

## Supporting information

### **Highly luminescent broadband phosphors based on acid solvent coordinated two-dimensional layered tin-based perovskites**

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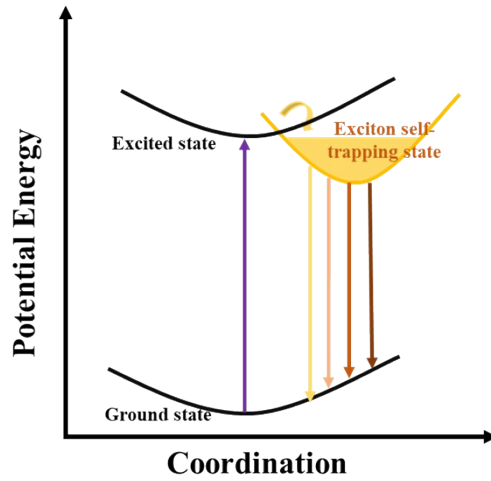


Fig. S1 Schematic of the process of emission from the exciton self-trapping state.

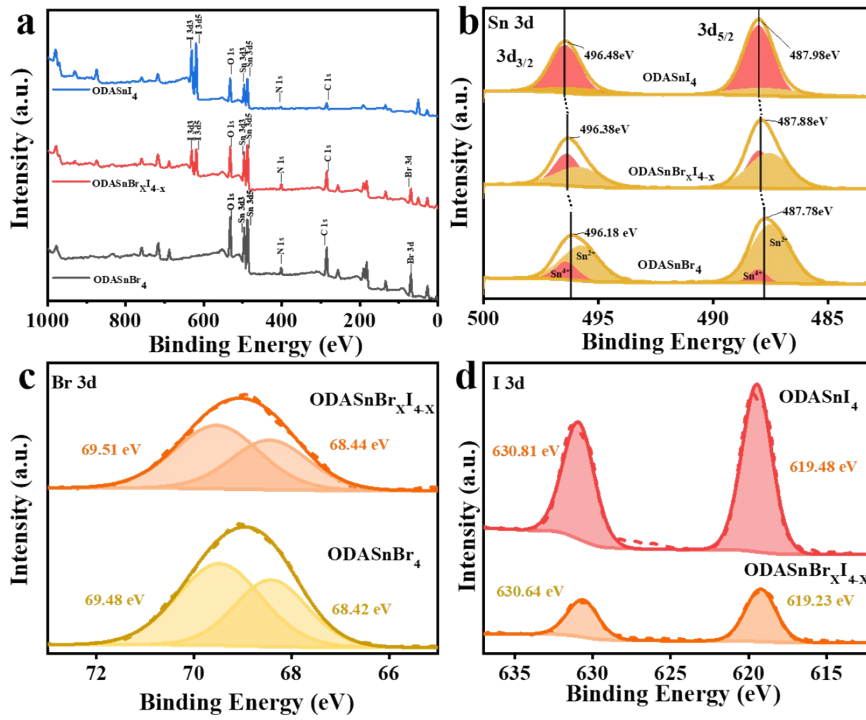


Fig. S2 X-ray photoelectron spectroscopy (XPS) of ODASnBr<sub>4</sub>, ODASnBr<sub>x</sub>I<sub>4-x</sub> and ODASnI<sub>4</sub>. (a) survey spectra of ODASnBr<sub>4</sub>, ODASnBr<sub>x</sub>I<sub>4-x</sub> and ODASnI<sub>4</sub>; XPS spectra of (b) Sn3d, (c) Br3d, and (d) I3d core levels.

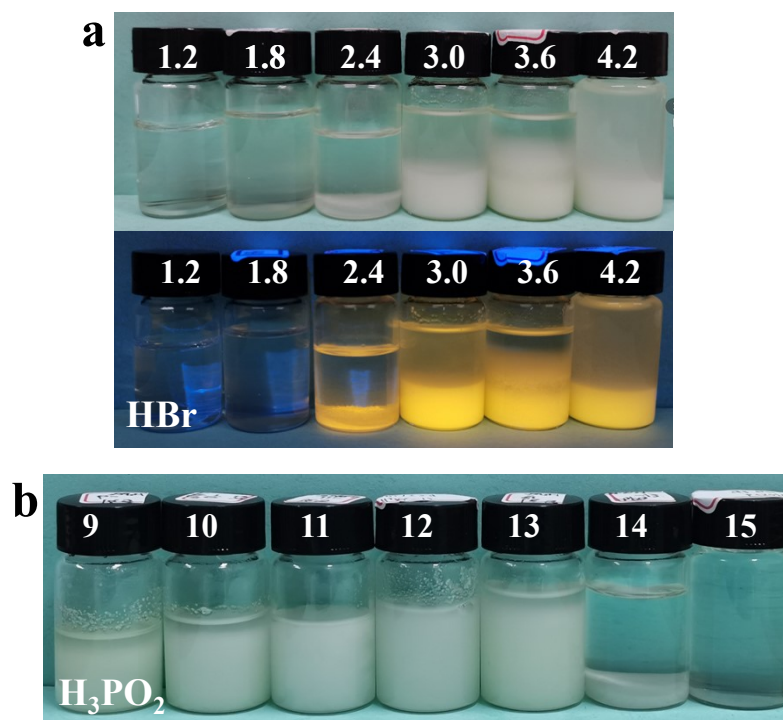


Fig. S3 Photographs of the ODASnBr<sub>4</sub> suspension synthesized by different volumes of (a) HBr and (b) H<sub>3</sub>PO<sub>2</sub> respectively.

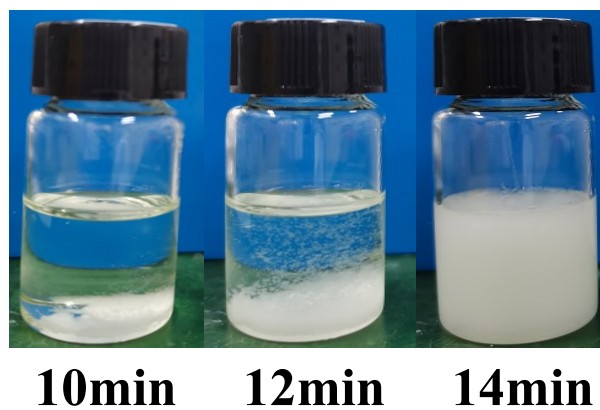


Fig. S4 The production process of sample with H<sub>3</sub>PO<sub>2</sub> of 13 mL. The time at the bottom of the picture represents the time after the solution cooled naturally from 80 °C.

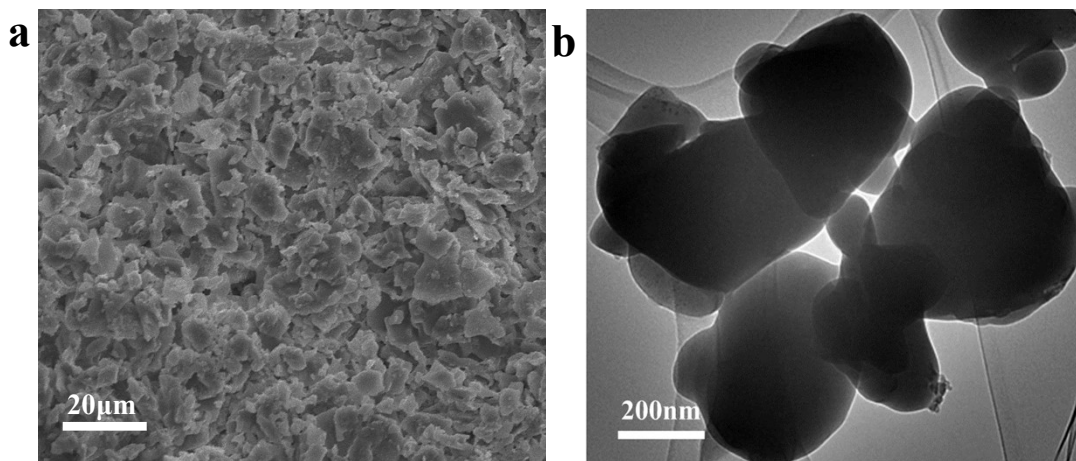


Fig. S5 (a) SEM images and (b) TEM images of ODASnBr<sub>4</sub> (samples HBr 3.0 mL).

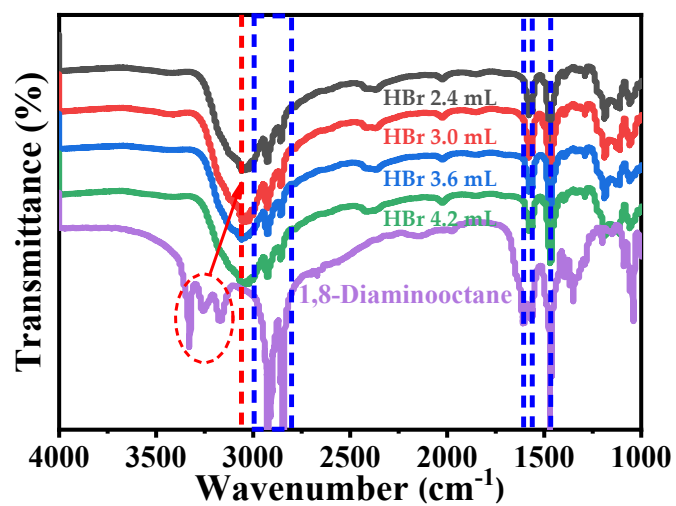


Fig. S6 FTIR spectra of ODASnBr<sub>4</sub> perovskite and the reactants used in its synthesis.

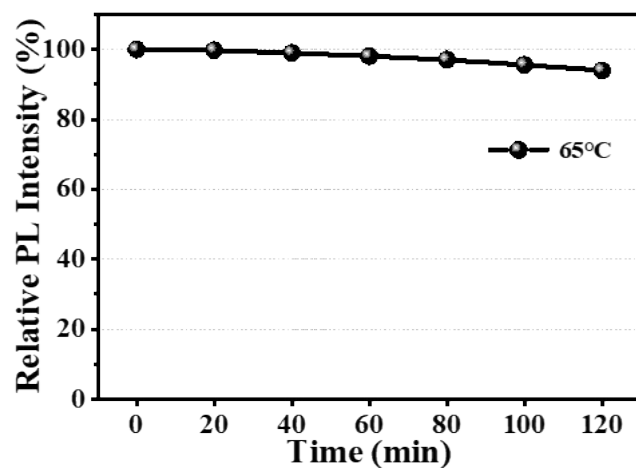


Fig. S7 Relative PL intensity of ODASnBr<sub>4</sub> at 65 °C for 120 mins.

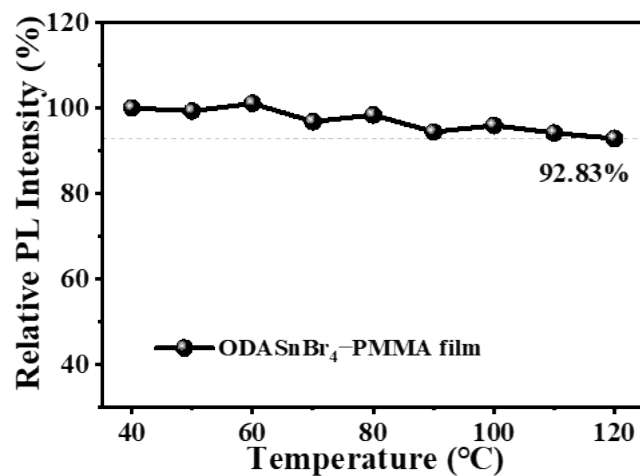


Fig. S8 Relative PL intensity of ODASnBr<sub>4</sub>-PMMA film heat from 40 to 120 °C.

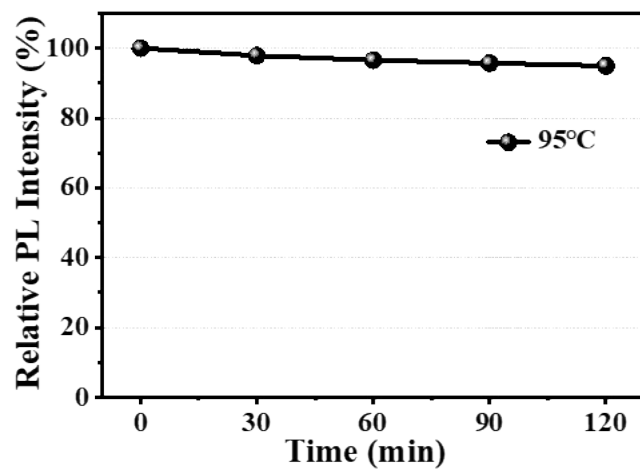


Fig. S9 Relative luminance efficiency of film at 95 °C for 120 min.

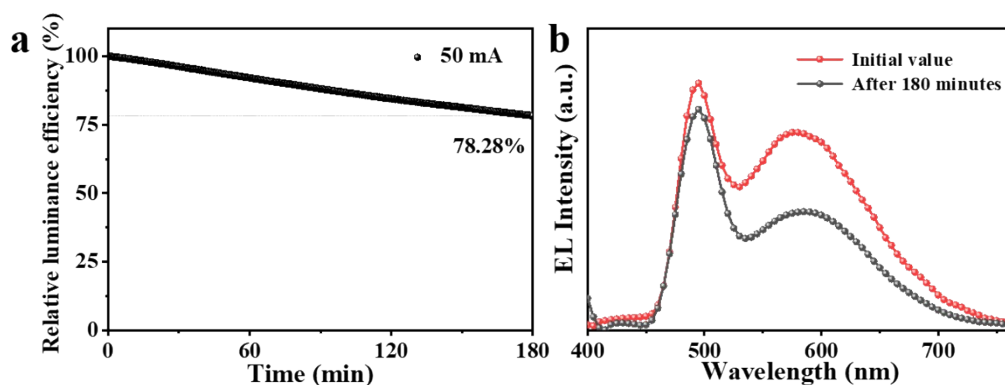


Fig. S10 (a) Relative luminance efficiency of as-fabricated WLEDs and (b) EL spectra of different times.

Table S1. The comparison of the XRD with other reported results.

Chemical formula	$2\theta/^\circ$			Ref.
	(0 0 2)	(0 0 4)	(0 0 8)	
ODASnBr <sub>4</sub>	6.20	12.47	24.77	S1
ODASnBr <sub>4</sub> [2%-DCM]	6.18	12.42	-	S1
ODASnBr <sub>4</sub> [2%-CFM]	6.13	12.30	23.7	S1
ODASnBr <sub>4</sub>	6.71	13.40	-	S2
Calculated ODASnBr <sub>4</sub>	6.56	13.15	-	S2
ODASnBr <sub>4</sub> [EtOH]	6.70	13.36	29.9	S2
ODASnBr <sub>4</sub> [PrOH]	6.41	12.80	-	S2
ODASnBr <sub>4</sub> [BuOH]	6.45	12.86	-	S2
ODASnBr <sub>4</sub> [PeOH]	6.42	12.80	-	S2
ODASnBr <sub>4</sub> [alcohol]	6.12-6.28	-	-	S3
ODASnBr <sub>4</sub>	6.36	12.72	25.44	This work

Table S2. Comparison of microstructure and spectroscopic properties of ODASnBr<sub>4</sub>.

Chemical formula	Experiment method	Microstructure	Spectroscopic properties			Ref.
			PL peak (nm)	PLQY (%)	$\tau$ ( $\mu$ s)	
ODASnBr <sub>4</sub> [2% CFM]	ligand assisted re-precipitation method followed by CFM/DCM treatment	spherical microcrystals	570-598	88 $\pm$ 4	>2.4	S1
ODASnBr <sub>4</sub> [2% DCM]			592-608	83 $\pm$ 4	>2.9	
ODASnBr <sub>4</sub> [EtOH]	argon gas assisted saturation recrystallization method followed by grinding with different alcohol dopants	irregular layered structure (20-50 nm)	616	86	3.61	S2
ODASnBr <sub>4</sub> [PrOH]		regular layered structure (>100 nm)	611	83	3.37	
ODASnBr <sub>4</sub> [BuOH]		crumby structure	613	87	3.24	
ODASnBr <sub>4</sub> [PeOH]		crumby structure	615	87	3.32	
ODASnBr <sub>4</sub> [alcohol]	wet milling method	2D sheet	572-601	88 $\pm$ 1	>2.9	S3
ODASnBr <sub>4</sub>	saturation recrystallization method in ambient environment	flake microcrystals (5-20 $\mu$ m)	586	98.22	3.166	This work

Table S3. The performance parameter of the as-fabricated WLEDs.

LED	drive current (mA)	luminous flux (lm)	luminance efficiency (lm/W)	CRI	color coordinate
Blue: Yellow = 1:2	50	0.9612	6.408	89.4	(0.3421, 0.3384)
Blue: Yellow = 1:2	60	1.1817	6.565	89.9	(0.3429, 0.3466)
Blue: Yellow = 1:2	70	1.4036	6.684	90.3	(0.3439, 0.3526)
Blue: Yellow = 1:2	80	1.6341	6.808	90.3	(0.3452, 0.3547)
Blue: Yellow = 1:2	90	1.8650	6.908	90.1	(0.3466, 0.3614)
Blue: Yellow = 1:4	50	0.9744	6.496	86.1	(0.3227, 0.3209)
Blue: Yellow = 1:4	60	1.1872	6.596	86.5	(0.3212, 0.3228)
Blue: Yellow = 1:4	70	1.4084	6.707	86.9	(0.3198, 0.3250)
Blue: Yellow = 1:4	80	1.6457	6.857	87.4	(0.3162, 0.3291)
Blue: Yellow = 1:4	90	1.8722	6.934	87.1	(0.3134, 0.3352)

**References:**

- S1. S. Wang, J. Popović, S. Burazer, A. Portniagin, F. Liu, K. Low, Z. Duan, Y. Li, Y. Xiong, Y. Zhu, S. V Kershaw, A. B. Djurišić and A. L. Rogach, *Adv. Funct. Mater.* 2021, 31, 2102182.
- S2. S. Wang, S. V Kershaw and A. L. Rogach, *Chem. Mater.*, 2021, 33, 5413-5421.
- S3. J. Qi, S. Wang, A. Portniagin, S. V Kershaw and A. L. Rogach, *Nanomaterials*, 2021, 11, 2738-2748.