

Table 1 - Examples of common biocides according to their chemical groups, common application, solubility, and limitations. (Supplementary material)

<i>General group</i>	<i>Applications</i>	<i>Limitations</i>	<i>Examples</i>	<i>Physical state and Octanol/water partition coefficient as log Pow</i>
<i>Alcohols</i>	Disinfectants, antiseptics, antimycotics, deodorants, and preservatives ^{1,2}		Ethanol	Polar liquid; Non-polar liquid
			isopropanol etc.	
<i>Aldehydes</i>	Disinfection and sterilization in many areas such as in the leather tanning industry, poultry industry, cosmetic field, micro-biological field, food industry and medical area ^{1,3,4}	Usually commercially obtained as an acidic solution and it is activated (made alkaline) before the use; incompatible with strong oxidizers and strong bases	Glutaraldehyde (pentanedial)	Liquid -0.22
		Less active under acid conditions	Formaldehyde (methanal)	Flammable gas (often stored in aqueous solution) 0.35
		Less active under acid conditions	Ortho-phthalaldehyde (benzene-1,2-dicarboxaldehyde)	Solid 0.99 @ 30°C
<i>Amphoteric surfactants</i>	Commonly available household cleaning/hygiene agents ⁵	Less active under acid conditions. Ideal: Neutral/alkaline conditions	Cocamidopropyl betaine	Liquid -1.28 to -3.63 @ 25°C
			Dodecyl-di (aminoethyl)-glycine	
			N-Alkyl aminopropyl glycine	
<i>Anionic surfactants</i>	Commonly available household cleaning/hygiene agents ⁵	Effective only at low pH; Neutralized by detergent residues; Corrosive to soft metals, mild steel	Sodium dodecyl sulfate	
			alkylbenzene sulfonate (LAS)	
			Sodium lauryl sulfate (SLS)	Solid 1.6
<i>Cationic surfactants</i>	Disinfectants; highly effective for the inactivation of SARS-CoV-2 ⁶	Neutralized by certain surfactants. Ideal: Neutral/alkaline conditions	Sodium dodecyl benzene sulfonate (SDBS)	0.45
			benzalkonium chloride	0.96-2.93 (C8-C12)
			Methyldimethyl ammonium chloride	
<i>Halogens (or halogen releasing compounds)</i>	Food industries and sanitization of equipment; water treatment ⁷⁻¹²	Acid conditions; Vapor phase corrosion	Chlorine	Gas (stable active halogen liquid; chlorine tablets)
			Sodium hypochlorite (SH)	liquid - 3.42 @ 20°C
			Chlorine dioxide (CD)	Gas (tablets) -3.22
		Stabilized form slower and less		

		active under alkaline/neutral conditions	Sodium dichloroisocyanurate (NaDCC)	Solid -0.0056
			Sodium bromide	water-soluble solid
			Bromine-chlorine-dimethylhydantoin (<i>BCDMH</i>)	Solid 0.32-0.40
		Corrosive in presence of chlorine ions; Rapid decomposition at high temperatures; rapid decomposition by metals, organic matter. Ideal: Acid conditions	Hydrogen peroxide	Liquid -1.36
<i>Peroxygens</i>	Cleaning; antiseptics; air, water, and surface disinfection (including device and food surfaces) ^{1,9,11,13}		Peracetic acid	Liquid -0.52
	Preservative for industrial emulsions, adhesives, polishes, glues, household products, paper products, cutting oils ^{1,14,15}		1,2-benzisothiazolin-3-one (BIT)	Solid 0.7 @ 20°C
<i>Isothiazolinones</i>			2-methyl-4-isothiazolin-3-one (MIT)	Solid -0.49
			5-Chloro-2-methyl-4-isothiazolin-3-one (CMIT)	Solid 0.53
			2-n-Octyl-4-isothiazolin-3-one (OIT)	Liquid >3.1 (20°C)
			4,5-dichloro-2-octyl-4-isothiazolin-3-one (DCOIT)	Solid 4.68
	Disinfectants (including aerial ones), antiseptics, surgical scrubs, toilet soaps, cosmetics, textiles and cutting oils, herbicides and fungicide ^{1,16,17}		2,4-dichlorophenoxyacetic acid (2,4-D)	Solid 2.81
<i>Phenolics</i>		Some surfactants can "over-solubilize" them reducing their effective availability	2,4-Dichlorophenol	Solid 3.06
			2,4,5-trichlorophenol	Solid 3.72
			5-chloro-2-(2,4-dichlorophenoxy)phenol (triclosan)	Solid 4.8
			Hexachlorophene [2,2'-methylenebis(3,4,6-trichlorophenol)]	Solid 7.54
<i>Polymeric biguanides</i>	Disinfectant and antiseptic	Incompatible with anionic surfactants	1,6-bis(4-chloro-phenyl biguanide)hexane (chlorhexidine)	Solid -1.1
			Alexidine	Water soluble solid