Electronic Supplementary Information

Development of a panchromatic photosensitizer and its application to photocatalytic CO₂ reduction

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	E / eV	contributes		
		mbip	Os	mtpy
LUMO+3	-2.62	52	4	43
LUMO+2	-2.70	46	0	54
LUMO+1	-2.79	0	11	89
LUMO	-2.83	87	11	2
НОМО	-5.71	28	57	15
HOMO-1	-6.05	11	63	26
HOMO-2	-6.08	29	61	10

Table S1. DFT-calculated characteristics of frontier molecular orbitals of Os in ground state.



Figure S1. Single-photon counting plots of transient emission at 795 nm from DMA solution of **Os** (\bullet) and its single-exponential fitting (red line). LED pulse (456 nm; pulse duration: 1.3 ns) was used for the excitation light source.



Figure S2. (a) Emission spectra of CO_2 -saturated DMA solutions containing **Os** and BI(OH)H at five different concentrations. (b) Stern-Volmer plot of emission intensity from **Os** against BI(OH)H concentration. Excitation wavelength was 550 nm.



Figure S3. Emission spectrum of **Os** measured in frozen DMA by liquid nitrogen (red line) and its Franck-Condon fitting (blue line).



Figure S4. Normalized absorption (green line) and emission (red line) spectra of Os in DMA.



Figure S5. Cyclic voltammogram of BI(OH)H (5 mM) (red line) in an Ar-saturated DMA solution containing Et_4NBF_4 (0.1 M) as a supporting electrolyte with a Ag/AgNO₃ (10 mM) reference electrode and micro glassy-carbon working electrode ($\phi = 33 \mu m$). Scan rate was 200 V·s⁻¹. Black line shows the background current.



Figure S6. Photocatalytic formation of HCOOH: CO₂-saturated DMA solutions (3.5 mL) containing **Os** (50 μ M), **Ru(CO)** (50 μ M) and BI(OH)H (0.2 M) were irradiated at 480-nm light. Light intensity was 7.5 × 10⁻⁸ einstein s⁻¹.



Figure S7. Visible absorption spectra of **Os** (50 μ M, green line), **Ru(CO)** (50 μ M, red line) and BI(OH)H (0.2 M, black line) in DMA. Light-pass length was 10 mm.