Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A. This journal is © The Royal Society of Chemistry 2021

Electronic Supplementary Information for

Carbon black nanoparticle trapping: A strategy to realize the true energy storage potential of redox-active conjugated microporous polymers

Chang Wan Kang,^a Yoon-Joo Ko,^b Sang Moon Lee,^c Hae Jin Kim,^c Jaewon Choi, *^d and Seung Uk Son *^a

^a Department of Chemistry, Sungkyunkwan University, Suwon 16419, Korea

E-mail: sson@skku.edu

^b Laboratory of Nuclear Magnetic Resonance, National Center of Inter-University Research Facilities (NCIRF), Seoul

National University, Seoul 08826, Korea

^cKorea Basic Science Institute, Daejeon 34133, Korea

^d Department of Chemistry and Research Institute of Natural Science,

Gyeongsang National University, Jinju 52828, Korea

E-mail: cjw0910@gnu.ac.kr

Fig. S1. SEM images of CB, CB@CMP-DMBs, CMP-DMB, CB-O, CB@CMP-BQs, and CMP-BQ.



Fig. S2. TEM images of CB@CMP-DMB1, CB@CMP-DMB2, CB@CMP-BQ1, and CB@CMP-BQ2, compared with those of CB, CB@CMP-DMB3, CMP-DMB, CB-O, CB@CMP-BQ3, and CMP-BQ.



Fig. S3. TEM images of a mixture of CMP-DMB and CB@CMP-DMB that prepared by the Sonogashira coupling of 1,3,5-triethynylbenzene (0.13 mmol) and 1,4-diiodo-2,5-dimethoxybenzene (0.20 mmol) in the presence of CB (12.5 mg).



Fig. S4. N₂ adsorption-desorption isotherm curves (obtained at 77K) and pore size distribution diagrams (based on the DFT method) of CB@CMP-DMB1, CB@CMP-DMB2, CB@CMP-BQ1, and CB@CMP-BQ2.



Fig. S5. IR spectra of CB@CMP-DMB1, CB@CMP-DMB2, CB@CMP-BQ1, and CB@CMP-BQ2.







Fig. S6. Solid state ¹³C NMR spectra of CB@CMP-DMB1, CB@CMP-DMB2, CB@CMP-BQ1, and CB@CMP-BQ2.

Fig. S7. PXRD patterns of CB@CMP-DMBs, CB@CMP-BQs, CMP-DMB, CMP-BQ, CB, and CB-O.





Fig. S8. TGA curves of CB@CMP-DMBs, CB@CMP-BQs, CMP-DMB, and CMP-BQ.

Fig. S9. (a) Cyclic voltammograms and (b) charge-discharge profiles of CB@CMP-BQ1 and CB@CMP-BQ2.



Fig. S10. Scan rate-dependent cyclic voltammograms of CB@CMP-BQ1, CB@CMP-BQ2, CB@CMP-BQ3, CB-O, and CMP-BQ.



Fig. S11. Current density-dependent charge-discharge profiles of CB@CMP-BQ1, CB@CMP-BQ2, CB@CMP-BQ3, CB-O, and CMP-BQ.



Fig. S12. (a) An equivalent circuit model used for the analysis of EIS results, (b) the fitted Nyquist plots, (c) equivalent circuit parameters of CB-O, CB@CMP-BQs, and CMP-BQ.



Fig. S13. (a) Photographs of pellets and (b) conductivity measurements of CB-O, CB@CMP-BQs, and CMP-BQ.





Fig. S14. IR spectra of CB@CMP-BQ3 before and after 10000 cycles.

Entry	Materials	$S_{\rm BET}{}^{ m a}$	V _{mic} ^b	
		(m^2/g)	(cm^3/g)	
1	СВ	68	0.00	
2	CB@CMP-DMB1	363	0.07	
3	CB@CMP-DMB2	452	0.13	
4	CB@CMP-DMB3	524	0.15	
5	CMP-DMB	609	0.16	
6	CB-O	78	0.00	
7	CB@CMP-BQ1	173	0.02	
8	CB@CMP-BQ2	229	0.02	
9	CB@CMP-BQ3	301	0.07	
10	CMP-BQ	313	0.07	

Table S1. Porous parameters of CB, CB-O, CMP-DMB, CMP-BQ, CB@CMP-DMBs, and CB@CMP-BQs.

^a Surface areas based on BET theory. ^b Micropore volumes based on t-plot.

			Specific Capacitances (F/g)					Measurment Type	Ref		
Entry	Materials	Type of Materials	Current Density (A/g)								
			0.1	0.2	0.5	1	2	5	10		
1	Aza-CMP@350	CMP				461		397	378	three electrode	1
2	СМР	CMP						142		three electrode	2
3	aG-PTEPE-TBPE-C	CMP/C composite		179						three electrode	3
4	TpDAB	CMP				400				three electrode	4
5	N3-CMP-1	Carbonization	175			164			149	three electrode	5
6	G-TEPA-TPA-C	Carbonization		268						two electrode	6
7	NPCM-1	Carbonization	264							two electrode	7
8	TAT-CMP-2	CMP				183	173	158	137	three electrode	8
9	Fc-GMP	CMP/C composite			231				134	two electrode	9
10	H-CMP-BPPB	CMP			220	193			120	two electrode	10
11	PAQTA	CMP				168				two electrode	11
12	H-NCB-900	Carbonization				286			224	two electrode	12
13	POP _{M-TFP}	РОР			178	137.4	130.5			three electrode	13
	POP _{M-TFP}	РОР		91.7						two electrode	
14	N-CMP-BZ	CMP				189			138	two electrode	14
15	GT-POP-1	CMP			324					two electrode	15
16	CB@CMP-BO3	CMP/C composite			424	373			280	two electrode	This work

Table S2. Electrochemical performance of the recent CMP-based electrode materials for supercapacitors.

S1 Y. Kou, Y. Xu, Z. Guo and D. Jiang, Angew. Chem. Int. Ed. 2011, 50, 8753-8757.

S2 H. Zhang, Y. Zhang, C. Gu and Y. Ma, Adv. Energy Mater. 2015, 5, 1402175.

S3 K. Yuan, P. Guo-Wang, T. Hu, L. Shi, R. Zeng, M. Forster, T. Pichler, Y. Chen and U. Scherf, Chem. Mater. 2015, 27, 7403-7411.

S4 B. C. Patra, S. Khilari, L. Satyanarayana, D. Pradhan and A. Bhaumik, Chem. Commun. 2016, 52, 7592-7595.

- S5 J. -S. M. Lee, T. -H. Wu, B. M. Alston, M. E. Briggs, T. Hasell, C. C. Hu and A. I. Cooper, J. Mater. Chem. A 2016, 4, 7665-7673.
- S6 K. Yuan, T. Hu, Y. Xu, R. Graf, G. Brunklaus, M. Forster, Y. Chen and U. Scherf, ChemElectroChem 2016, 3, 822-828.

S7 Y. Xu, S. Wu, S. Ren, J. Ji, Y. Yue and J. Shen, RSC Adv. 2017, 7, 32496-32501.

S8 X. -C. Li, Y. Zhang, C. -Y. Wang, Y. Wan, W. -Y. Lai, H. Pang and W. Huang, Chem. Sci. 2017, 8, 2959-2965.

S9 A. M. Khattak, H. Sin, Z. A. Ghazi, X. He, B. Liang, N. A. Khan, H. R. Alanagh, A. Iqbal, L. Li and Z. Tang, J. Mater. Chem. A 2018, 6, 18827-18832.

S10 J. Choi, J. H. Ko, C. W. Kang, S. M. Lee, H. J. Kim, Y. -J. Ko, M. Yang and S. U. Son, J. Mater. Chem. 2018, 6, 6233-6237.

S11 Y. Liao, H. Wang, M. Zhu and A. Thomas, Adv. Mater. 2018, 30, 1705710.

S12 J. Lee, J. Choi, D. Kang, Y. Myung, S. M. Lee, H. J. Kim, Y. -J. Ko, S. -K. Kim and S. U. Son, ACS Sustainable Chem. Eng 2018, 6, 3525-3532.

S13 L. Xu, R. Liu, F. Wang, S. Yan, X. Shi and J. Yang, RSC Adv. 2019, 9, 1586-1590.

S14 S. Y. Park, C. W. Kang, S. M. Lee, H. J. Kim, Y. -J. Ko, J. Choi and S. U. Son, Chem. Eur. J. 2020, 26, 12343-12348.

S15 H. Zhang, X. Tang and C. Gu, J. Mater. Chem. A 2021, 9, 4984-4989.