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Supporting Information

Novel CuBr-assisted graphdiyne synthesis strategy and application

for efficient photocatalytic hydrogen evolution

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Fig.S1 XRD pattern of (a) CuI, (b)GDY(CI), GDY-CuI, and CuI.



Fig.S2 Raman pattern of (a) GDY(CI), the insert picture is GDY-CuI, (b)FT-IR spectra of GDY(CI) and GDY-CuI.



After adding sacrificial reagent

Ultrasonic treatment

After light irradition for 5h

Fig.S3 Color change of CuBr and CuI in the photocatalytic hydrogen evolution process.



Fig.S4 (a) Photoluminescence spectroscopy of EY, GDY, GDY-CuI, and GDY-CuBr, (b) time-resolved photoluminescence spectra of EY, GDY-CuI, and GDY-CuBr.

Tuble 51 Exponential out to mang parameters of emission about 161 me samples.				
Samples	τ_1 [ns]	τ_2 [ns]	χ^2	$\tau_{\rm av}[{\rm ns}]$
EY	0.21(B ₁ =100%)		0.85	0.21
GDY-CuI	4.52(B ₁ =0.54%)	0.20(B ₂ =99.46%)	0.89	0.20
GDY-CuBr	3.87(B ₁ =0.94%)	0.17(B ₂ =99.06%)	1.01	0.17

Table S1 Exponential curve fitting parameters of emission decay for the samples.



Fig.S5 (a) Electrochemical impedance spectroscopy of GDY, and GDY-CuI, (b) Linear sweep voltammograms of GDY, CuI, GDY-CuI, and GDY-CuBr.