Supporting Information

ITO-Free Large Area PDLC Smart Windows: A Cost-Effective Fabrication Using Spray Coated SnO² on Invisible Al Mesh

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Supporting Information Table S1. Component wise price distribution for a 1 ft^2 ITO/PDLC/ITO device.

 τ_{ON} = Response time; τ_{OFF} = Recovery time; V_{th} = Threshold voltage

Supporting Information Table S2. Hybrid mesh-based LC device comparison with the ITObased standard device.

Figure S1. (a) Schematic representation of metal mesh fabrication by crackle lithography: coating crackle precursor and drying to form crackle template, and metal evaporation followed by template removal. Polarizing Optical microscopy images of a Au mesh/PDLC/Au mesh device at applied voltage of (b) 0 V, (c) 20 V, (e) 35 V and (f) 75 V (transmission mode). Digital photograph of the device at (d) OFF $(0 V)$ and (g) ON $(75 V)$ state.

Figure S2. (a) FESEM image of the crackle template and (b) Au mesh. A charging effect is observed in the Au mesh image due to the presence of non-conducting regions between the mesh polygon.

Figure S3. (a) 3-dimensional potential distribution modeling in the LC film of 10 µm thickness sandwiched between the metal network and a conducting thin film counter electrode. (b) Binary image of a metal network. Potential distribution at the (c) surface of the metal network and (d) surface of the thin film electrode. The potential distribution modeling is performed by a selfwritten finite element method approach coded in python.

Figure S4. (a) low and (b) high magnification FESEM images of the hybrid TCE; the line mapping indicates the uniform coating of $SnO₂$ film across the bare glass and Au mesh coated region. (c) XRD plot indicating the presence of peaks corresponding to both Au and $SnO₂$. (d) Transmission plot for Au mesh, pristine $SnO₂$ film and the hybrid electrode.

Figure S5. FESEM image showing a seamless junction of the Au mesh.

Figure S6. EDS mapping of (a) Au, (b) O and (c) Sn confirming the uniform coating of SnO₂ on Au mesh.

Figure S7. (a) X-ray photoelectron spectroscopic (XPS) measurement of SnO₂ film. Highresolution (b) Sn_{3d} and (c) O_{1s} XPS spectra of SnO_2

Figure S8. Polarizing optical microscopic images in (a) OFF and (b) ON state of a PDLC device fabricated using ITO electrodes. The clear contrast difference indicates a proper device switching. The arrow marks indicate the direction of the crossed polarisers.

Figure S9. Polarizing optical microscopic images of ITO/PDLC/Au SnO₂ device in its (a) OFF and (b) ON state.

Figure S10. Polarizing optical microscopic images of Gel devices fabricated using (a)-(b) ITO/ITO and (c)-(d) Au_SnO₂ /Au_SnO₂ electrodes in their OFF and ON states. (e) Electrooptic response and recovery plots. (f) Freedericksz transformation behavior: normalized capacitance versus applied voltage profiles for various combinations of electrodes.

Figure S11. Optical microscopy image of an interconnected Al mesh electrode.

Figure S12. Optical microscopic images of Al_ SnO₂/PDLC/Al_SnO₂ device in its (a) OFF and (b) ON state. The bottom electrode mesh is also visible in the ON-state, representing the PDLC layer becoming transparent upon applying the electric field.

Figure S13. (a) Electro-optic response recovery of Al_SnO₂/PDLC/Al_SnO₂ device and (b) Freedericksz transformation behavior.

Figure S14. Photographs of a smart window-cum-display device (daylight view) in its fully opaque (a), partially transparent (b)-(g), and fully transparent states (h).

Figure S15. (a) Optical profilometry image of a ITO thin film and (b) the corresponding thickness profile. Large-area Al_ITO hybrid TCE on (c) glass and (d) flexible PET substrate.

Figure S16. PDLC device made of Al ITO as the transparent electrodes in its (a) OFF and (b) ON state.

Figure S17. (a) Low and (b) high mag. FESEM images of PEDOT:PSS coated Al mesh electrode. A PDLC device comprised of the Al_PEDOT:PSS hybrid electrodes in its (c) OFF and (d) ON states.