8O)$ : C, 70.02; H, 7.83; N, 3.63. Found: C, 68.99; H, 7.70; N, 3.66.



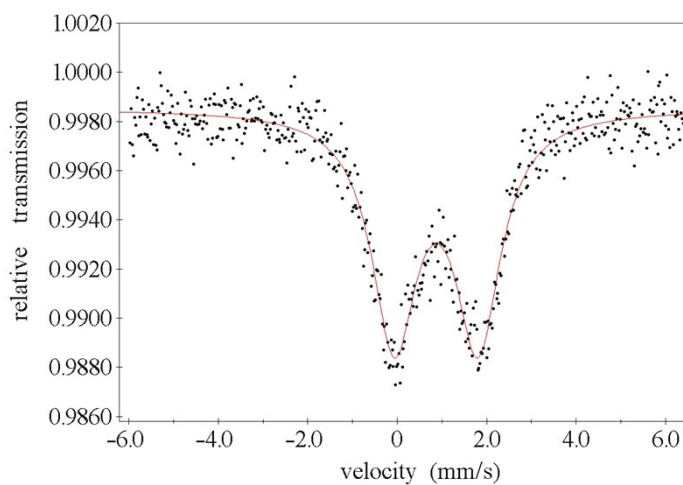
**Fig. S1** Cyclic voltammogram of 2,7-*t*Bu<sub>2</sub>-PTO in THF ( $1 \times 10^{-4}$  M, 0.1 M <sup>n</sup>Bu<sub>4</sub>NPF<sub>6</sub>, Ag/Ag<sup>+</sup> electrode, 298 K) was measured at 100 mV•s<sup>-1</sup>.



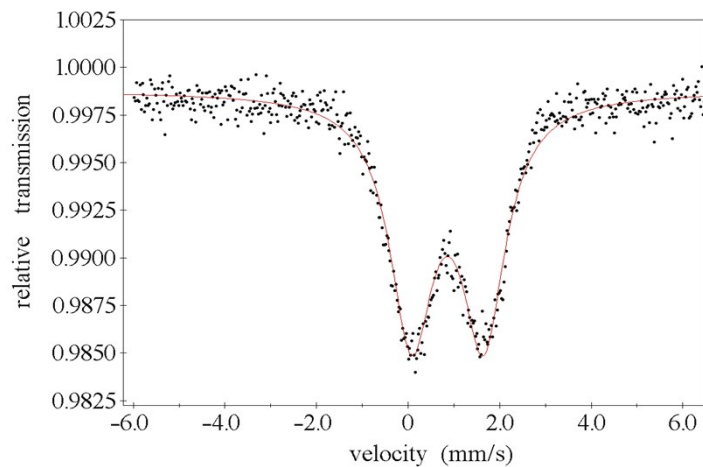
**Fig. S2** Cyclic voltammogram of **1** in THF ( $1 \times 10^{-4}$  M, 0.1 M <sup>n</sup>Bu<sub>4</sub>NPF<sub>6</sub>, Ag/Ag<sup>+</sup> electrode, 298 K) was measured at 100 mV•s<sup>-1</sup>.



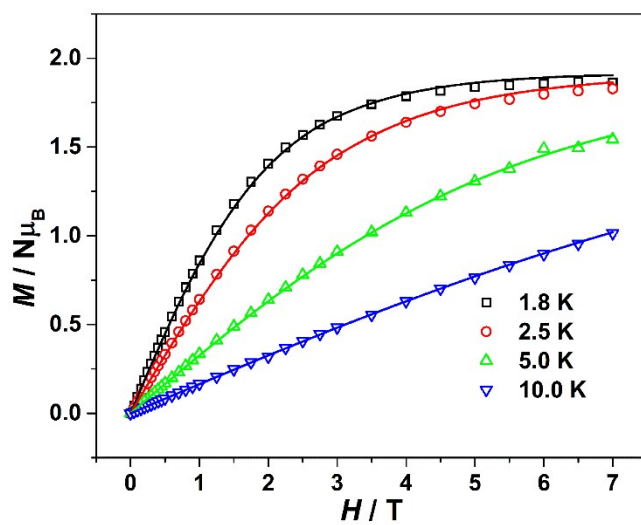
**Fig. S3** Cyclic voltammogram of **2** in THF ( $1 \times 10^{-4}$  M, 0.1 M  $n\text{Bu}_4\text{NPF}_6$ , Ag/Ag<sup>+</sup> electrode, 298 K) was measured at  $100 \text{ mV} \cdot \text{s}^{-1}$ .



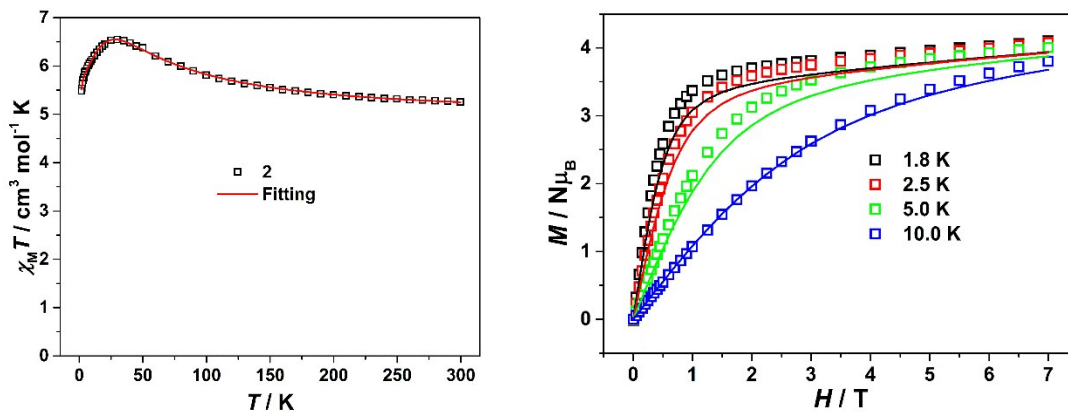
**Fig. S4** Zero-field  $^{57}\text{Fe}$  Mössbauer spectrum of **2** recorded at 80K.



**Fig. S5** Zero-field  $^{57}\text{Fe}$  Mössbauer spectrum of  $2\mathbf{K}_2$  recorded at 80K.



**Fig. S6** Isothermal magnetization at different temperatures for  $\mathbf{1}$ . The solid lines are the best fitting results using PHI program.



**Fig. S7**  $\chi_M T$ - $T$  (left) and  $M$ - $H$  (right) plots of **2** with the fitting result (solid line) using the PHI program, in which the constant  $J$  between radicals is fixed at  $165 \text{ cm}^{-1}$  same as that in compound **1** for comparison due to the very similar structure.

The temperature and field dependent magnetizations were fitted to quantify the anisotropy parameters based on equation 1 using the *PHI* program:

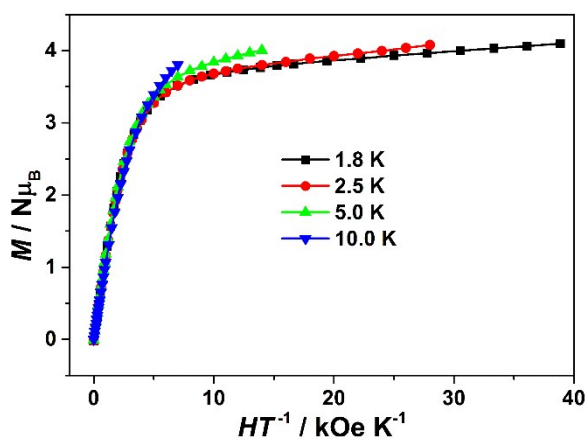
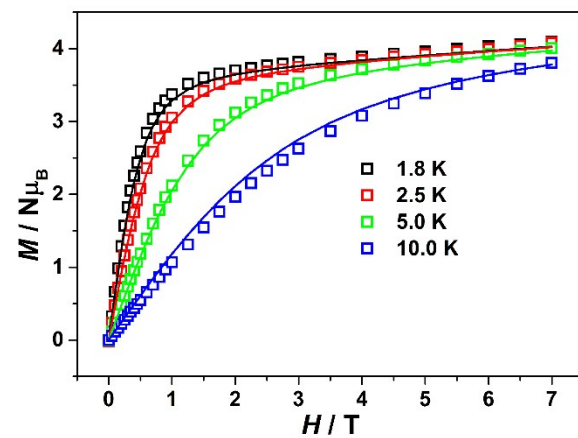
$$H = -2J_1(\mathcal{S}_{Fe1}\mathcal{S}_{Rad1} + \mathcal{S}_{Rad2}\mathcal{S}_{Fe2}) - 2J\mathcal{S}_{Rad1}\mathcal{S}_{Rad2} - 2J_2(\mathcal{S}_{Fe1}\mathcal{S}_{Rad2} + \mathcal{S}_{Rad1}\mathcal{S}_{Fe2}) + D\left[\mathcal{S}_z^2 - \frac{S(S+1)}{3}\right] + g\mu_B\mathcal{S}H \quad (1)$$

)

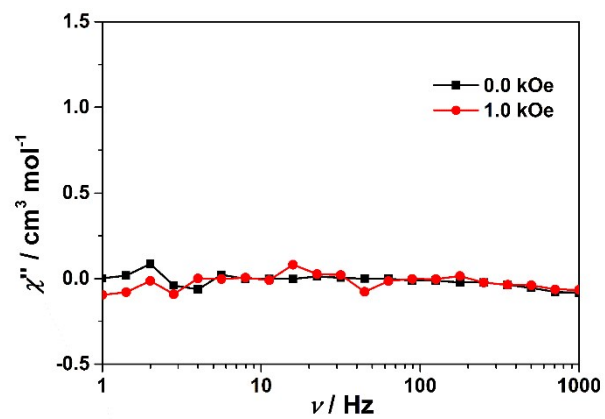
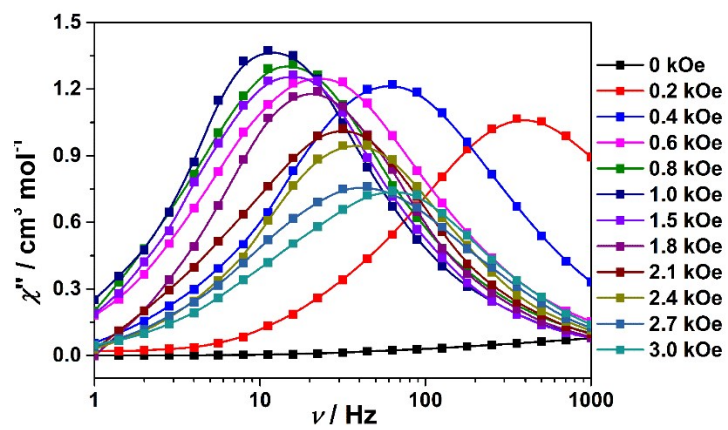
where  $J_1$  and  $J_2$  are the magnetic coupling constants between the spins of radicals and  $\text{Fe}^{\text{II}}$  ions (Fig. 7, inset). A reasonable fitting gives  $D = -12.00(1) \text{ cm}^{-1}$ ,  $g = 2.25(1)$ ,  $J_1 = -1092.5$ ,  $J = 165$  (fixed),  $J_2 = -2.51(4)$  and  $TIP = 5.93 \times 10^{-4} \text{ cm}^3 \text{ mol}^{-1}$ .

We can find that the fitting results for the  $M$ - $H$  curves show a large deviation with the experimental data. It can be ascribed to the electronic structure change from **1** to **2**. For example, the angle O1-Mg1-O2 is  $83.87^\circ$  in **1** but  $80.97^\circ$  in **2**. These reduced angle will lead to maximum electron cloud orientation shifts in radicals and the overlap decrease of the magnetic orbitals. Such that, the coupling between radicals may be weakened somewhat. So, the results in text are thought to be acceptable rather than here, although the fittings of  $\chi_M T$ - $T$  here are as beautiful as in the maintext.

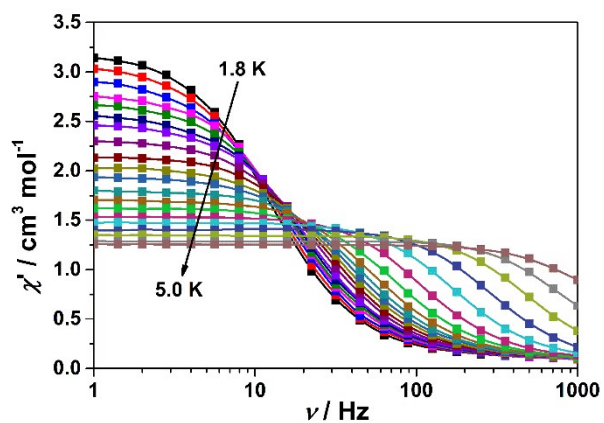




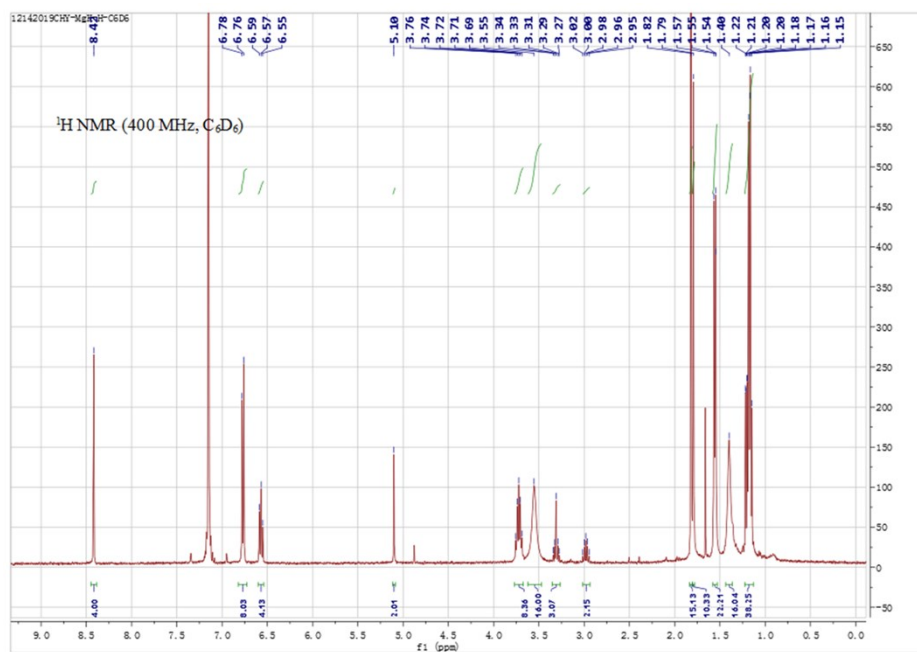
**Fig. S8** Top: Isothermal magnetization at different temperatures for **2**. The solid line is the fitting value; Down: Experimental  $M$  vs  $H/T$  plots at different temperatures for **2**.



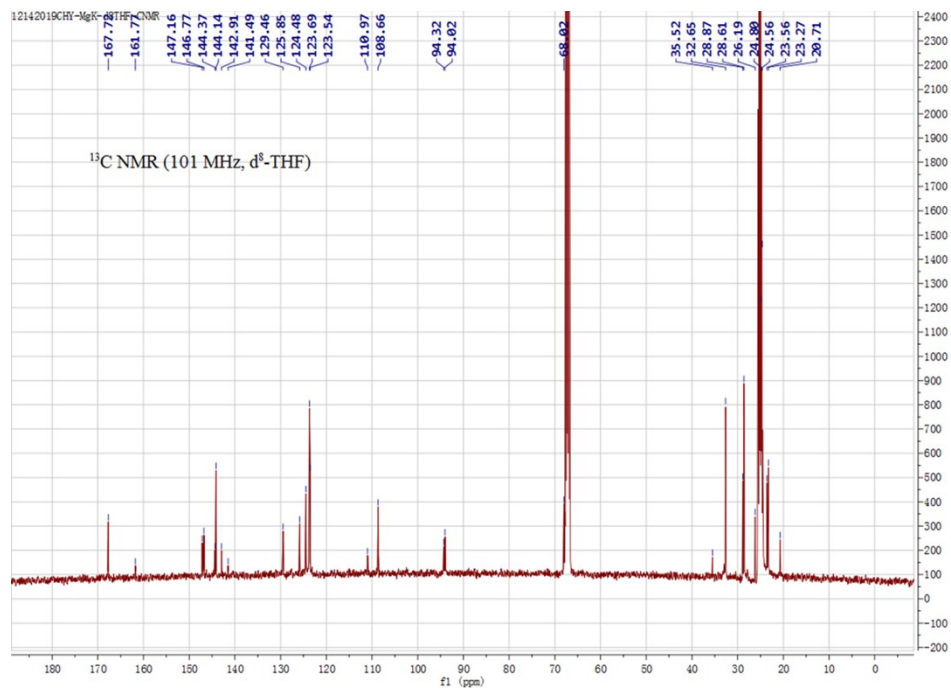
**Fig. S9** Isothermal field sweep measurement performed on polycrystalline sample of complex **2** (up) and  $2\mathbf{K}_2$  (down).



**Fig. S10** Frequency-dependence of the in-phase ( $\chi_M'$ ) *ac*-susceptibilities for **2** at different temperatures (1.8 to 5.0 K).



**Fig. S11**  $^1\text{H}$  NMR spectrum of **1K<sub>2</sub>** in  $\text{C}_6\text{D}_6$  at room temperature.



**Fig. S12** <sup>13</sup>C NMR spectrum of **1K<sub>2</sub>** in d<sup>8</sup>-THF at room temperature.

**Table S1** Crystal data and structure refinements. Severely disordered solvent molecules in the four crystals were squeezed.

	<b>1</b>	<b>1K<sub>2</sub></b>	<b>2</b>	<b>2K<sub>2</sub></b>
Formula	C <sub>82</sub> H <sub>104</sub> Mg <sub>2</sub> N <sub>4</sub> O <sub>4</sub> • C <sub>14</sub> H <sub>16</sub>	C <sub>82</sub> H <sub>104</sub> Mg <sub>2</sub> K <sub>2</sub> N <sub>4</sub> O <sub>4</sub> • 4(C <sub>4</sub> H <sub>8</sub> O)	C <sub>82</sub> H <sub>104</sub> Fe <sub>2</sub> N <sub>4</sub> O <sub>4</sub>	C <sub>82</sub> H <sub>104</sub> Fe <sub>2</sub> K <sub>2</sub> N <sub>4</sub> O <sub>4</sub> •4(C <sub>4</sub> H <sub>8</sub> O)
Formula weight	1442.57	1624.92	1321.39	1688.00
Temp. (K)	193(2)	296(2)	193(2)	296(2)
Crystal system	Monoclinic	Triclinic	Monoclinic	Triclinic
Space group	<i>P</i> 21/ <i>n</i>	<i>P</i> -1	<i>P</i> 21/ <i>n</i>	<i>P</i> -1
<i>a</i> (Å)	13.0089(3)	12.3217(14)	12.9263(5)	13.1826(7)
<i>b</i> (Å)	18.5762(4)	14.4239(17)	18.6784(6)	13.9086(7)
<i>c</i> (Å)	19.9212(5)	15.785(2)	19.8163(7)	19.5230(11)
$\alpha$ (°)	90	90.292(4)	90	92.382(2)
$\beta$ (°)	101.5820(10)	93.572(5)	100.993(2)	107.892(2)
$\gamma$ (°)	90	92.232(4)	90	116.504(2)
<i>V</i> [Å <sup>3</sup> ]	4716.05(19)	2797.8(6)	4696.7(3)	2980.6(3)
<i>Z</i>	2	1	2	1
$\rho_{\text{calcd}}$ (g·cm <sup>-3</sup> )	1.016	0.964	0.934	0.940
$\mu$ (mm <sup>-1</sup> )	0.376	0.142	1.910	0.357
<i>F</i> (000)	1560	878	1416	906
Collected data	33139	21163	33952	21664
Unique data	8282 [ <i>R</i> (int)=0.0624]	9776 [ <i>R</i> (int) = 0.0540]	8279 [ <i>R</i> (int)= 0.0522]	10360 [ <i>R</i> (int)=0.0380]
GOF on <i>F</i> <sup>2</sup>	1.097	1.019	1.049	1.054
Final <i>R</i> indexes	<i>R</i> <sub>1</sub> = 0.0673 <i>wR</i> <sub>2</sub> = 0.1251	<i>R</i> <sub>1</sub> = 0.0603 <i>wR</i> <sub>2</sub> = 0.1586	<i>R</i> <sub>1</sub> = 0.0488 <i>wR</i> <sub>2</sub> = 0.1544	<i>R</i> <sub>1</sub> = 0.0520 <i>wR</i> <sub>2</sub> = 0.1532
[ <i>I</i> > 2σ( <i>I</i> )]				
<i>R</i> indexes (all data)	<i>R</i> <sub>1</sub> = 0.1221 <i>wR</i> <sub>2</sub> = 0.1374	<i>R</i> <sub>1</sub> = 0.0870 <i>wR</i> <sub>2</sub> = 0.1756	<i>R</i> <sub>1</sub> = 0.0739 <i>wR</i> <sub>2</sub> = 0.1686	<i>R</i> <sub>1</sub> = 0.0609 <i>wR</i> <sub>2</sub> = 0.1616
Completeness	0.994	0.990	0.997	0.985

**Table S2** The fit parameters obtained from analyses of the ac susceptibilities of **2** under 1.0 kOe dc field.

$T / K$	$\chi_T / \text{cm}^3 \text{mol}^{-1}$	$\chi_S / \text{cm}^3 \text{mol}^{-1}$	$\ln(\tau / \text{s})$	$a$	$R^2$
1.8	3.21573	0.10315	-4.3924	0.08205	0.01167
1.9	3.08632	0.10317	-4.47789	0.08539	0.00733
2	2.94477	0.10424	-4.57138	0.08637	0.00497
2.1	2.80046	0.10893	-4.68429	0.07571	0.00896
2.2	2.697	0.10748	-4.76074	0.08127	0.00679
2.3	2.58363	0.10592	-4.84572	0.08285	0.00608
2.4	2.48313	0.10953	-4.93303	0.07604	0.00749
2.6	2.32297	0.10759	-5.08536	0.08163	0.00704
2.8	2.1637	0.10704	-5.23846	0.07735	0.00693
3	2.03814	0.10611	-5.39157	0.07952	0.00745
3.2	1.93825	0.10642	-5.53091	0.0789	0.00724
3.4	1.81168	0.10625	-5.70331	0.072	0.00471
3.6	1.71163	0.10616	-5.91042	0.06361	0.00561
3.8	1.62372	0.10013	-6.18304	0.06141	0.00319
4	1.53993	0.1047	-6.55372	0.04069	0.00167
4.2	1.47738	0.09343	-7.00998	0.03939	0.00178
4.4	1.40842	0.08986	-7.52761	0.02738	0.0016
4.6	1.34757	0.09915	-8.05962	0.01936	7.54E-04
4.8	1.28868	0.08134	-8.62206	0.01705	1.06E-03
5	1.25309	1.60E-14	-9.23172	0.03985	4.78E-04

## Computational details:

All the calculations were performed at Gaussian 09 program suite.<sup>4,5</sup> The ground-state structures of the studied compound **1** were optimized using density functional theory (DFT) at the (U)B3LYP/6-31G(d) level of approximation, and no imaginary frequency was found, which confirmed the local minimum of the optimized structures.

## Coordinates of complex **1**

### Close-shell singlet state

Mg	0.3376	11.0548	11.0804
N	1.6182	10.5059	9.5145
N	1.6581	12.582	11.6408
O	-0.448	11.0178	12.9651
O	-0.5313	9.2036	11.0973
C	-1.0128	9.9089	13.2541
C	-1.0621	8.9082	12.2209
C	-1.7078	7.6417	12.4765
C	-1.7497	6.6428	11.4952
H	-1.3218	6.798	10.6603
C	-2.3984	5.4327	11.7064
C	-3.0098	5.2252	12.9391
H	-3.4702	4.4095	13.0973
C	-2.9634	6.1895	13.9502
C	-2.3162	7.414	13.737
C	-2.4018	4.3715	10.5888
C	-3.2417	3.182	10.9624
H	-4.1739	3.4624	11.0736
H	-2.9115	2.7987	11.8004
H	-3.1881	2.5089	10.2527
C	-2.912	4.9924	9.3007
H	-2.947	4.3082	8.5995
H	-2.3094	5.7131	9.0242
H	-3.8108	5.3551	9.446
C	-0.9655	3.9043	10.3525
H	-0.5875	3.5716	11.1933
H	-0.4269	4.655	10.0247
H	-0.9614	3.1852	9.6855
C	2.5004	11.3738	8.992
C	2.8151	12.6188	9.5276
C	2.5173	13.1288	10.7886
C	1.5302	9.2245	8.9646
C	0.6949	8.9252	7.8413
C	0.6904	7.5266	7.4765
H	0.2004	7.2763	6.704
C	1.3109	6.5826	8.1292
H	1.1842	5.675	7.8784
C	2.1107	6.8867	9.1553
H	2.6208	6.2136	9.5886
C	2.184	8.2108	9.5715
C	-0.1295	10.0561	7.2
H	0.169	10.9335	7.5762
C	0.0901	10.0624	5.6636
H	0.9201	9.5856	5.4499
H	0.152	10.9874	5.3473

H	-0.6648	9.6157	5.2247
C	-1.6195	9.8327	7.5563
H	-2.1809	10.3887	6.9748
H	-1.7722	10.0782	8.4912
H	-1.8488	8.8887	7.4252
C	3.0427	8.4152	10.7971
H	2.6625	9.2182	11.2589
C	3.1039	7.3524	11.8432
H	2.2448	7.3064	12.3135
H	3.8152	7.563	12.4845
H	3.2912	6.4876	11.4213
C	4.49	8.7937	10.4465
H	5.0032	8.9267	11.2703
H	4.4934	9.6221	9.9221
H	4.8977	8.073	9.9221
C	1.658	13.0417	13.0118
C	0.7231	14.0221	13.3995
C	0.7434	14.3484	14.8104
H	0.1497	15.012	15.141
C	1.6069	13.7148	15.6883
H	1.5686	13.9207	16.6147
C	2.4786	12.831	15.2551
H	3.0883	12.443	15.8707
C	2.534	12.4414	13.8983
C	-0.2271	14.6382	12.4959
H	-0.0915	14.2549	11.581
C	-1.6802	14.3404	12.9463
H	-1.8745	14.8346	13.77
H	-1.7814	13.379	13.1087
H	-2.3036	14.6208	12.2422
C	-0.052	16.1714	12.4161
H	-0.8753	16.6101	12.7154
H	0.1397	16.4312	11.4926
H	0.6916	16.4454	12.9947
C	3.5349	11.4435	13.4804
H	3.3868	11.2408	12.513
C	3.4218	10.1463	14.2603
H	3.5389	10.3285	15.218
H	4.1143	9.5223	13.961
H	2.5377	9.7504	14.1121
C	4.9691	11.9821	13.6446
H	5.0658	12.8136	13.1344
H	5.6078	11.3184	13.3111
H	5.1465	12.1579	14.5909
O	-4.1252	4.821	15.5384
O	-4.0419	6.6352	17.4062
C	-3.5604	5.9299	15.2494
C	-3.511	6.9306	16.2826
C	-2.8654	8.1971	16.0269
C	-2.8234	9.196	17.0083
H	-3.2514	9.0408	17.8432
C	-2.1747	10.4061	16.7971
C	-1.5633	10.6136	15.5643
H	-1.103	11.4293	15.4061
C	-1.6098	9.6493	14.5533
C	-2.257	8.4248	14.7665



C	-2.1713	11.4673	17.9147
C	-1.3315	12.6568	17.541
H	-0.3993	12.3764	17.4299
H	-1.6616	13.0401	16.703
H	-1.3851	13.3299	18.2508
C	-1.6612	10.8464	19.2028
H	-1.6262	11.5306	19.904
H	-2.2638	10.1257	19.4793
H	-0.7624	10.4837	19.0574
C	-3.6077	11.9345	18.151
H	-3.9856	12.2672	17.3102
H	-4.1462	11.1838	18.4788
H	-3.6118	12.6536	18.818
Mg	-4.9108	4.784	17.423
N	-6.1913	5.3329	18.989
N	-6.2313	3.2568	16.8627
C	-7.0736	4.465	19.5115
C	-7.3883	3.22	18.9759
C	-7.0905	2.71	17.7149
C	-6.1033	6.6143	19.5388
C	-5.2681	6.9136	20.6622
C	-5.2636	8.3122	21.027
H	-4.7736	8.5625	21.7995
C	-5.8841	9.2562	20.3743
H	-5.7574	10.1638	20.6251
C	-6.6839	8.9521	19.3482
H	-7.194	9.6252	18.9149
C	-6.7571	7.628	18.932
C	-4.4436	5.7827	21.3035
H	-4.7422	4.9053	20.9272
C	-4.6632	5.7764	22.8398
H	-5.4932	6.2532	23.0536
H	-4.7251	4.8514	23.1562
H	-3.9083	6.2231	23.2788
C	-2.9537	6.0061	20.9472
H	-2.3923	5.4501	21.5287
H	-2.8009	5.7606	20.0123
H	-2.7244	6.9501	21.0783
C	-7.6158	7.4236	17.7064
H	-7.2357	6.6206	17.2446
C	-7.6771	8.4864	16.6603
H	-6.818	8.5324	16.19
H	-8.3884	8.2758	16.0189
H	-7.8644	9.3512	17.0821
C	-9.0631	7.0451	18.0569
H	-9.5764	6.9121	17.2332
H	-9.0666	6.2167	18.5814
H	-9.4709	7.7658	18.5814
C	-6.2311	2.7971	15.4916
C	-5.2963	1.8167	15.104
C	-5.3166	1.4904	13.6931
H	-4.7229	0.8268	13.3624
C	-6.1801	2.124	12.8152
H	-6.1418	1.9181	11.8888
C	-7.0517	3.0078	13.2484
H	-7.6615	3.3958	12.6327

C	-7.1071	3.3974	14.6052
C	-4.346	1.2006	16.0075
H	-4.4817	1.5839	16.9225
C	-2.8929	1.4984	15.5572
H	-2.6986	1.0042	14.7334
H	-2.7918	2.4598	15.3947
H	-2.2696	1.218	16.2612
C	-4.5211	-0.3326	16.0874
H	-3.6978	-0.7713	15.7881
H	-4.7129	-0.5924	17.0109
H	-5.2648	-0.6066	15.5087
C	-8.108	4.3953	15.023
H	-7.96	4.598	15.9904
C	-7.995	5.6925	14.2432
H	-8.1121	5.5103	13.2855
H	-8.6875	6.3165	14.5425
H	-7.1109	6.0884	14.3914
C	-9.5423	3.8567	14.8589
H	-9.639	3.0252	15.3691
H	-10.181	4.5204	15.1923
H	-9.7197	3.6809	13.9125
H	3.361	13.2744	8.8818
C	3.2355	14.4189	11.2261
H	4.2054	14.4574	10.7758
H	2.6639	15.2683	10.9152
H	3.3363	14.4273	12.2913
C	3.2212	10.9574	7.6964
H	2.6536	11.2862	6.851
H	4.1929	11.4046	7.6693
H	3.3179	9.8922	7.6686
H	-7.9342	2.5644	19.6217
C	-7.7945	4.8814	20.8071
H	-8.7616	4.4248	20.8397
H	-7.221	4.5631	21.6525
H	-7.9017	5.9458	20.8292
C	-7.8087	1.4199	17.2774
H	-7.2456	0.5703	17.6028
H	-8.7849	1.3894	17.7144
H	-7.8947	1.4037	16.211

### Open-shell singlet state

Mg	5.13954	0.00003	-0.0208
N	6.52559	-1.51878	-0.02706
N	6.52547	1.51896	-0.02534
O	3.61488	0.00077	-1.34093
O	3.62518	-0.0008	1.31377
C	2.47432	0.00041	-0.73573
C	2.48038	-0.00045	0.71736
C	1.22434	-0.00085	1.44236
C	1.20549	-0.00164	2.84373
H	2.16436	-0.00194	3.35076
C	0.00939	-0.00201	3.57788
C	-1.18839	-0.0016	2.85596
H	-2.14489	-0.00185	3.36325
C	-1.21333	-0.00084	1.45002
C	0.00179	-0.00044	0.71742

C	0.0564	-0.00284	5.11651
C	-1.35063	-0.00317	5.74323
H	-1.92602	-0.89055	5.45604
H	-1.926	0.88454	5.45699
H	-1.26581	-0.00375	6.8357
C	0.8049	-1.26579	5.6055
H	0.8557	-1.27609	6.70107
H	1.83051	-1.30722	5.22408
H	0.28989	-2.1771	5.28078
C	0.80491	1.25959	5.60685
H	0.28991	2.17124	5.2831
H	1.83052	1.30141	5.22547
H	0.85571	1.26873	6.70243
C	7.8415	-1.28339	-0.03968
C	8.42729	0.00018	-0.04704
C	7.8414	1.28369	-0.03839
C	6.03179	-2.87072	-0.01539
C	5.73242	-3.52179	-1.23927
C	5.15311	-4.79593	-1.19731
H	4.9167	-5.30136	-2.1302
C	4.87342	-5.42756	0.00994
H	4.4206	-6.41559	0.01977
C	5.18167	-4.78512	1.2045
H	4.96726	-5.28199	2.14725
C	5.76204	-3.51094	1.2211
C	6.02484	-2.88881	-2.59946
H	6.50968	-1.92379	-2.42317
C	6.9963	-3.7458	-3.43626
H	7.93207	-3.94252	-2.90122
H	7.24498	-3.23335	-4.37334
H	6.55546	-4.71524	-3.69682
C	4.73155	-2.61644	-3.39311
H	4.96898	-2.15043	-4.3578
H	4.0671	-1.94286	-2.84418
H	4.18695	-3.54605	-3.59813
C	6.08636	-2.86535	2.56807
H	6.57392	-1.90565	2.37157
C	4.81087	-2.57528	3.38414
H	4.14001	-1.90343	2.8407
H	5.07062	-2.10077	4.33892
H	4.26424	-3.49888	3.60977
C	7.06872	-3.71947	3.39492
H	7.34136	-3.19713	4.31979
H	7.9912	-3.93045	2.84249
H	6.62627	-4.68195	3.67754
C	6.03156	2.87085	-0.01229
C	5.73281	3.52349	-1.23548
C	5.15333	4.7975	-1.19218
H	4.91732	5.3041	-2.12454
C	4.87295	5.42756	0.01573
H	4.42004	6.41553	0.02659
C	5.18063	4.78361	1.20963
H	4.96573	5.27926	2.1529
C	5.76107	3.50944	1.22489
C	6.02589	2.89223	-2.59633
H	6.51169	1.9275	-2.42104

C	4.73283	2.61936	-3.39019
H	4.18727	3.54866	-3.5941
H	4.0691	1.94451	-2.84195
H	4.97066	2.15469	-4.35541
C	6.99642	3.75104	-3.43232
H	6.5544	4.72009	-3.69232
H	7.24593	3.23956	-4.3697
H	7.93182	3.94863	-2.89695
C	6.08488	2.86222	2.5712
H	6.5717	1.90234	2.3737
C	4.80923	2.5723	3.38707
H	4.26335	3.49606	3.61381
H	5.06871	2.09652	4.34129
H	4.13781	1.90157	2.84296
C	7.06797	3.71475	3.39881
H	7.99078	3.92513	2.8467
H	7.33992	3.19151	4.32338
H	6.62644	4.6775	3.68198
O	-3.61486	-0.00075	1.34079
O	-3.62516	0.00062	-1.3139
C	-2.47429	-0.00044	0.7356
C	-2.48036	0.00034	-0.7175
C	-1.22433	0.00076	-1.4425
C	-1.20547	0.00154	-2.84388
H	-2.16433	0.00181	-3.35091
C	-0.00936	0.00196	-3.57802
C	1.18842	0.00159	-2.8561
H	2.14492	0.00189	-3.36338
C	1.21335	0.0008	-1.45016
C	-0.00177	0.00037	-0.71756
C	-0.05636	0.00279	-5.11665
C	1.35067	0.00314	-5.74336
H	1.92604	0.89054	-5.45616
H	1.92606	-0.88455	-5.4571
H	1.26587	0.00371	-6.83582
C	-0.80487	1.26573	-5.60564
H	-0.85565	1.27604	-6.70122
H	-1.8305	1.30714	-5.22425
H	-0.28989	2.17704	-5.28091
C	-0.80484	-1.25965	-5.607
H	-0.28982	-2.1713	-5.28326
H	-1.83046	-1.30151	-5.22564
H	-0.85564	-1.26878	-6.70258
Mg	-5.13954	0.00003	0.02068
N	-6.52558	1.51882	0.02702
N	-6.52547	-1.51892	0.02545
C	-7.84149	1.28344	0.04013
C	-8.42728	-0.00012	0.04749
C	-7.84139	-1.28364	0.0389
C	-6.03177	2.87075	0.0152
C	-5.73308	3.52238	1.23895
C	-5.15365	4.79645	1.19672
H	-4.91766	5.30227	2.12951
C	-4.87333	5.42757	-0.01065
H	-4.42048	6.41557	-0.02067
C	-5.18099	4.78463	-1.20509

H	-4.96615	5.2811	-2.14795
C	-5.76132	3.51042	-1.22144
C	-6.02609	2.88995	2.59927
H	-6.51217	1.92551	2.42319
C	-6.99623	3.74825	3.43624
H	-7.93161	3.94674	2.90116
H	-7.24583	3.2359	4.37312
H	-6.55384	4.71687	3.69721
C	-4.73295	2.61603	3.39265
H	-4.97072	2.15063	4.35754
H	-4.06951	1.94145	2.84373
H	-4.18709	3.54502	3.59721
C	-6.08512	2.86432	-2.56828
H	-6.57157	1.90409	-2.37159
C	-4.80951	2.57561	-3.38463
H	-4.13779	1.90459	-2.84124
H	-5.06898	2.10063	-4.33925
H	-4.26396	3.49976	-3.6106
C	-7.06867	3.71727	-3.39492
H	-7.34048	3.19487	-4.32
H	-7.99152	3.92664	-2.84251
H	-6.6276	4.68052	-3.67712
C	-6.03157	-2.87081	0.01237
C	-5.73333	-3.52379	1.2355
C	-5.15382	-4.79779	1.19207
H	-4.91813	-5.30462	2.12438
C	-4.87304	-5.42756	-0.0159
H	-4.42016	-6.41554	-0.02685
C	-5.1803	-4.78331	-1.20974
H	-4.96514	-5.27875	-2.15307
C	-5.76063	-3.50909	-1.22489
C	-6.0268	-2.89286	2.5964
H	-6.51347	-1.92855	2.42123
C	-4.73384	-2.61891	3.39006
H	-4.18742	-3.54777	3.59368
H	-4.07081	-1.94337	2.84181
H	-4.9719	-2.15461	4.35542
C	-6.99641	-3.75256	3.43256
H	-6.5533	-4.72101	3.69293
H	-7.2466	-3.24108	4.36977
H	-7.93152	-3.95144	2.89716
C	-6.08408	-2.86158	-2.57114
H	-6.56998	-1.90125	-2.37356
C	-4.80835	-2.57288	-3.38731
H	-4.26341	-3.49713	-3.6143
H	-5.06762	-2.09676	-4.34142
H	-4.13616	-1.90283	-2.84331
C	-7.06817	-3.71324	-3.39848
H	-7.99125	-3.92226	-2.84632
H	-7.33948	-3.19001	-4.32324
H	-6.62777	-4.67662	-3.68125
H	9.51014	0.00023	-0.05839
C	8.80879	2.45475	-0.04338
H	9.84538	2.11256	-0.05886
H	8.64186	3.09696	-0.91408
H	8.66456	3.08512	0.84004

C	8.809	-2.45436	-0.04506
H	8.63945	-3.09878	-0.91357
H	9.8455	-2.1121	-0.06444
H	8.66758	-3.0825	0.84043
H	-9.51013	-0.00017	0.05916
C	-8.80896	2.45443	0.04685
H	-9.84555	2.11214	0.06015
H	-8.64321	3.09427	0.91954
H	-8.66361	3.08721	-0.83462
C	-8.80878	-2.4547	0.04503
H	-8.64491	-3.0932	0.91908
H	-9.84542	-2.11247	0.05567
H	-8.66145	-3.08881	-0.83514

### Triplet state

Mg	0.3376	11.0548	11.0804
N	1.6182	10.5059	9.5145
N	1.6581	12.582	11.6408
O	-0.448	11.0178	12.9651
O	-0.5313	9.2036	11.0973
C	-1.0128	9.9089	13.2541
C	-1.0621	8.9082	12.2209
C	-1.7078	7.6417	12.4765
C	-1.7497	6.6428	11.4952
H	-1.3218	6.798	10.6603
C	-2.3984	5.4327	11.7064
C	-3.0098	5.2252	12.9391
H	-3.4702	4.4095	13.0973
C	-2.9634	6.1895	13.9502
C	-2.3162	7.414	13.737
C	-2.4018	4.3715	10.5888
C	-3.2417	3.182	10.9624
H	-4.1739	3.4624	11.0736
H	-2.9115	2.7987	11.8004
H	-3.1881	2.5089	10.2527
C	-2.912	4.9924	9.3007
H	-2.947	4.3082	8.5995
H	-2.3094	5.7131	9.0242
H	-3.8108	5.3551	9.446
C	-0.9655	3.9043	10.3525
H	-0.5875	3.5716	11.1933
H	-0.4269	4.655	10.0247
H	-0.9614	3.1852	9.6855
C	2.5004	11.3738	8.992
C	2.8151	12.6188	9.5276
C	2.5173	13.1288	10.7886
C	1.5302	9.2245	8.9646
C	0.6949	8.9252	7.8413
C	0.6904	7.5266	7.4765
H	0.2004	7.2763	6.704
C	1.3109	6.5826	8.1292
H	1.1842	5.675	7.8784
C	2.1107	6.8867	9.1553
H	2.6208	6.2136	9.5886
C	2.184	8.2108	9.5715
C	-0.1295	10.0561	7.2

H	0.169	10.9335	7.5762
C	0.0901	10.0624	5.6636
H	0.9201	9.5856	5.4499
H	0.152	10.9874	5.3473
H	-0.6648	9.6157	5.2247
C	-1.6195	9.8327	7.5563
H	-2.1809	10.3887	6.9748
H	-1.7722	10.0782	8.4912
H	-1.8488	8.8887	7.4252
C	3.0427	8.4152	10.7971
H	2.6625	9.2182	11.2589
C	3.1039	7.3524	11.8432
H	2.2448	7.3064	12.3135
H	3.8152	7.563	12.4845
H	3.2912	6.4876	11.4213
C	4.49	8.7937	10.4465
H	5.0032	8.9267	11.2703
H	4.4934	9.6221	9.9221
H	4.8977	8.073	9.9221
C	1.658	13.0417	13.0118
C	0.7231	14.0221	13.3995
C	0.7434	14.3484	14.8104
H	0.1497	15.012	15.141
C	1.6069	13.7148	15.6883
H	1.5686	13.9207	16.6147
C	2.4786	12.831	15.2551
H	3.0883	12.443	15.8707
C	2.534	12.4414	13.8983
C	-0.2271	14.6382	12.4959
H	-0.0915	14.2549	11.581
C	-1.6802	14.3404	12.9463
H	-1.8745	14.8346	13.77
H	-1.7814	13.379	13.1087
H	-2.3036	14.6208	12.2422
C	-0.052	16.1714	12.4161
H	-0.8753	16.6101	12.7154
H	0.1397	16.4312	11.4926
H	0.6916	16.4454	12.9947
C	3.5349	11.4435	13.4804
H	3.3868	11.2408	12.513
C	3.4218	10.1463	14.2603
H	3.5389	10.3285	15.218
H	4.1143	9.5223	13.961
H	2.5377	9.7504	14.1121
C	4.9691	11.9821	13.6446
H	5.0658	12.8136	13.1344
H	5.6078	11.3184	13.3111
H	5.1465	12.1579	14.5909
O	-4.1252	4.821	15.5384
O	-4.0419	6.6352	17.4062
C	-3.5604	5.9299	15.2494
C	-3.511	6.9306	16.2826
C	-2.8654	8.1971	16.0269
C	-2.8234	9.196	17.0083
H	-3.2514	9.0408	17.8432
C	-2.1747	10.4061	16.7971

C	-1.5633	10.6136	15.5643
H	-1.103	11.4293	15.4061
C	-1.6098	9.6493	14.5533
C	-2.257	8.4248	14.7665
C	-2.1713	11.4673	17.9147
C	-1.3315	12.6568	17.541
H	-0.3993	12.3764	17.4299
H	-1.6616	13.0401	16.703
H	-1.3851	13.3299	18.2508
C	-1.6612	10.8464	19.2028
H	-1.6262	11.5306	19.904
H	-2.2638	10.1257	19.4793
H	-0.7624	10.4837	19.0574
C	-3.6077	11.9345	18.151
H	-3.9856	12.2672	17.3102
H	-4.1462	11.1838	18.4788
H	-3.6118	12.6536	18.818
Mg	-4.9108	4.784	17.423
N	-6.1913	5.3329	18.989
N	-6.2313	3.2568	16.8627
C	-7.0736	4.465	19.5115
C	-7.3883	3.22	18.9759
C	-7.0905	2.71	17.7149
C	-6.1033	6.6143	19.5388
C	-5.2681	6.9136	20.6622
C	-5.2636	8.3122	21.027
H	-4.7736	8.5625	21.7995
C	-5.8841	9.2562	20.3743
H	-5.7574	10.1638	20.6251
C	-6.6839	8.9521	19.3482
H	-7.194	9.6252	18.9149
C	-6.7571	7.628	18.932
C	-4.4436	5.7827	21.3035
H	-4.7422	4.9053	20.9272
C	-4.6632	5.7764	22.8398
H	-5.4932	6.2532	23.0536
H	-4.7251	4.8514	23.1562
H	-3.9083	6.2231	23.2788
C	-2.9537	6.0061	20.9472
H	-2.3923	5.4501	21.5287
H	-2.8009	5.7606	20.0123
H	-2.7244	6.9501	21.0783
C	-7.6158	7.4236	17.7064
H	-7.2357	6.6206	17.2446
C	-7.6771	8.4864	16.6603
H	-6.818	8.5324	16.19
H	-8.3884	8.2758	16.0189
H	-7.8644	9.3512	17.0821
C	-9.0631	7.0451	18.0569
H	-9.5764	6.9121	17.2332
H	-9.0666	6.2167	18.5814
H	-9.4709	7.7658	18.5814
C	-6.2311	2.7971	15.4916
C	-5.2963	1.8167	15.104
C	-5.3166	1.4904	13.6931
H	-4.7229	0.8268	13.3624



C	-6.1801	2.124	12.8152
H	-6.1418	1.9181	11.8888
C	-7.0517	3.0078	13.2484
H	-7.6615	3.3958	12.6327
C	-7.1071	3.3974	14.6052
C	-4.346	1.2006	16.0075
H	-4.4817	1.5839	16.9225
C	-2.8929	1.4984	15.5572
H	-2.6986	1.0042	14.7334
H	-2.7918	2.4598	15.3947
H	-2.2696	1.218	16.2612
C	-4.5211	-0.3326	16.0874
H	-3.6978	-0.7713	15.7881
H	-4.7129	-0.5924	17.0109
H	-5.2648	-0.6066	15.5087
C	-8.108	4.3953	15.023
H	-7.96	4.598	15.9904
C	-7.995	5.6925	14.2432
H	-8.1121	5.5103	13.2855
H	-8.6875	6.3165	14.5425
H	-7.1109	6.0884	14.3914
C	-9.5423	3.8567	14.8589
H	-9.639	3.0252	15.3691
H	-10.181	4.5204	15.1923
H	-9.7197	3.6809	13.9125
H	3.361	13.2744	8.8818
C	3.2355	14.4189	11.2261
H	4.2054	14.4574	10.7758
H	2.6639	15.2683	10.9152
H	3.3363	14.4273	12.2913
C	3.2212	10.9574	7.6964
H	2.6536	11.2862	6.851
H	4.1929	11.4046	7.6693
H	3.3179	9.8922	7.6686
H	-7.9342	2.5644	19.6217
C	-7.7945	4.8814	20.8071
H	-8.7616	4.4248	20.8397
H	-7.221	4.5631	21.6525
H	-7.9017	5.9458	20.8292
C	-7.8087	1.4199	17.2774
H	-7.2456	0.5703	17.6028
H	-8.7849	1.3894	17.7144
H	-7.8947	1.4037	16.211

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