

## Supplementary information

### **Large electrocaloric effect with high thermal and electric field cycling stability in solution-processed Y:HfO<sub>2</sub> thin films**

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Table S1: The references for the data plotted in the comparison in Figure 1 in the article.

Code used	Material	T <sub>m</sub> (K)	ΔT (K)	Pb-free	Reference
Single Crystals					
BT	BaTiO <sub>3</sub>	288	1.33	✓	[1]
SBN	Ce-Sr <sub>x</sub> Ba <sub>1-x</sub> Nb <sub>2</sub> O <sub>6</sub>	334	0.85	✓	[2]
BT	BaTiO <sub>3</sub>	402	0.9	✓	[3]
PMN-PT	0.7Pb(Mg <sub>1/3</sub> Nb <sub>2/3</sub> )O <sub>3</sub> -0.3PbTiO <sub>3</sub>	403	2.7	×	[4]
BT	BaTiO <sub>3</sub>	283	1.4	✓	[5]
BT-BMT	BaTiO <sub>3</sub> -Bi(Mg <sub>1/2</sub> Ti <sub>1/2</sub> )O <sub>3</sub>	416	1.21	✓	[6]
BHT	BaHf <sub>0.1</sub> Ti <sub>0.89</sub> O <sub>3</sub>	343	0.35	✓	[7]
PST	Pb(Sc <sub>1/2</sub> Ta <sub>1/2</sub> )O <sub>3</sub>	300	3.7	×	[8]
BST	Ba <sub>0.65</sub> Sr <sub>0.35</sub> TiO <sub>3</sub>	343	2.1	✓	[9]
PLZST	Pb <sub>0.97</sub> La <sub>0.02</sub> (Zr <sub>0.80</sub> Sn <sub>0.14</sub> Ti <sub>0.06</sub> )O <sub>3</sub>	313	-12.9	×	[10]
PLZST	Pb <sub>0.97</sub> La <sub>0.02</sub> (Zr <sub>0.80</sub> Sn <sub>0.14</sub> Ti <sub>0.06</sub> )O <sub>3</sub>	323	-14.1	×	[10]
BNT-BT	0.94(Bi <sub>0.5</sub> Na <sub>0.5</sub> )TiO <sub>3</sub> -0.06BaTiO <sub>3</sub>	339	3.2	✓	[11]
PMN-PT	0.7[PbMg <sub>1/3</sub> Nb <sub>2/3</sub> O <sub>3</sub> ]-0.3[PbTiO <sub>3</sub> ]	429	2.7	×	[12]
Thin films					
HYO(ECE1)	Y-HfO <sub>2</sub>	358	24.84	✓	This work
HYO(ECE2)	Y-HfO <sub>2</sub>	358	24.35	✓	This work
HYO(ECE3)	Y-HfO <sub>2</sub>	358	19.41	✓	This work
PZT	PbZr <sub>0.95</sub> Ti <sub>0.05</sub> O <sub>3</sub>	495	12	×	[13]
PLZT(8/65/35)	(Pb <sub>0.88</sub> La <sub>0.08</sub> )(Zr <sub>0.65</sub> Ti <sub>0.35</sub> )O <sub>3</sub>	318	40	×	[14]
PBZ*	Pb <sub>0.8</sub> Ba <sub>0.2</sub> ZrO <sub>3</sub>	290	45.3	×	[15]
HSO	Si-HfO <sub>2</sub>	298	9.5	✓	[16]
HAO	Al-HfO <sub>2</sub>	295	5.7	✓	[17]
HGO	Gd-HfO <sub>2</sub>	295	3.1	✓	[17]
PMN-PT	0.65[PbMg <sub>1/3</sub> Nb <sub>2/3</sub> O <sub>3</sub> ]-0.35[PbTiO <sub>3</sub> ]	413	31	×	[18]
PLZST	Pb <sub>0.97</sub> La <sub>0.02</sub> (Zr <sub>0.75</sub> Sn <sub>0.2</sub> Ti <sub>0.07</sub> )O <sub>3</sub>	306	33	×	[19]
HZO	Hf <sub>0.2</sub> Zr <sub>0.8</sub> O <sub>2</sub>	298	13.4	✓	[20]
HZO	Hf <sub>0.3</sub> Zr <sub>0.7</sub> O <sub>2</sub>	448	8.9	✓	[20]
Polymers					
P(VDF-TrFE-CFE)	P(VDF-TrFE-CFE)	303	15	✓	[21]
P(VDF-TrFE)	P(0.55VDF-0.45TrFE)	353	12.6	✓	[22]
P(VDF-TrFE)	P(0.7VDF-0.3TrFE)	390	21.2	✓	[23]
P(VDF-TrFE-CFE)	P(0.562VDF-0.363TrFE-0.076CFE)	350	21.6	✓	[23]
Special form					
Cascade	P(VDF-TrFE-CFE)	300	8.7	✓	[24]
Nanocubes	P(VDF-TrFE-CFE)/ Ba <sub>0.67</sub> Sr <sub>0.33</sub> TiO <sub>3</sub>	316	9.1	✓	[25]
Nanowire	P(VDF-TrFE-CFE)/ Ba <sub>0.67</sub> Sr <sub>0.33</sub> TiO <sub>3</sub>	300	32	✓	[25]
Molecular	Cyclohexylmethylammonium bromide	364	4.2	✓	[26]

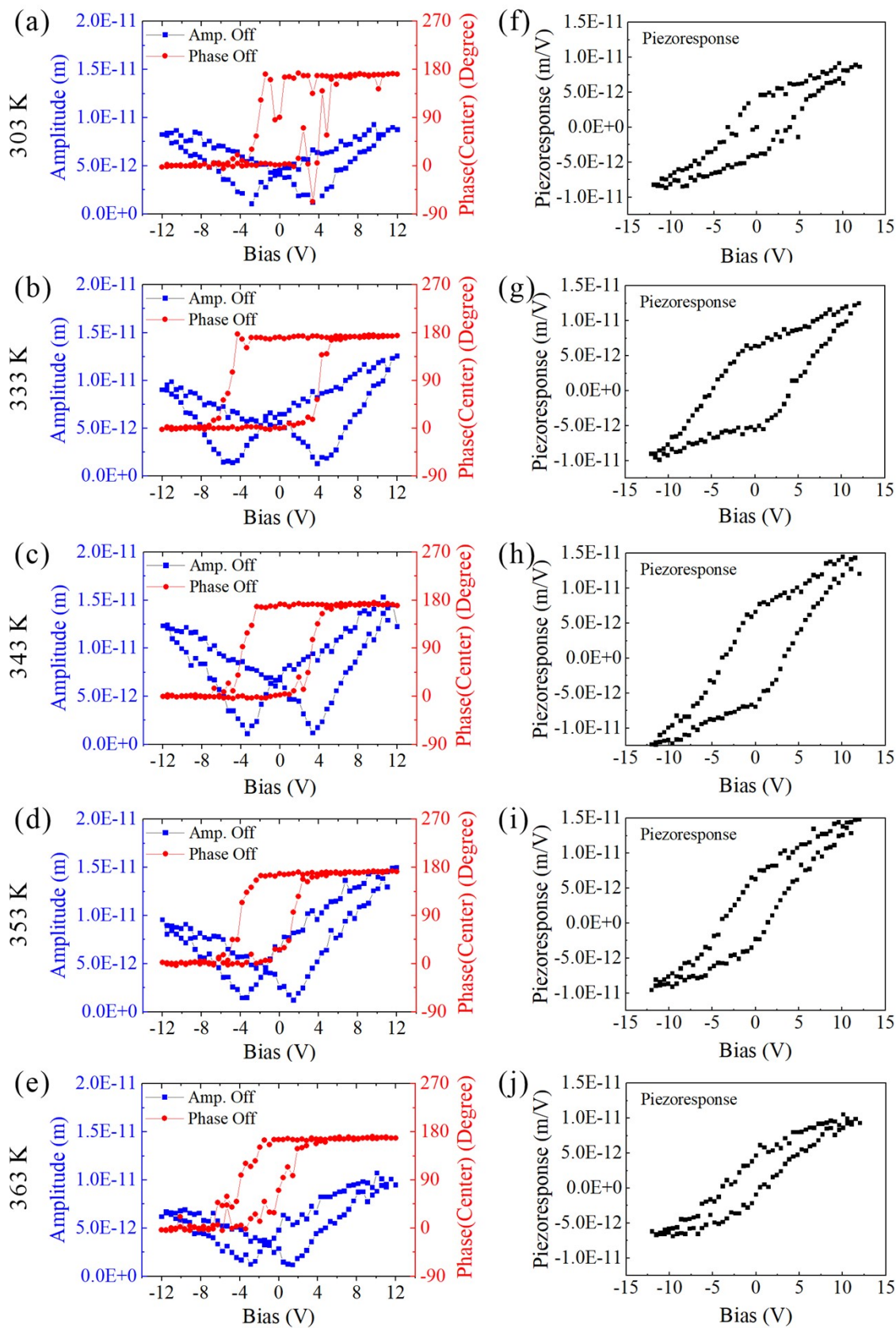


Figure S1: (a)-(e) PFM local hysteresis at temperature 303, 333, 343, 353 and 363 K. (f)-(j) Local piezo-response at those specified temperatures.

## References

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