

Analytical Methods

Electronic Supplementary Information – ESI †

Optimization and validation of liquid-liquid extraction with low-temperature purification (LLE-LTP) for determining fluopyram fungicide in water samples using HPLC-DAD

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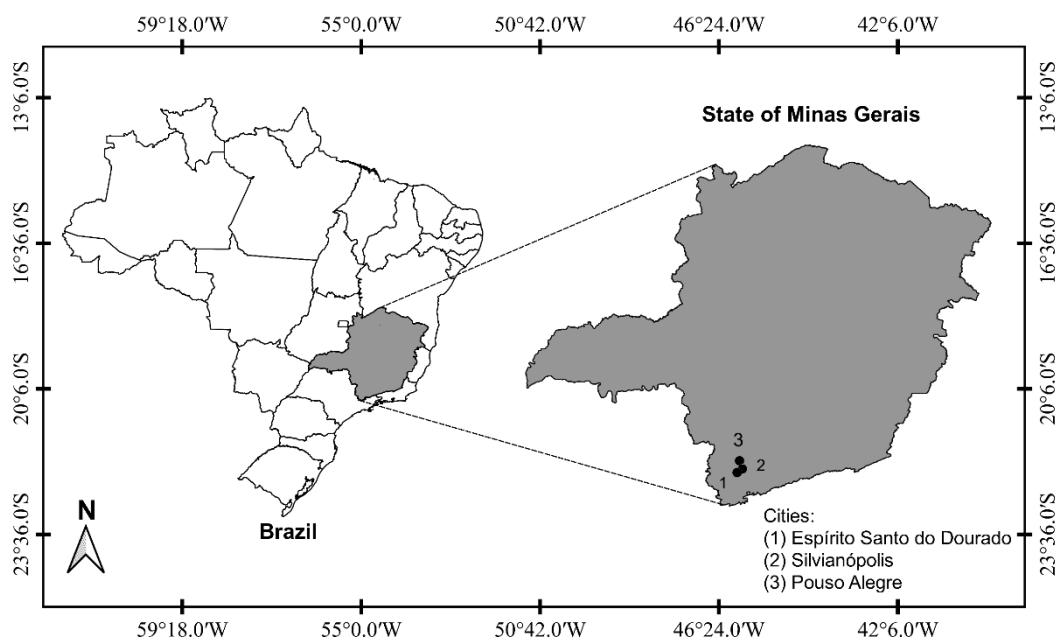


Fig. S1 Map showing the location of the three Brazilian cities where the water samples were collected.

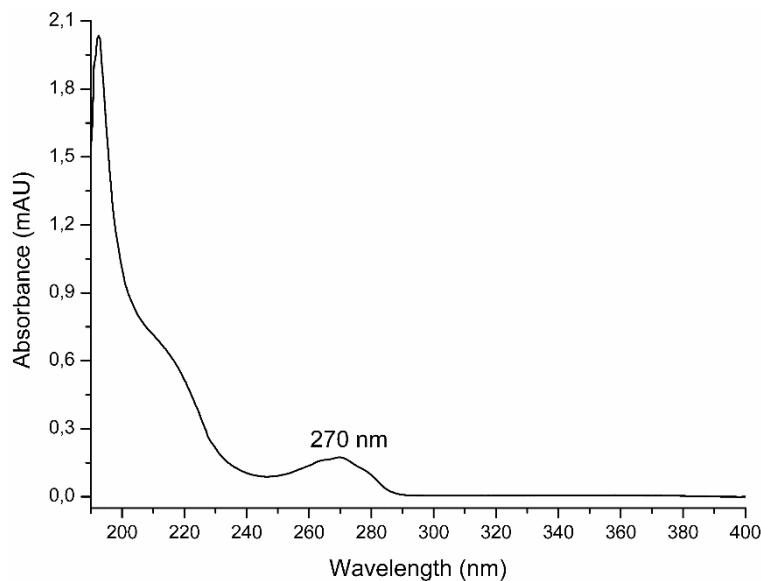


Fig. S2 Absorption spectrum in the ultraviolet and visible region of the fluopyram standard solution at 1.0 mg L^{-1} in acetonitrile.

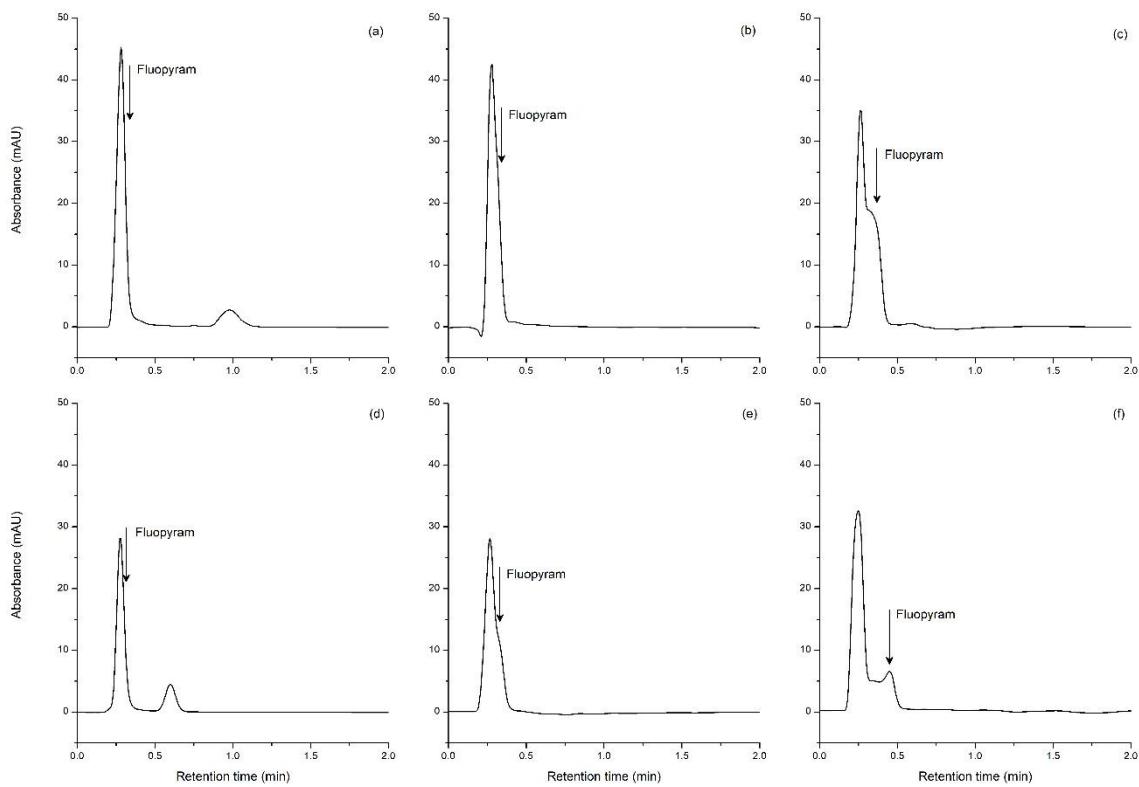


Fig. S3 Chromatograms of the fluopyram standard solution at 1.0 mg L^{-1} in acetonitrile. Zorbax column (2.1 mm x 50 mm, 1.8 μm); column temperature = 30 °C; $\lambda = 270 \text{ nm}$; flow rate = 0.5 mL min⁻¹; mobile phase (a) ACN:H₂O (100:0 v/v), (b) ACN:H₂O (90:10 v/v), (c) ACN:H₂O (80:20 v/v), (d) MeOH:H₂O (100:0 v/v), (e) MeOH:H₂O (90:10 v/v), and (f) MeOH:H₂O (80:20 v/v). The arrows in the figure indicate the retention time of the fluopyram.

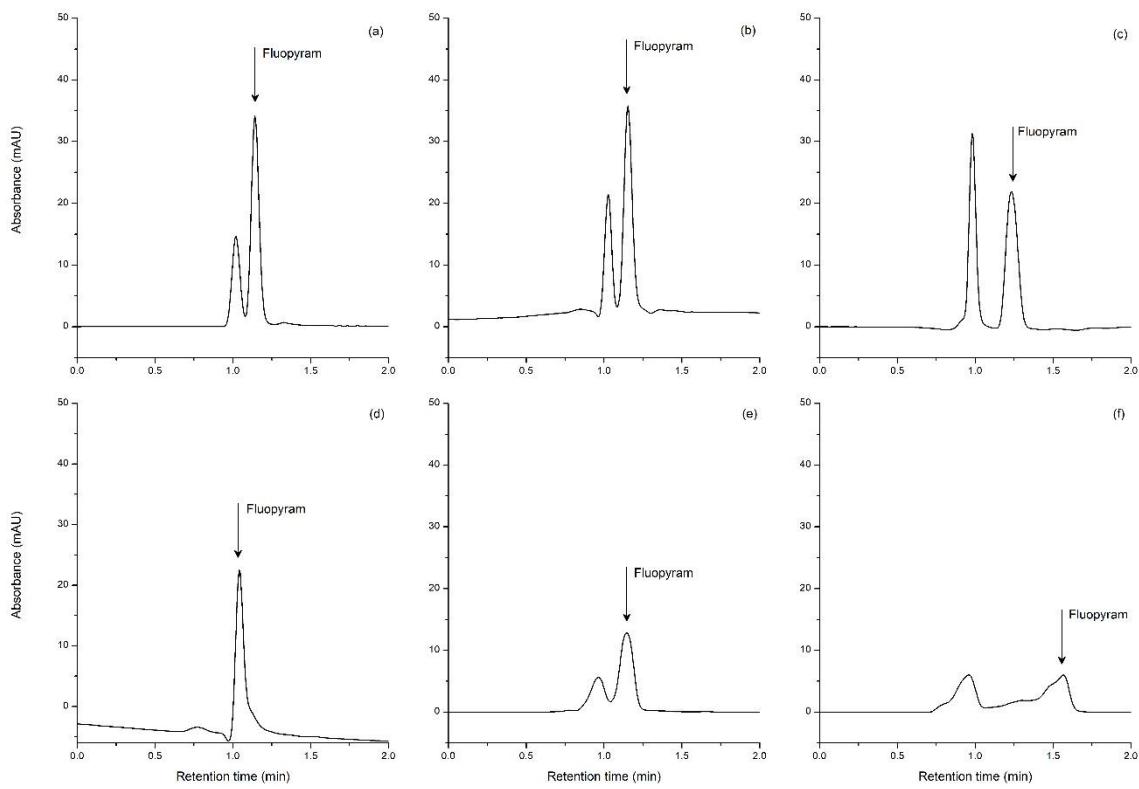


Fig. S4 Chromatograms of the fluopyram standard solution at 1.0 mg L^{-1} in Acetonitrile. Poroshell column (4.6 mm x 50 mm, 2.7 μm , Agilent Technologies); column temperature = 30 °C; $\lambda = 270 \text{ nm}$; flow rate = 0.5 mL min $^{-1}$; mobile phase (a) ACN:H₂O (100:0 v/v), (b) ACN:H₂O (90:10 v/v), (c) ACN:H₂O (80:20 v/v), (d) MeOH:H₂O (100:0 v/v), (e) MeOH:H₂O (90:10 v/v), and (f) MeOH:H₂O (80:20 v/v). The arrows in the figure indicate the retention time of the fluopyram.

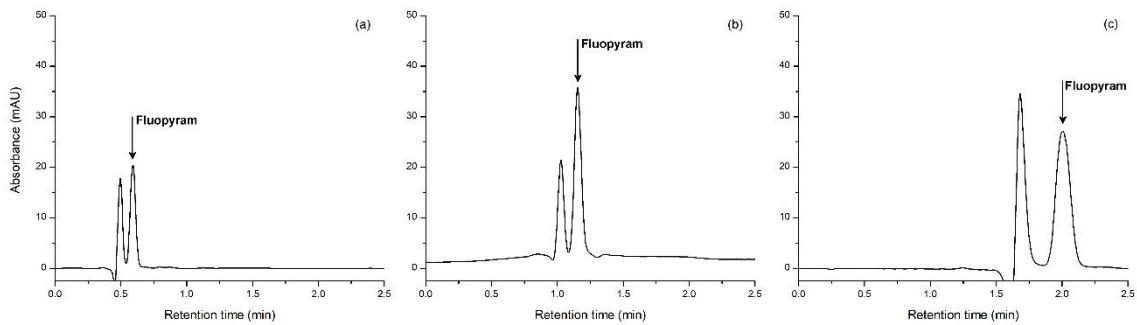


Fig. S5 Chromatograms of the fluopyram standard solution at 1.0 mg L^{-1} in Acetonitrile.

Poroshell column (4.6 mm x 50 mm, 2.7 μm , Agilent Technologies), $\lambda = 270 \text{ nm}$; column temperature = 30°C ; mobile phase ACN:H₂O (90:10 v/v); flow rate (a) 1.0 mL min^{-1} , (b) 0.5 mL min^{-1} , and (c) 0.3 mL min^{-1} .

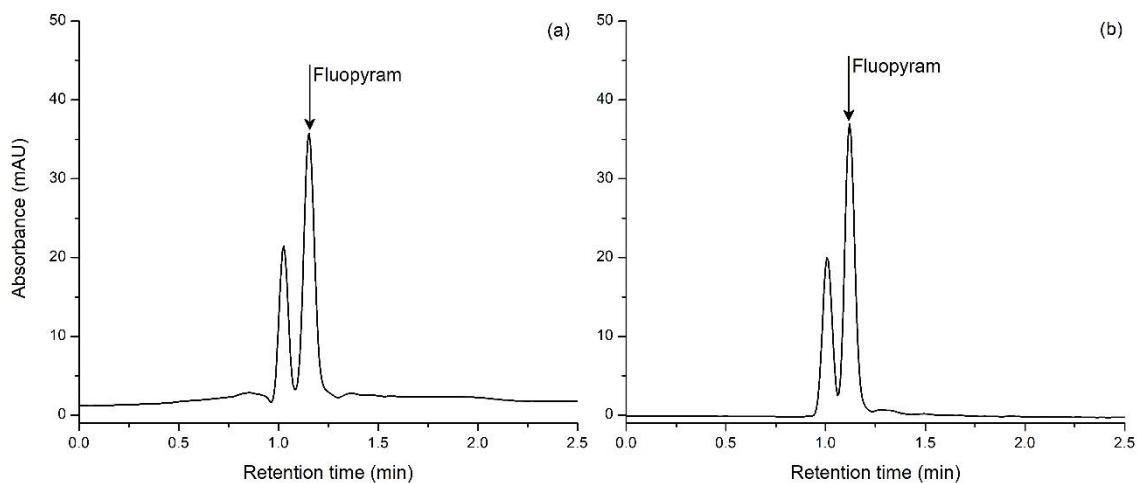


Fig. S6 Chromatograms of the fluopyram standard solution at 1.0 mg L^{-1} in Acetonitrile.

Poroshell column (4.6 mm x 50 mm, 2.7 μm , Agilent Technologies), $\lambda = 270 \text{ nm}$; mobile phase ACN:H₂O (90:10 v/v); flow rate = 0.5 mL min^{-1} ; column temperature (a) 30°C , and (b) 40°C .

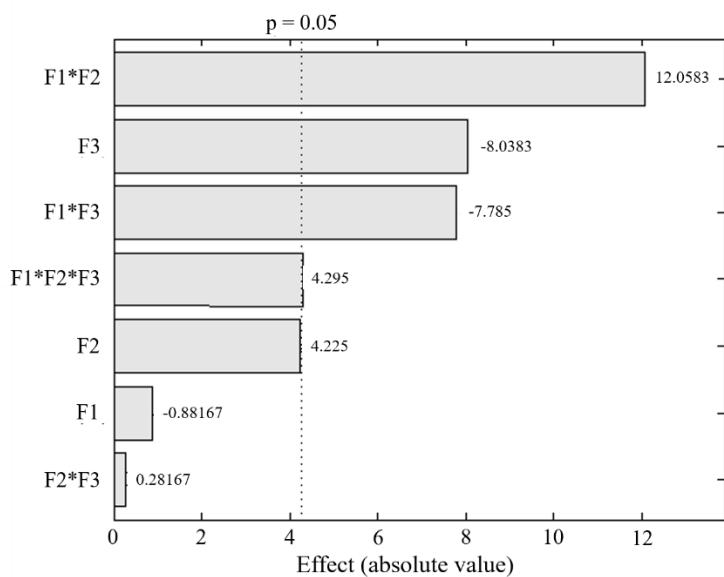


Fig. S7 Pareto chart of the effects and their interaction in the LLE-LTP of fluopyram at a probability level of 95% ($p \leq 0.05$).

F1, factor 1 (extraction phase: proportion of ACN:EtOAc); F2, factor 2 (ionic strength); F3, factor 3 (sample freezing time).

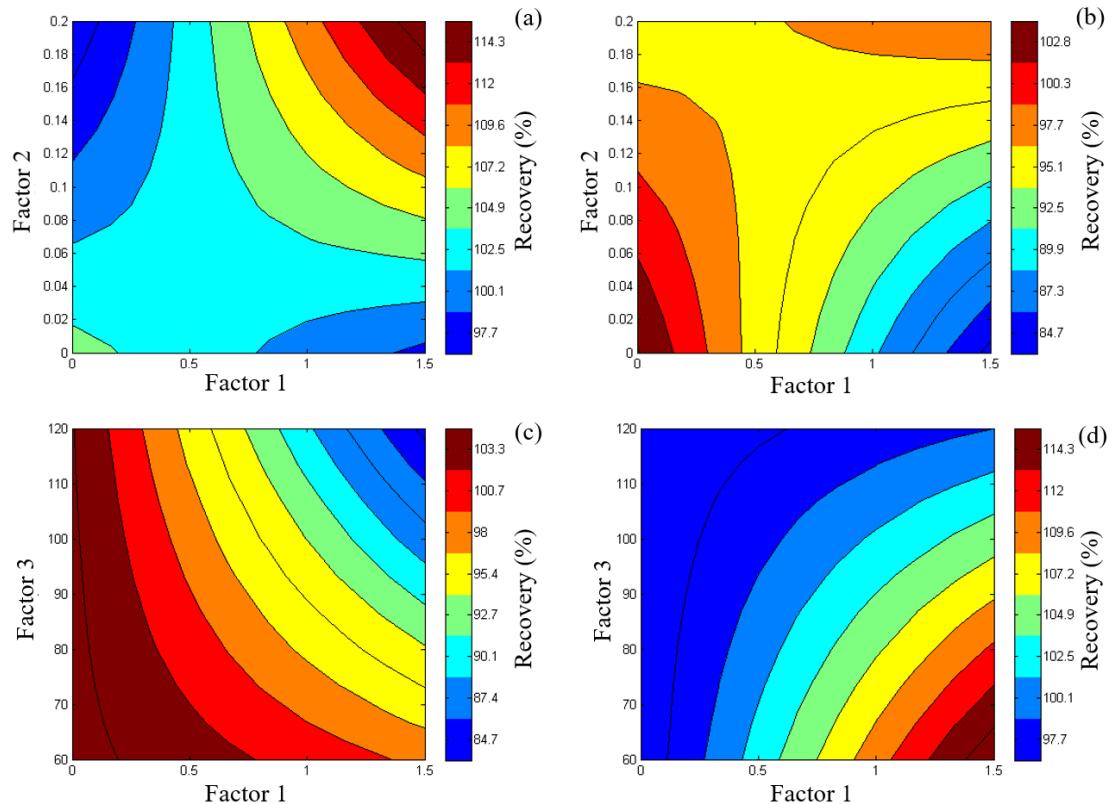


Fig. S8 Response surface plots for interactive effects between the extraction phase (factor 1) and ionic strength (factor 2) at the time of (a) 60 min and (b) 120 min; and the interactive effects between the extraction phase (factor 1) and the sample freezing time (factor 3) with the ionic strength at (c) level (+) and (d) level (-).

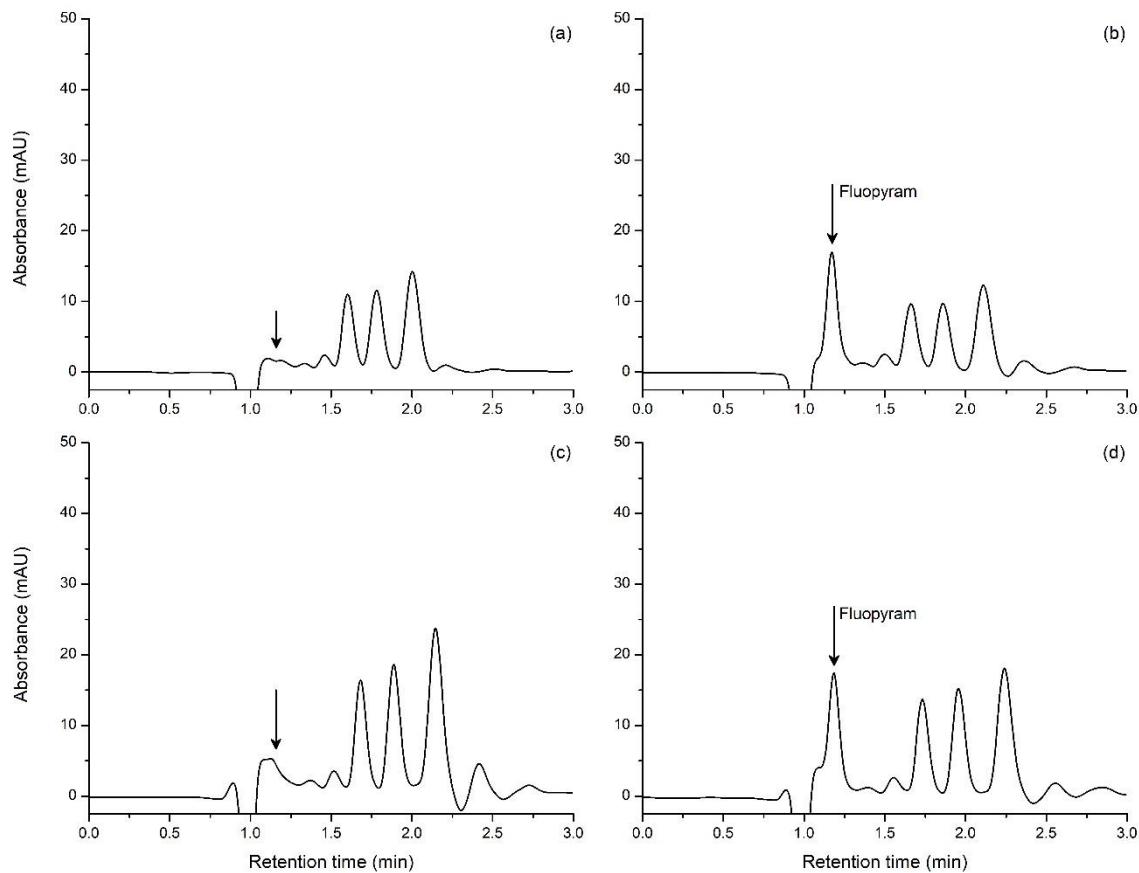


Fig. S9 Chromatograms of (a) the blank extract and (b) the fortified matrix extract obtained using the extraction phase composed by acetonitrile; and chromatograms of (c) the blank extract and (d) the fortified matrix extract obtained using the extraction phase composed by EtOAc.

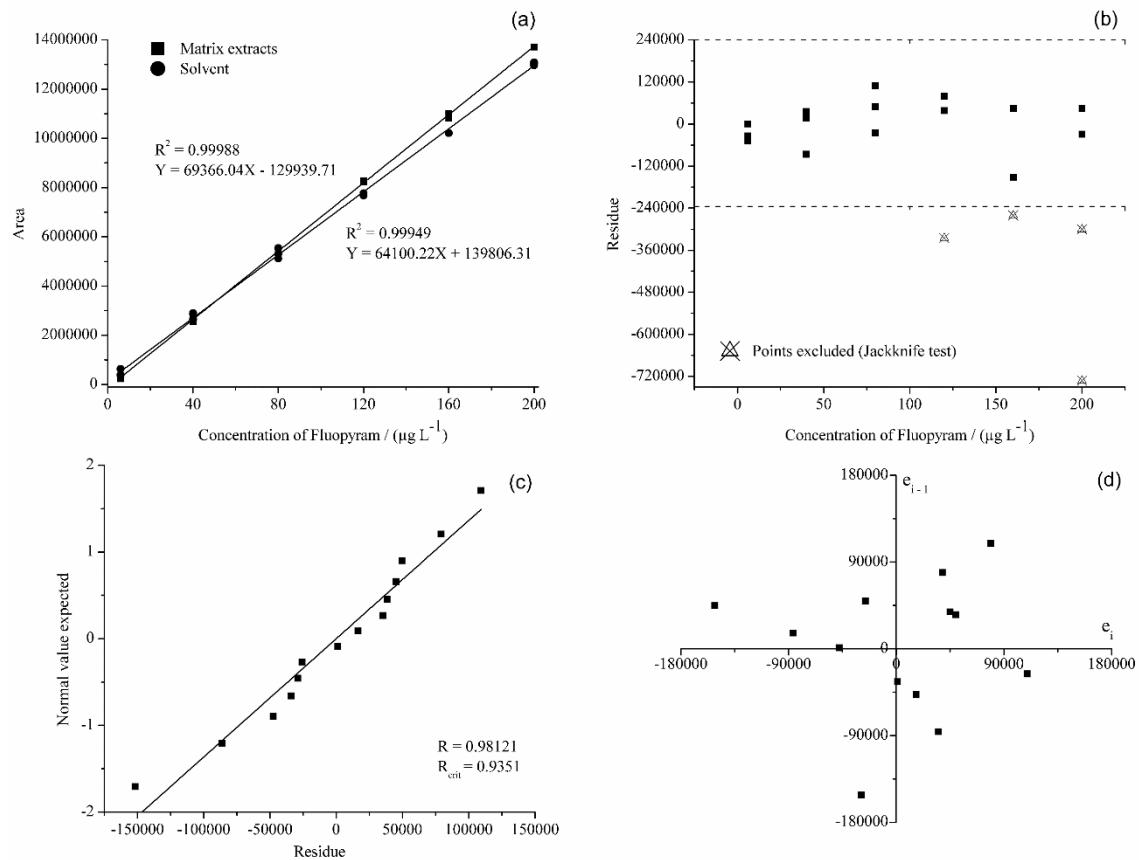


Fig. S10 (a) analytical curves of fluopyram solutions in solvent and matrix extracts; (b) linear regression residues after the exclusion of the outliers (Jackknife test); (c) normal probability of regression residues (Ryan-Joiner test); (d) autocorrelation of regression residues (Durbin-Watson test).

ei, residue; R^2 , determination coefficient; R, correlation coefficient of Ryan-Joiner test; d, Durbin-Watson statistics

Table S1 ANOVA for the significance of regression and deviation from linearity

	Regression	Deviation from linearity
n	14	14
F_{crit}	4.75	3.84
F	4.88×10^4	1.12

n, number of observations; F_{crit} , F critical; F, ratio between variances