

SUPPLEMENTARY INFORMATION

Face-On Oriented Hydrophobic Conjugated Polymers as Dopant-Free Hole-Transport Materials for Efficient and Stable Perovskite Solar Cells with Fill Factor Approaching 85%

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1. Experimental

1.1 Device fabrication and characterization

Materials: SnO₂ colloid precursor (tin (IV) oxide, 15% in H₂O colloidal dispersion) was purchased from Alfa Aesar. Lead iodide (99.99%) was purchased from TCI, lead bromide (99.999%) and cesium iodide (99.9%) were bought from Sigma-Aldrich. Formamidinium iodide (FAI) and methylammonium bromide (MABr) were synthesized by reacting

formamidine acetate (Aladdin) and hydroiodic acid (55.0-58.0 wt% in H₂O, Aladdin), methylamine (30-33 wt% in ethanol, Aladdin) and hydrobromic acid (48 wt% in H₂O, Aladdin), respectively. The particular experimental operations were referred to as previously reported methods.¹ spiro-OMeTAD was purchased from Derthon Optoelectronic Materials Science Technology Co LTD.

Fabrication of the devices: The ITO substrate was washed with deionized water, acetone, and anhydrous ethanol for 15 min, respectively. Then, after the dry clean substrates were treated by plasma for 10 min, a thin layer of SnO₂ nanoparticle film (SnO₂ colloid precursor diluted by deionized water, 1: 3, weight ratio) were spin-coated on the ITO substrates at 3000 rpm for 30 s and annealed in ambient air at 150 °C for 30 min. Next, a CsFAMA triple-cation “mixed” perovskite precursor solution was prepared by dissolving PbI₂ (1.2 M), FAI (1.1 M), PbBr₂ (0.2 M) and MABr (0.2 M) in a mixed solvent of anhydrous DMF and DMSO (4:1, volume ratio). After stirring for 30 min, 60 μL CsI which was previously dissolved as a 1.5 M stock solution in DMSO was added to the mixed perovskite precursor solution. Then, the perovskite precursor solution was spin-coated onto the substrates with SnO₂ via using a two steps procedure (the first step is under 1000 rpm for 10 s and the second step is under 6000 rpm for 30 s) in a nitrogen glove box. When entering the last 5 seconds before the second step finish, the anti-solvent (150 μL of chlorobenzene) were dropped on the spinning substrate.² The substrates were then annealed on a hotplate at 100 °C for 30 min. Afterwards, the substrates were cooled to room temperature, and the solution with different HTMs were spin-coated on different substrates containing perovskite layers. For spiro-OMeTAD solution, 72.3 mg of spiro-OMeTAD was dissolved 1 mL of chlorobenzene with additives of 17.5 μL of bis(trifluoromethylsulfonyl)imide lithium salt (Li-TFSI, Sigma-Aldrich) solution (520 mg mL⁻¹ in acetonitrile), 28.5 μL of 4-tert-butylpyridine (TBP, Sigma-Aldrich), the spin-coating condition: 4000 rpm for 30 s. For HTMs of PBZ-1Si, PBZ-2Si and PBZ-3Si, the optimized 4 mg mL⁻¹, 6 mg mL⁻¹ and 6 mg mL⁻¹ were prepared in chlorobenzene, respectively, the spin-

coating condition: 2500 rpm for 30 s. Finally, 80 nm of the gold counter electrode was thermally evaporated under high vacuum.

Film and device characterization: The grazing incident wide-angle X-ray scattering (GIWAXS) measurements were performed at BL46XU beamline of SPring-8. PBZ-3Si and spiro-OMeTAD samples were prepared on Si substrates or the surface of ITO/SnO₂/perovskite film. The sample was irradiated with an X-ray energy of 12.39 keV ($\lambda = 1 \text{ \AA}$) at a fixed-incident angle on the order of 0.12° through a Huber diffractometer. The GIWAXS pattern was recorded with a two-dimensional image detector (Pilatus 300 K). The surface morphology was characterized by scanning electron microscopy (SEM) and atomic force microscopy (AFM). The cross-sectional view morphology of the perovskite films was observed using a field-emission scanning electron microscope (SEM, Apreo S). The roughness of the films was recorded using atomic force microscopy (AFM, Multimode-8J, America). Contact angles were characterized on an Optical contact angle measuring and contour analysis systems (OCA40 Micro). Photoluminescence was measured at room temperature on a fluorescence spectrophotometer (FLS980, Edinburgh Instruments). The steady-state PL emissions were measured under 700 nm light source excitation using monochromatized Xe lamp, while the time-resolved PL decays were carried out with a pulsed diode laser excitation source. The X-ray photoelectron spectrum (XPS) was performed using an X-ray photoelectron spectroscopy system (Axis Supra, Shimadzu) with Al K α X-ray radiation (1486.6 eV) as the X-ray source, and C-C (C1s) peak at 284.8eV is used as a charge correction reference. The photovoltaic performance of devices was measured with a Keithley 2401 source meter using a solar simulator (SS-F5-3A, Enlitech) at 100 mW cm⁻² illumination (AM 1.5 G) equipped with a calibrated Si reference cell (SRC-2020, Enlitech). The J–V curves of the PSCs were measured in reverse scan (from 1.2 to -0.1 V) or forward scan (from -0.1 to 1.2 V) modes at a scan speed of 200 mV s⁻¹. The active area of devices was defined by a metal shadow mask of 0.1 cm². External quantum efficiency (EQE) was obtained on a computer-controlled quantum efficiency

measuring instrument (QE-R, Enlitech). Electrochemical impedance spectroscopy (EIS) measurements were carried out under 10 mW cm^{-2} illumination at applied voltages (-0.85 V) using an electrochemical workstation (Zennium Zahner, Germany) with an AC perturbation of 10 mV ranging from 100 mHz to 1 MHz , at room temperature with $40\text{-}50\%$ humidity. Transient photocurrent decay (TPC) was performed at short circuit condition under white light bias generated from an array of light-emitting diodes equal to 1 sun intensity. Transient photovoltage (TPV) measurements was performed at open circuit condition under white light bias generated from an array of light-emitting diodes equal to 1 sun intensity. Stability test: The dark long-term stability assessment of the perovskite solar cells was carried out by repeating the J - V characterizations over various times. The unsealed devices were stored under ambient atmosphere with $20 \pm 5\%$ relative humidity at room temperature. The high thermal and humidity long-term stability was carried out (devices without encapsulation) on a hot plate at the temperature of $85 \text{ }^\circ\text{C}$ under $70 \pm 10\%$ RH ambient atmosphere. Space-charge-limited current (SCLC) measurement based on the hole-only devices (ITO/PEDOT:PSS/HTM/MoO₃/Ag); the dark J-V characteristics of the devices were measured by a Keithley 2401 source meter in the range of $0\text{-}5 \text{ V}$. The hole mobility (μ) can be extracted from the trap-free SCLC regime, according to Mott-Gurney law:

$$J_D = \frac{9\mu\epsilon_r\epsilon_0V_b^2}{8L^3} \quad (\text{S1})$$

Where J_D is the current density, V_b is applied voltage, L is the thickness of HTM film, ϵ_r is the relative dielectric constant of organic materials ($\epsilon_r=3$), ϵ_0 is the vacuum permittivity, and q is the electron charge.³

1.2 DFT calculation

The ground-state geometry optimization was calculated using density functional theory (DFT) method at the B3LYP/6-31G(d) level of theory with the Gaussian 09 program package.⁴

Stationary points were verified by frequency analysis. The optimized structures were found to

be stable. The calculated molecular electronic static potential (ESP) results were obtained with the Multiwfn 3.7 program.⁵

2. Additional Figures and Data

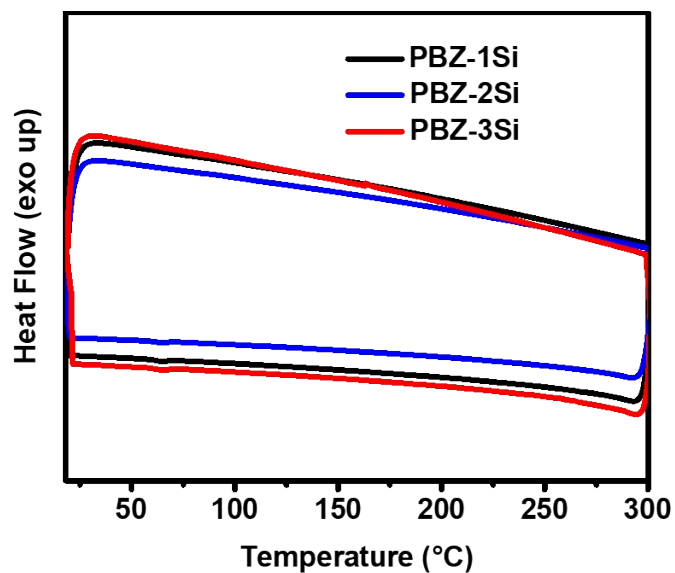


Figure S1. Differential scanning calorimetry curves during the second heating scan for PBZ-Si.

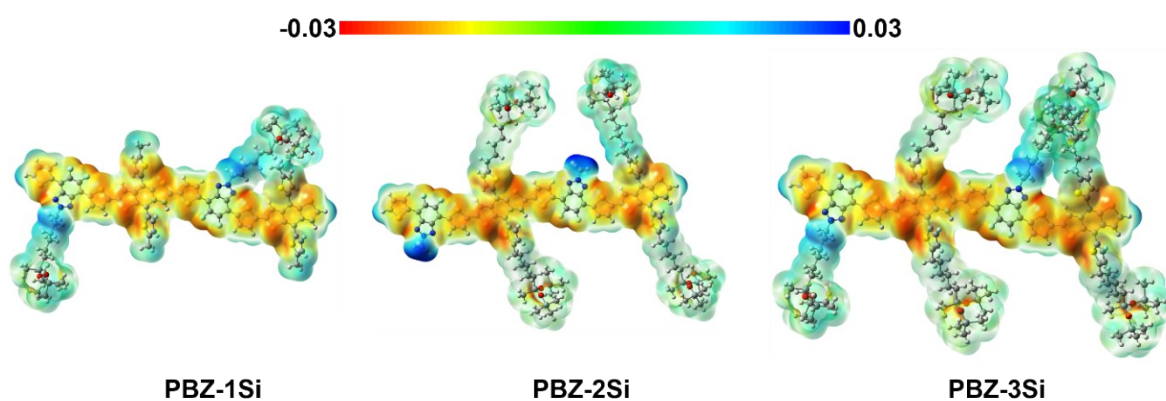


Figure S2. Calculated ESP profiles of PBZ-Si polymers.

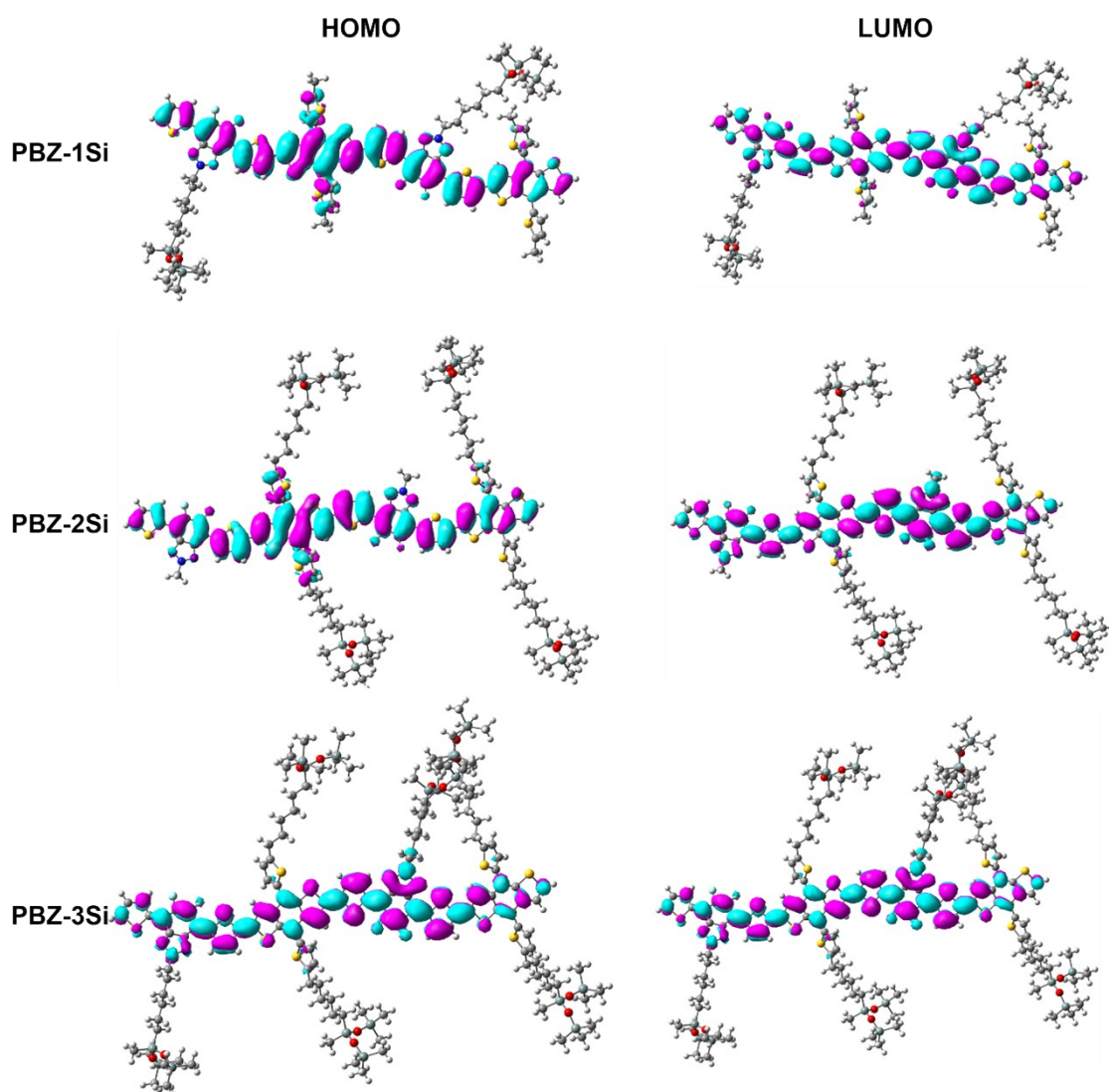


Figure S3. HOMO and LUMO of PBZ-Si polymers obtained from DFT calculations.

Table S1. Calculated HOMO and LUMO energy levels of polymers.

Samples	HOMO (eV)	LUMO (eV)
PBZ-1Si	-4.74	-2.40
PBZ-2Si	-4.74	-2.42
PBZ-3Si	-4.71	-2.38

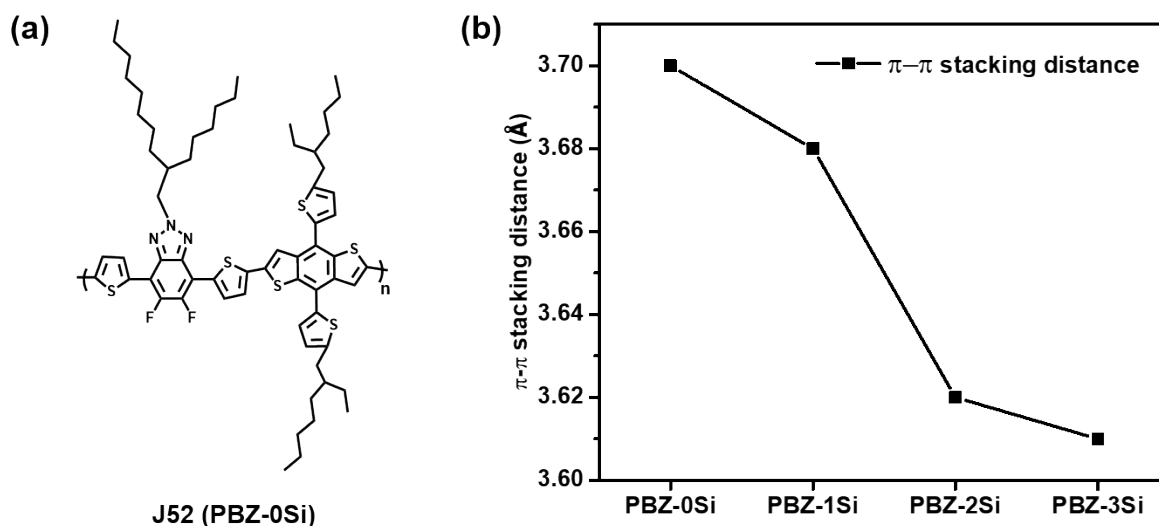


Figure S4. (a) The structure of J52 molecular⁶ which represents the PBZ-0Si. (b) π - π distance of polymers.

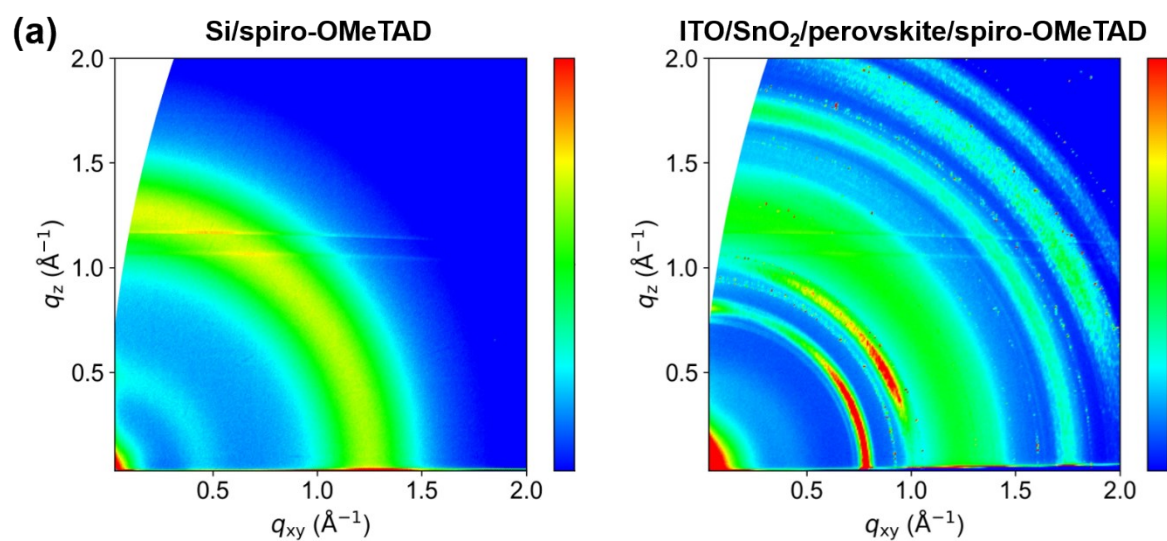


Figure S5. GIWAXS patterns of the spiro-OMeTAD films coated on different substrates: (a) Si/spiro-OMeTAD, and (b) ITO/SnO₂/perovskite/spiro-OMeTAD.

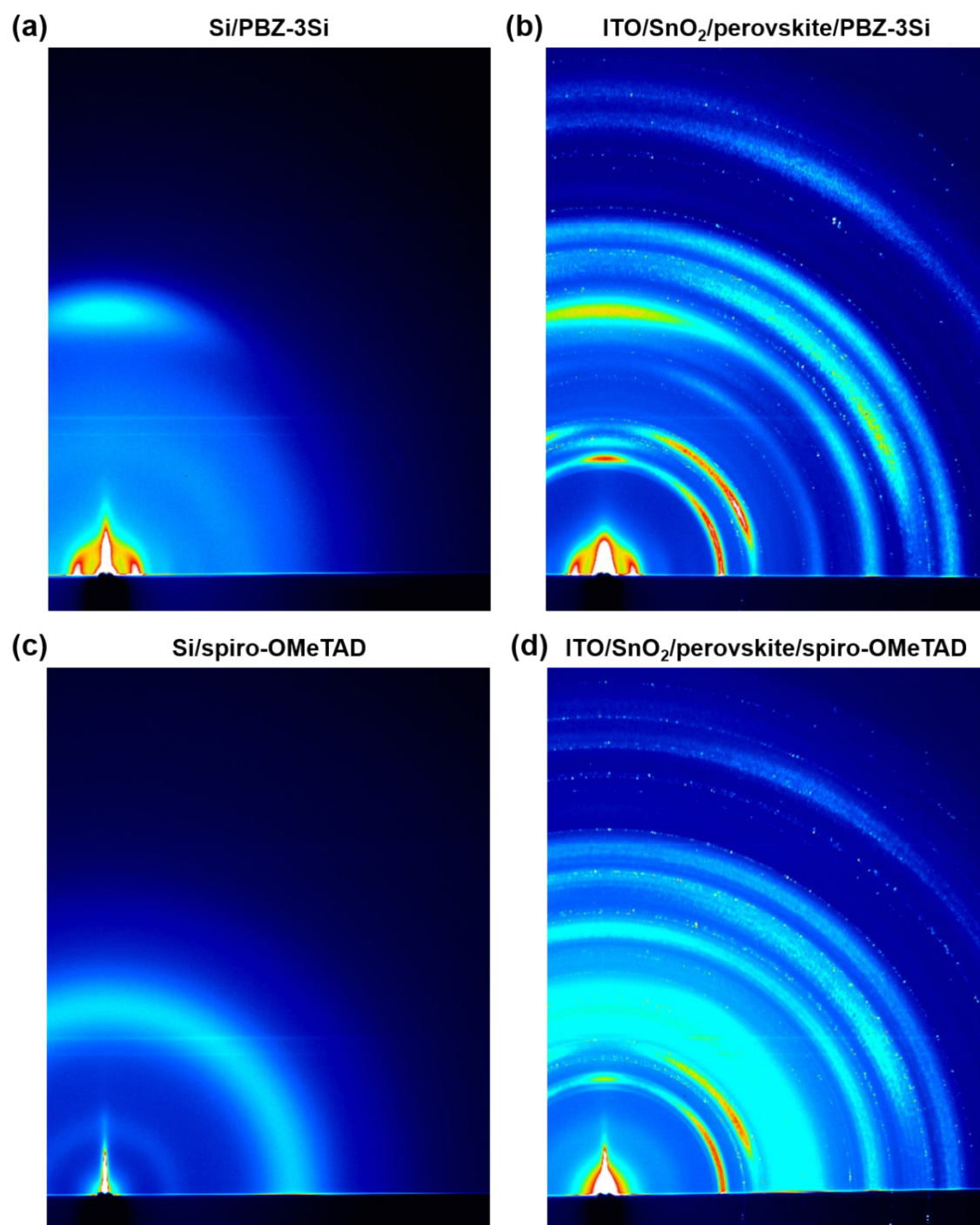


Figure S6. Image data of 2D X-ray detector for GIWAXS measurements (a) Si/PBZ-3Si, (b) ITO/SnO₂/perovskite/PBZ-3Si, (c) Si/spiro-OMeTAD, and (d) ITO/SnO₂/perovskite/spiro-OMeTAD.

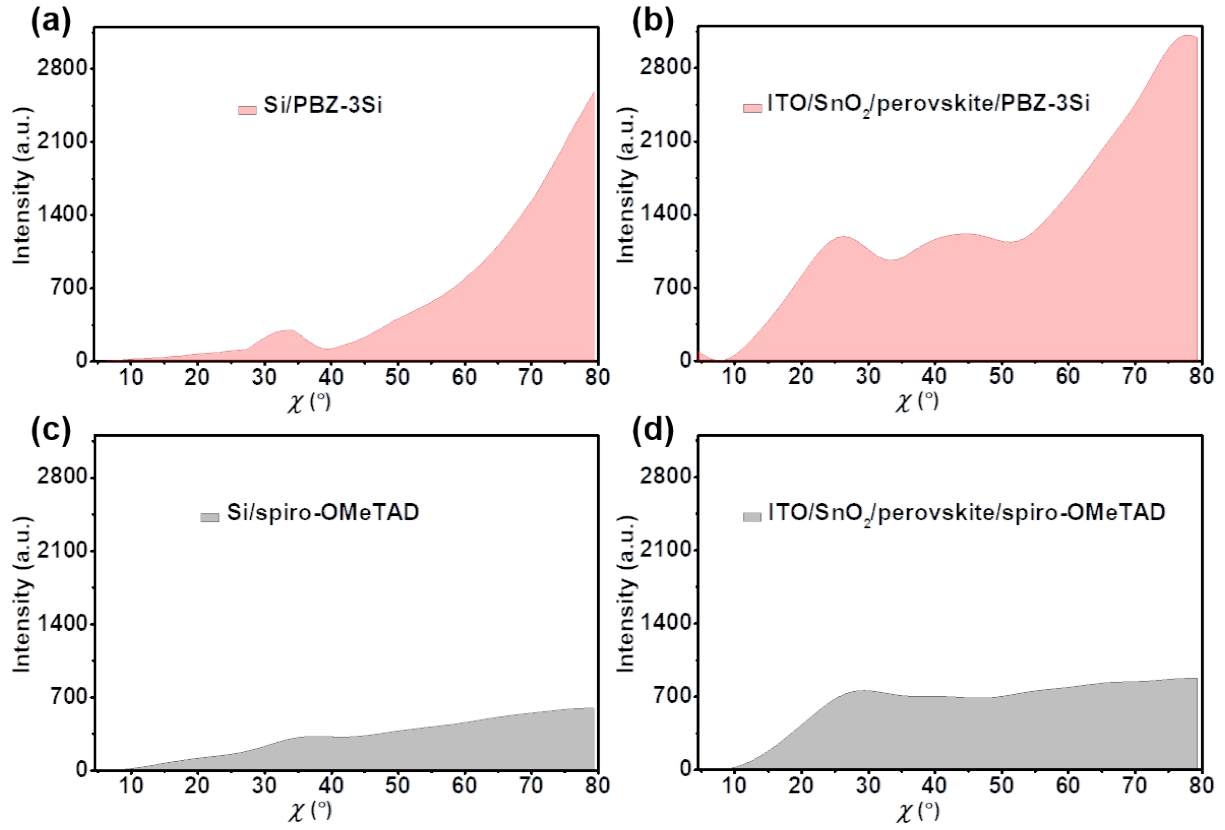


Figure S7. Integration of the azimuthal intensity along the 010 reflex ($\sim 1.75 \text{ \AA}^{-1}$) in GIWAXS for different films.

Table S2. Statistics of all parameters of TRPL.

Samples	τ_1 (ns)	τ_2 (ns)	A_1	A_2	$\tau_{\text{avg}}^{\text{a)}$ (ns)
Perovskite	7.17	483.78	0.04	0.96	483.48
Perovskite/ PBZ-1Si	5.62	26.53	0.64	0.36	20.85
Perovskite/ PBZ-2Si	5.44	58.24	0.40	0.60	55.12
Perovskite/ PBZ-3Si	3.49	81.14	0.58	0.42	76.78

^{a)} τ_{avg} : average lifetime which is calculated as following: $\tau_{\text{avg}} = (A_1\tau_1^2 + A_2\tau_2^2)/(A_1\tau_1 + A_2\tau_2)$.

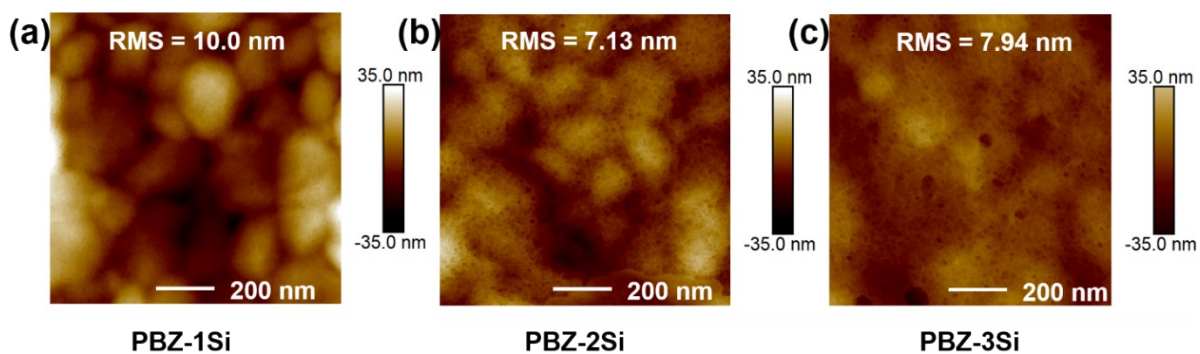


Figure S8. AFM images of PBZ-Si films.

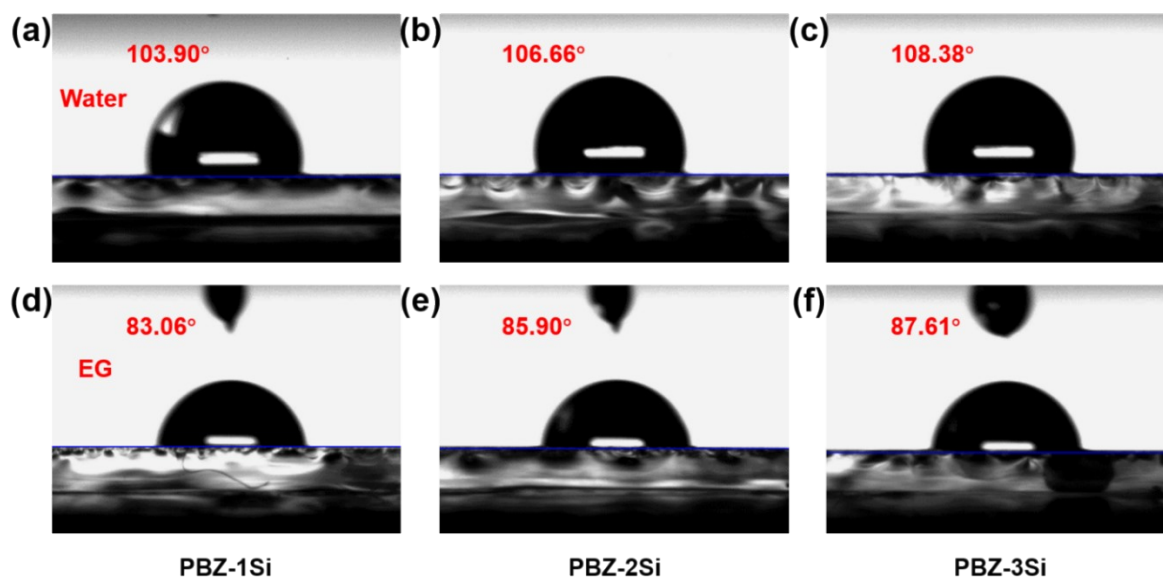


Figure S9. Contact angles (CAs) of water ((a)-(c)) and ethylene glycol (EG) ((d)-(f)) on PBZ-Si films.

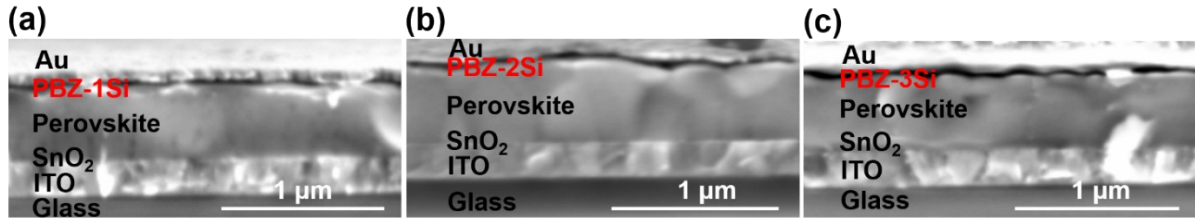


Figure S10. Cross sectional SEM images of the planar n-i-p structured PSCs device with different HTMs, scale bar 1 μm .

Table S3. The J - V performance of PSCs based on PBZ-1Si HTM films with different precursor concentrations.

Concentration (mg mL^{-1})	V_{OC} (V)	J_{SC} (mA cm^{-2})	FF (%)	PCE (%)
4	1.03	21.37	78.32	17.24
6	1.00	21.15	70.92	15.00
8	1.00	20.47	65.02	13.31

Table S4. The J - V performance of PSCs based on PBZ-2Si HTM films with different precursor concentrations.

Concentration (mg mL^{-1})	V_{OC} (V)	J_{SC} (mA cm^{-2})	FF (%)	PCE (%)
4	1.05	21.61	74.17	16.83
6	1.06	22.22	81.31	19.15
8	1.04	21.56	71.16	16.00

Table S5. The J - V performance of PSCs based on PBZ-3Si HTM films with different precursor concentrations.

Concentration (mg mL ⁻¹)	V_{OC} (V)	J_{SC} (mA cm ⁻²)	FF (%)	PCE (%)
4	1.08	21.43	80.37	18.60
6	1.09	21.80	84.93	20.18
8	1.08	20.79	81.95	18.40

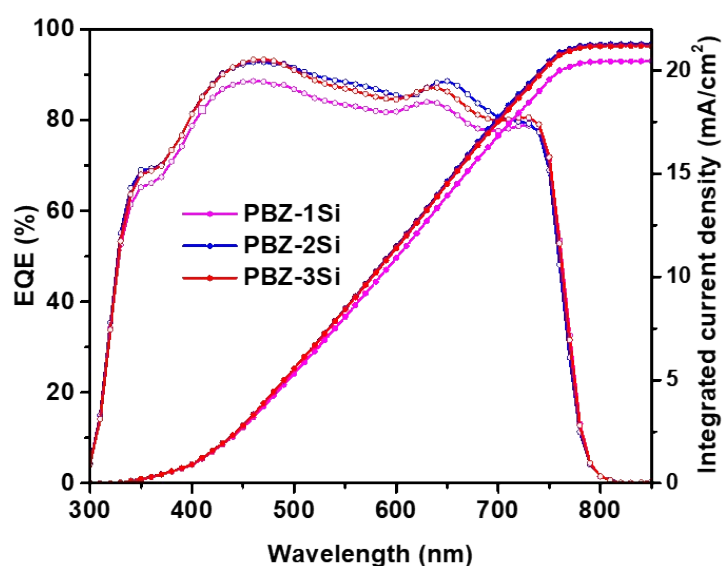


Figure S11. The EQE and the corresponding J_{SC} derived from the PBZ-Si based PSCs.

Table S6. A summary of high FFs and their corresponding efficiencies obtained in various dopant-free HTMs published in recent years and from this work.

HTM	Type	FF (%)	PCE (%)	Device structure	Ref.	Year
P3HT	polymer	82.5	23.2	FTO/SnO ₂ /(FAPbI ₃) _{0.95} (MAPbBr ₃) _{0.05} (DHA)/HTM/Au	7	2021
P3	polymer	80.0	20.3	FTO/mp-SnO ₂ /MAPbI ₃ /HTM/Au	8	2019

PCDTB T1	polymer	78.2	19.1	ITO/TiO ₂ /PC ₆₀ BM/MAPbI ₃ /HTM/MoO ₃ / Au	9	2018
PTEG	polymer	77.0	19.8	FTO/SnO ₂ /CsFAMAPbI _{3-x} Br _x /HTM/Au	10	2018
PBDTT	polymer	76.67	20.28	ITO/SnO ₂ /CsFAMAPbI _{3-x} Br _x /HTM/ MoO ₃ /Ag	11	2020
alkoxy- PTEG	polymer	79.8	21.2	FTO/SnO ₂ /Cs _{0.06} FA _{0.78} MA _{0.16} Pb _{0.94} I _{2.4} Br 0.48/HTM/Au	12	2020
DTB	polymer	69.85	19.68	FTO/c-TiO ₂ /m-TiO ₂ /FA _{0.85} MA _{0.15} PbI _{2.55} Br _{0.45} /HTM/Au	13	2018
PC3	polymer	80.0	20.8	FTO/mp-SnO ₂ /MAPbI ₃ (EACl)/HTM/ Au	14	2020
PBDT[2 F]T	polymer	72.60	17.52	FTO/SnO ₂ /MAPbI ₃ /HTM/Ag	15	2020
PBDT- N20	polymer	75	18.9	FTO/TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.85} (MAPb Br ₃) _{0.15} /HTM/Au	16	2020
2DP- TDB	polymer	79.57	22.17	ITO/SnO ₂ /FA _{0.85} MA _{0.15} PbI ₃ /HTM/MoO ₃ /Ag	17	2021
DTB(3% DEG)	polymer	72.2	20.10	FTO/TiO ₂ /mp-TiO ₂ /Cs _{0.05} FA _{0.81} MA _{0.14} PbI _{2.55} Br _{0.45} /HTM/Au	18	2019
PDCBT	polymer	80.0	21.2	ITO/C ₆₀ -SAM/SnO _x /PC ₆₀ BM/FA _{0.83} MA 0.17Pb _{1.1} Br _{0.50} I _{2.80} /HTM/Ta-WO _x /Au	19	2017
TCTA- BVP	polymer	78.7	18.27	FTO/TiO ₂ /C ₆₀ /MAPbI ₃ /HTM/Ag	20	2016
P25NH	polymer	83.2	17.3	ITO/SnO ₂ /(MA _{0.8} FA _{0.2})Pb(I _{0.93} Cl _{0.07}) ₃ /H TM/Ag	21	2020
PBT1-C	polymer	81.22	19.06	FTO/TiO ₂ /mp-TiO ₂ /CsFAMA(xGUAPb I ₃)/HTM/MoO ₃ /Ag	22	2019
FBA3	small molecula r	79.9	19.27	ITO/C ₆₀ /MAPbI _x Cl _{3-x} /HTM/MoO ₃ /Ag	23	2019
TPA- ANT- TPA	small molecula r	79.6	17.5	FTO/TiO ₂ /mp-TiO ₂ /MAPbI ₃ /HTM/Ag	24	2018
DTP- C6Th	small molecula r	79.9	21.04	FTO/SnO ₂ /C ₆₀ -SAM/MA _{0.7} FA _{0.3} Pb(I _{0.925} Br _{0.075}) ₃ /PMMA/HTM/Au	25	2019
OMe- TPA- CuPc	organom etallic	76.7	19.67	FTO/SnO ₂ /Cs _{0.05} (MA _{0.13} FA _{0.87}) _{0.95} Pb(I _{0.8} 7Br _{0.13}) ₃ /HTM/Au	26	2019
CuSCN	inorganic	78.2	20.4	FTO/TiO ₂ /mp-TiO ₂ /CsFAMAPbI _{3-x} Br _x / HTM/rGO/Au	27	2017
PBZ-3Si	polymer	84.93	20.18	ITO/SnO₂/CsFAMAPbI_{3-x}Br_x/HTM/ Au		This work

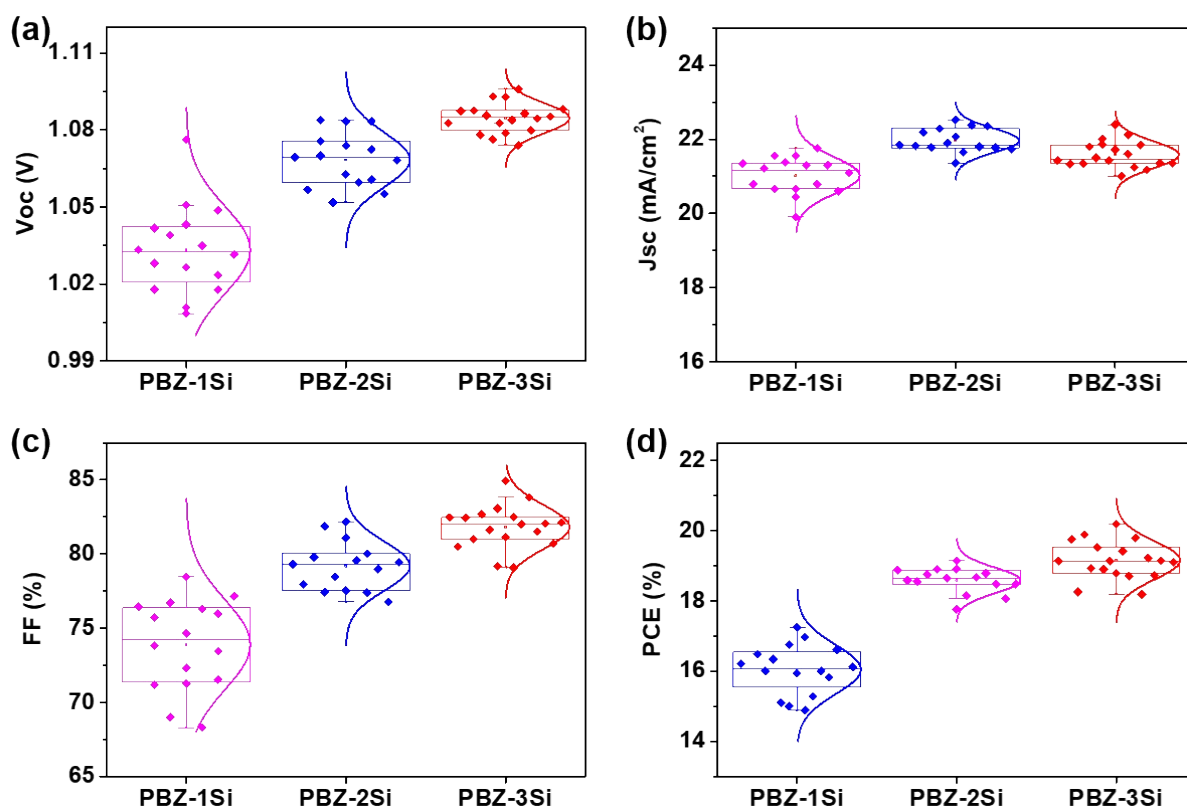


Figure S12. Statistical box plots of the photovoltaic parameters (V_{oc} , J_{sc} , FF, and PCE, respectively) of the devices with different HTMs in the forward scanning direction.

Table S7. Average photovoltaic parameters over 15 devices with different HTMs

HTM	V_{oc} (V)	J_{sc} (mA cm ⁻²)	FF (%)	PCE (%)
PBZ-1Si	1.03 ± 0.02	21.02 ± 0.49	73.89 ± 3.04	16.04 ± 0.70
PBZ-2Si	1.07 ± 0.01	21.96 ± 0.32	79.18 ± 1.65	18.58 ± 0.36
PBZ-3Si	1.08 ± 0.01	21.58 ± 0.36	81.80 ± 1.46	19.15 ± 0.54

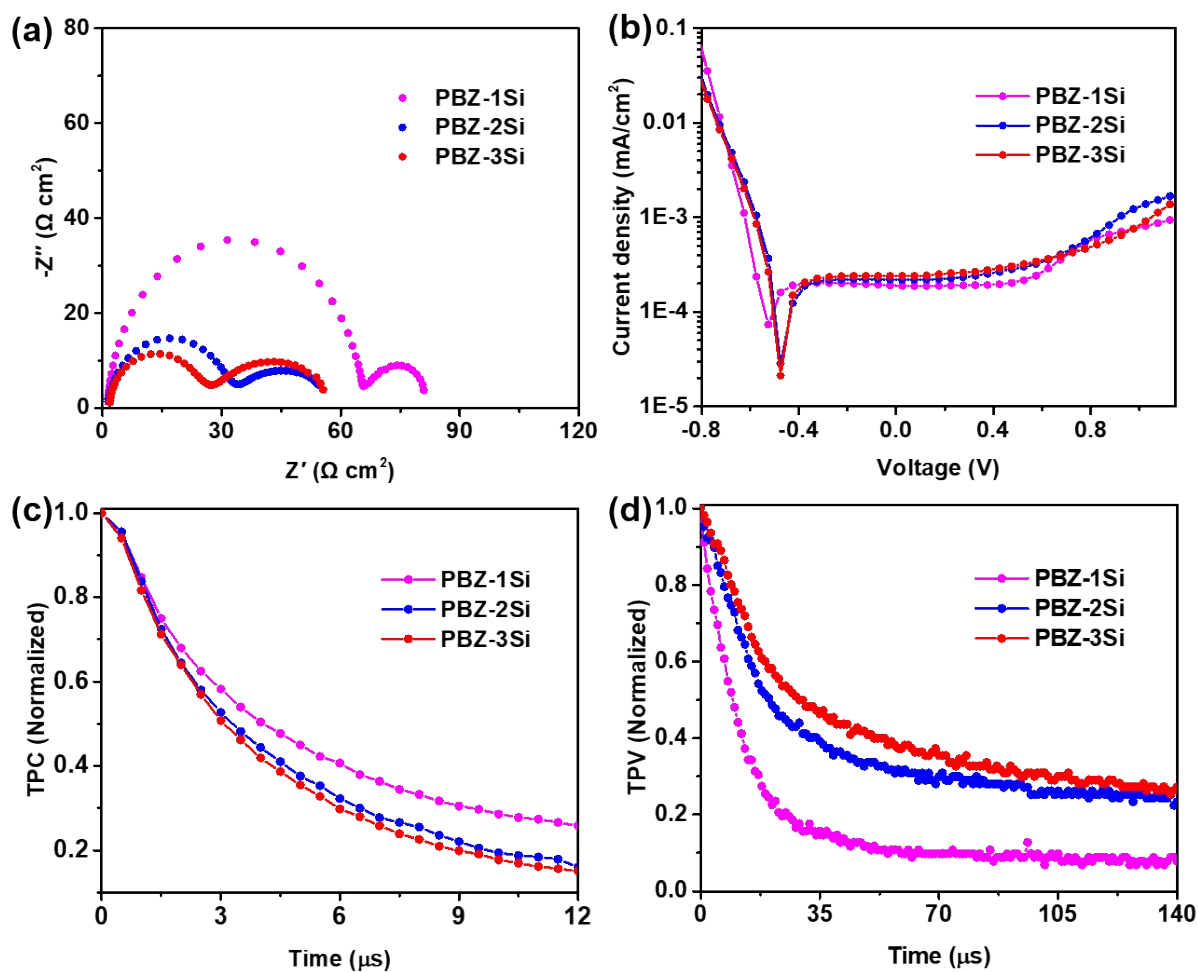


Figure S13. (a) Electrical impedance spectroscopy (EIS), Nyquist plot from the PBZ-Si based devices. (b) Dark J - V curves for devices based on PBZ-Si. (c) Transient photocurrent decay for devices based on PBZ-Si. (d) Transient photovoltage decay for devices based on PBZ-Si.

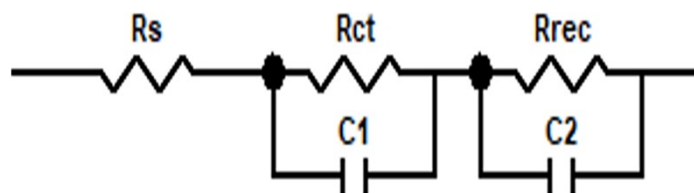


Figure S14. Equivalent circuit model for the Nyquist plots. R_s : series resistance. R_{ct} : transfer/transport resistance. R_{rec} : resistance of interfacial recombination. C : capacitance.

Table S8. EIS derived series resistance (R_s), transport resistance (R_{ct}) and recombination resistance (R_{rec}) for the different devices under 10 mW cm^{-2} illumination at applied voltages (-0.85V).

Samples	R_s ($\Omega \text{ cm}^2$)	R_{ct} ($\Omega \text{ cm}^2$)	R_{rec} ($\Omega \text{ cm}^2$)
PBZ-1Si	1.32	65.16	16.26
PBZ-2Si	1.54	31.37	19.18
PBZ-3Si	1.88	25.18	24.14

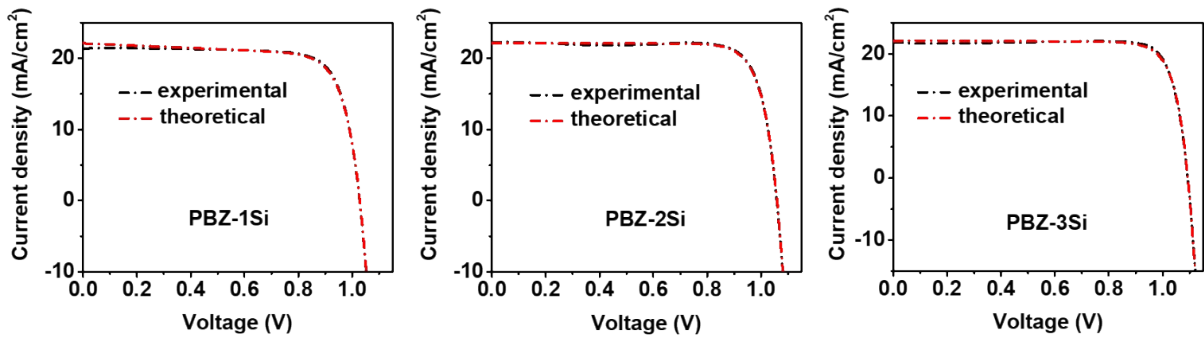


Figure S15. The comparison of J - V curves between the theoretical simulation and experimental measurement. Theoretically fitted J - V curves by modified detailed balance model and experimentally measured J - V characteristics for PBZ-1Si, PBZ-2Si and PBZ-3Si based devices with the champion efficiency, respectively.

Table S9. The parameters retrieved from the J - V curves of different cases.

Samples	γ_{bulk} (s^{-1})	U_{surf} ($\text{nm cm}^3 \text{ s}^{-1}$)	R_s ($\Omega \text{ cm}^2$)	R_{sh} ($\Omega \text{ cm}^2$)
PBZ-1Si	8.29×10^7	2.90×10^{-6}	7.45	1.15×10^4
PBZ-2Si	4.08×10^7	1.13×10^{-5}	10.38	59.41×10^4
PBZ-3Si	1.53×10^7	6.65×10^{-6}	10.13	9.25×10^4

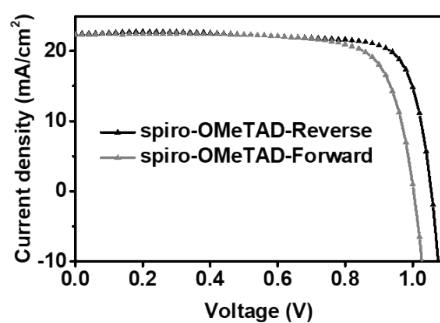


Figure S16. The J–V curves of PSCs with doped spiro-OMeTAD as HTMs. The concentration spiro-OMeTAD is typically 72.3 mg/mL with 17.5 μ L of Li-TFSI solution (520 mg mL⁻¹) in acetonitrile and 28.5 μ L of TBP as dopants.

Table S10. Photovoltaic parameters of PSCs based on doped spiro-OMeTAD as HTMs

HTM	Scan	V_{OC} (V)	J_{SC} (mA cm ⁻²)	FF (%)	PCE (%)
spiro-OMeTAD	Forward	1.00	22.32	76.57	17.09
	Reverse	1.05	22.50	79.53	18.79

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