

Supplementary information

**Boosting activity toward oxygen reduction reaction of a mesoporous FeCuNC catalyst
via heteroatom doping-induced electronic state modulation**

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Table S1. Specific surface area, micro-, meso-, and total pore volume of FeCu_{0.5}NC, FeCu_{1.0}NC,

	$S_{\text{BET}}^{\text{a}}$ (m ² /g)	$V_{\text{micro}}^{\text{b}}$ (cm ³ /g)	$V_{\text{meso}}^{\text{c}}$ (cm ³ /g)	V_{p}^{d} (cm ³ /g)
FeCu _{0.5} NC	950	0.21	0.70	0.91
FeCu _{1.0} NC	910	0.18	0.76	0.94
FeCu _{1.5} NC	900	0.21	0.60	0.81

and FeCu_{1.5}NC.

a: Specific surface area; b: Micropore volume; c: Mesopore volume; d: Total pore volume

Table S2. Composition of FeNC, FeCu_{1.0}NC, and CuNC from ICP-OES, EA, and XPS survey.

	ICP (wt.%)			EA (wt.%)			XPS (at%)		
	Fe	Cu	C	N	S	C	N	P	S
FeNC	2.9	-	78.1	4.1	-	94.8	2.19	0.09	-
FeCu _{1.0} NC	3.2	2.5	77.1	4.2	0.5	94.5	2.21	0.08	0.17
CuNC	-	2.7	71.1	3.9	0.4	95.9	2.17	0.14	0.15

Table S3. Composition of N species in FeNC, FeCu_{1.0}NC, and CuNC.

at%	Pyridinic N	M-N	Pyrrolic N	Graphitic N	Oxidized N
FeNC	29.8	10.6	6.5	40.4	12.7
FeCu _{1.0} NC	30.0	11.7	7.4	41.1	9.8
CuNC	29.2	10.0	5.9	40.7	14.2

Table S4. EXAFS fitting parameters of FeNC, FeCu_{1.0}NC, and CuNC.

Sample	Path	<i>N</i>	<i>R</i> (Å)	σ^2 (Å ²)	ΔE_0 (eV)	<i>R</i> , %	<i>R</i> range	<i>k</i> range
FeNC	Fe-N	3.10	2.03	0.007	2.91	0.1 (0.001)	1.0 – 2.4 Å	2.0 – 8.0 Å
	Fe-C1	1.09	2.05	0.002				
	Fe-C2	3.73	2.63	0.015				
FeCu _{1.0} NC_Fe	Fe-N	3.09	1.98	0.013	-0.02	0.8 (0.008)	0.9 – 2.0 Å	2.5 – 9.4 Å
	Fe-C	1.17	2.01	0.008				
FeCu _{1.0} NC_Cu	Cu-N	2.90	1.85	0.002	-11.98	0.9 (0.009)	1.0 – 2.0 Å	2.1 – 10.5 Å
	Cu-C	1.04	1.86	0.002				
CuNC	Cu-N	2.94	1.91	0.005	-9.79	0.4 (0.004)	1.0 – 2.0 Å	2.5 – 9.7 Å
	Cu-C	1.05	1.95	0.003				

❖ Amplitude reduction factor (*S*O²): Fe (0.78); Cu (0.86)

Table S5. Comparison of H₂/O₂ AEMFC and PGM-free cathodes based on previously reported literature.

Cathode catalyst	Catalyst loading (mg/cm ²)		Membrane	Current density (A/cm ²) @0.6 V	Peak power density (W/cm ²)	Operating temperature (°C)
	Cathode (NPMC)	Anode (PGM)				
FeCu _{1.0} NC (This study)	2.2	0.7	FAA-3-50	0.49	0.294	70
CF-VC ¹	2.4	0.7	LDPE	1.45	1.35	70
FeCoPc/C ²	0.3	1.0	LDPE	1.61	1.26	80
Fe-N-Gra ³	2.0	0.8	HMT-PMBI	0.34	0.243	60
CoFe-N-CDC/CNT ⁴	0.75	0.74	ETFE	1.69	1.12	60
Fe-N-CDC/CNT ⁴	0.71	0.74	ETFE	1.13	1.06	60
N-C-CoO _x ⁵	2.4	0.7	LDPE-BTMA	1.32	1.05	65
Fe/N/CNT ⁶	2.0	0.4	aQAPS-S ₈	0.47	0.49	60
Pyrolysed KB/FePc ⁷	2.0	0.8	HMT-PMBI	0.22	0.186	60
FePc/C ⁸	1	0.4	Tokuyama A901	0.19	0.120	55
Fe _{0.5} -NH ₃ ⁹	0.9	0.6	HDPE	1.78	1.4	65
New Fe-N-C ¹⁰	1	0.125	HDPE	1.70	1.3	80
New Fe-N-C ¹⁰	1	0.6	HDPE	2.72	2.05	80

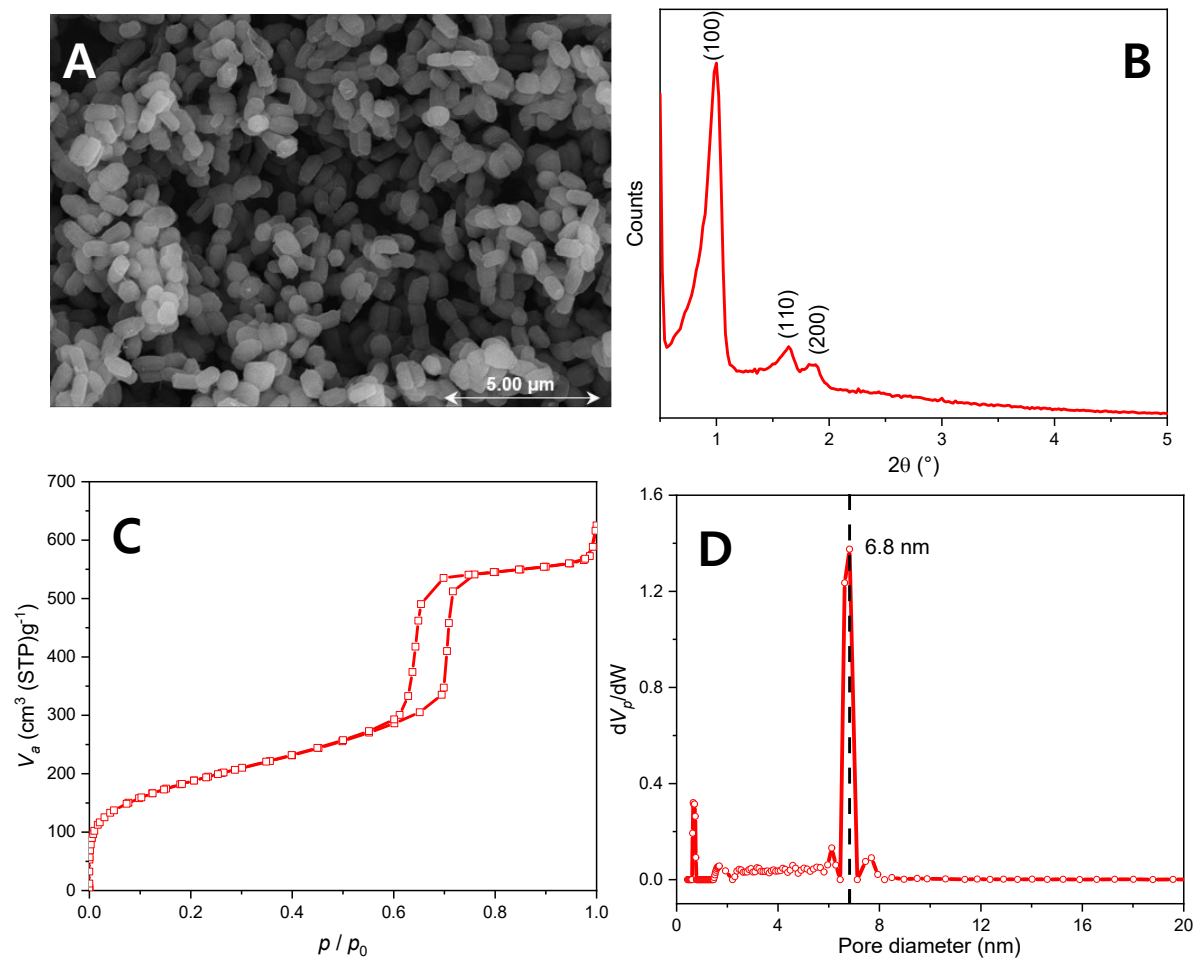


Figure S1. SEM image (A), XRD pattern (B), N_2 adsorption isotherm, and pore size distribution of grain-shaped SBA-15.

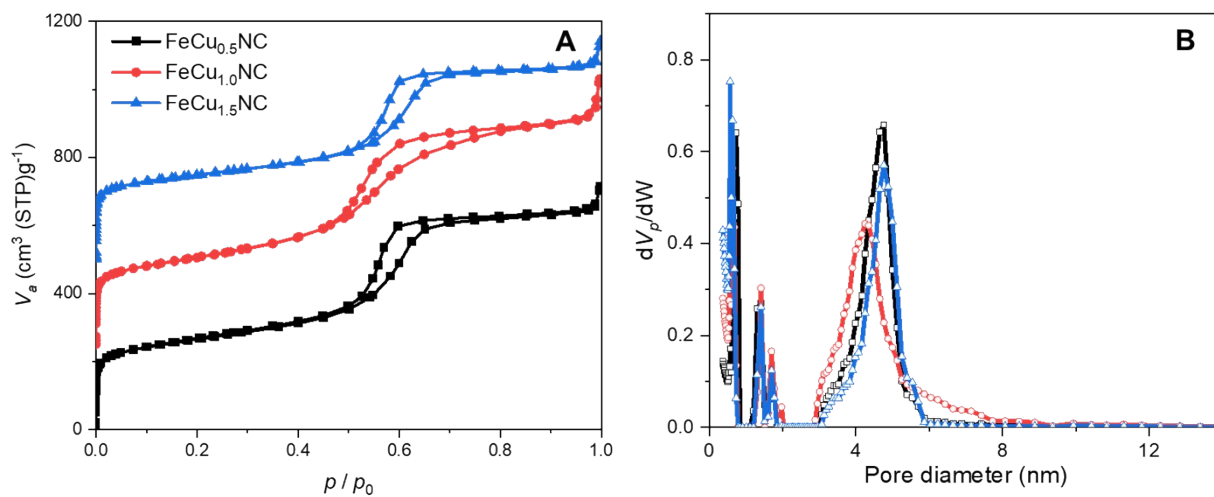


Figure S2. N₂ adsorption curves (A) and pore size distributions (B) of FeCu_{0.5}NC, FeCu_{1.0}NC, and FeCu_{1.5}NC.

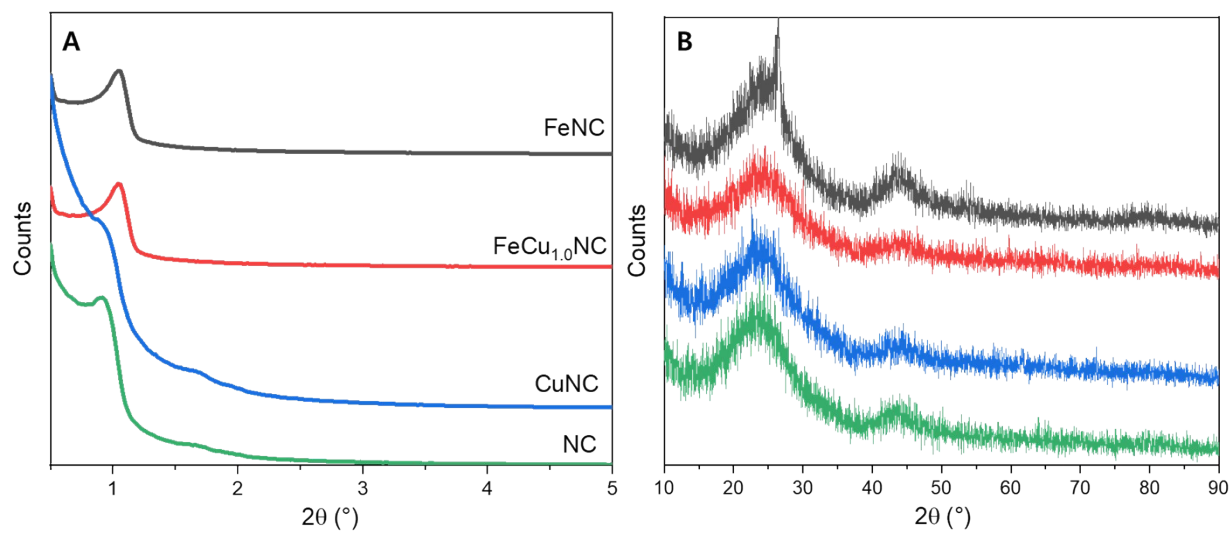


Figure S3. X-ray patterns at low-angle range (0.5–5°) (A) and mid-angle range (10–90°) (B) of FeNC, FeCu_{1.0}NC, CuNC, and NC.

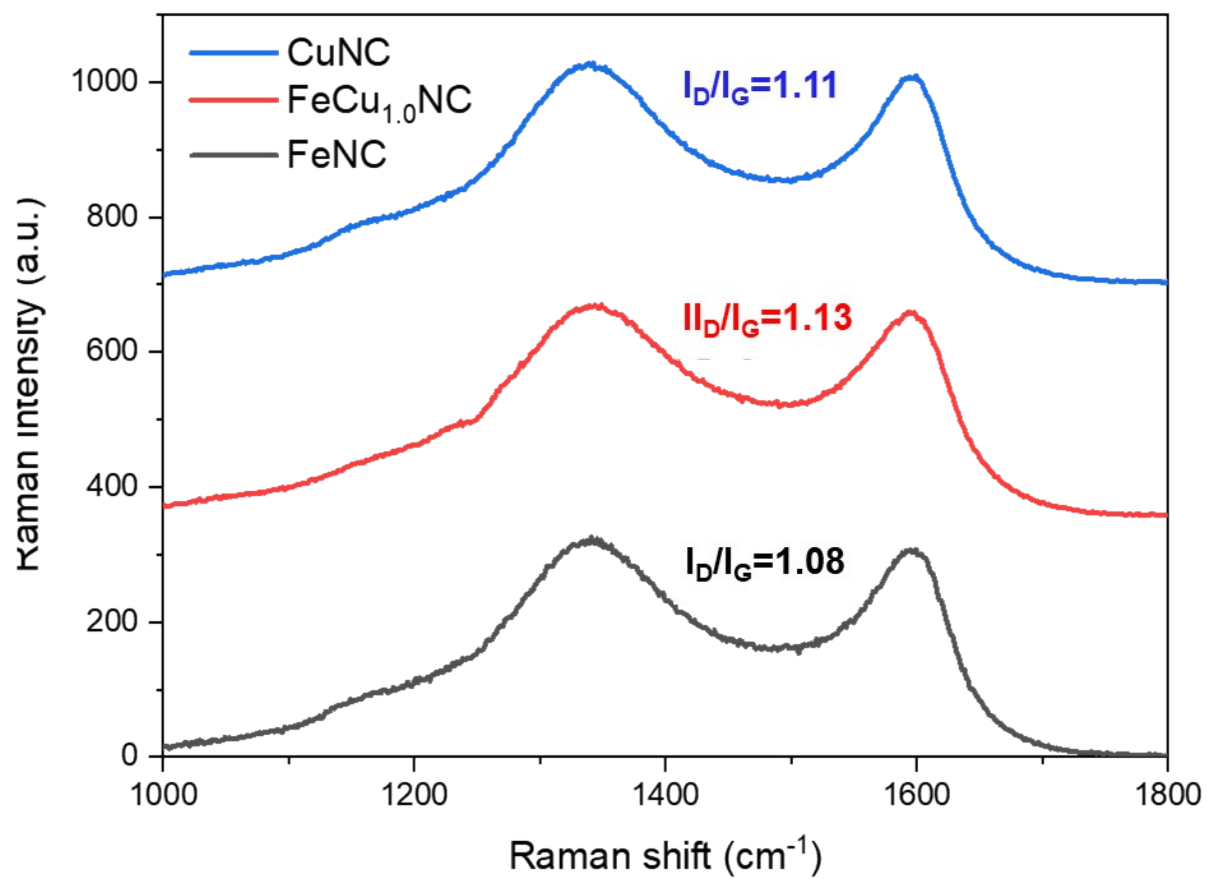


Figure S4. Raman spectra of FeNC, FeCu_{1.0}NC, and CuNC.

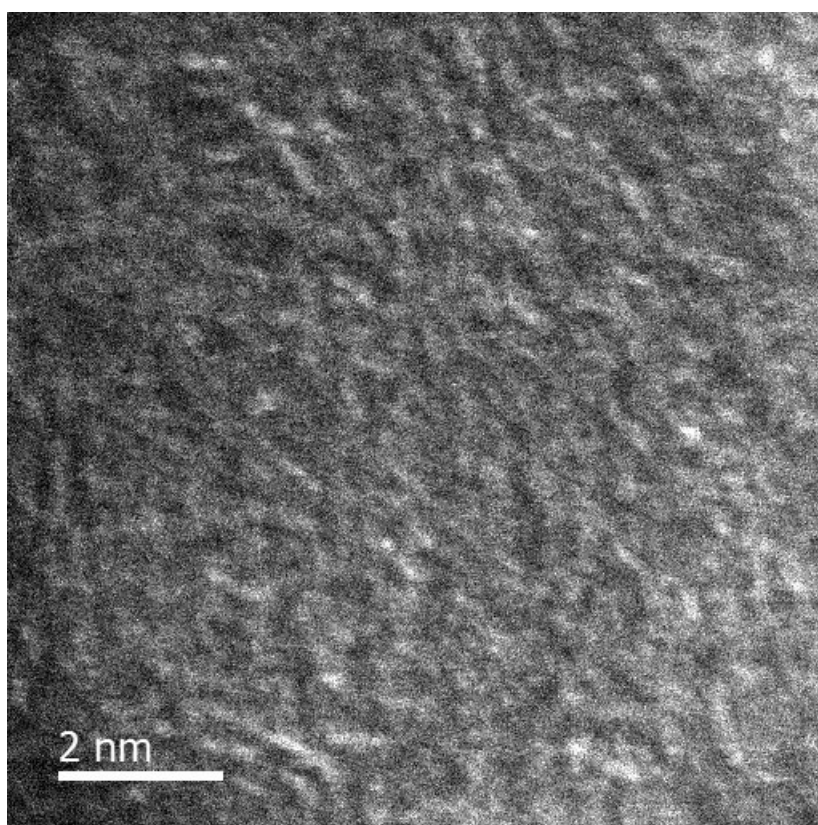


Figure S5. HAADF-STEM image of FeNC at atomic resolution.

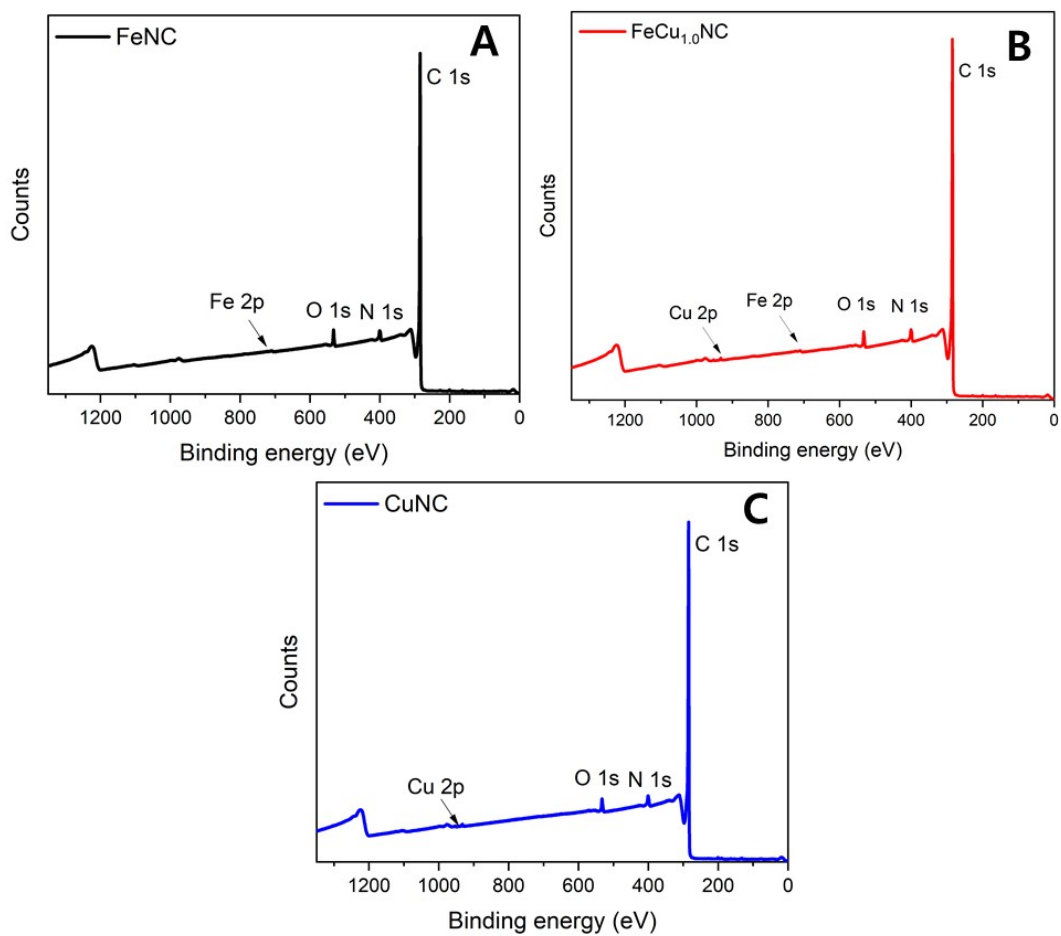


Figure S6. XPS survey of FeNC (A), FeCu_{1.0}NC (B), CuNC (C).

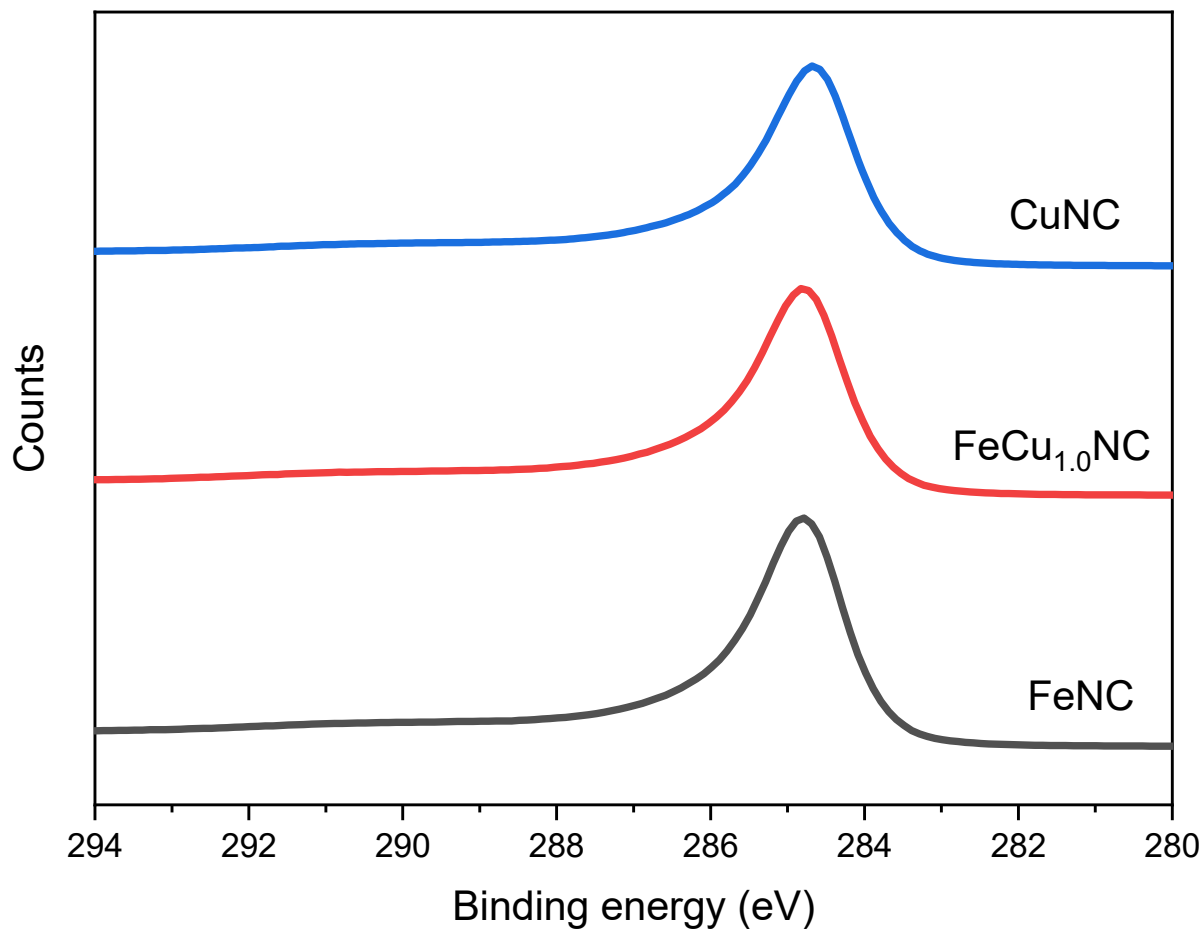


Figure S7. C1s spectra of FeNC, FeCu_{1.0}NC, CuNC.

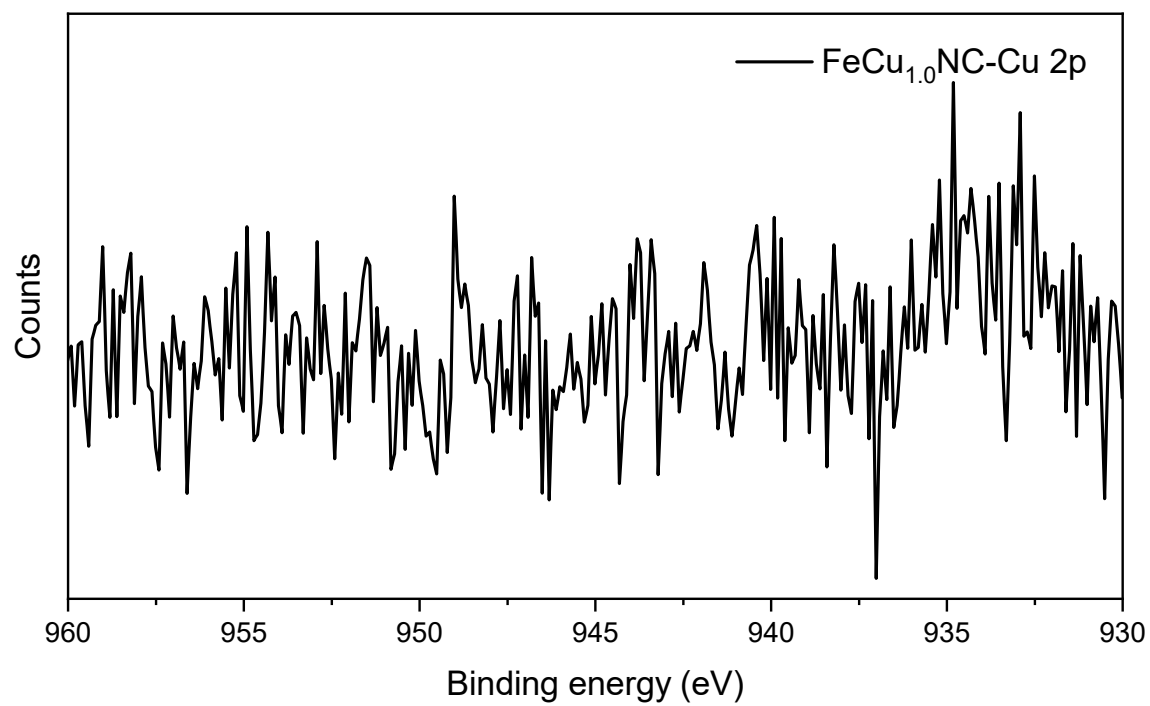


Figure S8. Cu 2p spectrum of FeCu_{1.0}NC .

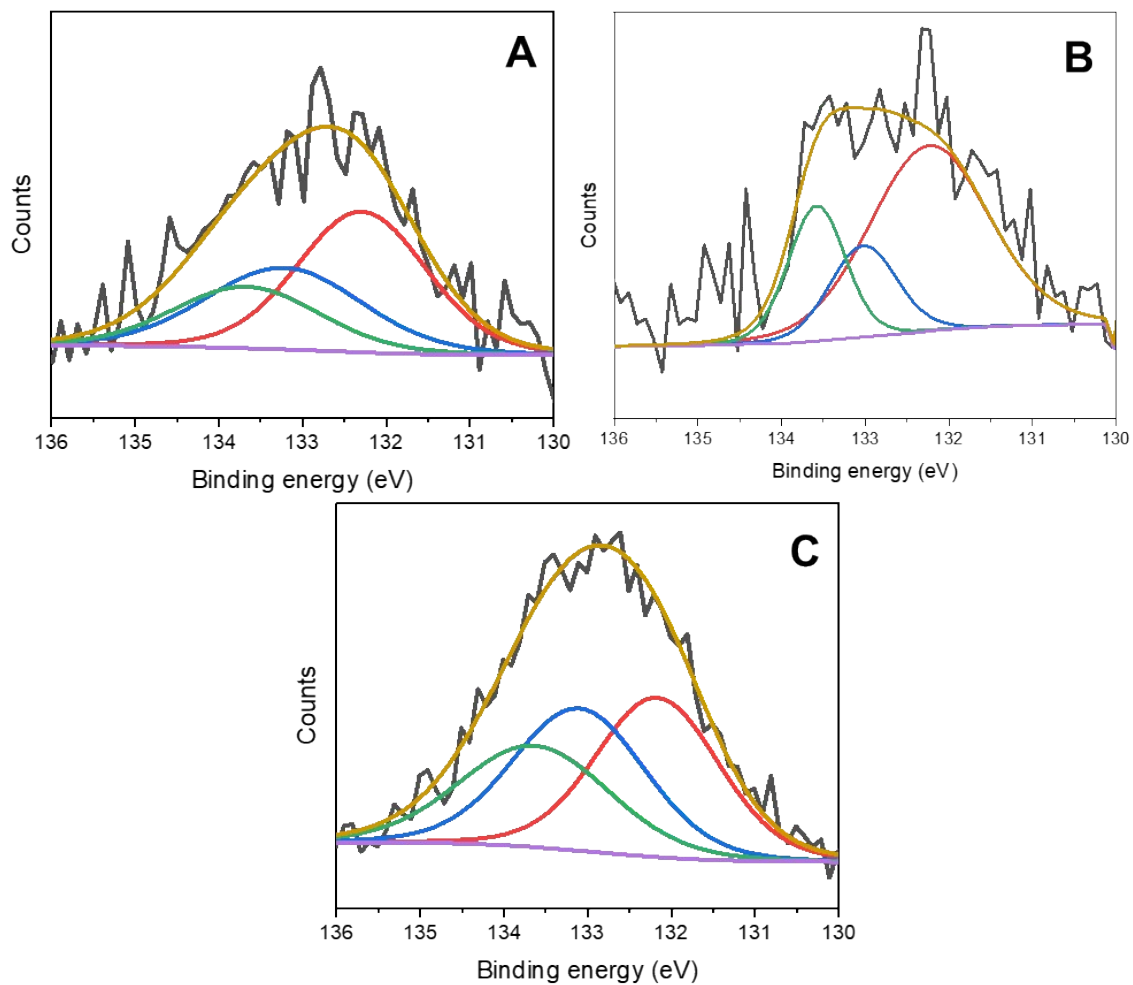


Figure S9. P 2p spectra of FeNC (A), FeCu_{1.0}NC (B), CuNC (C).

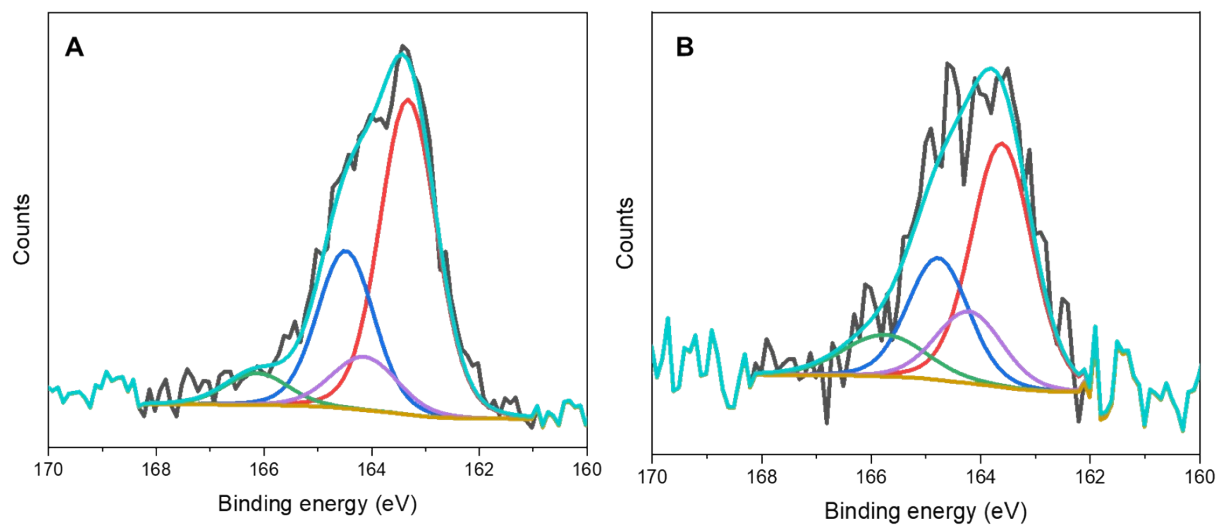


Figure S10. S 2p spectra of FeCu_{1.0}NC (A), CuNC (B).

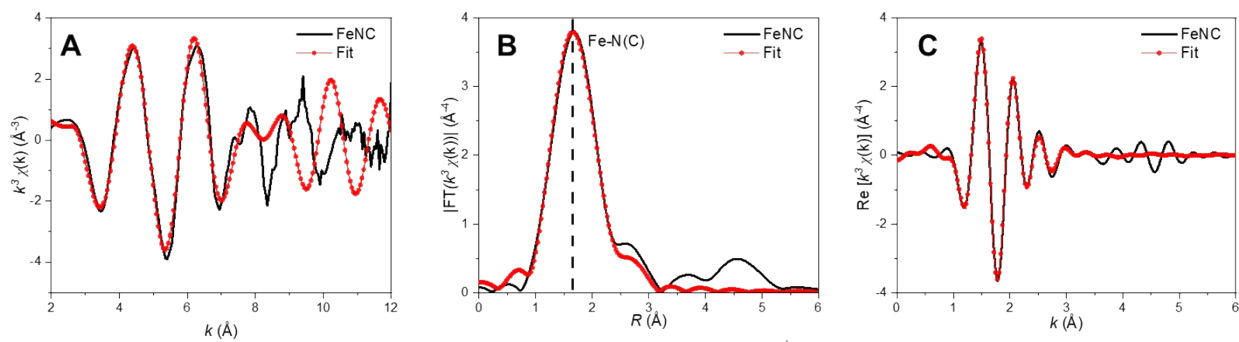


Figure S11. Raw data and their fit in k space (A), $R_{\text{magnitude}}$ (B), and R_{real} (C) part of FeNC in the Fe K-edge.

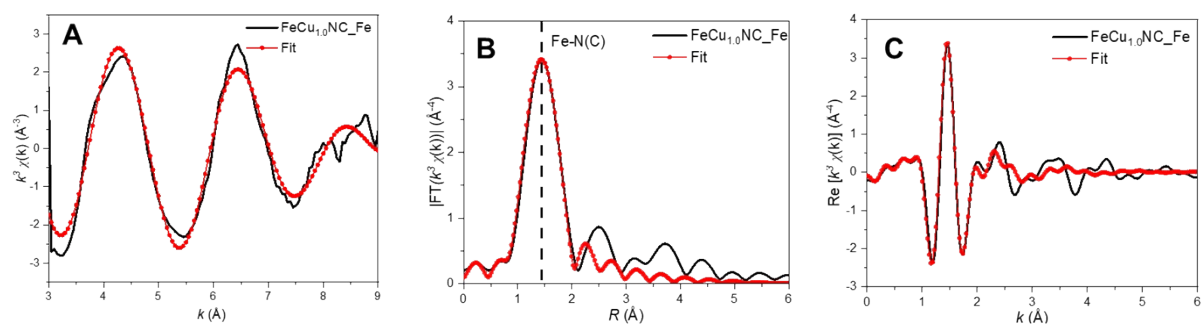


Figure S12. Raw data and their fit in k space (A), $R_{\text{magnitude}}$ (B), and R_{real} (C) part of FeCuNC in the Fe K-edge.

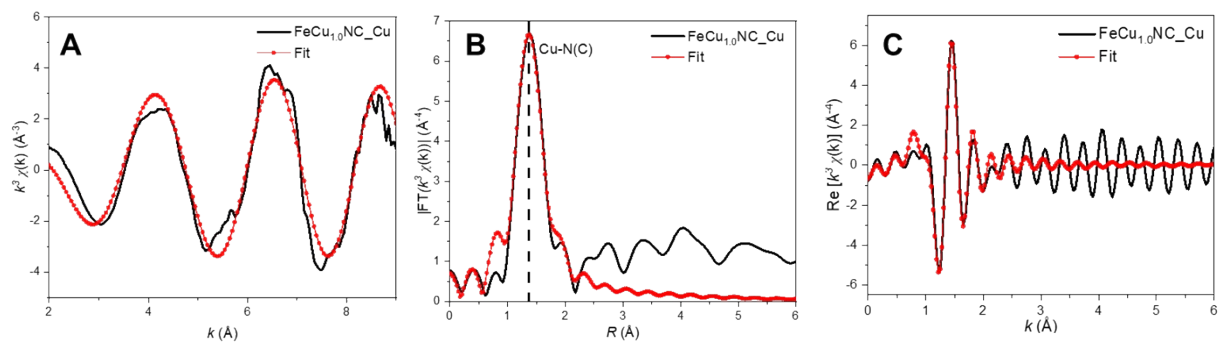


Figure S13. Raw data and its fit in k space (A), $R_{\text{magnitude}}$ (B), and R_{real} (C) part of FeCu_{1.0}NC in the Cu K-edge.

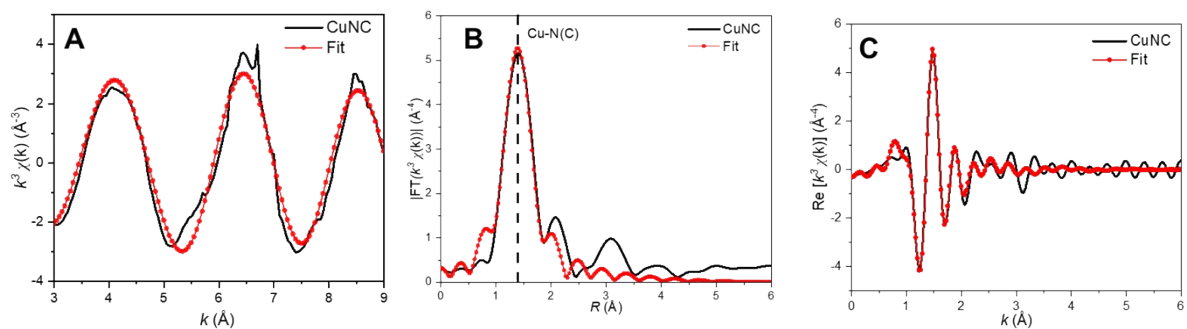


Figure S14. Raw data and its fit in k space (A), $R_{\text{magnitude}}$ (B), and R_{real} (C) part of CuNC in the Cu K-edge.

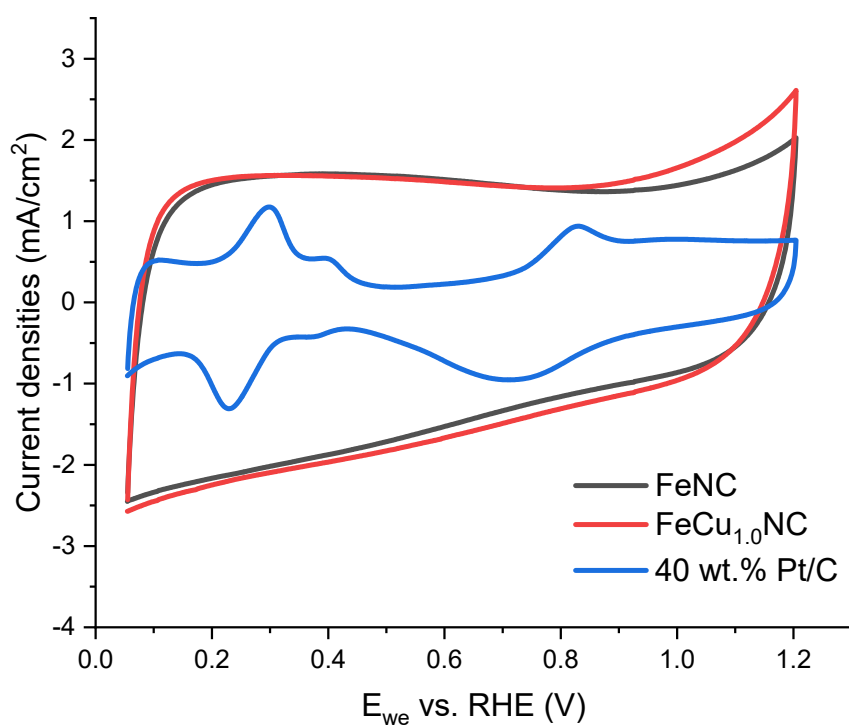


Figure S15. Cyclic voltammograms of FeNC, FeCu_{1.0}NC, and 40 wt.% Pt/C in 0.1 M KOH.

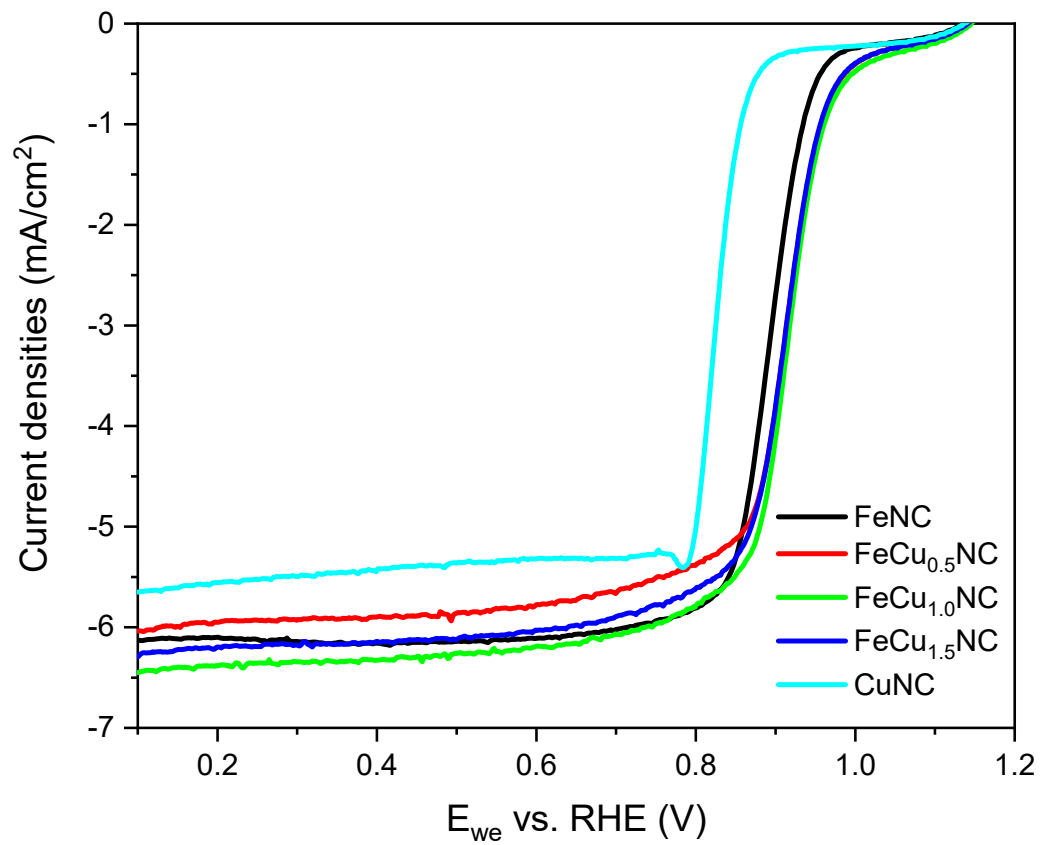


Figure S16. Linear sweep voltammogram of FeNC, FeCu_{0.5}NC, FeCu_{1.0}NC, FeCu_{1.5}NC, and CuNC in 0.1 M KOH at 10 mV/s.

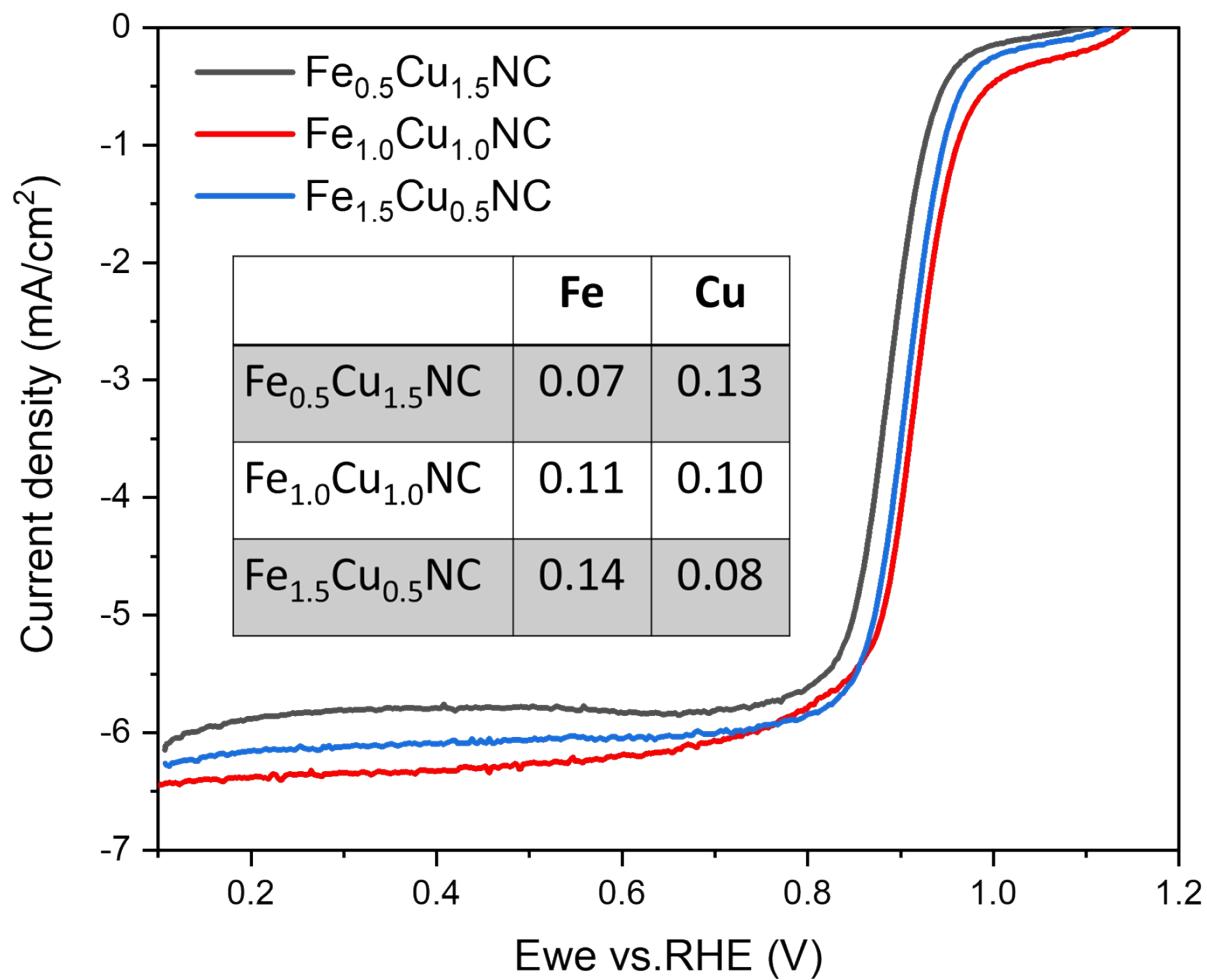


Figure S17. Linear sweep voltammogram of FeCuNC catalysts with different Fe to Cu atomic ratio in 0.1 M KOH at 10 mV/s. Inset table is amount of Fe and Cu (at%) measured from XPS.

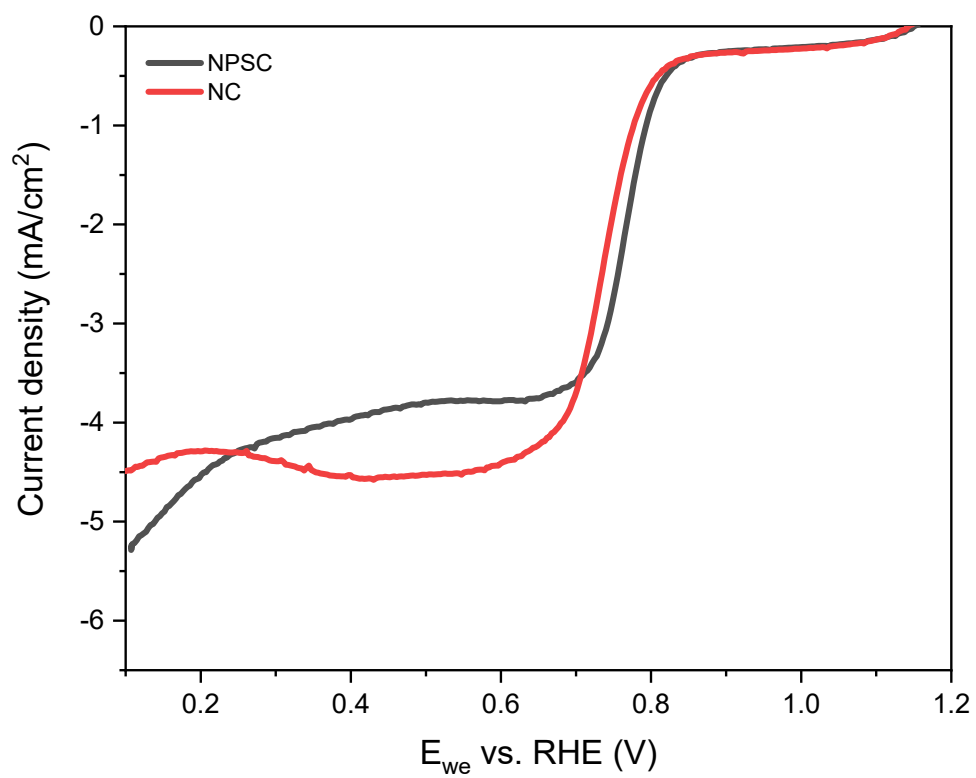


Figure S18. Linear sweep voltammogram of NC and NPSC in 0.1 M KOH at 10 mV/s.

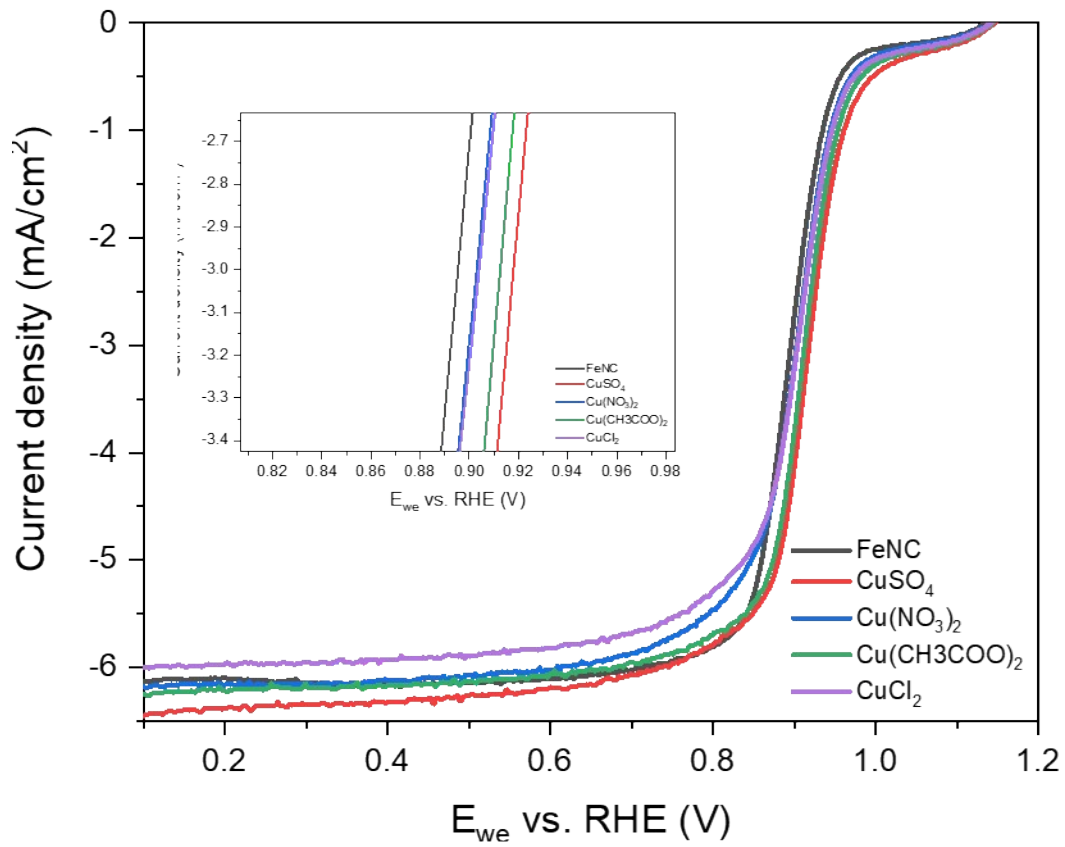


Figure S19. Linear sweep voltammogram of FeCuNC catalysts with different Cu precursors in 0.1 M KOH at 10 mV/s

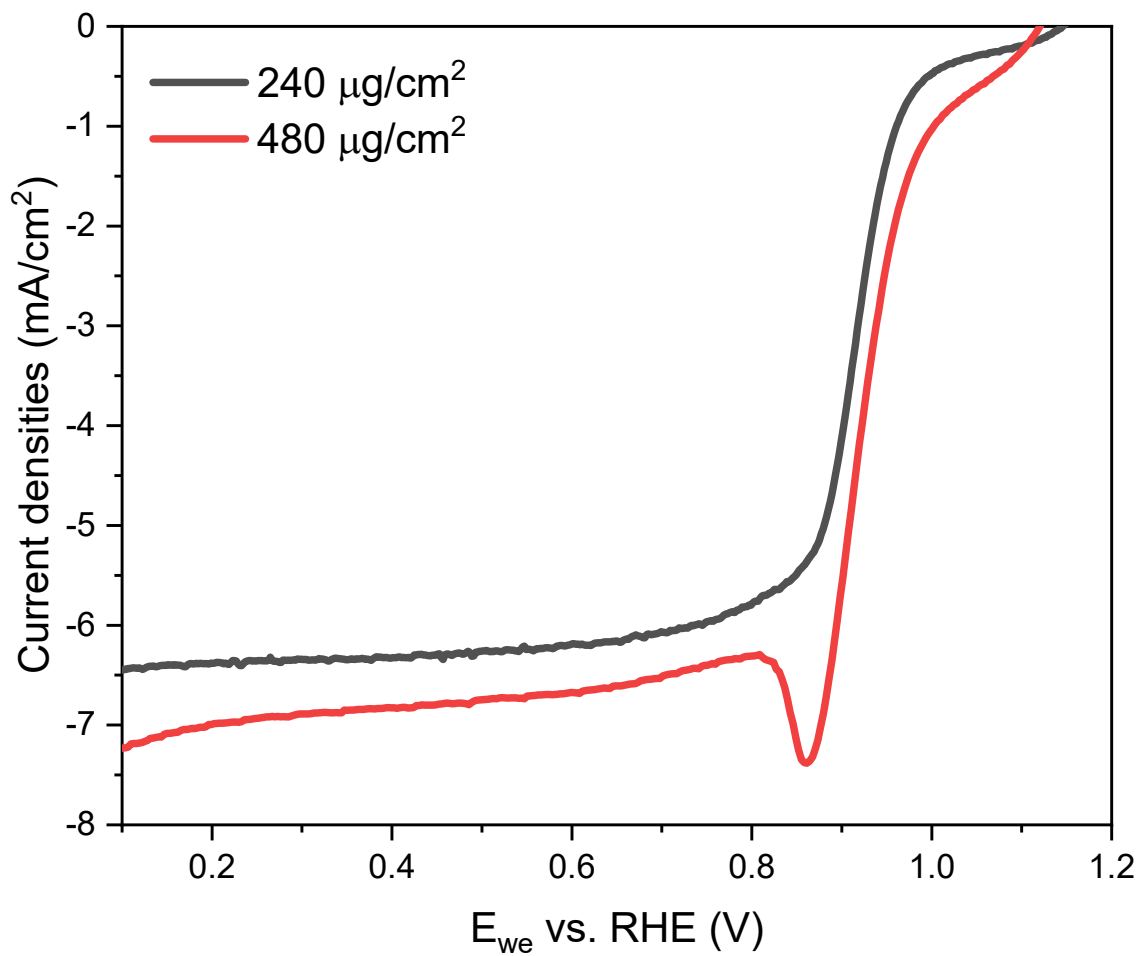


Figure S20. Linear sweep voltammogram of FeCu_{1.0}NC with different loading amounts on a glassy carbon working electrode in 0.1 M KOH at 10 mV/s.

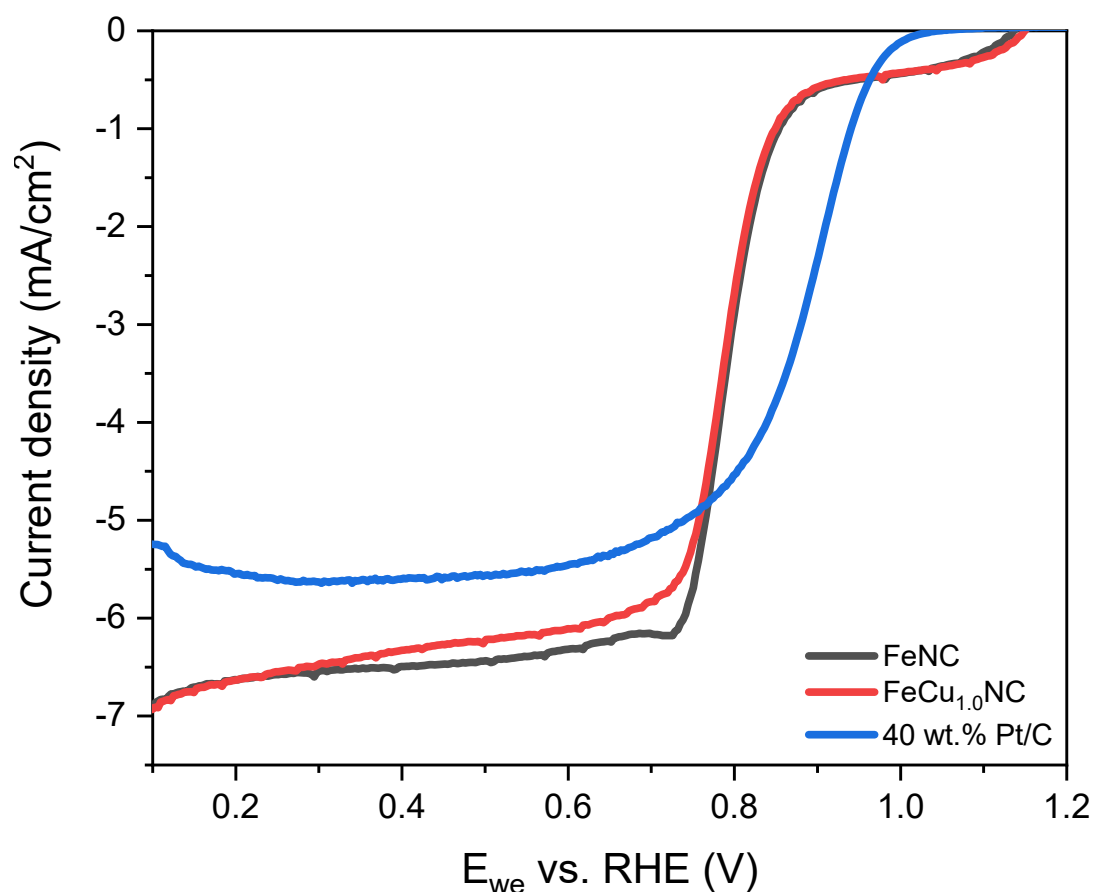


Figure S21. Linear sweep voltammogram of FeNC, FeCu_{1.0}NC, and Pt/C in 0.1 M HClO₄ at 10 mV/s.

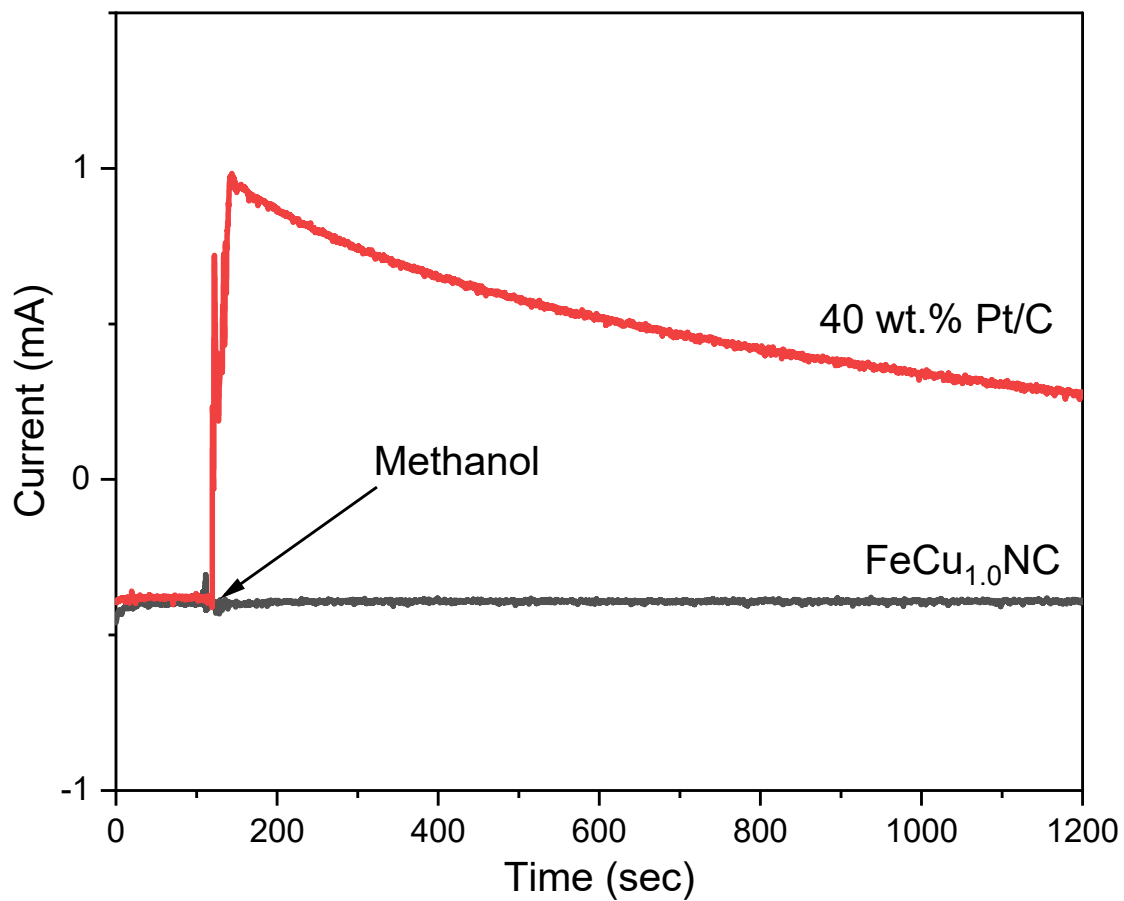


Figure S22. Methanol resistance of Pt/C and FeCu_{1.0}NC.

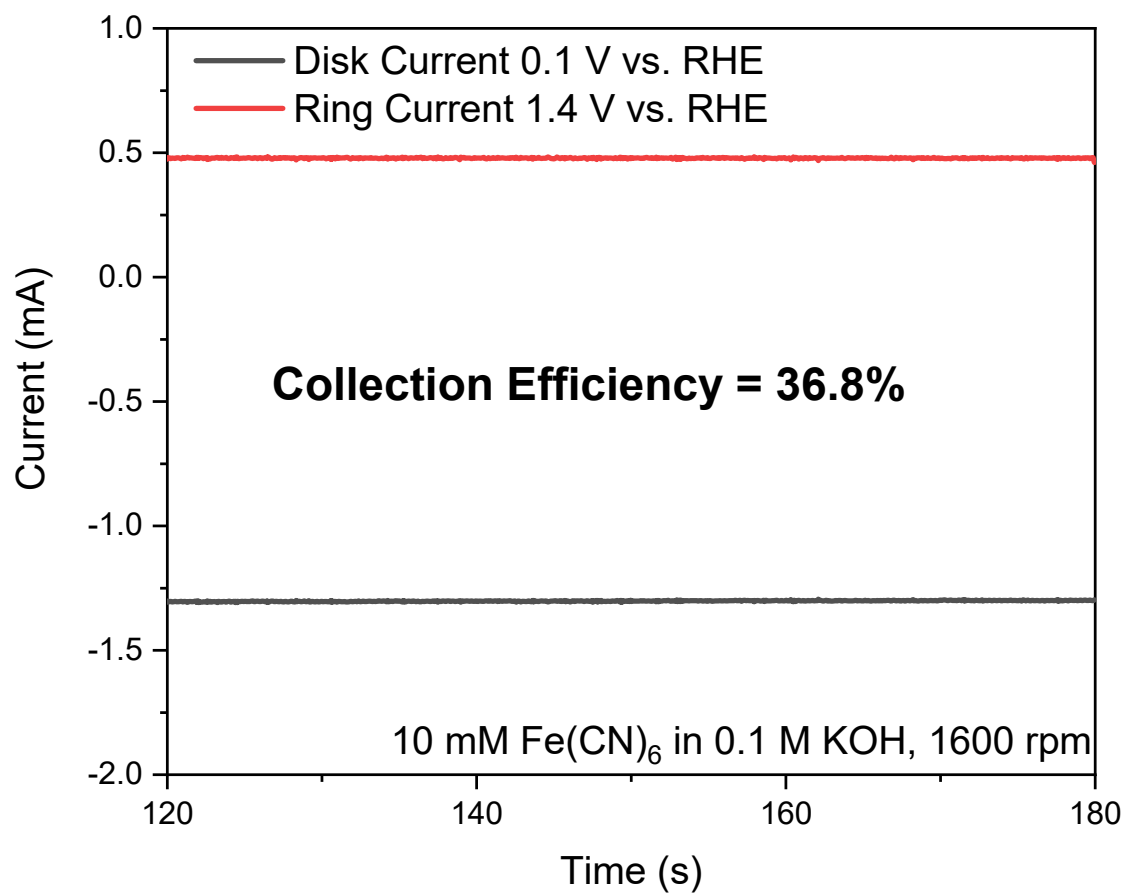


Figure S23. Result of collection efficiency experiment for FeCu_{1.0}NC. Collection efficiency was calculated by averaging current densities in the last 60 s.

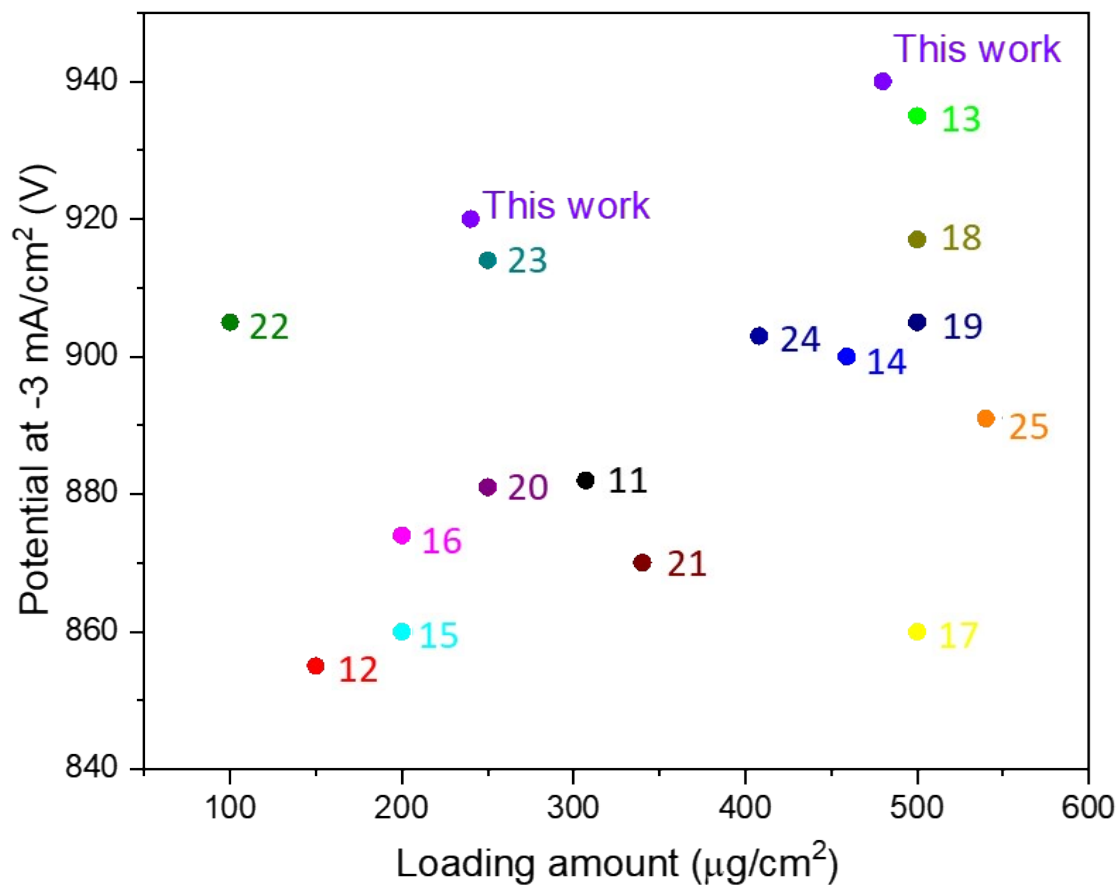


Figure S24. Comparison of the ORR (0.1 M KOH) catalytic activities (potential at -3 mA/cm²) determined in previous studies.

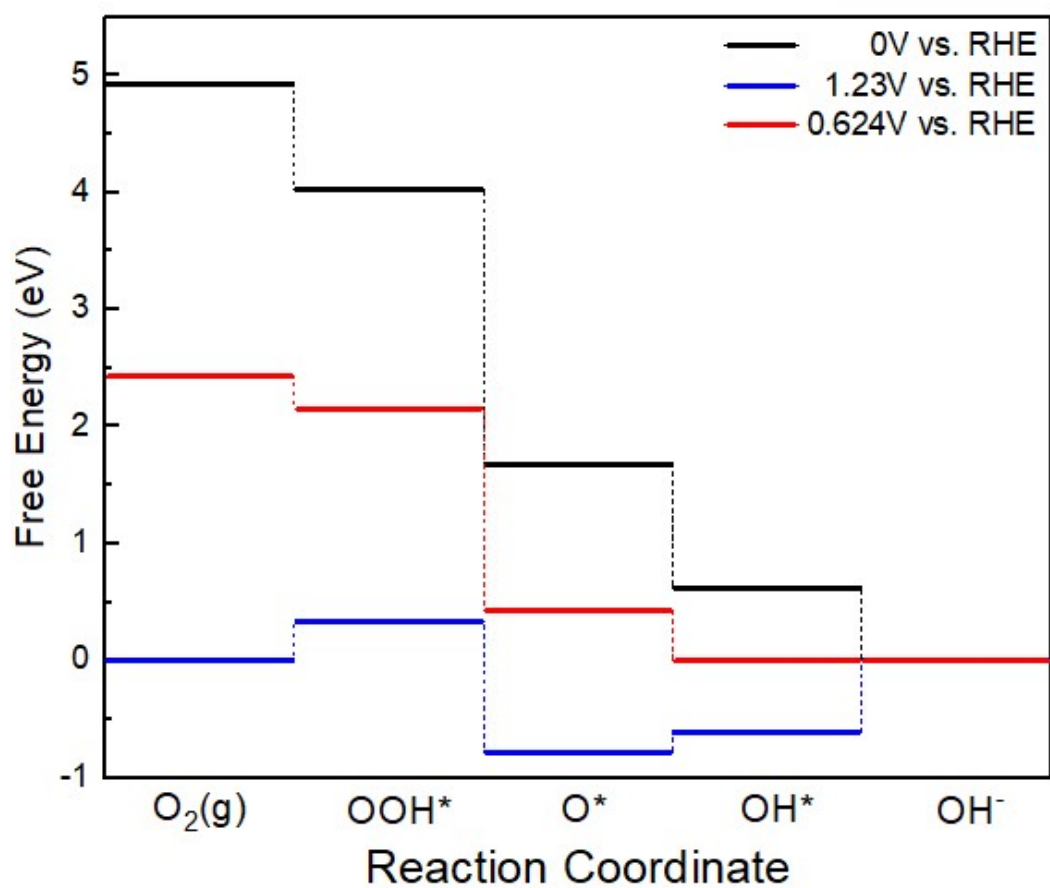


Figure S25. Calculated potential energy diagram at various applied potentials U for Fe-N₃-C-P-S model, such as open circuit voltage (0 V), equilibrium potential (1.23 V), and onset potential versus RHE, which is displayed in black, blue, and red bars, respectively.

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