

Supporting Information

A Facile and Broadly Applicable CdBr₂-Passivating Strategy for Halide Migration-Inhibiting Perovskite Films and High- Performance Solar Cells

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Experimental section

Characterizations

Thin Film Characterization: UV-vis absorption spectra of perovskite film on the glass were recorded using a Perkin Elmer Lambda 950 UV/VIS spectrophotometer over the 300-900 nm wavelength range. PL and TRPL were obtained on an Edinburgh Instruments FLS 980 with an excitation wavelength of 470 nm. The crystallinity of the perovskite films was characterized using powder X-ray diffractometer (Rigaku, D/max 2500 VB25/PC) with CuK α radiation ($\lambda = 0.15406$ nm) at a scan rate of $5^\circ \cdot \text{min}^{-1}$. The perovskite film morphology was analyzed using scanning electron microscope (SEM, JEOL-7610F) and atom force microscopy (AFM, Bruker Metrology Nanoscope III-D). The energy-dispersive spectroscopy (EDS) was measured using an X-ray energy dispersive spectrometer mounted on a JEOL-7610F microscope. The XPS and UPS measurements were carried out by Kratos X-ray photoelectron spectrometer (ThermoFisher ESCALAB 250Xi) equipped with monochromatic Al K α (1,486.6 eV) and non-monochromatic HeI (21.22 eV) sources.

Device Characterization: The current density-voltage (J - V) characteristics were measured using a solar simulator (Oriel Sol3A) with the Keithley 2400 source meter under AM1.5G. The J - V curves were measured in the ambient atmosphere with a scan rate of 50 mV s^{-1} . The external quantum efficiency (EQE) spectrum was measured using an EQE system (Oriel instruments) with 150 W xenon lamp (USHIO, Japan) as a source of monochromatic light. The electrochemical impedance spectroscopy (EIS) measurement and other electrochemical testing were measured by using an electrochemical workstation (Zennium 400147). The EIS measurements were recorded on a Zahner electrochemical work station set to a frequency ranging from 10^6 Hz to 10

Hz in the dark state. All SCLC tests were measured at room temperature under dark conditions.

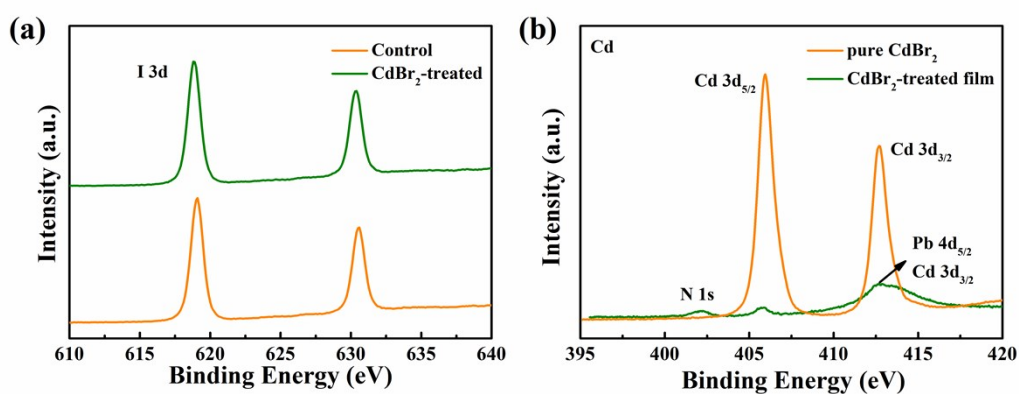


Fig. S1 (a) I 3d for perovskite films with and without the CdBr₂ treatment. (b) Cd 3d for CdBr₂-treated perovskite film and pure CdBr₂.

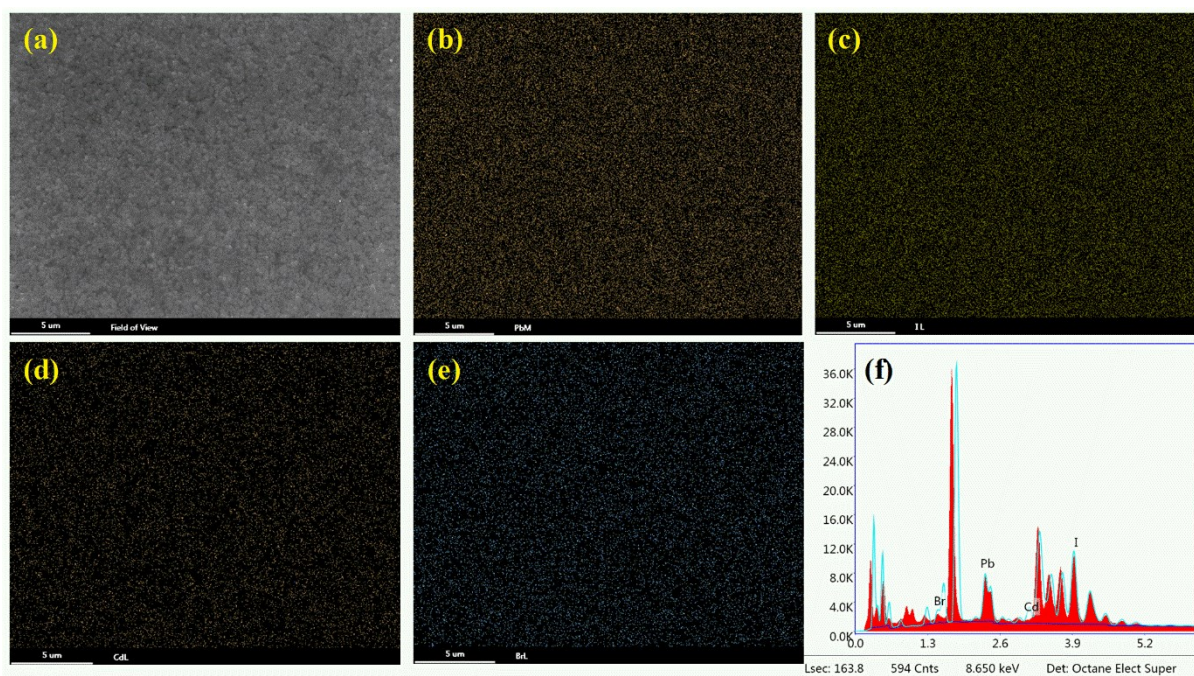


Fig. S2 SEM-EDS mappings of Pb, I, Cd and Br elements for the CdBr₂ modified perovskite film deposited onto the ITO substrate.

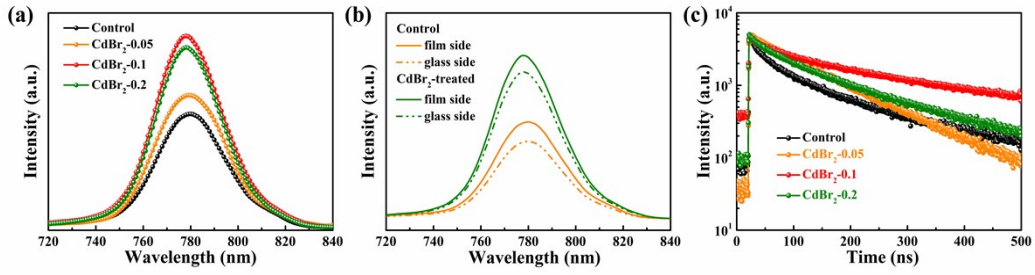


Fig. S3 (a) Steady state PL spectra of the perovskite films depending on CdBr₂ concentration. (b) Steady state PL spectra collected from the film and glass side of the perovskite film without and with CdBr₂-treated. (c) time-resolved PL decay spectra of the perovskite films depending on CdBr₂ concentration.

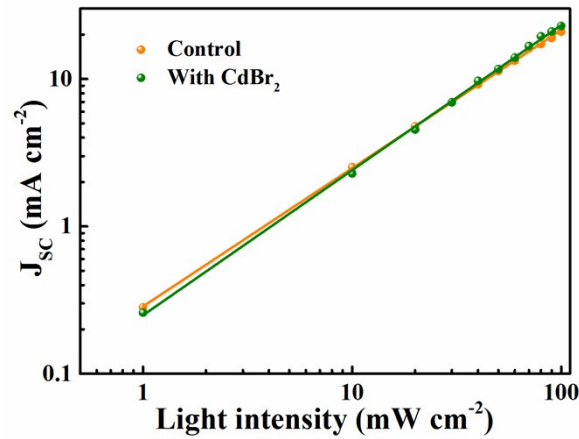


Fig. S4 J_{SC} versus light intensity for the PSCs with and without CdBr₂ treatment.

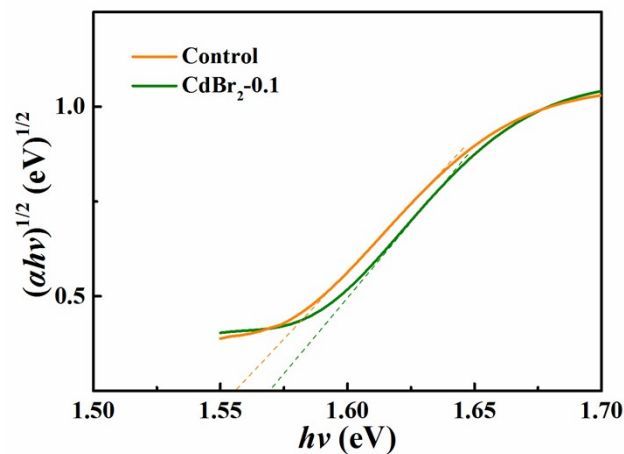


Fig. S5 $(\alpha hv)^{1/2}$ as a function of photo energy of perovskite films with and without CdBr_2 -treated.

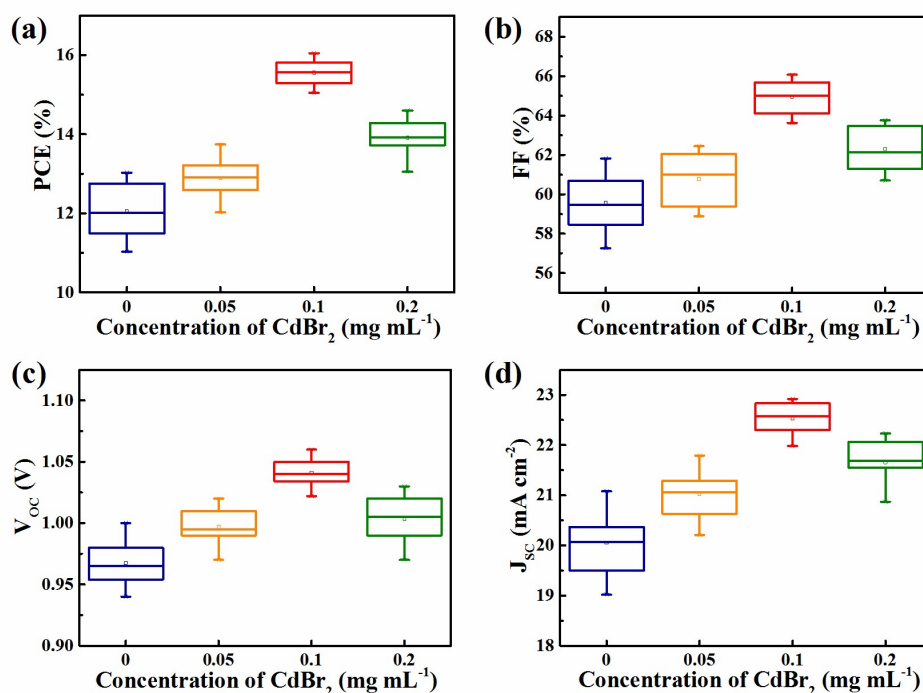


Fig. S6 Photovoltaic parameters statistics distribution of (a) PCE, (b) FF, (c) V_{oc} , (d) J_{sc} for the devices prepared from the four different perovskite films. (40 devices were collected from the different batch).

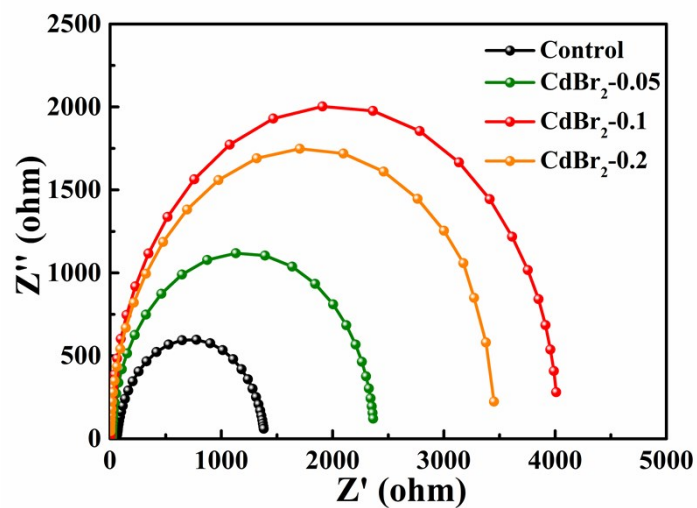


Fig. S7 Nyquist plots at bias 0.80 V of the perovskite devices with different concentration CdBr₂-treated under dark conditions.

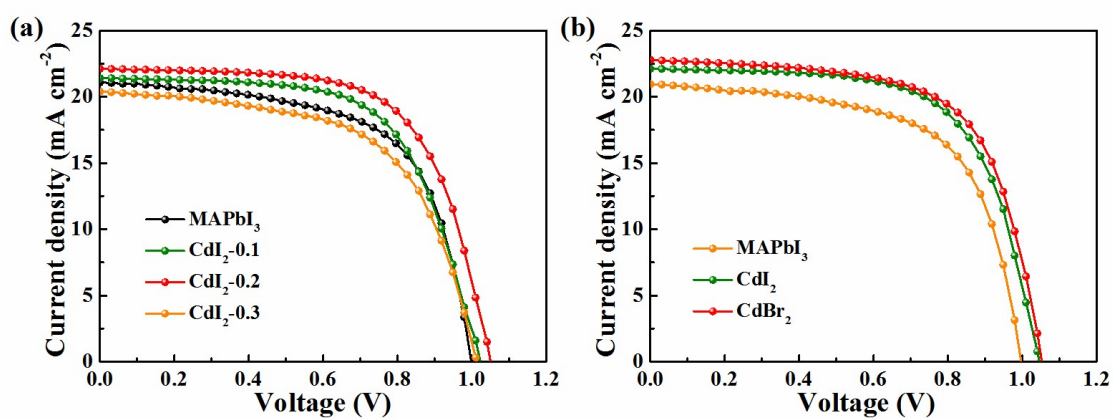


Fig. S8 (a) J - V curves of PSCs with different concentration CdI₂-treated. (b) J - V curves of the champion PSCs with CdBr₂ and CdI₂ treatment.

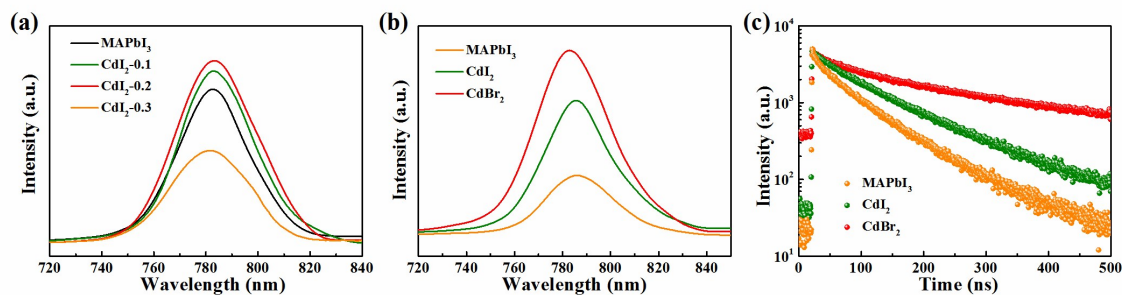


Fig. S9 (a) PL spectra of the perovskite films depending on CdI₂ concentration. (b) PL spectra of the control, CdI₂ and CdBr₂-treated perovskite films. (c) TRPL spectra of the perovskite films depending on CdBr₂ concentration.

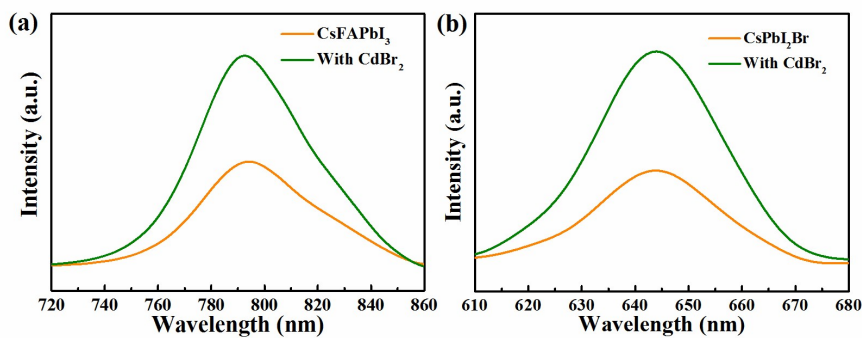


Fig. S10 PL spectra of the control and CdBr₂-treated (a) CsFAPbI₃ perovskite film and (b) CsPbI₂Br perovskite film.

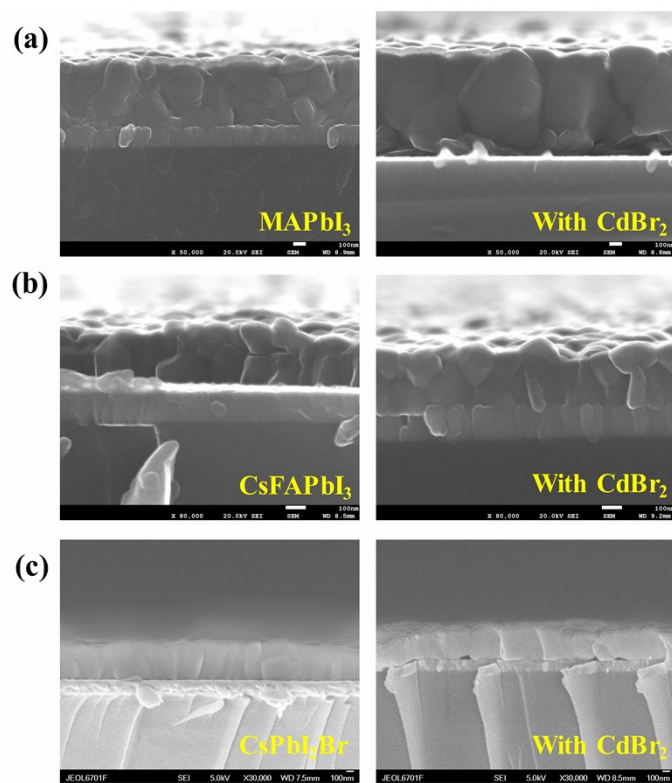


Fig. S11 The cross-sectional SEM images of (a) MAPbI₃ films, (b) CsFAPbI₃ films and (c) CsPbI₂Br films without or with CdBr₂-treated.

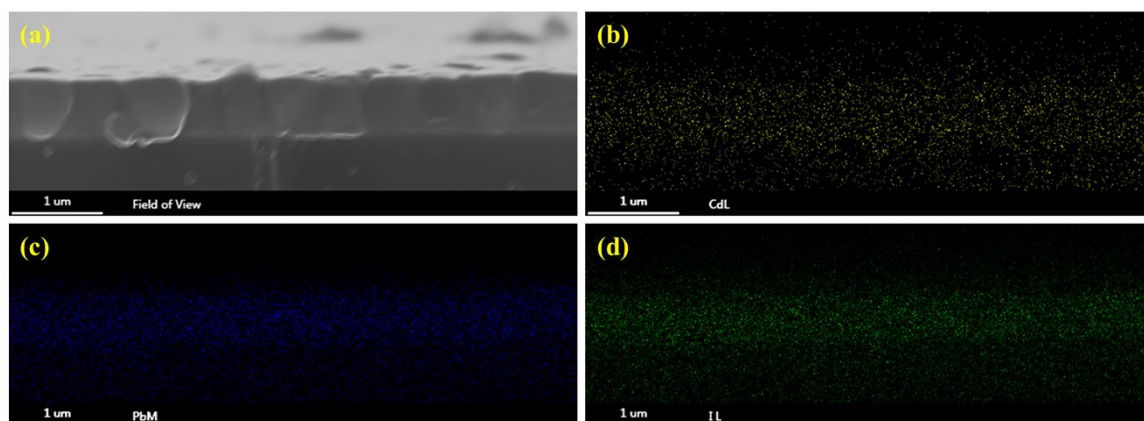


Fig. S12 Cross-sectional SEM-EDS mappings of Cd, Pb and I elements for the CdBr₂-treated perovskite film.

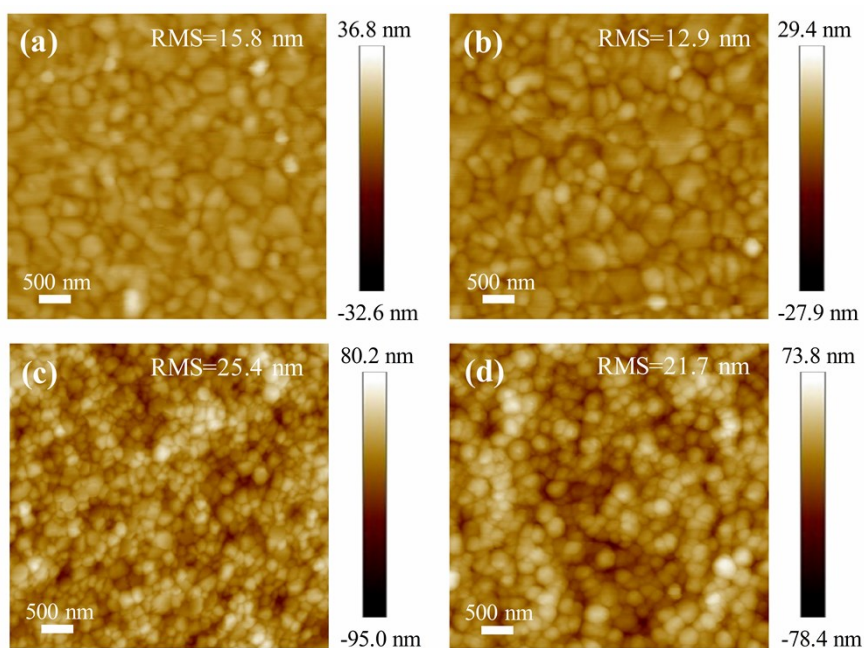


Fig. S13 AFM images of (a and b) MAPbI₃ perovskite films without and with CdBr₂-treated, (c and d) CsFAPbI₃ perovskite films without and with CdBr₂-treated.

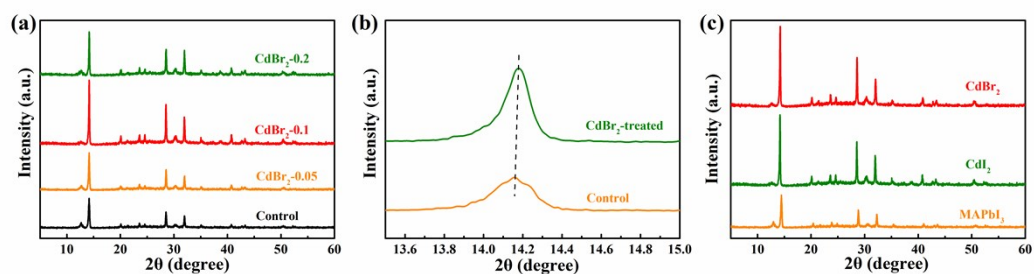


Fig. S14 (a) XRD patterns of the different concentration CdBr₂-treated perovskite films. (b) The (110) peak of CdBr₂-treated perovskite film. (c) XRD patterns of the control and CdBr₂- or CdI₂-treated perovskite films.

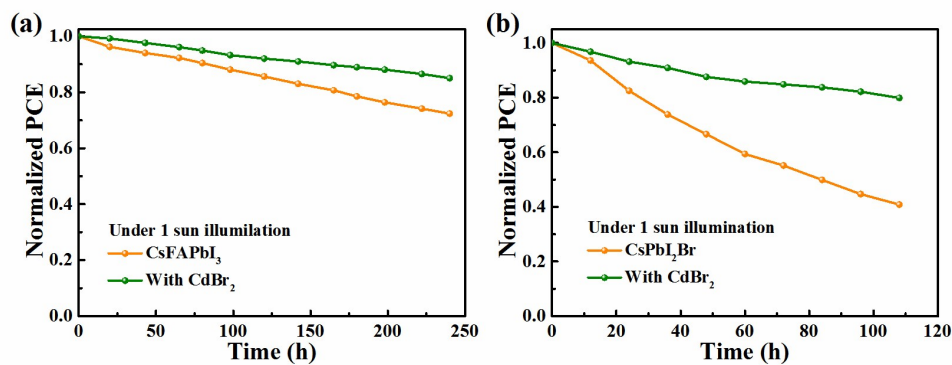


Fig. S15 Long-term stability measurements of (a) CsFAPbI₃-based devices and (b) CsPbI₂Br-based devices without encapsulation under one sun illumination at room temperature in N₂.

Table S1 The specific percentage of X/Pb atomic ratio.

Samples	Name	Atomic%
Control	Pb	15.33
	I	39.24
Treated	Pb	15.58
	I+Br	43.78

Table S2 Parameters of the TRPL spectroscopy based on the perovskite films with different concentrations of CdBr₂-treated.

Samples	τ_1	% of τ_1	τ_2	% of τ_2	τ_{ave}
Control	28.26	14.34	135.90	85.66	120.46
CdBr₂-0.05	43.90	6.65	118.09	93.35	113.16
CdBr₂-0.1	36.11	6.61	221.47	93.39	209.22
CdBr₂-0.2	35.58	10.19	151.49	89.81	139.68

Table S3 Photovoltaic parameters of C-PSCs with or without CdBr₂ treatment measured in both reverse and forward scan.

PSCs		V_{oc} (V)	J_{sc} (mA cm ⁻²)	FF (%)	η (%)	HI
Control	RS	1.00	21.08	61.82	13.03	14.2%
	FS	0.96	20.06	58.14	11.19	
CdBr ₂ -0.1	RS	1.06	22.92	66.08	16.05	5.2%
	FS	1.05	22.30	64.95	15.21	

*The hysteresis index (HI) was calculated to quantify the hysteresis level according to the following equation:

$$HI = \frac{PCE_{RS} - PCE_{FS}}{PCE_{RS}}$$

Table S4 Photovoltaic parameters of PSCs with different concentration CdI₂ treatment.

PSCs	V_{oc} (V)	J_{sc} (mA cm ⁻²)	FF (%)	η (%)
Control	1.00	20.94	61.70	12.91
CdI ₂ -0.1	1.02	21.41	63.20	13.86
CdI ₂ -0.2	1.04	22.24	64.35	14.88
CdI ₂ -0.3	1.01	20.39	59.72	12.54

Table S5 Parameters of the TRPL spectroscopy based on the different perovskite films without or with CdBr₂-treated.

Samples	τ_1	% of τ_1	τ_2	% of τ_2	τ_{ave}
CsFAPbI ₃	14.27	33.61	33.38	66.39	26.96
CsFAPbI ₃ -CdBr ₂	21.79	37.77	44.11	62.23	35.68
CsPbI ₂ Br	5.60	12.39	43.84	87.61	39.10
CsPbI ₂ Br-CdBr ₂	1.29	0.47	61.09	99.53	60.80

Table S6 Summary of key related information about the PSCs (structure similar to this work)

Perovskite component	PSC structure	PCE (%)	Date	Ref
MAPbI ₃	ITO/SnO ₂ /PVK/Co ₃ O ₄ @NC/C	14.63%	2021.02	1
Cs _{0.05} (FA _{0.85} MA _{0.15}) _{0.95} Pb(I _{0.85} Br _{0.15}) ₃	FTO/SnO ₂ /PVK/MWCNT(1 cm ²)	11.2%	2021.02	2
MAPbI ₃	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	11.91%	2020.06	3
MAPbI ₃	FTO/TiO ₂ /PVK/C	14.10%	2020.07	4
MAPbI ₂ Cl	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	11.70%	2019.01	5
CsFA _{0.83} MA _{0.17} PbI _{2.53} Br _{0.47}	FTO/SnO ₂ /PVK/CuSCN/C	15.3%	2020.03	6
MAPbI ₃	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	15.21%	2019.03	7
FA _{0.8} Cs _{0.2} PbI _{2.64} Br _{0.36}	FTO/c-TiO ₂ /m-TiO ₂ /PVK/PEO/C	14.9%	2019.04	8
FA _{0.85} MA _{0.15} PbI _{2.85} Br _{0.15}	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	14.41%	2019.01	9
MAPbI ₃	ITO/HMB-C ₆₀ /PVK/C	16.03%	2019.06	10
MAPbI ₃	FTO/c-TiO ₂ /m-TiO ₂ /PVK/NiO/C	13.6%	2019.05	11

$\text{Cs}_{0.05}\text{FA}_{0.81}\text{MA}_{0.14}\text{PbI}_{2.55}\text{Br}_{0.45}$	ITO/SnO ₂ (QDs)/PVK/C	13.64%	2020.03	12
MAPbI ₃	FTO/TiO ₂ /PVK-SWCNT/SWCNT-C	15.73%	2019.01	13
$\text{Cs}_x(\text{MA}_{0.7}\text{FA}_{0.3})_{1-x}\text{PbI}_3$	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	15.03%	2019.01	14
$\text{Cs}_{0.05}\text{FA}_{0.81}\text{MA}_{0.14}\text{PbI}_{2.55}\text{Br}_{0.45}$	FTO/c-TiO ₂ /m-TiO ₂ /PVK/C	15.09%	2020.06	15
MAPbI ₃	FTO/ZnO-RGO-CuInS ₂ /PVK/Au	15.74%	2020.06	16
MAPbI ₃	ITO/SnO ₂ /PVK/CdBr ₂ /C	16.05%		This work

Notes and references

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