Novel core-shell structured MgH₂/AlH₃@CNTs nanocomposites with extreme high dehydriding-rehydriding properties derived from nanoconfinement



Fig. S2 temperature programmed desorption of a) MgH_2 , b) AlH_3 . Samples were heated from room temperature to 500 °C with a rate of 10 °C/min.





Fig. S3 SEM images of various CNTs: a)CNTs4, b)CNTs1, c)CNTs2, d)CNTs3.



Fig. S5 XRD patterns of as-milled MgH₂/AlH₃@CNTs1 obtained by milling with different ball to powder ratio and rotating speed for 1h



Fig. S6 SEM images of MgH₂/AlH₃ wrapped with various CNTs:a)MgH₂/AlH₃@CNTs4, b)MgH₂/AlH₃@CNTs1, c)MgH₂/AlH₃@CNTs2, d)MgH₂/AlH₃@CNTs3

As can be seen in Fig. S6a, most MgH₂/AlH₃particles wrapped with CNTs4 after milling for 1 h have a size of 60-80 nm, which suggests that the CNTs4 leads to particle size refining of the nanocomposite. With the CNTs transforming from 4 to 1, it is obvious that powder agglomeration were observed, with the size of most particles being in the range of 120-150 nm, as shown in Fig. S6b. For MgH₂/AlH₃@CNTs2 and MgH₂/AlH₃@CNTs3, severe agglomeration or aggregation, a phenomenon somewhat like sintering of the powder particles was observed, the morphology of the particles became irregular and difficult to identify, and the size of most particles was estimated to be in the range of 200-250 nm, as shown in Fig. S6c and d.



Fig. S7 SEM images of MgH₂/AlH₃ nanoparticles





Fig. S10 Absorption property of commercial MgH2 at 250 °C under 5MPa of H2.