

Supporting Information

Improving the Efficiency and Stability of Inorganic Red Perovskite Light Emitting Diodes by Traces of Zinc Ions

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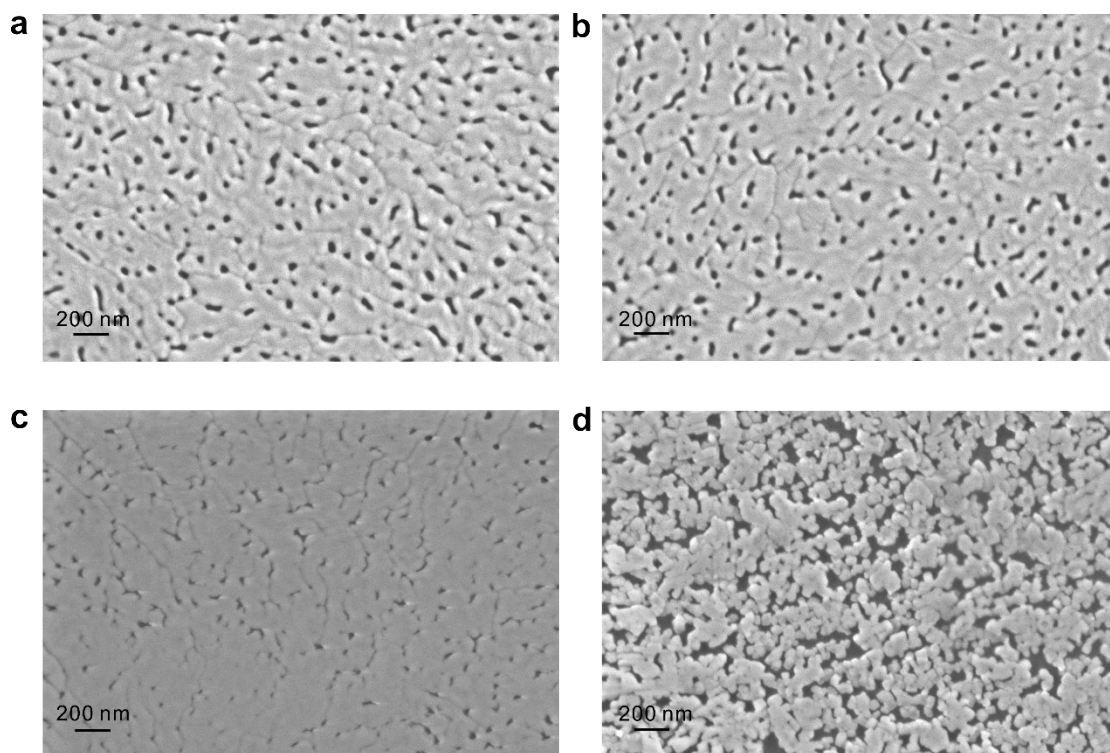


Figure S1. SEM images of the prepared $\text{CsPbI}_{3-x}\text{Br}_x$ films by adding different amounts of ZnI_2 into the precursor solution. (a) 0.0%, (b) 0.5%, (c) 1.0%, and (d) 1.5% molar ratio of Zn/Pb.

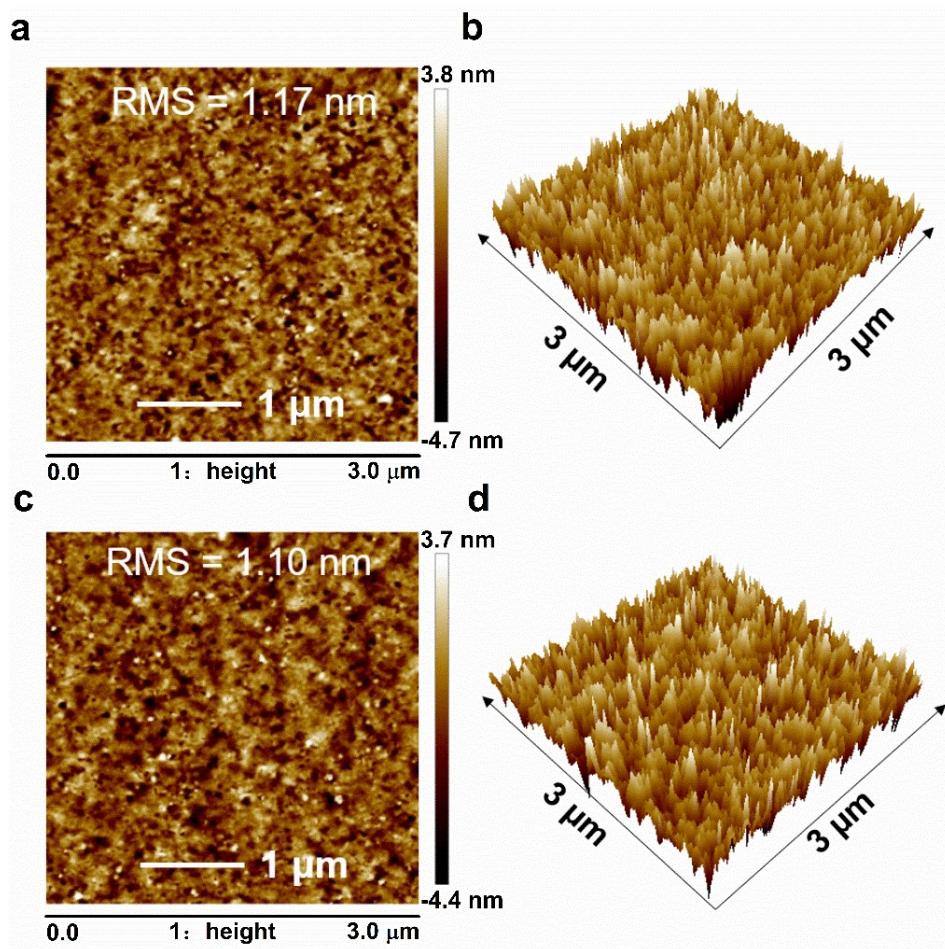


Figure S2. AFM images of the prepared perovskite films. Tapping-mode AFM images of the pristine $\text{CsPbI}_{3-x}\text{Br}_x$ film (a)(b). The film RMS roughness is about 1.17 nm, the scan area is $3 \mu\text{m} \times 3 \mu\text{m}$. Tapping-mode AFM images of the $\text{CsPbI}_{3-x}\text{Br}_x$ thin film incorporating with ZnI_2 (c)(d). The film RMS roughness is about 1.10 nm, the scan area is $3 \mu\text{m} \times 3 \mu\text{m}$.

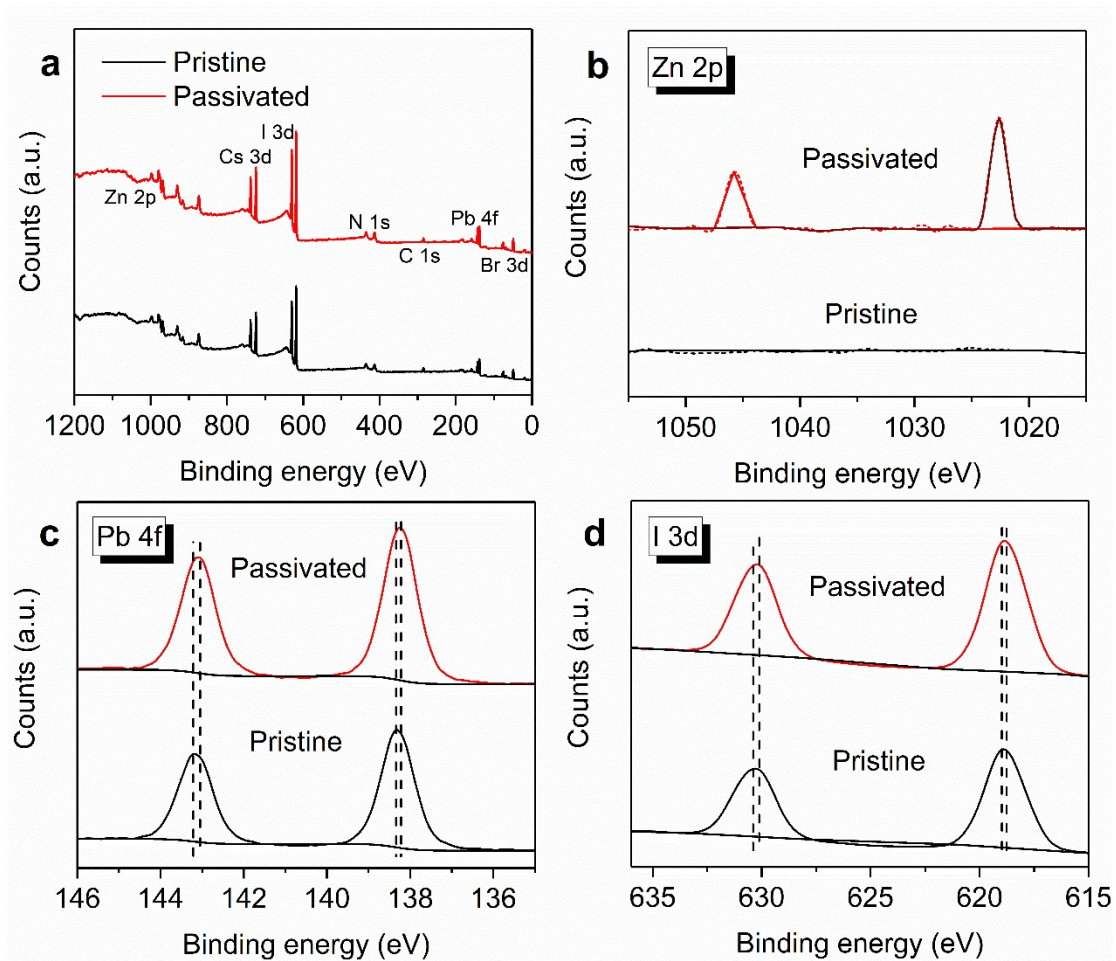


Figure S3. (a) XPS spectra of pristine and ZnI₂-passivated CsPbI_{3-x}Br_x films. XPS analysis of pristine and ZnI₂-passivated films for (b) Zn 2p, (c) Pb 4f and (d) I 3d.

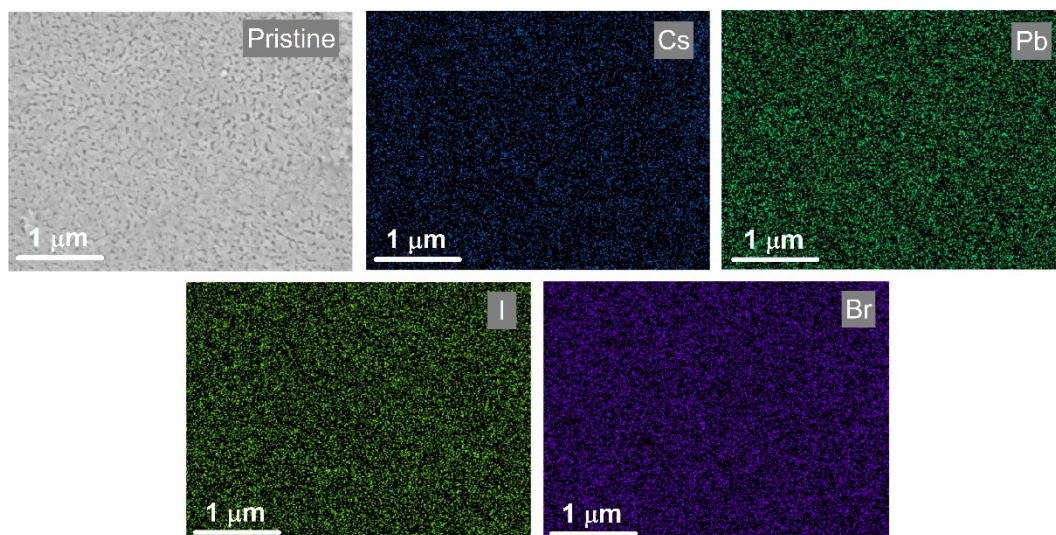


Figure S4. SEM image and corresponding elemental mapping images of pristine $\text{CsPbI}_{3-x}\text{Br}_x$ perovskite thin film. Scale bar: 1 μm .

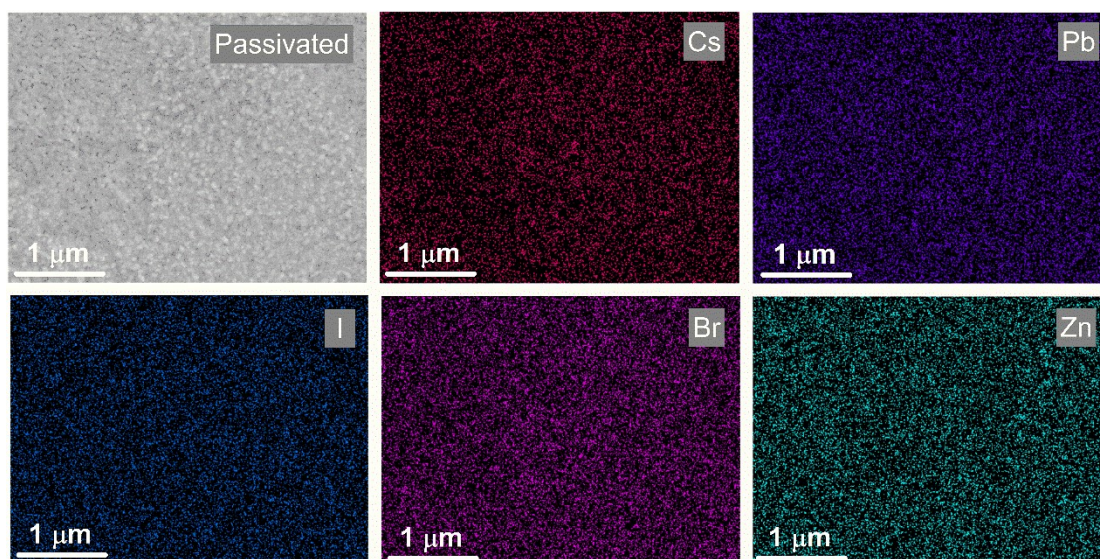


Figure S5. SEM image and corresponding elemental mapping images of passivated $\text{CsPbI}_{3-x}\text{Br}_x$ perovskite thin film. Scale bar: 1 μm .

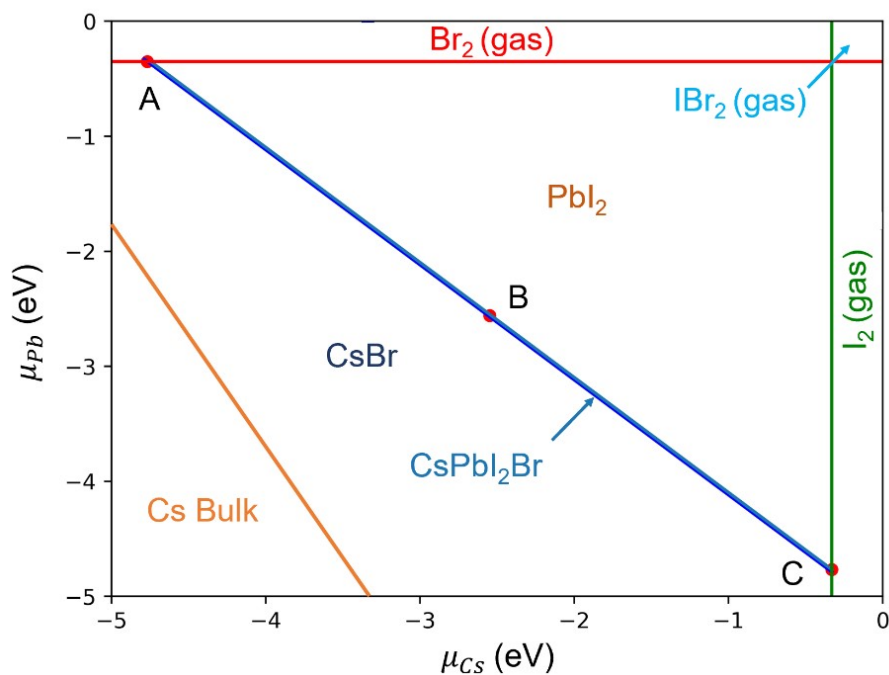


Figure S6. Stability regions of various compounds against Cs and Pb chemical potentials in $CsPbI_2Br$. The region indicates the available equilibrium chemical potential region for $CsPbI_2Br$. Three representative points (A: Br-rich/I-poor, B: moderate, C: I-rich/Br-poor conditions) are selected for the formation energy calculations.

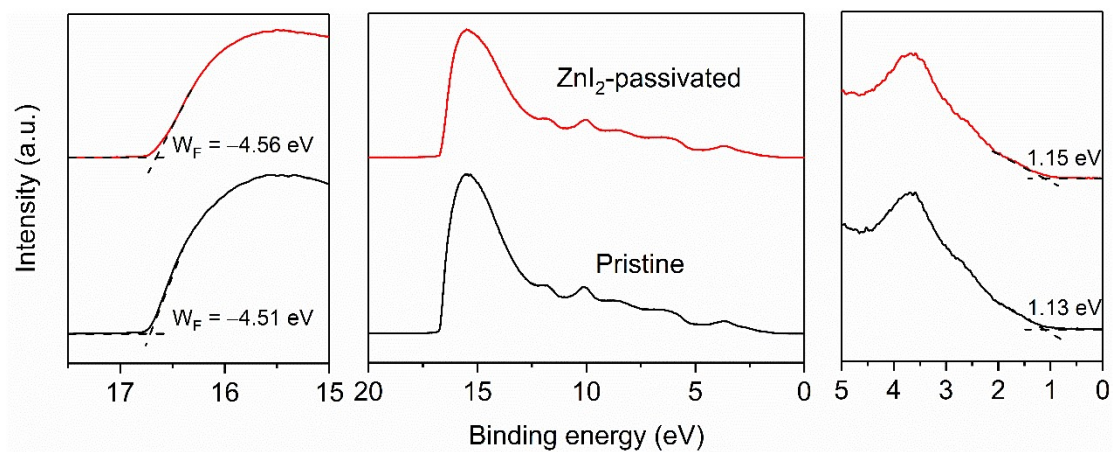


Figure S7. UPS spectra of the pristine $\text{CsPbI}_{3-x}\text{Br}_x$ and ZnI_2 -passivated $\text{CsPbI}_{3-x}\text{Br}_x$ films. Their W_F s are -4.51 eV and -4.56 eV, respectively. Their HOMOs are -5.71 eV and -5.64 eV, respectively.

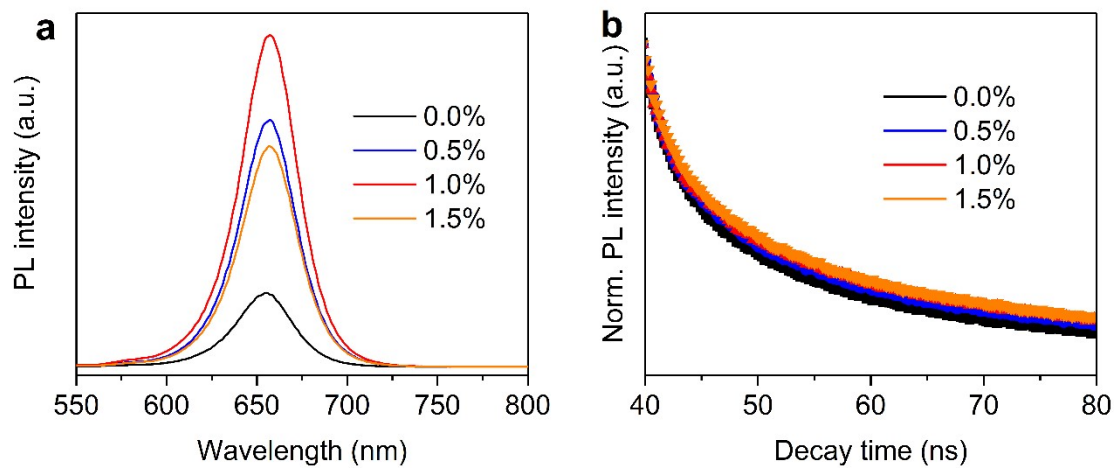


Figure S8. PL spectra (a) and TRPL spectra (b) of pristine and ZnI₂-passivated CsPbI_{3-x}Br films with different molar ratio of Zn/Pb on PEDOT:PSS films.

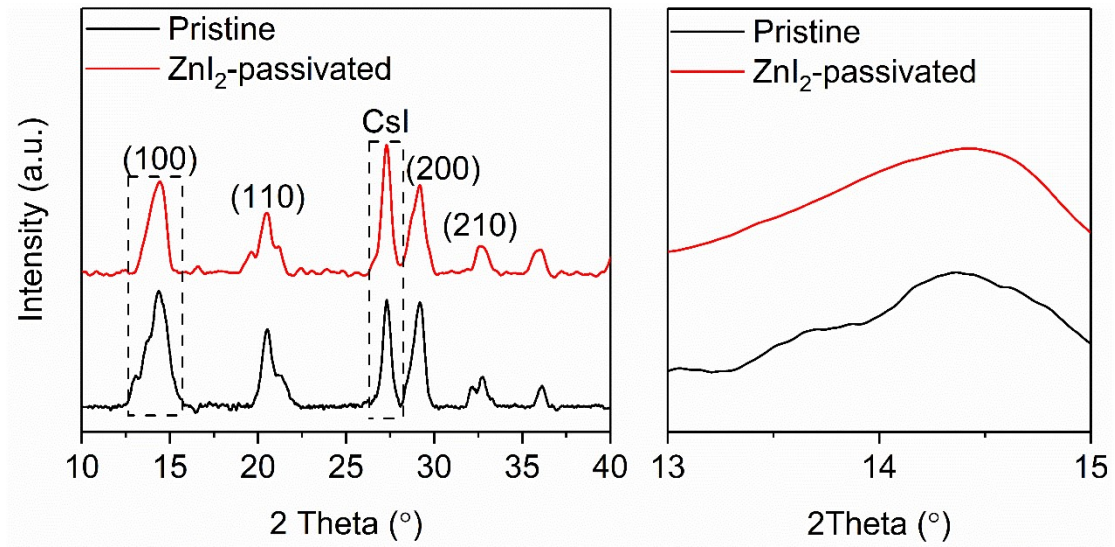


Figure S9. Comparison of the perovskite crystallinity. XRD patterns of the CsPbI_{3-x}Br_x films with and without ZnI₂ passivation.

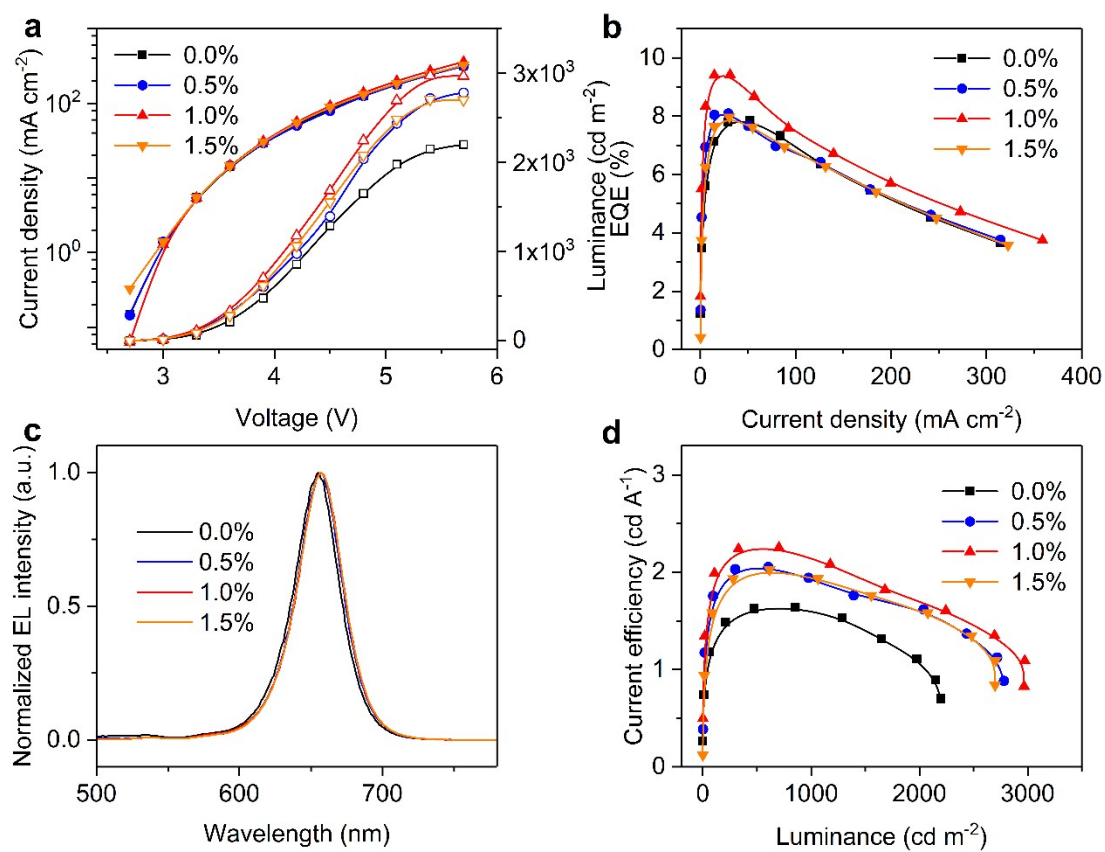


Figure S10. (a) J-V-L curves, (b) EQE versus current density curves, (c) EL spectra and (d) CE versus luminance of CsPbI_{3-x}Br_x PeLEDs with different molar ratio of Zn/Pb.

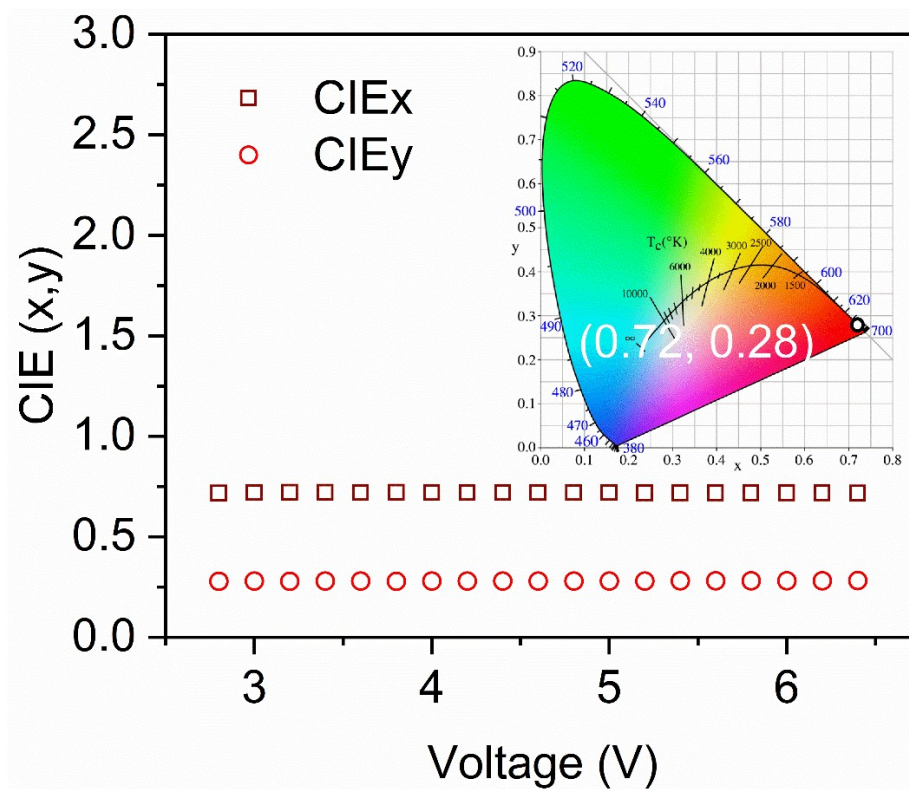


Figure S11. CIE coordinates of the ZnI_2 -passivated $\text{CsPbI}_{3-x}\text{Br}_x$ PeLEDs under different bias voltages. Inset is CIE color coordinates under an applied voltage of 3.6 V.

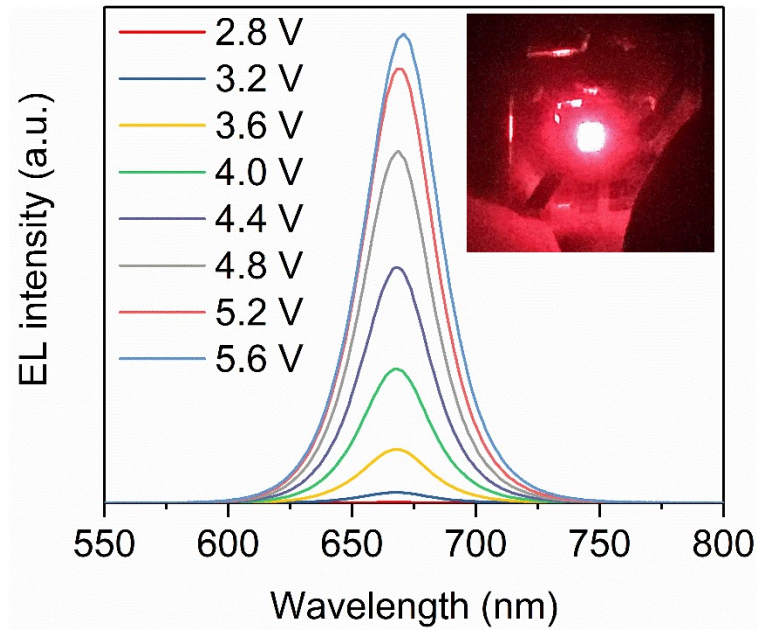


Figure S12. EL spectra of the ZnI₂-passivated CsPbI_{3-x}Br_x based PeLEDs at the bias voltages from 2.8 V to 5.6 V. Inset is a photograph of red emission PeLED operating under 5.4 V forward bias.

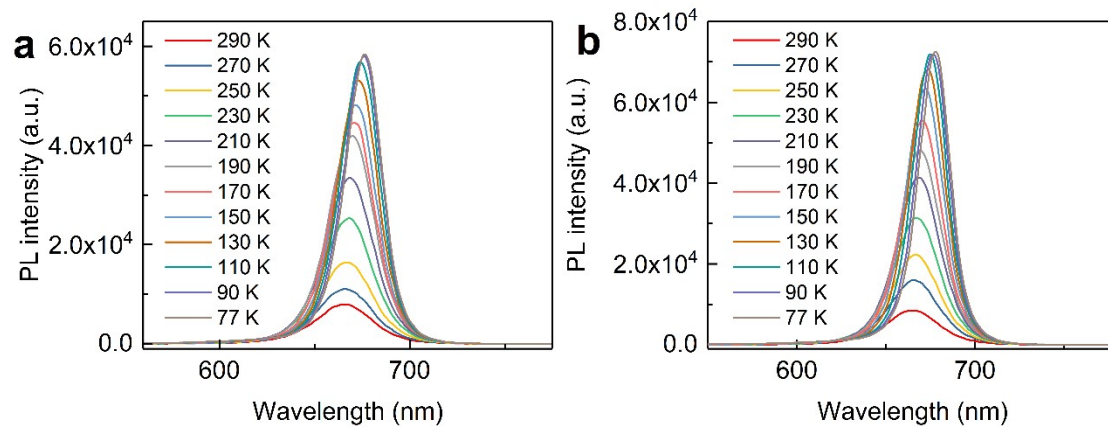


Figure S13. Temperature-dependent PL spectra of (a) pristine and (b) passivated $\text{CsPbI}_{3-x}\text{Br}_x$.

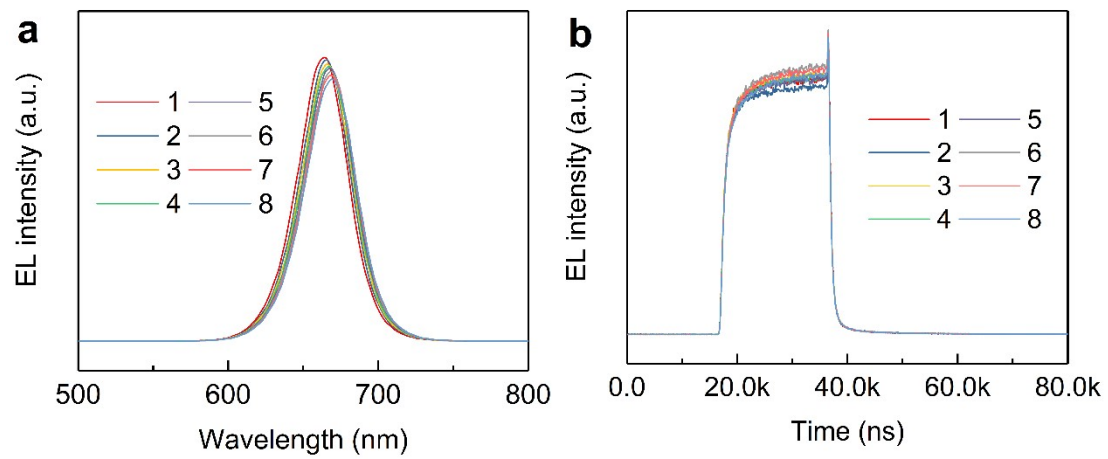


Figure S14. (a) Time-dependent EL spectrum and (b) time-dependent TRPL of passivated device. (testing every two minutes)

Table S1. Fitting parameters of the PL decay lifetime for the pristine CsPbI_{3-x}Br_x and ZnI₂-passivated CsPbI_{3-x}Br_x films with different molar ratio of Zn/Pb.

Perovskite	τ_1 [ns]	τ_2 [ns]	τ_3 [ns]	τ_{ave} [ns]
0.0 % (Pristine)	4.83	30.15	146.27	75.81
0.5%	5.19	33.34	168.70	96.82
1.0%	5.27	33.98	167.48	100.92
1.5%	4.10	26.48	142.02	94.51

Table S2. Device performances of red emission PeLEDs based on ZnI₂-passivated CsPbI_{3-x}Br_x perovskite films.

Devices	V _{on} [V]	EL peak [nm]	L _{max} [cd m ⁻²]	EQE _{max} [%]	CE _{max} [cd A ⁻¹]	T ₅₀ * [h]
Pristine	2.75	666	2020	6.94	1.46	2269.0
ZnI ₂ -passivated	2.70	666	4258	9.93	2.66	2667.0

* The initial luminance is 100 cd m⁻².

Table S3. Calculated formation energies (in eV) for defects in Zn doped CsPbI₂Br at the three selected points A, B and C in **Figure S6**.

	A	B	C
Cs substitution	0.84	3.060814	5.279681
Pb substitution	4.30	2.09	-0.12

Table S4. The summarization of the performances of all-inorganic red emission PeLEDs with perovskite NCs or perovskite films emitters (the peak wavelength is from 630 nm to 700 nm).

Perovskite Material	Device Structure	Device Performance Parameters					Ref.
		EL (nm)	EQE (%)	L_{max} (cd m ⁻²)	η_{CE} (cd A ⁻¹)	V_{on} (V)	
CsPb(Br/I) ₃ NCs	ITO/ZnO-PEI /Perovskite NCs/CBP/TCTA/MoO _x /Au	648	6.30	2216	6.42	1.9	(1)
NMA ₂ Cs _{n-1} Pb _n I _{3n+1}	ITO/ ZnO/ PEIE/Perovskite/ TFB/MoO _x /Au	689	3.7	440	-	2.0	(2)
PBA ₂ Cs _{n-1} Pb _n I _{3n+1}	ITO/NiO/TFB/PVK/Perovskite/TP Bi/Ca/Al	683	7.3	~100	-	~3.2	(3)
PEO/ α -CsPbI ₃	ITO/ZnO/PEI/Perovskite/ Poly-TPD/WO ₃ /Al	695	1.12	101	-	4.7	(4)
CsPbI ₃ NCs (IDA-treated)	ITO/PEDOT:PSS/Poly-TPD/ Perovskite NCs/TPBi/LiF/Al	688	5.02	748	-	4.1	(5)
CsPb(Br/I) ₃ NCs	ITO/PEDOT:PSS/Poly-TPD/ Perovskite NCs/TPBi/Liq/Al	645	14.1	794	11.6	2.7	(6)
		653	21.3	500	10.6	2.8	
CsPb(Br/I) ₃ NCs	ITO/PEDOT:PSS/Poly-TPD/ Perovskite NCs/TPBi/Liq/Al	637	3.55	2671	2.97	3.6	(7)
BA ₂ Cs _{n-1} Pb _n I _{3n+1} / PEO	ITO/PEDOT:PSS/Poly-TPD/ Perovskite/BCP/LiF/Al	680	6.23	1392	1.71	~2.6	(8)
NMA ₂ Cs _{n-1} Pb _n I _{3n+1}	ITO/ ZnO/ PEIE/ Perovskite/TFB/MoO _x /Au	694	7.3	732	0.36	1.9	(9)
PEOXA/CsPbBr _{0.6} I _{2.4}	ITO/ZnMgO/Perovskite/ Poly-TPD/MoO ₃ /Ag	668	6.55	338	1.36	1.5	(10)
Ag-doped CsPbI ₃ NCs	Ag/PEDOT:PSS/Perovskite NCs/TCTA/MoO ₃ /Au	690	11.2	1106	-	2.5	(11)
PBA ₂ Cs _{n-1} Pb _n I _{3n+1}	ITO/PEDOT:PSS/Poly-TPD/ PVK/Perovskite/TPBi/LiF/Al	664	13.3	968	-	2.6	(12)
CsPb _{0.64} Zn _{0.36} I ₃ NCs	ITO/ZnO/ PEI/Perovskite NCs/TCTA/MoO ₃ /Au	668	15.1	2202	-	2.0	(13)
NEA/ α -CsPbI ₃	ITO/PEDOT:PSS/PVK/ Perovskite/TPBi/LiF/Al	682	8.65	210	-	2.8	(14)
CsPb(Br _{0.43} I _{0.57}) ₃	ITO/PTAA/PEDOT:PSS/ Perovskite/TPBi/LiF/Al	664	0.8	2765	0.84	2.5	(15)
4-F-PMAl/ CsPbI _{2.8} Br _{0.2}	ITO/Poly-TPD/Perovskite/ TPBi/LiF/Al	689	18.6	~60	-	~3.0	(16)

PEA ₂ Cs _{n-1} Pb _n I _{3n+1} (Addition of ZnI ₂)	ITO/PEDOT:PSS/Perovskite/ TPBi/LiF/Al	666	9.93	4258	2.66	2.8	This work
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