

Supplementary Information

Table SI-1 Methods for the determination of mefenamic acid

Method / reagents / conditions	Object	Linearity	Detection limit	Ref
Spectrophotometric methods				
Copper(II) ammine sulphate; extragent – chloroform; pH 9; $\lambda=430$ nm	Pure form and pharmaceuticals	6-48 $\mu\text{g/ml}$	–	18
FeCl_3 in metanol solution; pH 2.7; $\lambda=495$ nm	Pharmaceuticals	96- 579 $\mu\text{g/ml}$		19
Ce(III) reaction with arsenazo (III) reagent; pH 3; $\lambda=654$ nm	Pharmaceuticals	1 to 10 $\mu\text{g/25 ml}$		20
<i>p</i> -chloranilic acid in acetone soln; pH 8.2; $\lambda=520$ nm;	Pure form and pharmaceuticals	10-300 $\mu\text{g/ml}$	2.50 $\mu\text{g/ml}$	21
N-bromosuccinamide in methanol soln; $\lambda=362$ nm;		5-70 $\mu\text{g/ml}$	0.51 $\mu\text{g/ml}$	
3-methylbenzo-thiazolin-2-one hydrazone in methanol soln in the presence of FeCl_3 ; $\lambda=602$ nm		1-6 $\mu\text{g/ml}$	0.06 $\mu\text{g/ml}$	
Mef in hydrochloric acid in methanol; $\lambda=279, 351$ nm	Pharmaceuticals	–	–	22
Copper (II) complex with 5,7,7,12,14,14-hexamethyl -1,4,8,11-tetraazacyclo tetradeca-4,11-diene; extragent – chloroform; pH 9; $\lambda=545$ nm	Pharmaceuticals	–	0.1 mg/ml	23
Methylene violet; extragent – chloroform; pH 7.6; $\lambda=540$ nm	Pharmaceuticals	1-8 $\mu\text{g/ml}$	–	24
First derivative of the ratio spectra solution Mef in NaOH / methanol (1:9)	Synthetic mixtures, pharmaceuticals	2-10 $\mu\text{g/ml}$	1.15 $\mu\text{g/ml}$	25
<i>p</i> -dimethyl-aminocinnamaldehyde in acidified absolute methanol medium; $\lambda= 665$ nm	pharmaceutical formulations	1-8 $\mu\text{g/ml}$		26
Astrafloxin FF / pH 9–11/ extragent : isooctane-dichloroethane 64 : 36 (v/v) / $\lambda=533$	pharmaceutical formulations	2.0-21.0 $\mu\text{g/mL}$	0.72 $\mu\text{g/mL}$	27
Chromatographic methods				
LC; UV detection; $\lambda = 278$ nm; mobile phase: methanol-glacial acetic acid-water (85:2:15)	Pharmaceuticals	25 – 150 $\mu\text{g/ml}$		28
HPLC; UV detection; $\lambda=254$ nm; pH 5.0; mobile phase: acetonitrile-ammonium buffer-	Pharmaceuticals			29

tetrahydrofuran (23:20:7)				
GLC; flame ionisation detector, $t_{\text{detect.}} = 290\text{ }^{\circ}\text{C}$; mobile phase: N,N-dimethylacetamid- tetramethyl ammonium hydroxide-1- iodobutane (80:10:20)	Human serum	2 – 50 μg in 2 ml serum	1 mg/l using 2 ml serum	47
HPLC; UV detection; $\lambda=280\text{ nm}$, pH 2.6; mobile phase: phosphoric acid-acetonitrile (40:60)	Plasma	0.1 – 10 $\mu\text{g/ml}$	0.08 $\mu\text{g/ml}$ for 50 ml plasma	48
HPLC; UV detection; $\lambda=280\text{ nm}$; pH 3; mobile phase: acetonitrile-water (50:50)	Human serum	25– 2000 ng/ml	15ng/ml	49
HPLC; UV detection; $\lambda=254\text{ nm}$; mobile phase: acetonitrile- phosphate buffer (pH 3.5), methanol and tetrahydrofuran.	Human plasma	11.5–75 ng/ml	-	50
LC-MS/MS; $\lambda = 210\text{ nm}$; mobile phase: potassium phosphate buffer (pH 4.0)- acetonitrile (30:70)	River water	0.3–12 ng/l	0.15ng/l	52
Electrochemical methods				
Solid-contact voltamperometric graphite-paste sensor based on $\text{La}(\text{OH})_3$; pH 5.8	Pharmaceutica ls	$2 \times 10^{-11} -$ $4 \times 10^{-9}\text{ M}$	$6.1 \times 10^{-12}\text{ M}$	30
Solid-contact voltamperometric graphite-paste sensor based on metal complex with a Schiff base Fe (III)-4-bromo-2-[[2-(2-pyridin-2-yl- ethylsulfanyl)-ethylimino]methyl]phenol (putBrsal); pH 3.5	Pharmaceutica ls	0.02 – 150 $\mu\text{M/l}$	0.02 $\mu\text{M/l}$	31