

## 1. Chromatogram of BaP and calibration curve determined by HPLC (Agilent 1200)

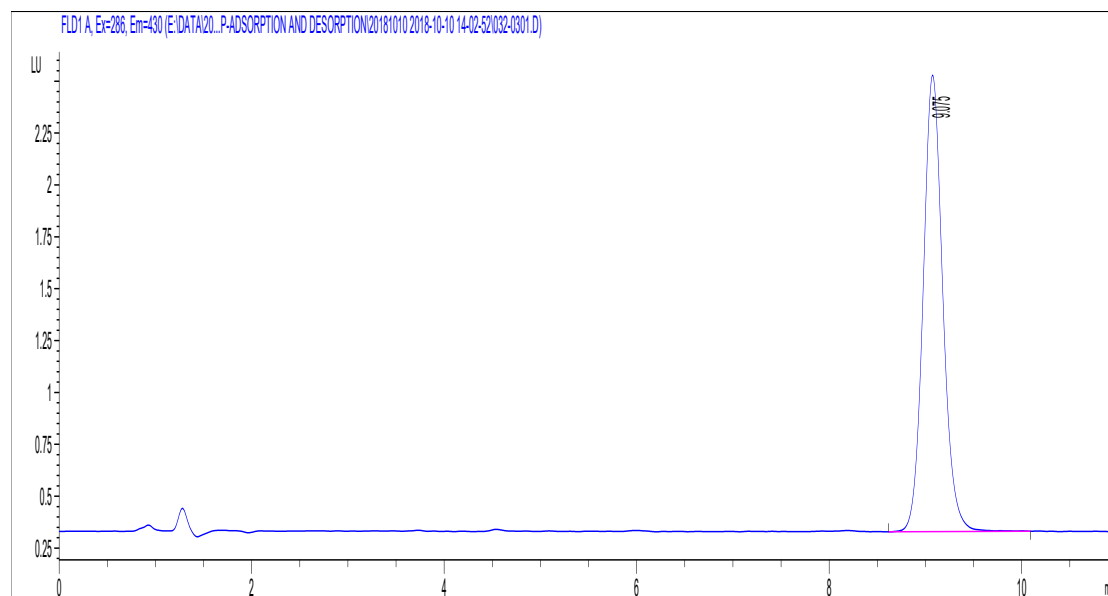


Fig.S1 the chromatogram of BaP

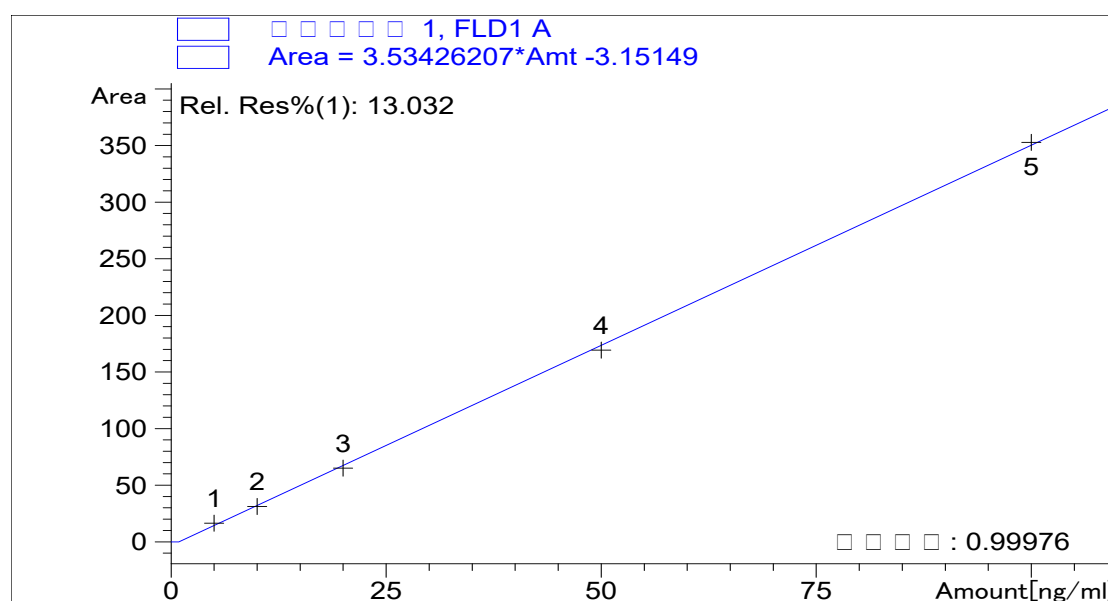


Fig.S2 The calibration curve of BaP

## 2. Adsorption mechanism equation

### 2.1 First order dynamic equation

$$q_t = q_e(1 - e^{-k_1 t}) \quad (\text{Eq. S1})$$

Where,  $q_t$  is amount of adsorption at time  $t$ ,  $q_e$  is adsorption amount at equilibrium,  $t$  is experimental time (in mins) and  $k_1$  is reaction rate constant.

### 2.2 Second order dynamic equation

$$\frac{t}{q_t} = \frac{1}{k_2 q_e^2} + \frac{1}{q_e} t \quad (\text{Eq. S2})$$

Where,  $q_t$ ,  $q_e$  and  $t$  stands for the same thing as Eq.1,  $k_2$  is the second order kinetic rate constant.

2.3 Intra particle diffusion equation

$$q_t = k_t t^{\frac{1}{2}} + c \quad (\text{Eq. S3})$$

Where  $k_t$  is intra particle diffusion rate constant and  $c$  is the constant related to the thickness and boundary of the adsorbent.

2.4 Langmuir equation

$$\frac{1}{q_e} = \frac{1}{q_m} + \frac{1}{k_l q_m c_e} \quad (\text{Eq. S4})$$

Where  $q_e$  is the concentration of adsorbents per unit mass,  $c_e$  is the concentration of equilibrium solution,  $q_m$  represents the maximum adsorption capacity of adsorbent monolayer, and the adsorption coefficient  $k_l$  represents the adsorption surface strength constant.

2.5 Freundlich equation

$$\lg c_s = \lg k_f + \frac{1}{n} \lg c_e \quad (\text{Eq. S5})$$

where  $c_s$  is the mass concentration of adsorption,  $c_e$  is the concentration of equilibrium solution,  $k_f$  is the constant of the model and  $\frac{1}{n}$  reflects the degree of nonlinearity of adsorption.

2.6 Temkin equation

$$q_e = B_T \ln \frac{K}{A} + B_T \ln c_e \quad (\text{Eq. S6})$$

Where  $B_T$  is the adsorption coefficient of heat of adsorbent (J/mole),  $A$  is the reciprocal of the adsorption coefficient,  $K$  is slope,  $T$  is the experimental temperature (298K) and  $R$  is the real value of gas constant (8.314J/mole/K) respectively.

**3 Graph of the fitting equation**

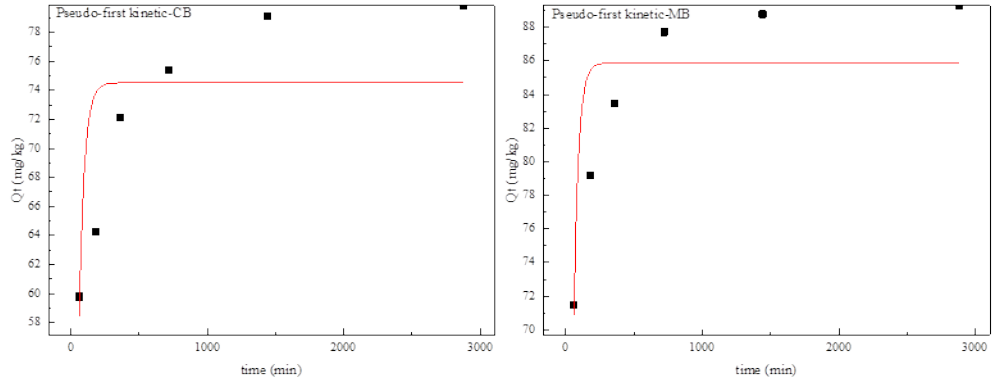


Fig.S3 Pseudo-first kinetic dynamic equation fitting curve

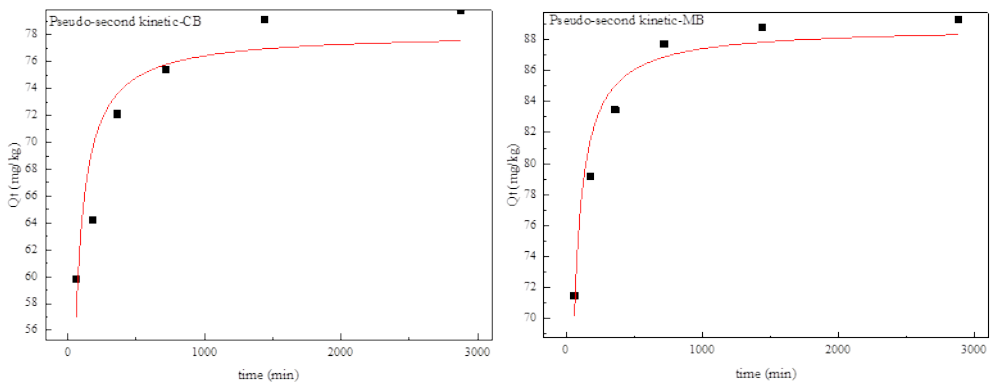


Fig.S4 Pseudo-second kinetic dynamic equation fitting curve

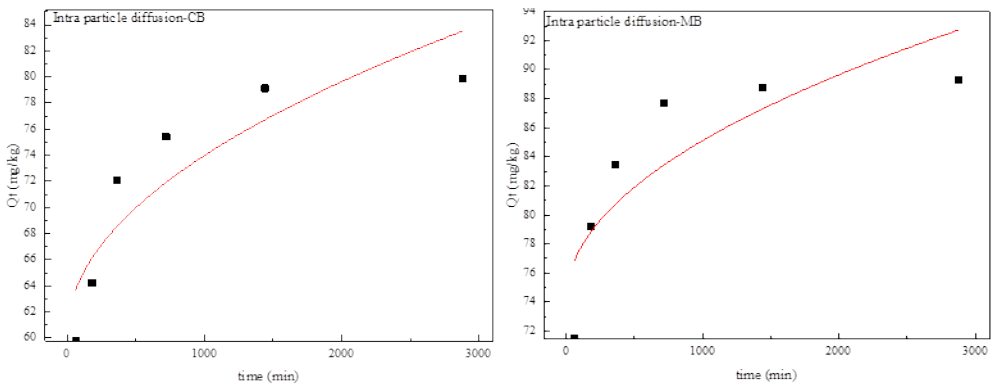


Fig.S5 Intra particle diffusion equation fitting curve

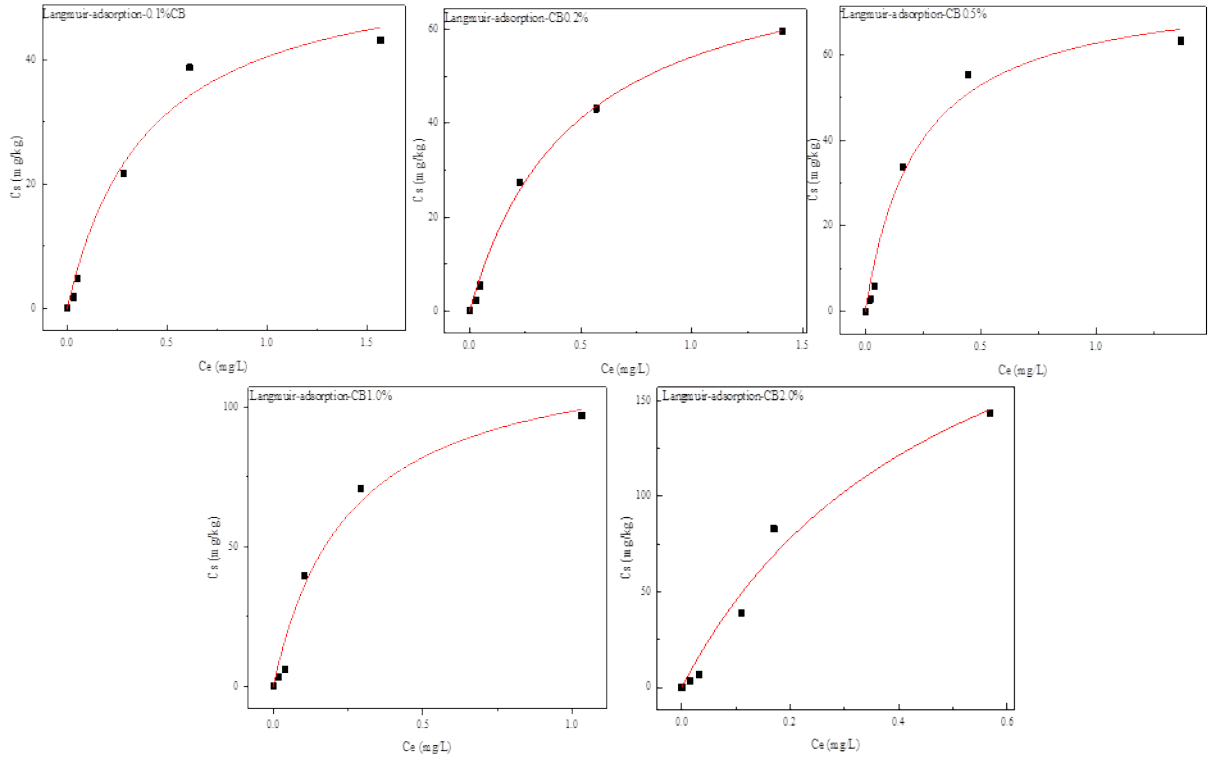


Fig.S6 Langmuir adsorption isotherm fitting curve of CB Biochar soil

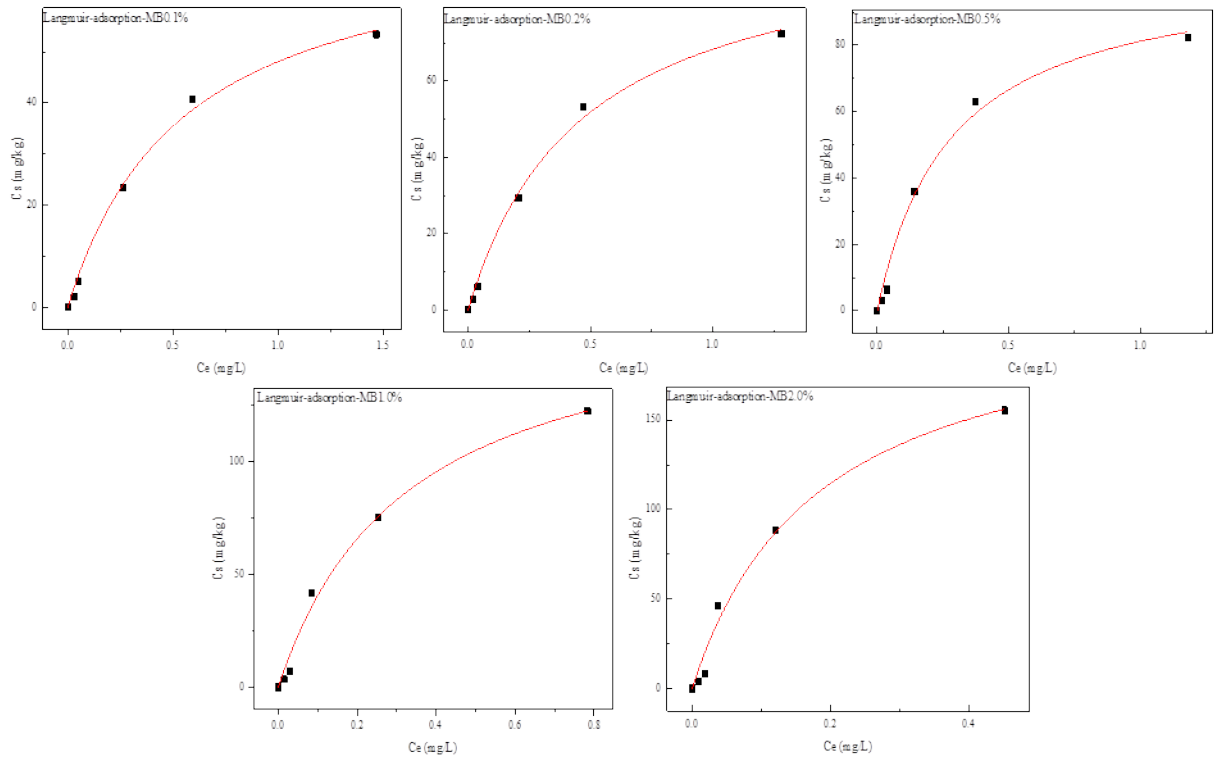


Fig.S7 Langmuir adsorption isotherm fitting curve of MB Biochar soil

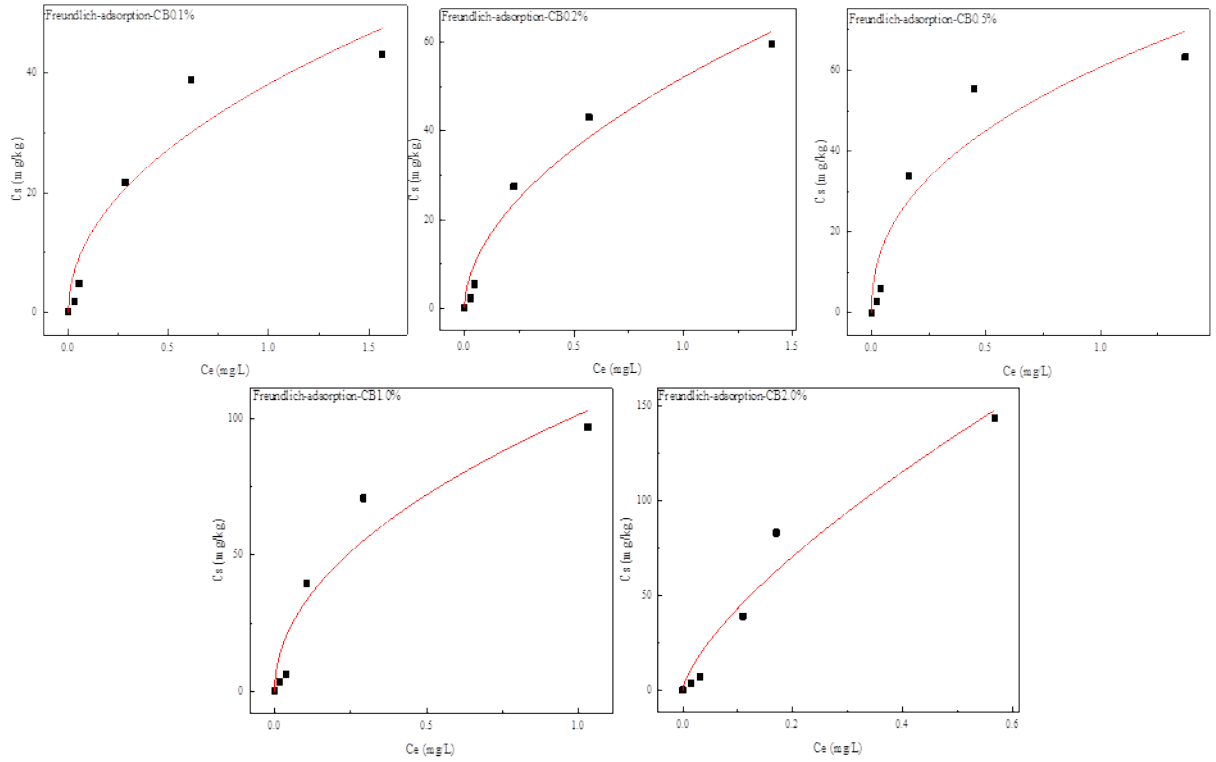


Fig.S8 Freundlich adsorption isotherm fitting curve of CB Biochar soil

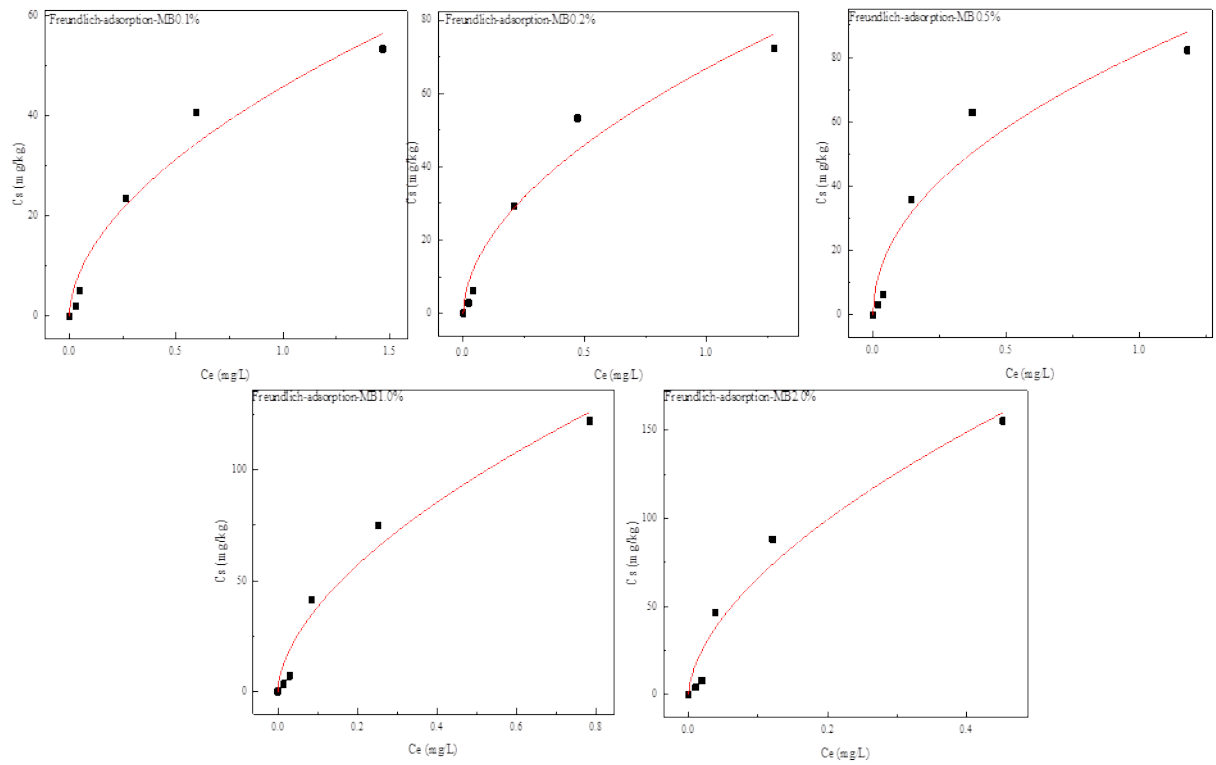


Fig.S9 Freundlich adsorption isotherm fitting curve of MB Biochar soil

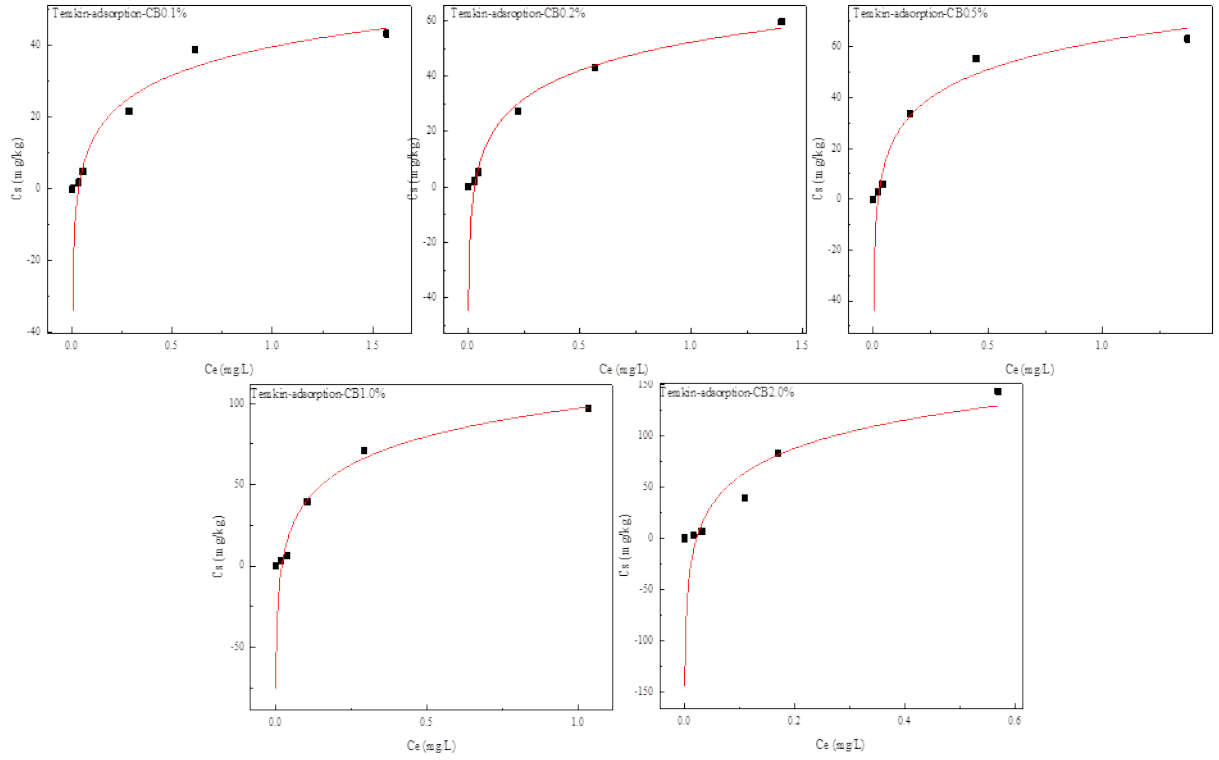


Fig.S10 Temkin adsorption isotherm fitting curve of CB Biochar soil

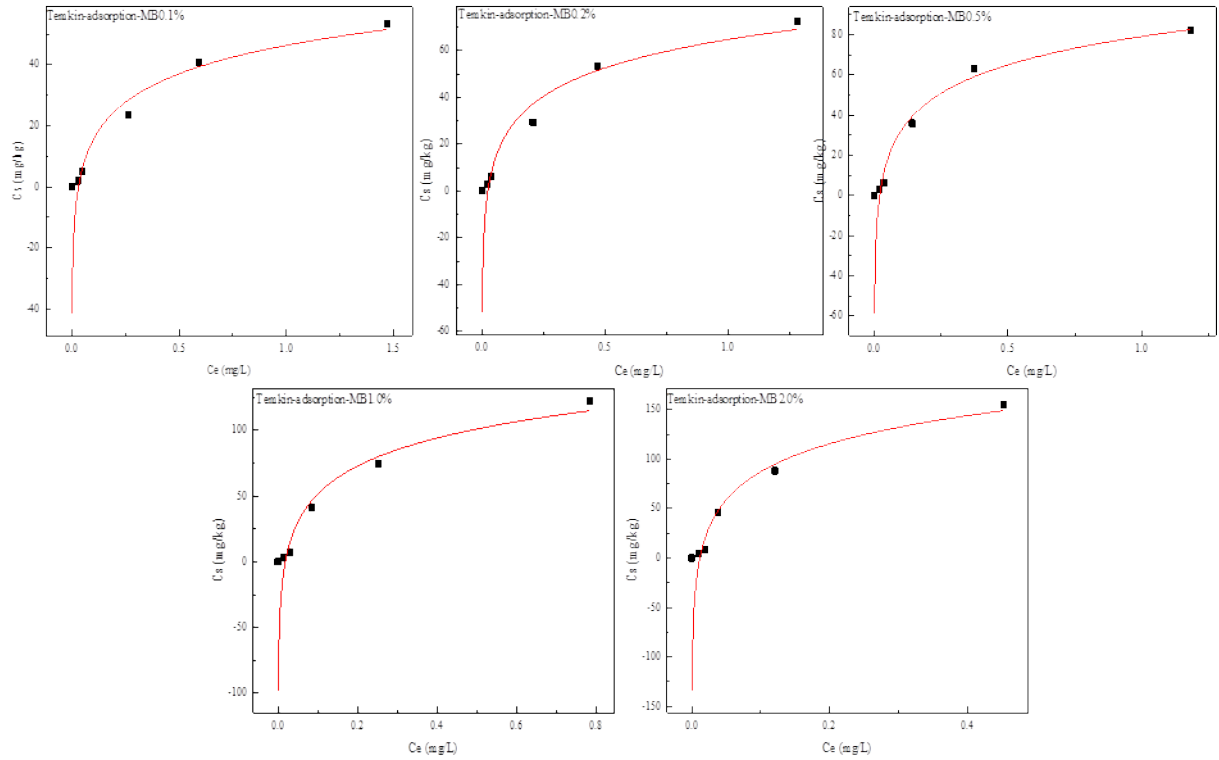


Fig.S11 Temkin adsorption isotherm fitting curve of MB Biochar soil

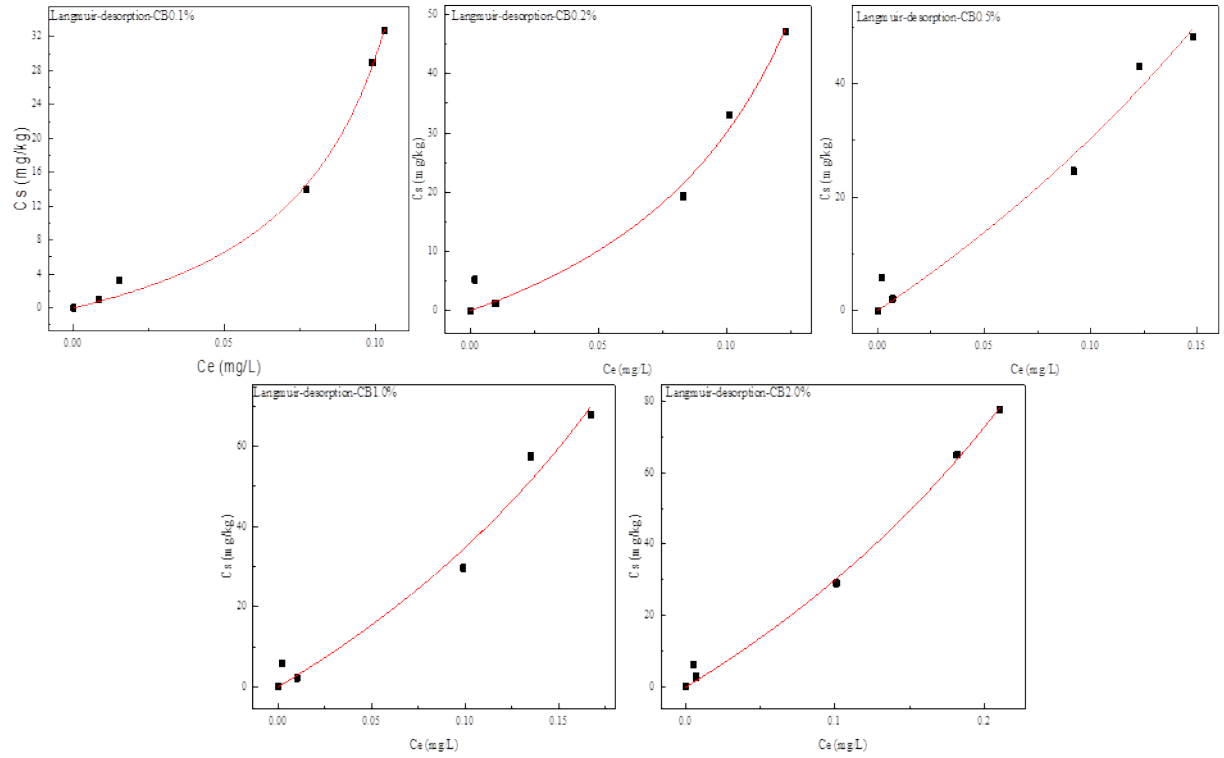


Fig.S12 Langmuir desorption isotherm fitting curve of CB Biochar soil

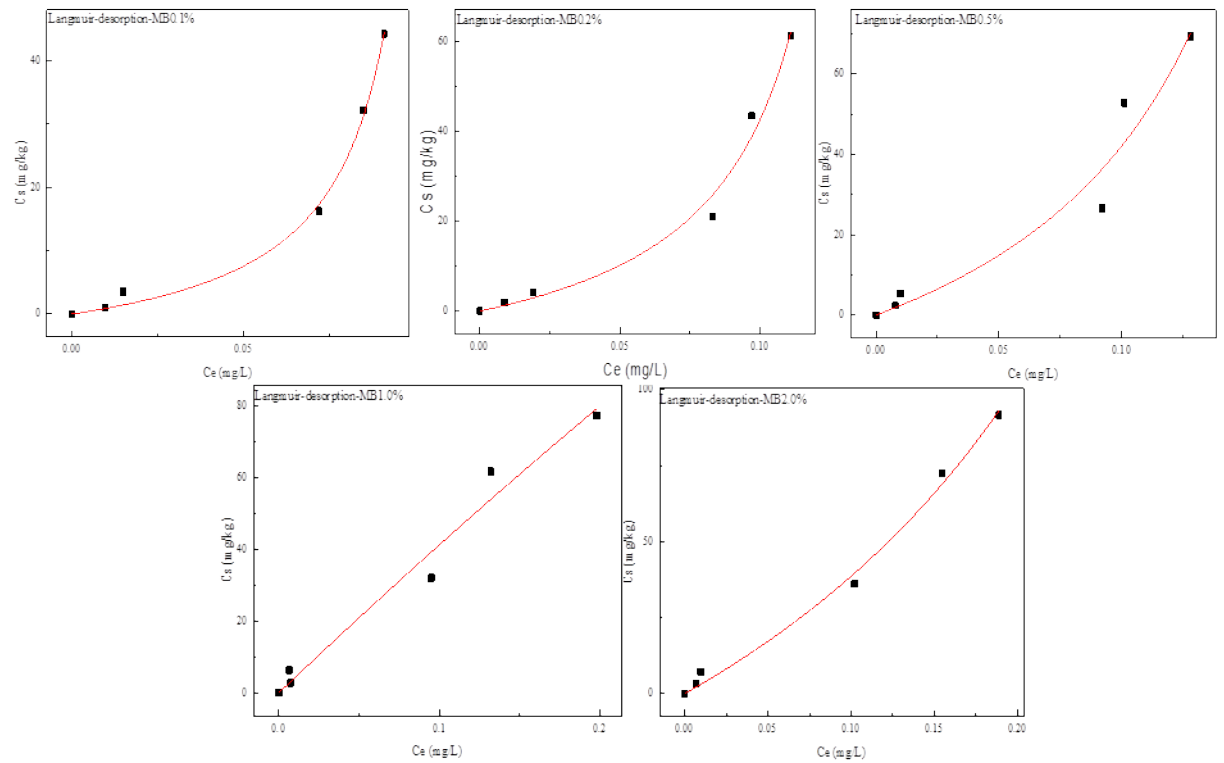


Fig.S13 Langmuir desorption isotherm fitting curve of MB Biochar soil

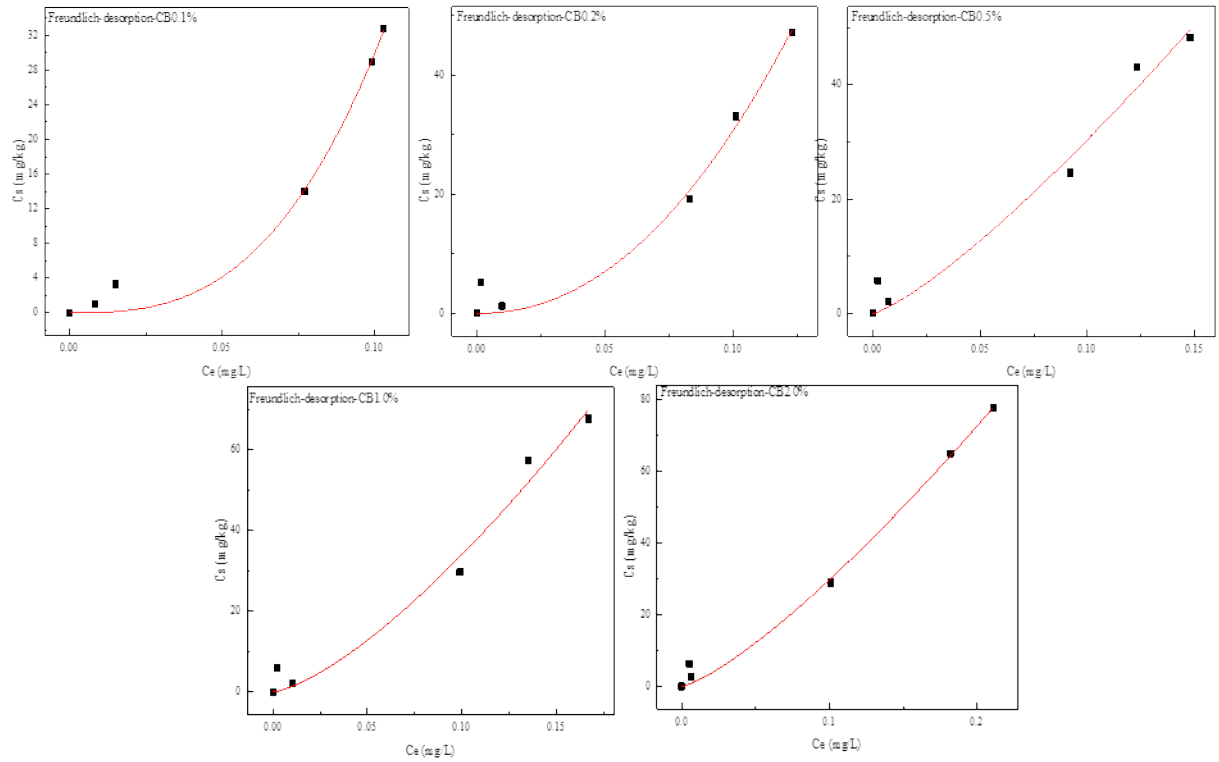


Fig.S14 Freundlich desorption isotherm fitting curve of CB Biochar soil

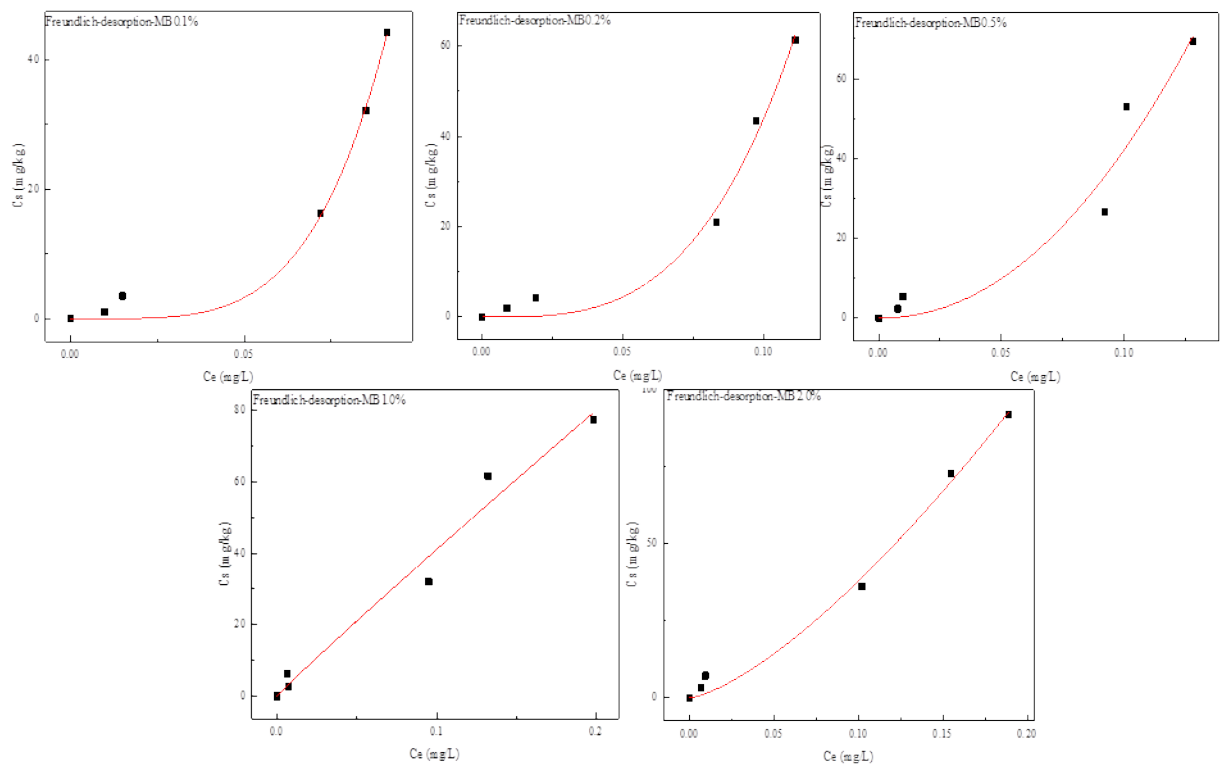


Fig.S15 Freundlich desorption isotherm fitting curve of MB Biochar soil



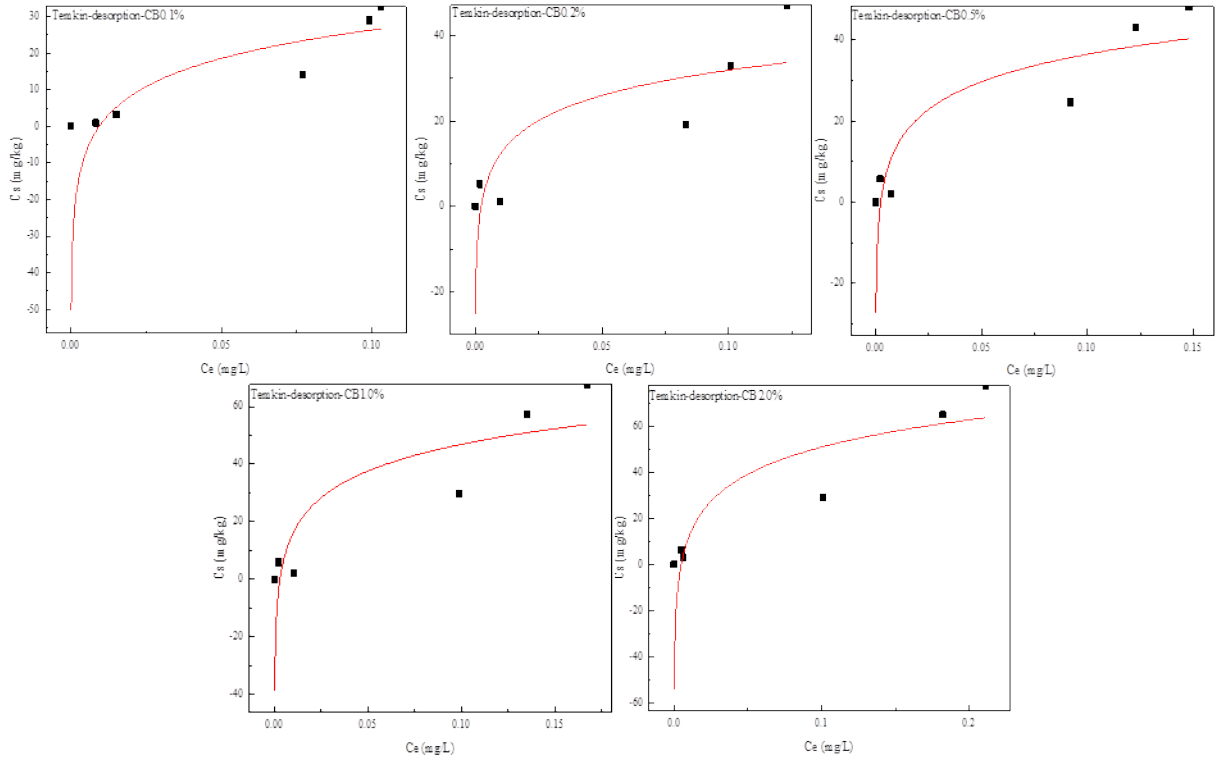


Fig.S16 Temkin desorption isotherm fitting curve of CB Biochar soil

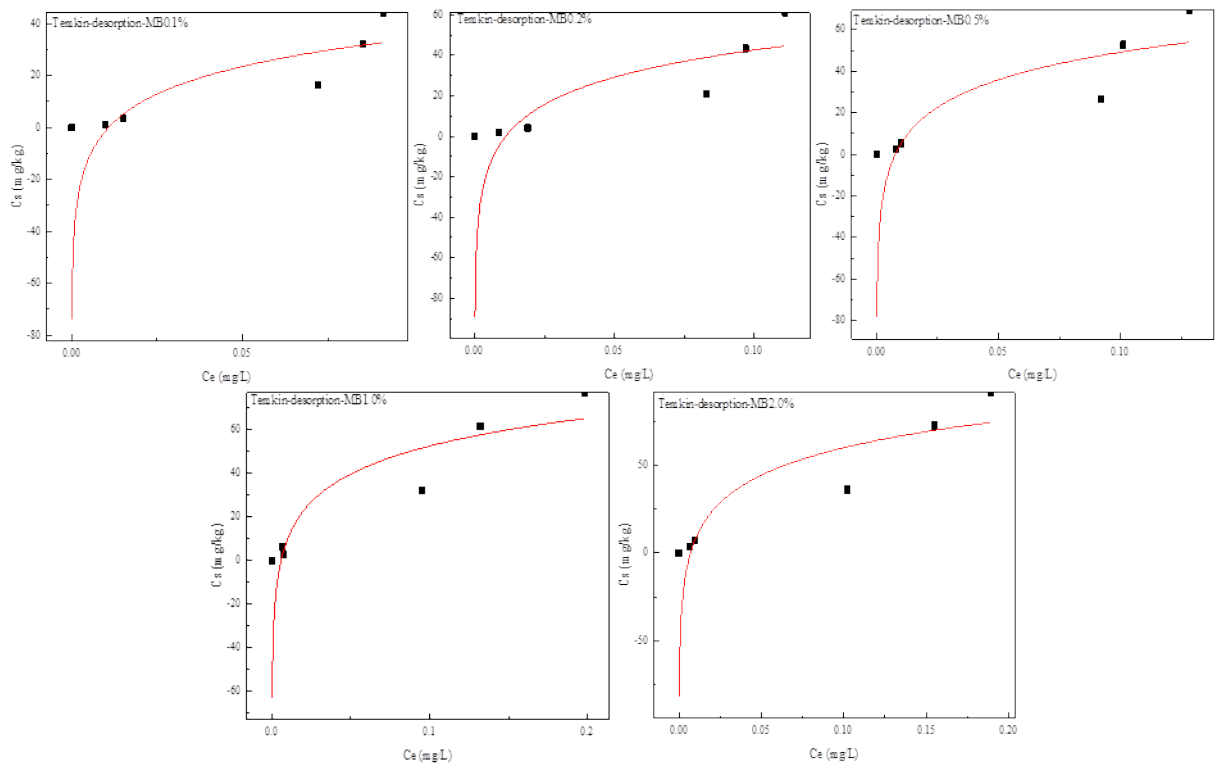


Fig.S17 Temkin desorption isotherm fitting curve of MB Biochar soil

