Supporting Information

Ultrapermeable Nanofiltration Membranes with Tunable Selectivity Fabricated with Polyaniline Nanofibers

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Figure S1. Surface and cross-sectional morphology of the PES microfiltration membrane used as the support membrane.



Figure S2. The crossflow device for evaluating the nanofiltration performance.



Figure S3. The morphology of the synthesized PANI nanofibers.



Figure S4. The thickness of PANI interlayer as a function of the PANI nanofiber loading.



Figure S5. Color change of the PANI interlayer during the interfacial polymerization process: (a, b, c) U-*i*TFC-1%, (d, e, f) U-*i*TFC-0.05%, (g, h, i) D-*i*TFC-1%, and (j, k, l) D-*i*TFC-0.05%.



Figure S6. Comparison of the stability when (a) the HCl molecules are dissociative, (b) the HCl molecules are all bonded to the PIP monomers, (c) the HCl molecules are bonded both to the PIP monomers and to the PANI and (d) HCl molecules are solely bonded to the PANI.



Figure S7. Additional SEM images showing the surface morphology of (a, b) D-*i*TFC-1% and (c, d) D-*i*TFC-0.05%.



Figure S8. PA films supported on silicon wafers for thickness measurements.



Figure S9. Elemental mapping of the positions marked with a red dot.

EL	AN	Series	Unn.	С	С
				norm.	Atom.
			[wt.%]	[wt.%]	[at.%]
С	6	K-	26.47	66.58	81.02
		series			
S	16	K-	10.66	26.82	12.22
		series			
Ν	7	K-	2.20	5.54	5.78
		series			
0	8	K-	0.42	1.07	0.97
		series			
		Total:	39.76	100.00	100.00

Table S1. EDX Elemental Distributions from Fig. S9a

EI	ΛN	Series	Unn	С	С
LL		Series	Unii.	C	
				norm.	Atom.
			[wt.%]	[wt.%]	[at.%]
С	6	К-	25.45	67.62	80.70
		series			
S	16	К-	8.92	23.69	10.59
		series			
Ν	7	К-	2.74	7.29	7.46
		series			
Ο	8	К-	0.53	1.41	1.26
		series			
		Total:	37.64	100.00	100.00

Table S2. EDX Elemental Distributions from Fig. S9b



Figure S10. Surface and cross-sectional morphology of an *i*TFC membrane with the PANI interlayer prepared using a 2 mg cm⁻² loading of PANI nanofibers.

Fig. S10 presents the membrane morphology (1% w/v PIP concentration) of the undoped interlayer built with a reduced areal loading of 2 mg cm⁻², the result indicates that the formation of the nanostructured PA layer can also be achieved with a varied thickness of the PANI interlayer.



Figure S11. C1s high resolution XPS spectra of TFC-C, U-*i*TFC-1%, D-*i*TFC-1%, U-*i*TFC-0.05%, and D-*i*TFC-0.05% in sequence.



Figure S12. Surface and cross-sectional morphology of (a, b, c) TFC-C-UF and (d, e, f) U-*i*TFC-1%-UF.



Figure S13. EDX measurement of (a) emeraldine base and (b) emeraldine salt PANI nanofibers with (c) and (d) as the elemental mapping, respectively.

EL	AN	Series	Unn.	С	С
				norm.	Atom.
			[wt.%]	[wt.%]	[at.%]
С	6	K-	21.87	62.62	76.01
		series			
S	16	K-	8.54	24.45	11.11
		series			
Ν	7	K-	3.83	10.97	11.41
		series			
0	8	K-	0.45	1.30	1.19
		series			
C1	17	K-	0.23	0.67	0.27
		series			
		Total:	34.93	100.00	100.00

Table S3. Elemental Distribution of Emeraldine Base PANI Nanofibers

Table S4. Elemental Distribution of Emeraldine Salt PANI Nanofibers

EL	AN	Series	Unn.	С	С
				norm.	Atom.
			[wt.%]	[wt.%]	[at.%]
С	6	K-	29.01	59.87	74.76
		series			
S	16	K-	9.22	19.03	8.90
		series			
Ν	7	K-	5.16	10.65	11.40
		series			
0	8	K-	0.49	1.01	0.95
		series			
Cl	17	K-	4.58	9.44	3.99
		series			
		Total:	48.46		



Figure S14. A comparison of Na_2SO_4 rejection and PWF of *i*TFC membranes with PANI interlayers in this work to *i*TFC membranes with other materials-based interlayers.



Figure S15. Long-term stability test of nanofiltration. Test conditions: 2,000 ppm Na₂SO₄ solution, 0.5 MPa, 25 °C.