

Supplementary Material

Carbon dots functionalized macroporous adsorption resin for bifunctional ultra-sensitive detection and fast removal of iron(III) ions

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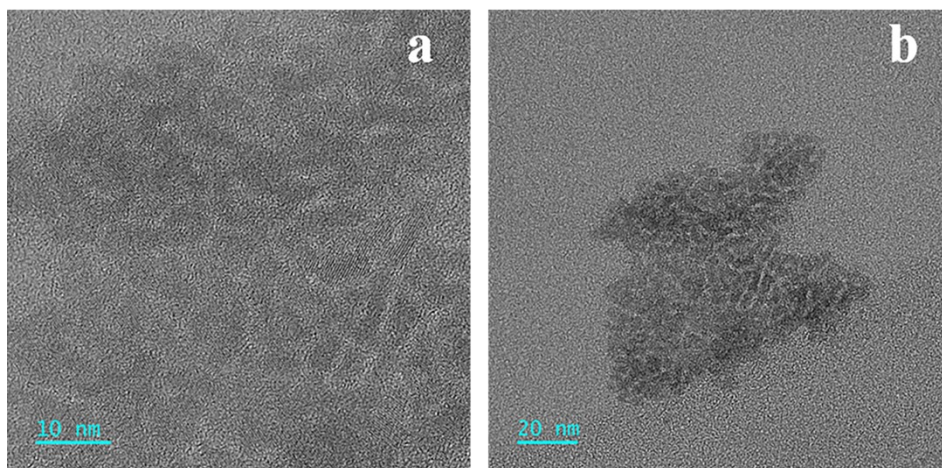


Fig. S1. TEM images of CDs with different magnification.

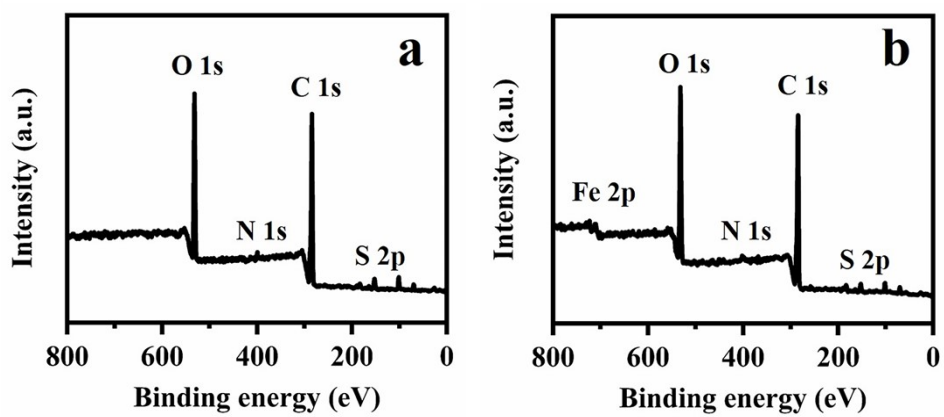


Fig. S2. XPS full spectra of MAR@poly(PA)@CD (a) before and (b) after adsorption of Fe³⁺.

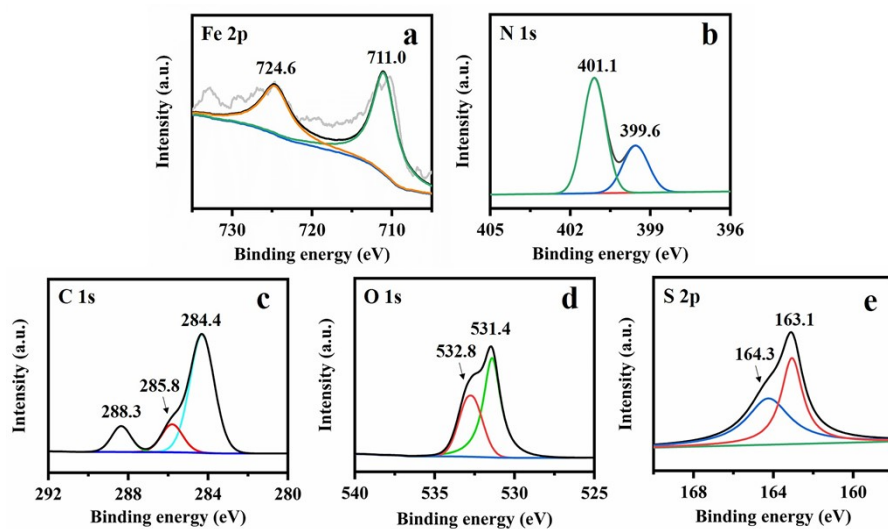


Fig. S3. High resolution XPS spectra of (a) Fe 2p, (b) N 1s, (c) C 1s, (d) O 1s and (e) S 2p after adsorption of Fe³⁺ of MAR@poly(PA)@CD.

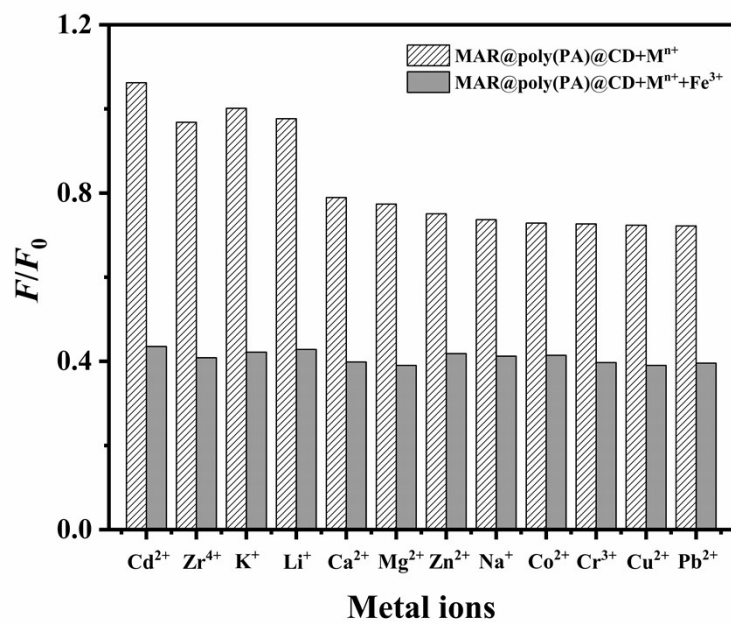


Fig. S4. Fluorescence responses of MAR@poly(PA)@CD toward 12 kinds of metal ions in the absence and presence of Fe³⁺.

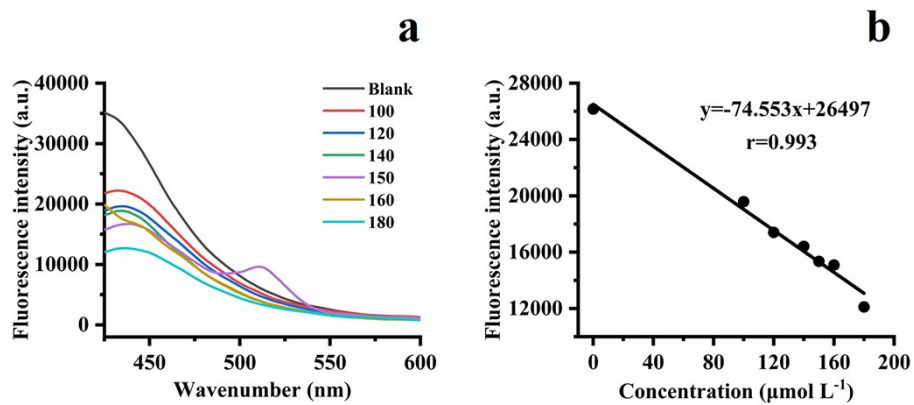


Fig. S5. (a) Fluorescence emission spectrum containing different concentrations of Fe^{3+} . (b) Linear response curve for detection of Fe^{3+} by CDs.

Table S1 Comparison of specific surface area, pore volume and average pore diameter of poly(GMA-*co*-EDMA) and MAR@poly(PA)@CD.

Materials	Specific surface area (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Average pore diameter (nm)
Poly(GMA- <i>co</i> -EDMA)	33.4	0.160	20.3
MAR@poly(PA)@CD	13.6	0.090	23.6

Table S2 Isothermal adsorption fitting parameters of MAR@poly(PA)@CD to Fe³⁺ by Langmuir and Freundlich models.

Langmuir isotherm model			Freundlich isotherm model		
Q_{\max} (mg g ⁻¹)	K_L (L g ⁻¹)	r_L	K_F (mg g ⁻¹)	$1/n$	r_F
24.15	3.451	0.9992	15.71	0.2549	0.9398

Table S3 Kinetic adsorption fitting parameters of MAR@poly(PA)@CD to Fe³⁺ by pseudo-first-order and pseudo-second-order models.

C_0 ($\mu\text{mol L}^{-1}$)	Q_e (mg g^{-1})	pseudo-first-order			pseudo-second-order		
		k_1 (min^{-1})	$Q_{1 \text{ cal}}$ (mg g^{-1})	r_1	k_2 ($\text{g mg}^{-1} \text{ min}^{-1}$)	$Q_{2 \text{ cal}}$ (mg g^{-1})	r_2
200.0	20.20	0.0382	8.492	0.7628	0.003658	22.68	0.9902