

## Supporting Information

### **A metal-organic framework approach to engineer mesoporous ZnMnO<sub>3</sub>/C towards enhanced lithium storage**

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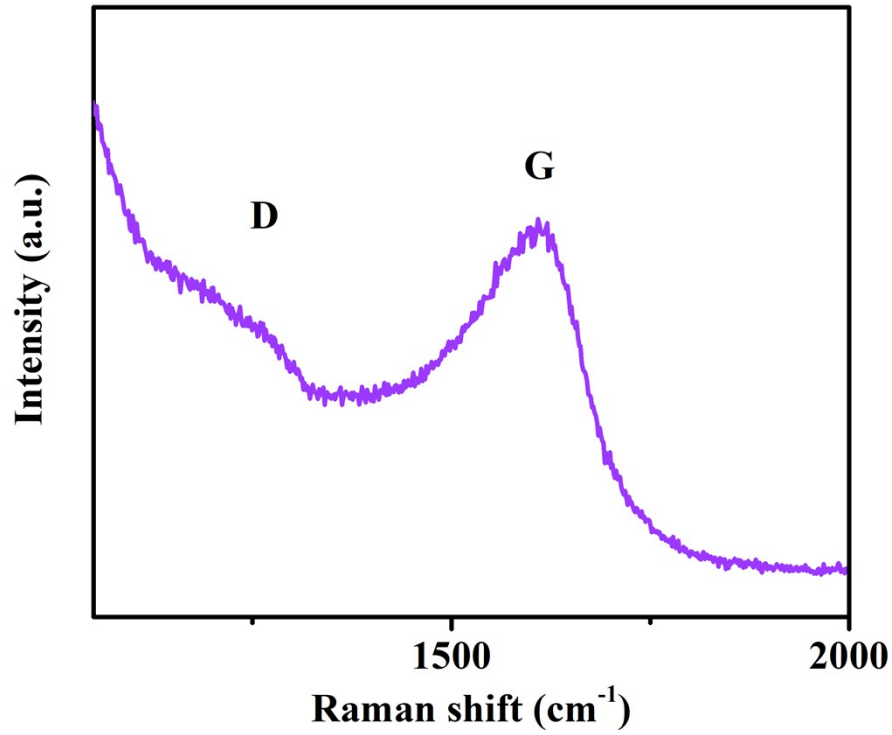


Fig. S1 Raman spectrum of ZnMnO<sub>3</sub>/C.

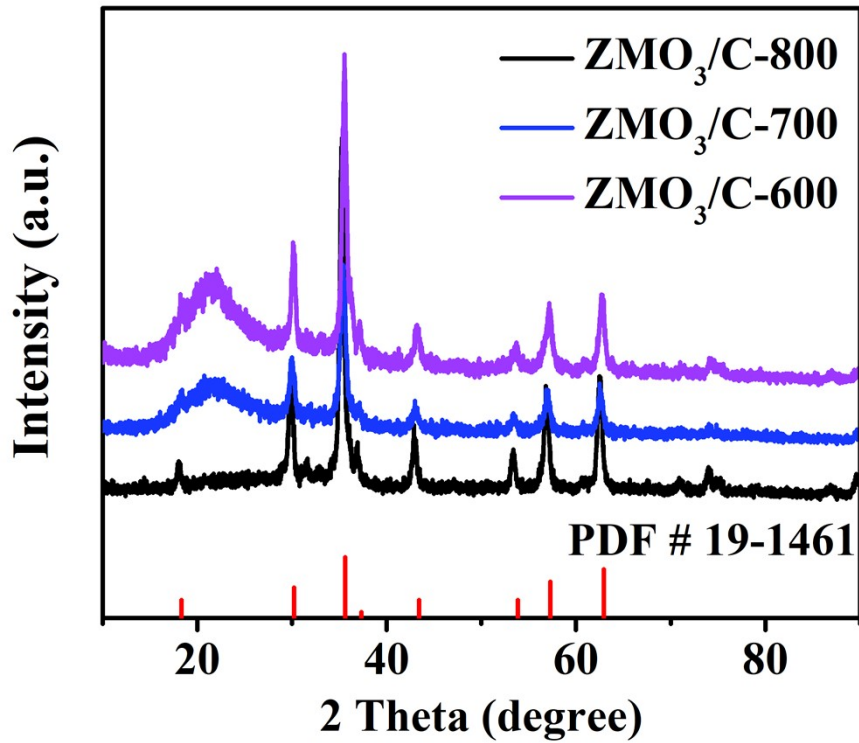


Fig. S2. XRD patterns of ZnMnO<sub>3</sub>/C calcinated at 600, 700 and 800 °C.

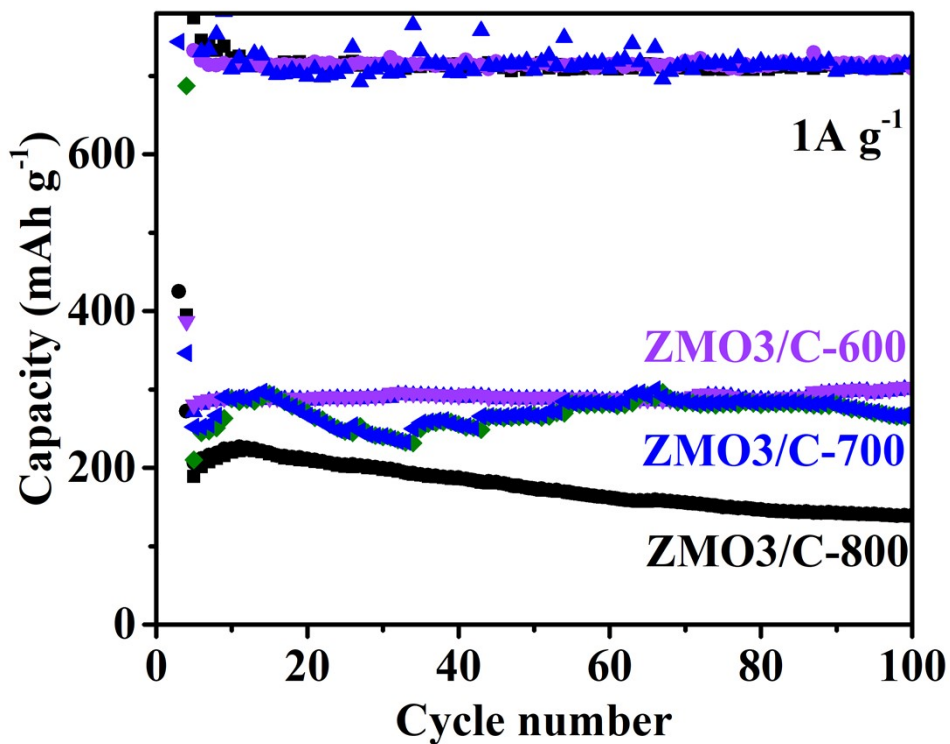


Fig. S3. Cycling performance of  $\text{ZnMnO}_3/\text{C}$  at  $1 \text{ A g}^{-1}$

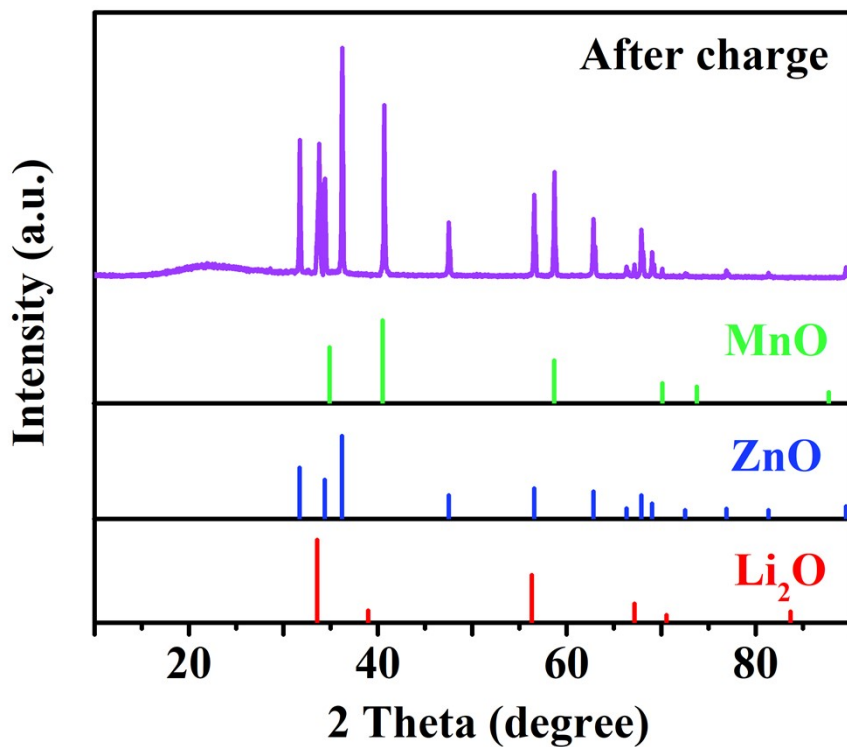


Fig. S4. XRD patterns of the  $\text{ZnMnO}_3/\text{C}$  electrode after 20 cycles of the charge process.

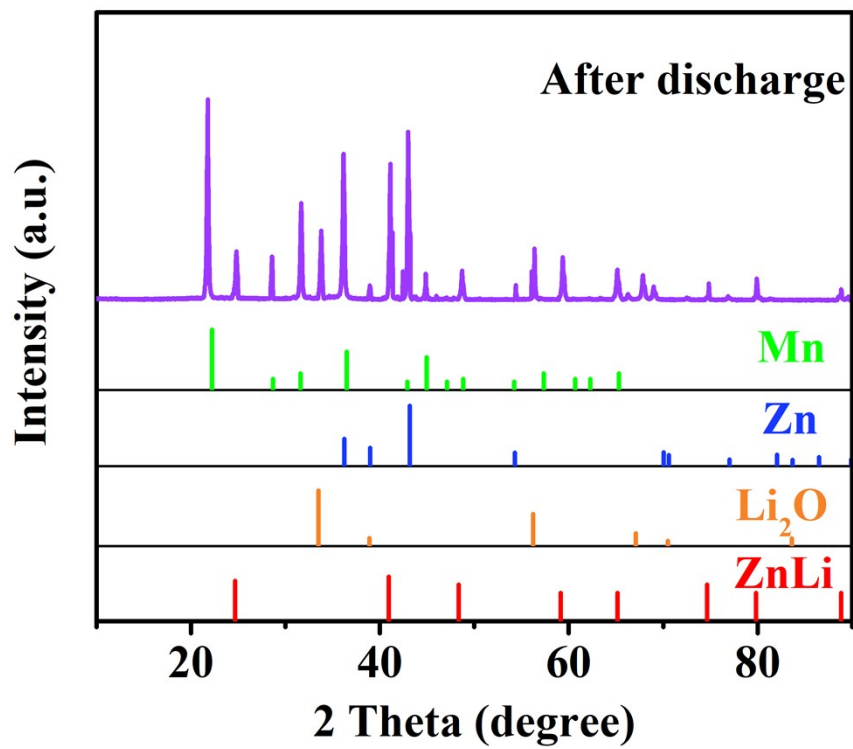


Fig. S5. XRD patterns of the ZnMnO<sub>3</sub>/C electrode after 20 cycles of the discharge process.

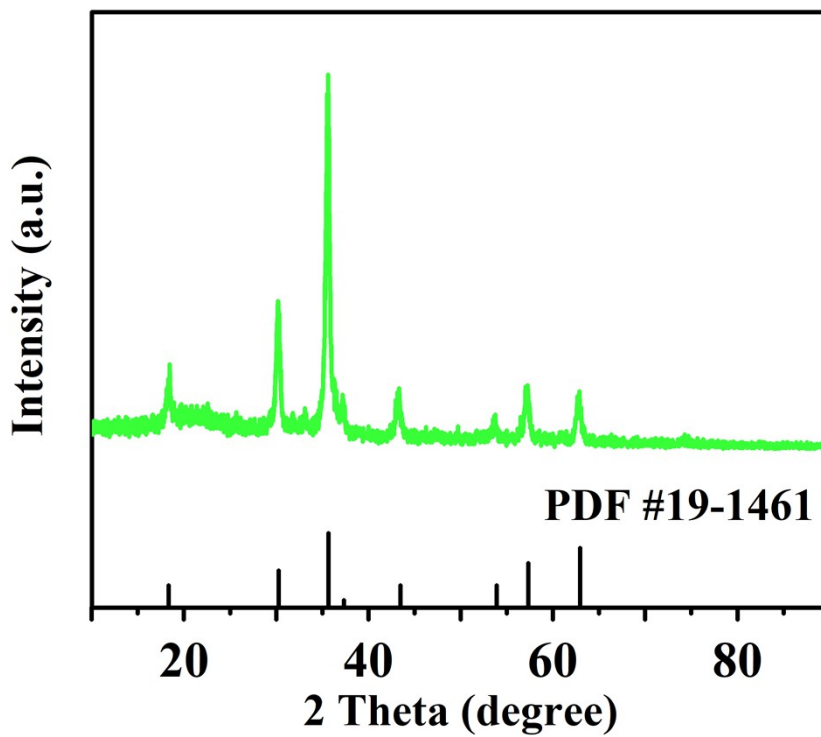


Fig. S6. XRD pattern of ZnMnO<sub>3</sub>.

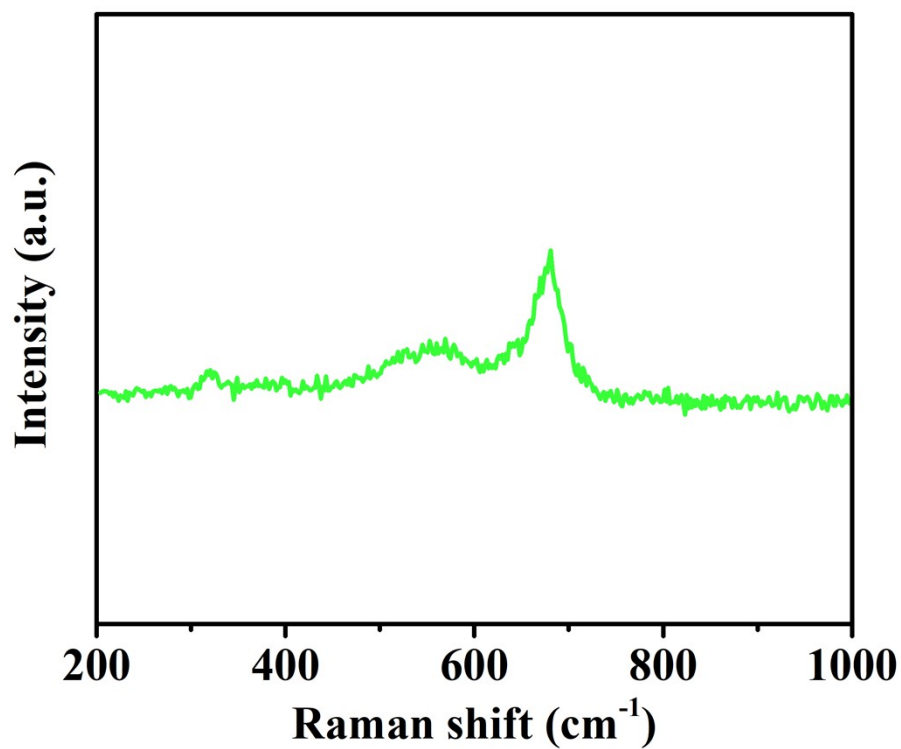


Fig. S7. Raman spectrum of ZnMnO<sub>3</sub>.

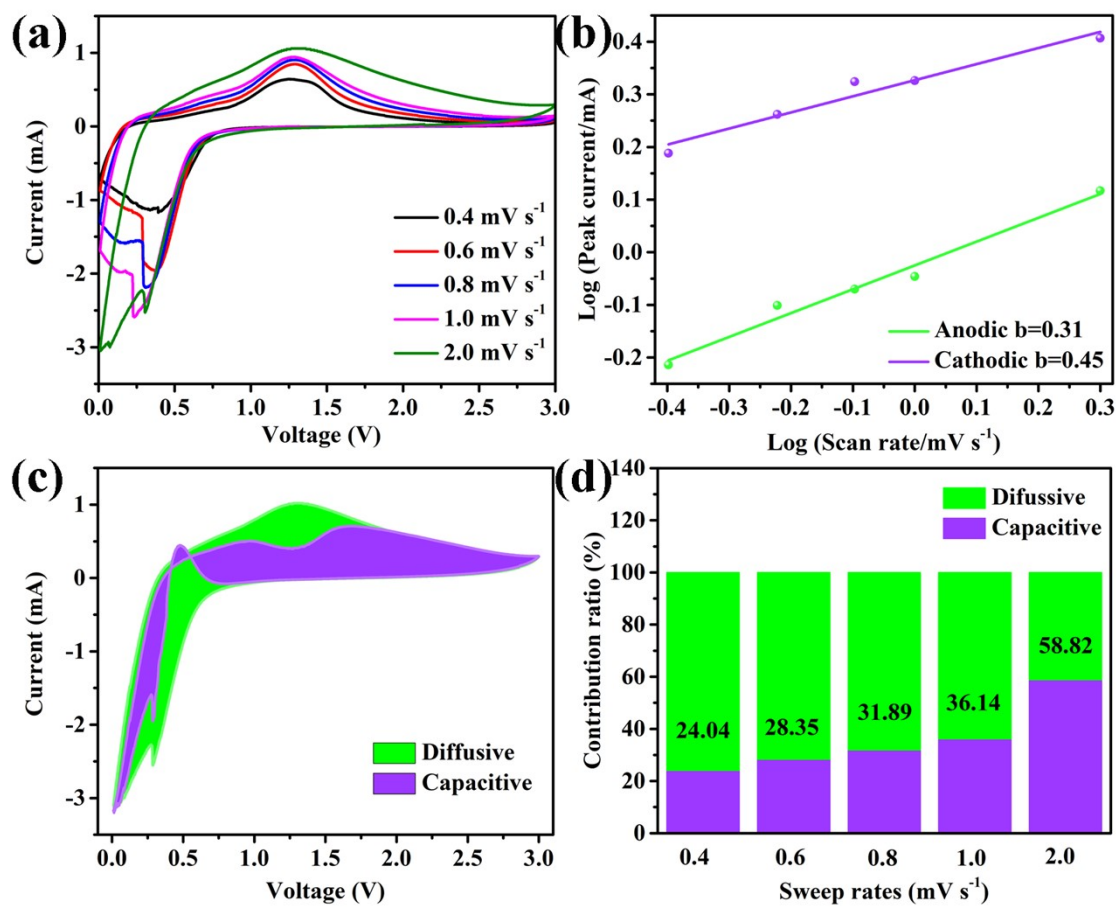


Fig. S8. (a) CV curves of ZnMnO<sub>3</sub> electrode at different sweep rates; (b) A linear relationship between log (scan rate) and log (peak current); (c) Typical capacitive contribution of ZnMnO<sub>3</sub> electrode at 2.0 mV s<sup>-1</sup>; (d) Contribution ratios of capacitance at different scan rates.

Table S1 The comparison of specific capacitance hybrid materials and other reported materials

Electrode materials	Cycle capacity (mAh g <sup>-1</sup> )	Cycle number	Ref
ZnMnO <sub>3</sub> porous spherulites	729/0.5 A g <sup>-1</sup>	50	S1
Hierarchical porous ZnMnO <sub>3</sub> yolk-shell microspheres	540/0.4 A g <sup>-1</sup>	300	S2
Porous ZnMnO <sub>3</sub>	560/0.4 A g <sup>-1</sup>	300	S3
Multi-shelled ZnMnO <sub>3</sub> hollow micro-spheres	290/0.4 A g <sup>-1</sup>	150	S4
1D ZnMnO <sub>3</sub>	382.9/0.8 A g <sup>-1</sup>	100	S5
MOF-derived ZnMnO <sub>3</sub> /C	460/1 A g <sup>-1</sup>	500	This work

#### References

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