

***In situ* Synthesis of Chiral AuNCs with Aggregation-Induced Emission Using  
Glutathione and Ceria Precursor Nanosheets for Glutathione Biosensing**

Mohamed Ibrahim Halawa <sup>a,b,c,d,e</sup>, Guoxing Wu <sup>a</sup>, Alaa Eldin Salem<sup>e</sup>, Lei Su <sup>b,\*</sup>, Bing Shi Li

<sup>a,\*</sup>, Xueji Zhang <sup>b,c,\*</sup>

<sup>a</sup> College of Chemistry and Environmental Engineering, Shenzhen University, Shenzhen 518060, China

<sup>b</sup> Guangdong Laboratory of Artificial Intelligence & Digital Economy (SZ), Shenzhen University, Shenzhen 518060, Peoples R China

<sup>c</sup> College of Biomedical Engineering, International Health Science Innovation Center, Shenzhen Key Laboratory for Nano-Biosensing Technology, Health Science center, Shenzhen University, Shenzhen 518060, China

<sup>d</sup> Department of Pharmaceutical Analytical Chemistry, Faculty of Pharmacy, Mansoura University, Mansoura, 35516, Mansoura, Egypt. Email: m\_halawa88@hotmail.com

<sup>e</sup> Department of chemistry, College of Science, United Arab Emirates University, Al Ain, United Arab Emirates

**\* Corresponding Author. E-mails:** [sulei@szu.edu.cn](mailto:sulei@szu.edu.cn), [phbingsl@szu.edu.cn](mailto:phbingsl@szu.edu.cn);

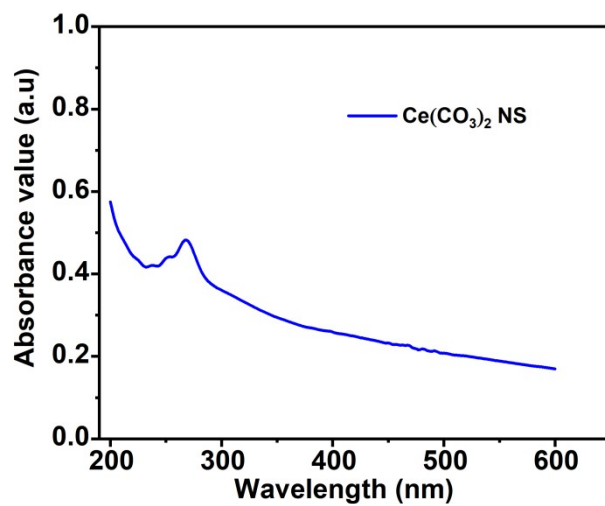
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## **Chemicals and Materials.**

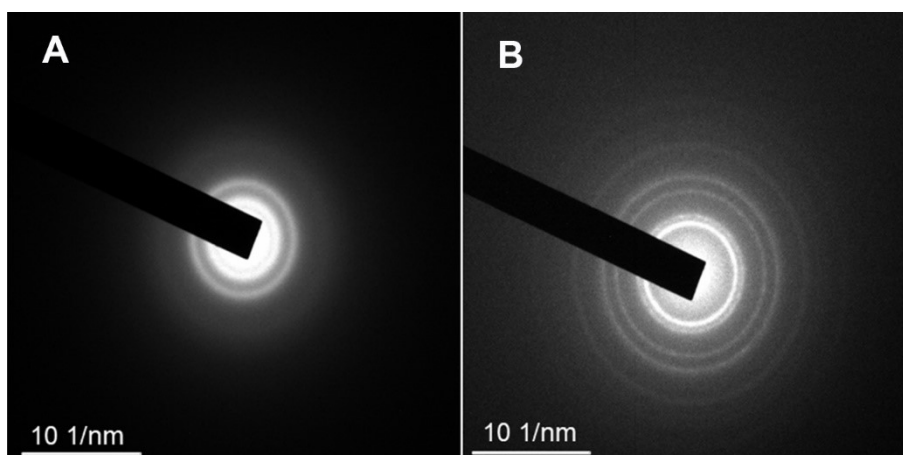
Cerium(III) nitrate hexahydrate ( $\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ), ammonium bicarbonate and sodium hydroxide were received from Alfa Aesar Co., Ltd., Aladdin Industrial Corporation and Xilong Scientific Co., Ltd; respectively. Gold(III) chloride trihydrate ( $\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$ ) and glutathione(GSH) were received from Energy Chemical Co., Ltd. 20.0 mg of the as-synthesized cerium carbonate  $\text{Ce}(\text{CO}_3)_2$  NS were dispersed in 2.0 mL distilled water by ultrasonication for preparing a colloidal stock solution of (10 mg/mL).

## **Instruments.**

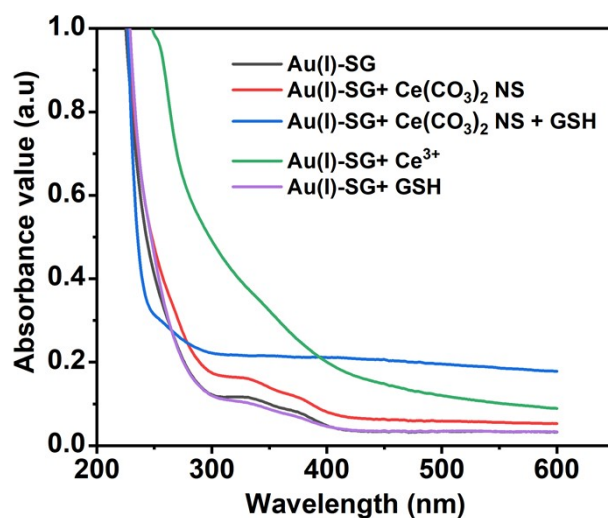
Photoluminescence (PL) and Circular Dichroism (CD) spectra of the NCs samples were recorded by a HITACHI F7000 and Bio-logic MOS-450 photospectrometer; respectively. High Resolution Transmission Electron Microscopic (HRTEM) images of AuNCs and  $\text{Ce}(\text{CO}_3)_2$  NS were obtained using on a JEOL JEM-2100 microscope biased at 200 kV. Atomic Force Microscope (AFM) and Scanning Electron Microscope (SEM) images of  $\text{Ce}(\text{CO}_3)_2$  NS were taken on a BRUKER Multi Mode 8 and JEOL JSM-7800F, respectively. X-ray photoelectron spectroscopy (XPS) measurements were performed on a Thermo Fisher Scientific K-Alpha+ spectrometer. X-ray powder diffraction (XRD) patterns were recorded on a PANalytical B.V. Empyrean X-ray diffractometer with Cu K $\alpha$  radiation ( $\lambda = 0.154056$  nm).



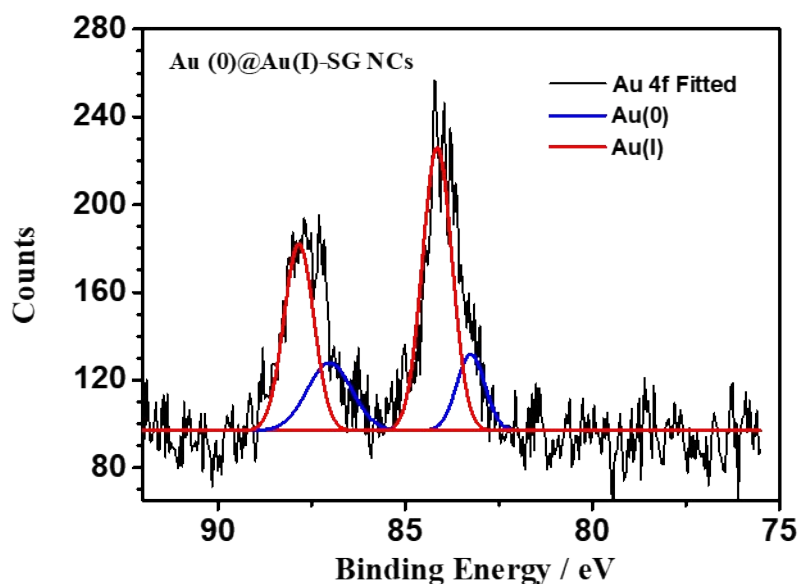
**Figure S1.** Absorption spectrum of  $\text{Ce}(\text{CO}_3)_2$  NS solution.



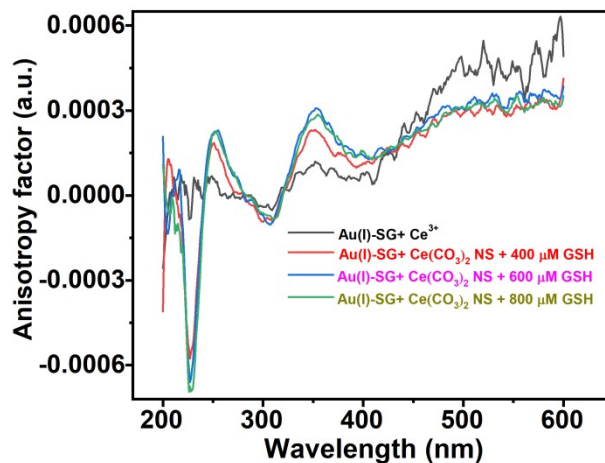
**Figure S2.** SAED images of insitu synthesized  $\text{Au}_0@Au(I)$ -SG NCs (A) and  $\text{Ce}(\text{CO}_3)_2$  NS (B)



**Figure S3.** Absorption spectra for solution of Au(I)-SG oligomers in the absence (black line) and presence of  $\text{Ce}^{3+}$  ions (green line), or GSH (light magenta line) and absorption spectra for solutions of the conjugate probe of Au(I)-SG/ $\text{Ce}(\text{CO}_3)_2$  NS in the absence (red line) and presence of GSH (purple line).



**Figure S4.** The Au ( $4f_{7/2}$ ) spectrum of the as-synthesized luminescent Au(0)@Au(I)-SG nanoclusters



**Figure S5.** Anisotropy factor spectra for solution of Au(I)-SG oligomers in presence of 50  $\mu\text{M}$   $\text{Ce}^{3+}$  ions and solutions of the conjugate probe of Au(I)-SG/ $\text{Ce}(\text{CO}_3)_2$  NS in presence of GSH (400, 600, 800  $\mu\text{M}$ ).

**Table S1.** XPS data of the synthesized  $\text{Ce}(\text{CO}_3)_2$  NS

Name	Peak BE	Height		Area (P)		Atomic %
		CPS	FWHM eV	CPS.eV	Area (N)	
C1s-Ce4s	284.8	14082.06	3.04	85384.39	1197.29	40.95
O1s	531.07	73645.64	3.19	246206.39	1427.36	48.82
Ce3d	884.41	66153.56	6.74	782059.17	298.78	10.22

Table S2: Repeatability and reproducibility results for GSH sensing using Au(I)-SG/ $\text{Ce}(\text{CO}_3)_2$  NS probe

GSH amount ( $\mu\text{M}$ )	% Recovery <sup>a</sup> $\pm$ % RSD	
	Intra-day precision	Inter-day precision
200	99.13 $\pm$ 1.90	98.84 $\pm$ 2.31
500	99.98 $\pm$ 1.75	101.35 $\pm$ 2.84
1000	100.50 $\pm$ 1.82	100.90 $\pm$ 2.62

<sup>a</sup> refers to average value of three assays.

**Table S3.** Comparison of AIE-based probe for GSH sensing with some previous methods.

Technique	Applied Materials*	Linear range ( $\mu\text{M}$ )	LOD ( $\mu\text{M}$ )	Ref.
Colorimetry	AgNPs	0-400	4.11	1
Colorimetry	Coumarin derivatives	0 -180	6.84	2
Colorimetry	$\text{Cu}^{2+}$ /Imidazole derivatives	7.5-37.5	2.98	3
Colorimetry	Cytidine-AuNCs	0-400	10	4
Colorimetry	NDP	$0-80 \times 10^3$	178	5
Chronoamperometry	AuNPs-PEDOT/ GCE	0.5-10	0.1	6
CV	AuNPs/ $\text{Al}_2\text{O}_3$ . $\text{TiO}_2$ NPs /GCE	5-50, 100-750		7
DPV	PDI-SH/ CPE	$(0.5-5) \times 10^3$	17	8
CV	Graphene modified-SPCE	1-100	8.01	9
HPLC-ED	Reversed C18- HPLC	5.1–325.4	2.3	10
Fluorimetry	Au(I)-SG/ $\text{Ce}(\text{CO}_3)_2$ NS	0-1000	1.02	Our work

\* AgNPs, NDP, AuNPs, PDI-SH, GCE, PEDOT, CPE and SPCE represent silver nanoparticles, naphthalene derivate containing piarselenole, gold nanoparticles, thiolated perylene diimides, glassy carbon electrode, poly(3,4)ethylene dioxythiophene, carbon paste electrode and screen printed carbon electrode; respectively.

**Table S4.** Comparison of AIE-based probe for GSH sensing with other previous luminescent approaches.

Technique	Applied Materials*	Linear range ( $\mu\text{M}$ )	LOD ( $\mu\text{M}$ )	Ref.
ECL	GO/CdTe QDs	24-214	8.3	11
ECL	CdSe/ZnS QDs	10-180	1.5	12
CL	Peroxidase/luminol- $\text{H}_2\text{O}_2$	0.75-30	0.75	13
Fluorimetry	N-GQDs/MoS <sub>2</sub>	400-4000	2.47	14
Fluorimetry	MnO <sub>2</sub> /UCNPs	N/A	0.9	15
Fluorimetry	TAT-probe	0-12	5.15	16
Fluorimetry	DTFN	0-500	1.03	17
Fluorimetry	AuNCs/MnO <sub>2</sub> NS	0-500	4	18
Fluorimetry	TP-N	0-50	1.53	19
Fluorimetry	N,S-CDs@Cu <sup>2+</sup>	10-150	3.74	20
Fluorimetry	NP-BO-HEM	0-200	1.37	21
Fluorimetry	Au(I)-SG/Ce(CO <sub>3</sub> ) <sub>2</sub> NS	0-1000	1.02	Our work

\* GO, QD, GQDs, UCNPs, TAT-probe, DTFN, TP-N and CDs, refer to, graphene oxide, quantum dots, graphene quantum dots, upconversion nanoparticles, two-photon biothiols probe, dual-targeting fluorescence nanoprobe, phthalazinetrione derivative, 2-(benzo[d]thiazol-2-yl)-4-hydroxyphthalazin-1(2H)-one hydrate-dimer, and carbon dots, respectively.

## References

1. I. Sanskriti and K. K. Upadhyay, *New J. Chem.*, 2017, **41**, 4316-4321.
2. K. Xiong, F. Huo, J. Chao, Y. Zhang and C. Yin, *Anal. Chem.*, 2019, **91**, 1472-1478.
3. M. S. Kim, J. M. Jung, J. H. Kang, H. M. Ahn, P.-G. Kim and C. Kim, *Tetrahedron*, 2017, **73**, 4750-4757.
4. C. Jiang, C. Zhang, J. Song, X. Ji and W. Wang, *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 2021, **250**, 119316.
5. X. Zeng, X. Zhang, B. Zhu, H. Jia, W. Yang, Y. Li and J. Xue, *Sensors Actuators B: Chem.*, 2011, **159**, 142-147.
6. R. Rajaram, P. Kanagavalli, S. Senthilkumar and J. Mathiyarasu, *Biotechnology and Bioprocess Engineering*, 2020, **25**, 715-723.
7. M. Cubukcu, F. N. Ertas and U. Anik, *Current Analytical Chemistry*, 2012, **8**, 351-357.
8. B. Perk, Y. T. Buyuksunetci, O. Hakli, C. Xue, Q. Li and U. Anik, *Chemistryselect*, 2021, **6**, 11648-11652.
9. W. T. Wahyuni, E. Rohaeti and D. R. Sari, *IOP Conference Series: Earth and Environmental Science*,

- 2018, **187**, 012078.
10. Z. Buchtova, Z. Lackova, J. Kudr, Z. Zitka, J. Skoda and O. Zitka, *Molecules*, 2018, **23**, 2504.
  11. Y. Wang, J. Lu, L. Tang, H. Chang and J. Li, *Analytical chemistry* 2009, **81**, 9710-9715.
  12. L. Dennany, M. Gerlach, S. O'Carroll, T. E. Keyes, R. J. Forster and P. Bertoncello, *Journal of Materials Chemistry* 2011, **21**, 13984-13990.
  13. T. Kamidate and H. Watanabe, *Talanta*, 1996, **43**, 1733-1738.
  14. S. Tang, C. Yu, L. Qian, C. Zhou, Z. Zhen, B. Liu, X. Cheng and R. Cheng, *Microchem. J.*, 2021, **171**.
  15. R. Deng, X. Xie, M. Vendrell, Y.-T. Chang and X. Liu, *Journal of the American Chemical Society*, 2011, **133**, 20168-20171.
  16. P. Su, Z. Zhu, Y. Tian, L. Liang, W. Wu, J. Cao, B. Cheng, W. Liu and Y. Tang, *Talanta*, 2020, **218**.
  17. H. Wang, P. Zhang, C. Zhang, S. Chen, R. Zeng, J. Cui and J. Chen, *Materials Advances*, 2020, **1**, 1739-1744.
  18. S. Lin, H. Cheng, Q. Ouyang and H. Wei, *Analytical Methods*, 2016, **8**, 3935-3940.
  19. N.-N. Li, N.-N. Shi, D. Yang, R.-X. Wu, C.-G. Xu, B. Zhu, F. Shao, X. Zhang, S.-Y. Bi and Y.-H. Fan, *J. Mol. Liq.*, 2021, **342**.
  20. X. Sun, C. Wang, P. Li, Z. Shao, J. Xia, Q. Liu, F. Shen and Y. Fang, *Food Chem.*, 2022, **372**.
  21. L. Jia, L.-Y. Niu and Q.-Z. Yang, *Anal. Chem.*, 2020, **92**, 10800-10806.