

Boosting electrochemical properties of CoCo_2O_4 porous nanowire arrays by microwave-assisted synthesis for battery-supercapacitor hybrid devices

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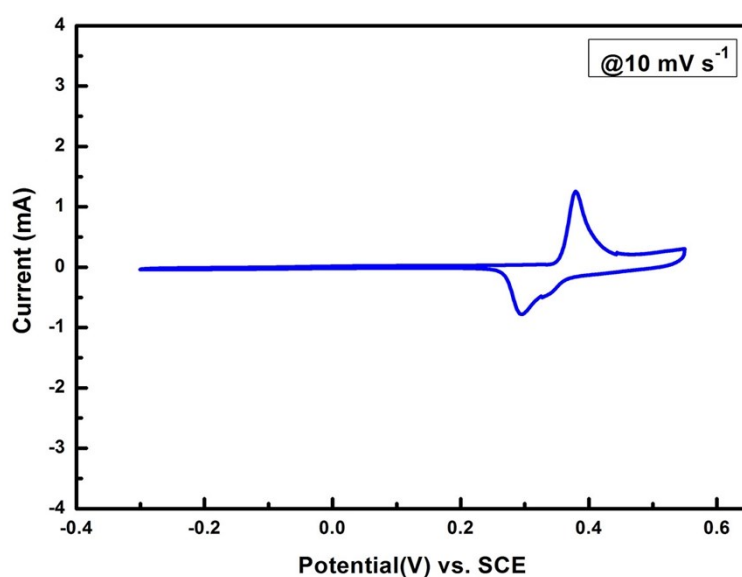


Fig. S1 CV curve at 10 mV s^{-1} of the bare Ni foam.

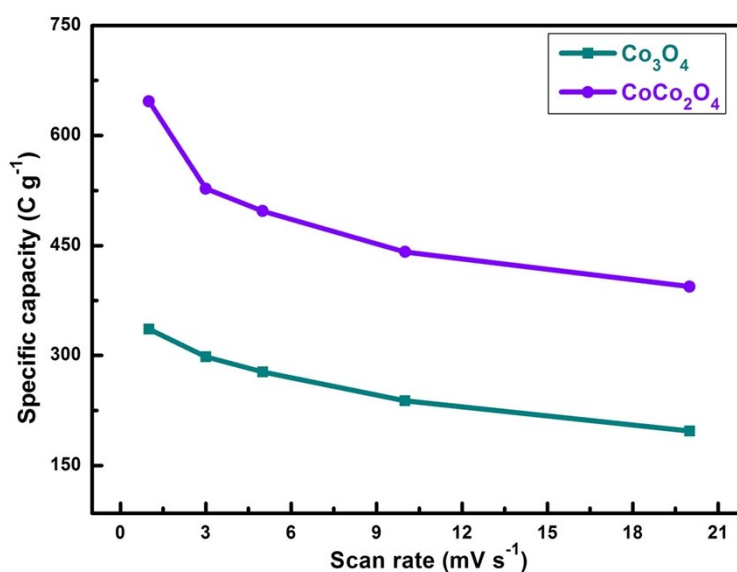


Fig.S2 Specific capacity measured of CoCo_2O_4 and Co_3O_4 electrodes at different scan rates.

The specific capacity (C g^{-1}) of the prepared cobalt oxide (Co_3O_4 and CoCo_2O_4) in

three-electrode system were measured by cyclic voltammetry (CV) using the following equation (S1).

$$Cs (C g^{-1}) = \frac{\int_{V_1}^{V_2} I(V) dV}{2mv} \quad (S1)$$

where I is the current; m is the mass; v is the potential scan rate and $\Delta V (V_2-V_1)$ is the potential window.

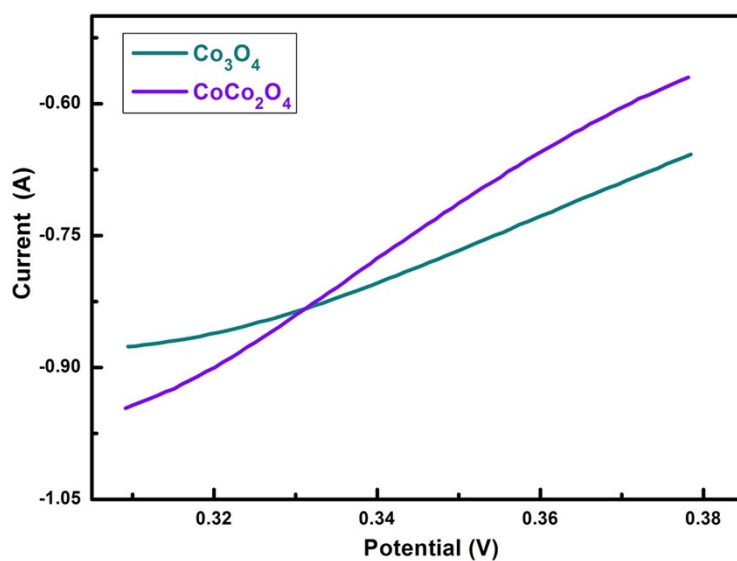


Fig.S3 I - V test curves of Co_3O_4 and $CoCo_2O_4$ electrode.

Tab.S1 Battery-supercapacitor hybrid device performances of Co₃O₄-based electrode materials prepared by different methods reported in the literature.

Single Electrode				Battery-supercapacitor hybrid device					
Structure of Co ₃ O ₄ based material	Method of synthesis	Specific capacitance	Retention	Assembly	Specific capacitance	Energy density (Wh/kg)	Power density (W/kg)	Retention	Reference
Co ₃ O ₄	Soak-adsorption	314 C g ⁻¹ /785 F g ⁻¹ (0.5A g ⁻¹)	93% (5000 cycles @ 5A g ⁻¹)	Co ₃ O ₄ //AC	/	13	257	85.9% (10000 cycles @ 5A g ⁻¹)	1
Co ₃ O ₄ /rGO-C	Template nanocasting technique	354.6 C g ⁻¹ / 1/709.1 F g ⁻¹ (1A g ⁻¹)	91.2% (6000 cycles @ 50mV s ⁻¹)	Co ₃ O ₄ /rGO-C//rGO	154.2 F g ⁻¹ (1A g ⁻¹)	48.2	750.5	88.7% (5000 cycles @ 50mV s ⁻¹)	2
C/ Co ₃ O ₄	Calcination	169.2 C g ⁻¹ / 1/423 F g ⁻¹ (1A g ⁻¹)	83% (2000 cycles @ 3A g ⁻¹)	C/Co ₃ O ₄ //AC	60.76 F g ⁻¹ (1A g ⁻¹)	21.1	790	/	3
Co ₃ O ₄ /NiCo ₂ O ₄	Hydrothermal	134.2 C g ⁻¹ / 1/335.5 F g ⁻¹ (0.5A g ⁻¹)	73.6% (5000 cycles @ 10A g ⁻¹)	Co ₃ O ₄ //AC	15 F g ⁻¹ (0.5A g ⁻¹)	5.2	26.5 W/kg	87.6% (3000 cycles @ 1A g ⁻¹)	4
Co ₃ O ₄ /MnO ₂	Annealing	292 F g ⁻¹ (4A g ⁻¹)	87.4% (5000 cycles @ 10A g ⁻¹)	Co ₃ O ₄ /MnO ₂ //AC	57.7 F g ⁻¹ (2A g ⁻¹)	46.2	/	84.5% (5000 cycles @ 10 A g ⁻¹)	5

Co ₃ O ₄	Hydrothermal	598.5 C g ⁻¹ (1A g ⁻¹)	93.7% (8000 cycles @ 10A g ⁻¹)	Co ₃ O ₄ //AC	99.0 F g ⁻¹ (1A g ⁻¹)	22.5	800	91.8% (10000 cycles @ 10 A g ⁻¹)	6
NiMn LDH@Co ₃ O ₄	Hydrothermal	243.2 C g ⁻¹ 1/607.9 F g ⁻¹ (0.5A g ⁻¹)	97% (1000 cycles @ 5A g ⁻¹)	NiMn LDH@Co ₃ O ₄ // activated graphene	97.3 F g ⁻¹ (0.5 A g ⁻¹)	26.49	350	73 % (5000 cycles @3 A g ⁻¹)	7
NiCo ₂ O ₄ / MWCNT/RGO	Hydrothermal	400.5 C g ⁻¹ 1//890 F g ⁻¹ (1A g ⁻¹)	91% (4000 cycles @ 5A g ⁻¹)	NiCo ₂ O ₄ / MWCNT/RG O//AC	94.2 F g ⁻¹ (1A g ⁻¹)	34.5	750	90.2% (5000 cycles @ 10A g ⁻¹)	8
Co ₃ O ₄	Electrochemical deposition	403 C g ⁻¹ /806 F g ⁻¹	90.8% (5000 cycles @ 15A g ⁻¹)	Co ₃ O ₄ //AC	81 F g ⁻¹ (1A g ⁻¹)	25.3	752	86.5% (4000 cycles @ 5A g ⁻¹)	9
Co ₃ O ₄ @MnO ₂	Hydrothermal	336 C g ⁻¹ /560 F g ⁻¹ (0.2A g ⁻¹)	95% (5000 cycles @ 5A g ⁻¹)	Co ₃ O ₄ @MnO ₂ //activated GO	49.8 F g ⁻¹ (1A g ⁻¹)	17.7	158	81.1% (10000 cycles @ 3A g ⁻¹)	10
CoCo ₂ O ₄	Microwave assisted hydrothermal	743.8 C g⁻¹ (1A g⁻¹)	89.7% (5000 cycles @ 15A g⁻¹)	CoCo₂O₄//m- CNT	80.2 F g⁻¹ (124.3 C g⁻¹) (1A g⁻¹)	26.8	775.4	82.3% (5000 cycles @ 3A g⁻¹)	This work

Tab.S2 Summary the values of different components in the fitting of EIS data.

Parameter	Value	
	Hydrothermal synthesis	Microwave synthesis
	Co ₃ O ₄	CoCo ₂ O ₄
R _s (Ω)	0.257	0.362
C _d (F)	1.137	1.085
R _{ct} (Ω)	1.119	0.712
W ₂ (Ω.s ^{-0.5})	1.710	0.979

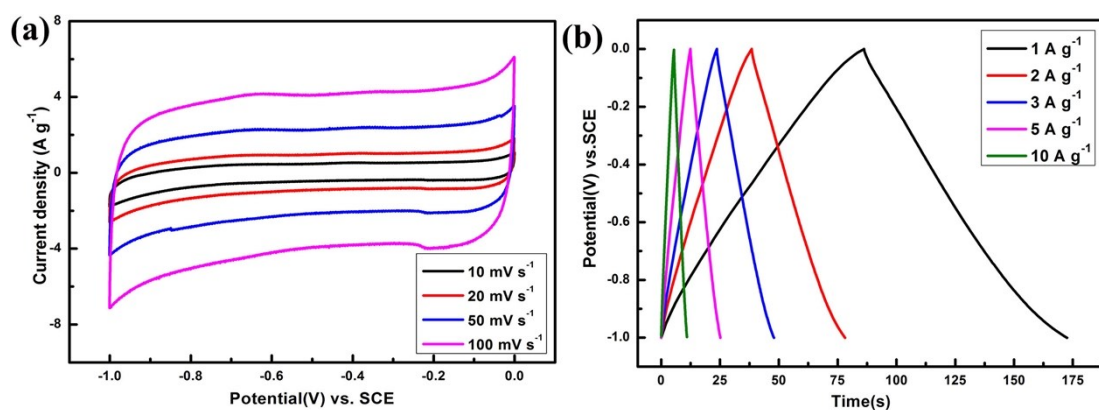


Fig.S4 CV curves at different scan rates (a), and GCD curves at different current densities (b) of the m-CNT electrode.

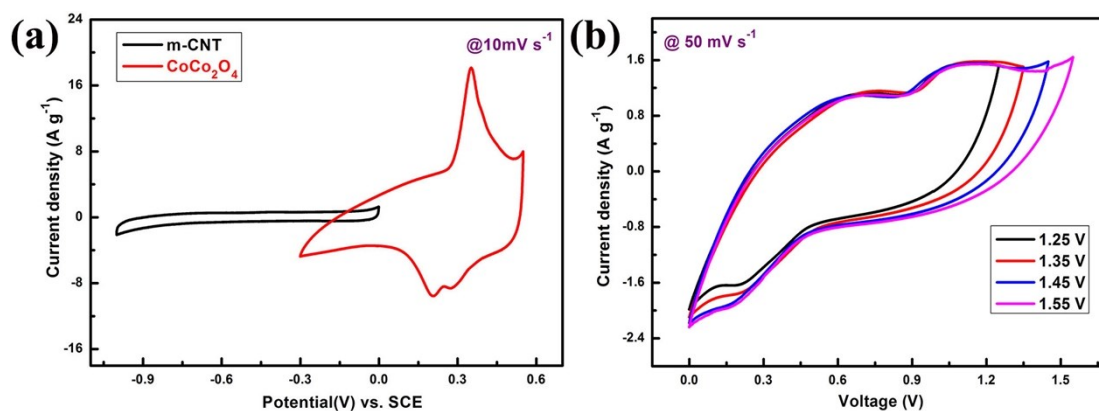


Fig.S5 (a) Comparative CV curves of m-CNT and CoCo₂O₄ electrodes in a three-electrode system at 10 mV s⁻¹. (b) CV curves of the BSH device measured at different potential windows at 50 mV s⁻¹.

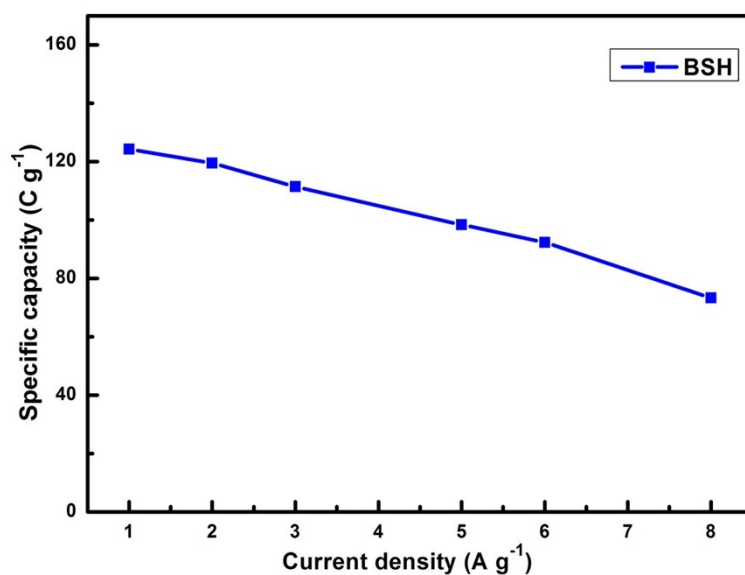


Fig.S6 Specific capacitance measured of CoCo₂O₄//m-CNT BSH at different current densities.

Tab.S3 Summary the values of different components in the fitting of EIS data.

Parameter	Value	Deviation
R ₁ (Ω)	0.591	0.126
C ₁ (F)	0.456	0.403
C ₂ (F)	0.041	0.097
R ₂ (Ω)	1.764	0.355
W ₂ (Ω.s ^{-0.5})	1.518	0.862

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