Supplementary Information (SI) for:

Self-Assembling Figure-of-Eight and Pseudoplectoneme Aromatic

Oligoamide Ribbons

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1. Methods

Nuclear Magnetic Resonance

NMR spectra were recorded on 2 different NMR spectrometers: (1) an Avance II NMR spectrometer (Bruker Biospin) with a vertical 7.05 T narrow-bore/ultrashield magnet operating at 300 MHz for ¹H observation and 75 MHz for ¹³C observation by means of a 5-mm direct BBO H/X probe with Z gradient capabilities; (2) a DPX-400 NMR spectrometer (Bruker Biospin) with a vertical 9.4 T narrow-bore/ultrashield magnet operating at 400 MHz for ¹H observation by means of a 5-mm direct QNP ¹H/¹³C/³¹P/¹⁹F probe with gradient capabilities; Chemical shifts are reported in parts per million (ppm, δ) relative to the ¹H residual signal of the deuterated solvent used. ¹H NMR splitting patterns with observed first-order coupling are designated as singlet (s), doublet (d), triplet (t), or quartet (q). Coupling constants (*J*) are reported in hertz. Samples were not degassed. Data processing was performed with Topspin 3.6 software.

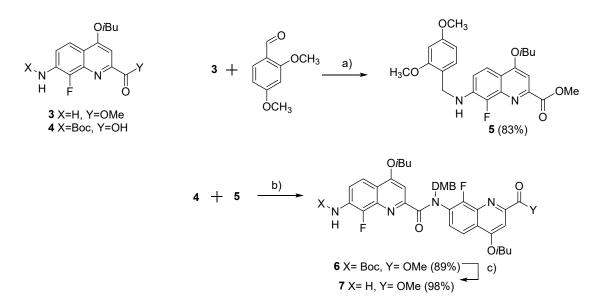
Crystallography

The diffraction data for compounds **1b** and **2b** were collected at the IECB X-ray facility (CNRS UAR 3033 – INSERM US001, University of Bordeaux) with a Rigaku FRX rotating anode (2.9 kW) diffractometer using CuKa wavelength with a partial chi goniometer (AFC11). The X-ray source is equipped with high flux Osmic Varimax mirrors and a Pixel Hybrid Dectris Eiger1M detector. Data were processed with the Rigaku Oxford Diffraction CrysalisPro software (version1.171.41.118a).^[1] The crystal structure of compound 1b was solved with the dual-space algorithm implemented into Shelxt^[2] while the structure of 2b with Shelxd^[2]. Both structures were refined by full-matrix least-squares method on F2 with Shelxl-2014^[2] within Olex2. ^[3] Only non-H atoms of the backbones and side chains observable in the electron density maps were refined with anisotropic displacement parameters. H-atoms were refined in the riding-model approximation, with Uiso(H)=1.2Ueq (CH, CH2, NH). DFIX, AFIX, RIGU and SIMU restraints were apply to model geometry of the molecules and thermal motion parameters. Some residual electron density peaks observed in the difference Fourier maps could not be modelled. The PLATON/SQUEEZE^[4] procedure was applied on both datasets. Highly disordered Chlorobenzene, methanol and water solvent molecules were squeezed in structures **1b** and **2b** corresponding to contributions of some 2279 and 1602 electrons respectively.

2. Materials and Methods for chemical synthesis

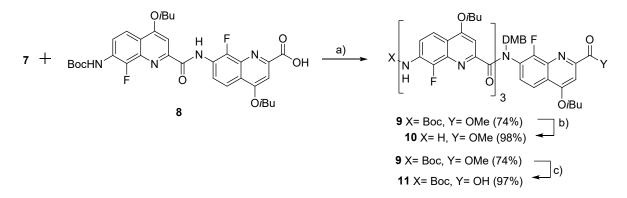
All reactions were carried out under a dry nitrogen atmosphere. Commercial reagents were purchased from SigmaAldrich, TCI Chemicals or Alfa-Aesar and were used without further purification. Tetrahydrofurane (THF) and dichloromethane (CH₂Cl₂) were dried over alumina columns (MBRAUN SPS-800 solvent purification system); chloroform (CHCl₃) and diisopropylethylamine (DIEA) were distilled over calcium hydride (CaH₂) prior to use. Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60-F254 plates and observed under UV light. Column chromatography purifications were carried out on Merck GEDURAN Si60 (40-63 µm). Radial centrifugally accelerated chromatography purifications were carried out on Chromatotron® with silica gel, Merck grade 7749, TLC grade with binder and fluorescent indicator. Preparative recycling GPC (gel permeation chromatography) were performed on JAIGEL 20*600 mm columns (Japan Analytical Industry) at a flow rate of 10 mL min⁻¹ with a mobile phase composed of 1% (vol/vol) Et₃N in chloroform. Monitoring by UV detection was carried out at 254 nm, 280 nm, 300 nm and 360 nm. ESI mass spectra were obtained from the Mass Spectrometry Laboratory at the European Institute of Chemistry and Biology (UMS 3033 & US01 - IECB), Pessac, France

2.1 Synthesis of dimers



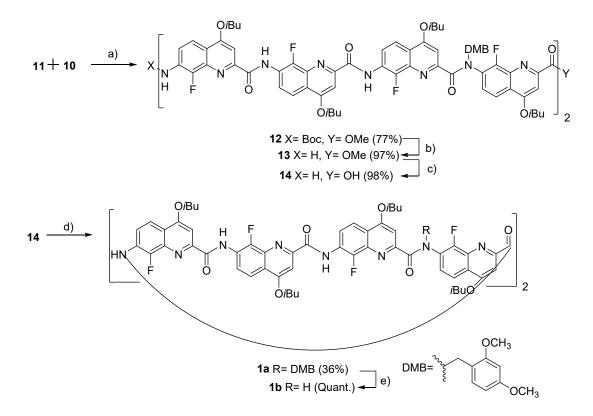
Scheme S1. Synthesis of DMB protected Q^F dimer **6** and **7**: a) sodium triacetoxyborohydride, 1,2-dichloroethane, room temperature, 16h; b) i) 1-chloro-N,N,2-trimethyl-1-propenylamine, CHCl₃, room temperature, 3 h ; ii) DIEA, CHCl₃, room temperature, 12 h; c) HCl 4N in dioxane, room temperature, 4h.

2.2 Synthesis of tetramers



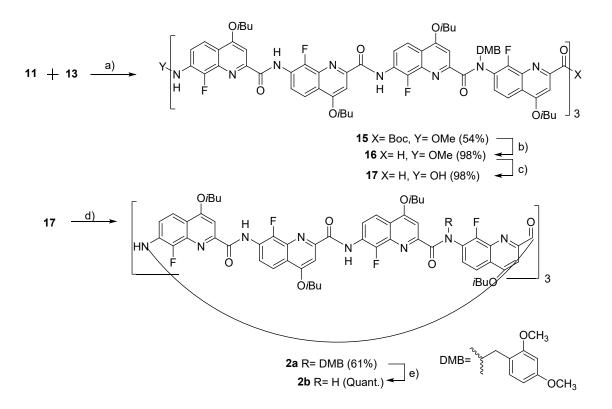
Scheme S2. Synthesis of Q^F tetramer **10** and **11**: a) i)1-chloro-N,N,2-trimethyl-1-propenylamine, CHCl₃, room temperature, 3 h ; ii) DIEA, CHCl₃, room temperature, 12 h; b) HCl 4N in dioxane, room temperature, 4 h. c) LiOH, THF, H₂O, room temperature, 4 h.

2.3 Synthesis of cyclo-octamers



Scheme S3. Synthesis of Q^F octamer macrocycle **1b**: a) i)1-chloro-N,N,2-trimethyl-1-propenylamine, CHCl₃, room temperature, 3 h; ii) DIEA, CHCl₃, room temperature, 12 h; b) HCl 4N in dioxane, room temperature, 4 h; c) LiOH, THF, H₂O, room temperature, 4 h; d) triphenylphosphine, trichloroacetonitrile, DIPEA, CHCl₃, room temperature, 48 h; e) trifluoroacetic acid, CHCl₃, 60°C, 2 h.

2.4 Synthesis of cyclco-dodecamers



Scheme S4. Synthesis of Q^F dodecamer macrocycle **2b**: a) i)1-chloro-N,N,2-trimethyl-1-propenylamine, CHCl₃, room temperature, 3 h ; ii) DIEA, CHCl₃, room temperature, 12 h; b) HCl 4N in dioxane, room temperature, 4 h; c) sodium hydroxide, THF, H₂O, room temperature, 4 h; d) triphenylphosphine, trichloroacetonitrile, DIPEA, CHCl₃, room temperature, 48 h; e) trifluoroacetic acid, CHCl₃, 60°C, 2 h.

3. Solution studies

3.1 NMR spectra of 1b

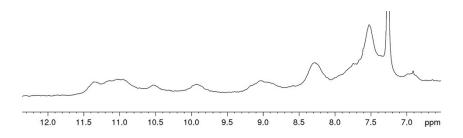


Figure S1. Part of ¹H NMR spectrum (400 MHz, 0.5mM, CDCl₃) of octamer macrocycle 1b.

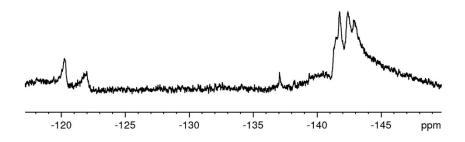
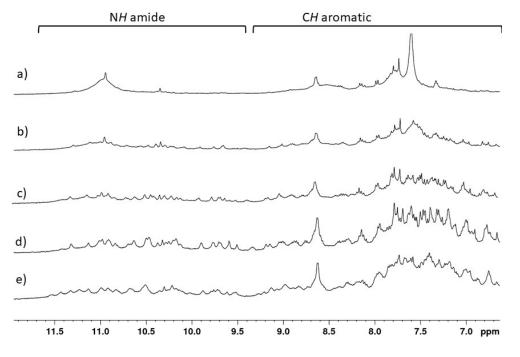


Figure S2. ¹⁹F NMR spectrum (400 MHz, 0.5mM, CDCl₃) of octamer macrocycle 1b.



3.2 Variable temperature experiment of 1b

Figure S3. Part of ¹H NMR spectra (400 MHz, C₂D₂Cl₄) for octamer macrocycle **1b** at variable temperature, a) 353 K, b) 333 K, c) 313 K, d) 298 K, e) 278 K.

3.3 NMR spectra of 2b

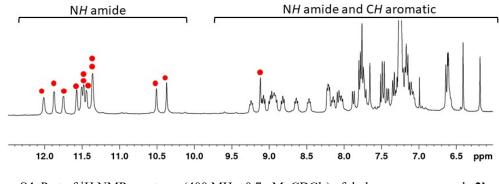


Figure S4. Part of ¹H NMR spectrum (400 MHz, 0.7mM, CDCl₃) of dodecamer macrocycle 2b.

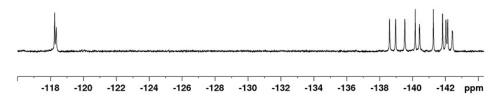


Figure S5. ¹⁹F NMR spectrum (400 MHz, 0.7mM, CDCl₃) of dodecamer macrocycle 2b.

3.4 Variable concentration experiment for 2b

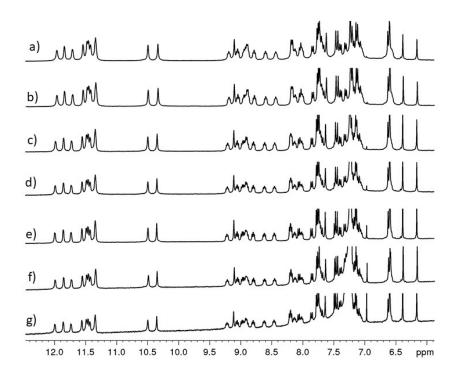


Figure S6. Part of ¹H NMR spectra (400 MHz, CDCl₃, 298 K) of dodecamer **2b** at different concentrations, a) 3.20 mM, b) 2.37 mM, c) 1.57 mM, d) 0.98 mM, e) 0.49 mM, f) 0.24 mM, g) 0.1 mM.

3.5 Variable temperature experiment for 2b

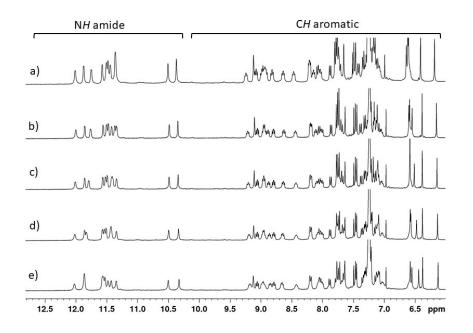


Figure S7. Part of ¹H NMR spectra (400 MHz, 0.5 mM, CDCl₃) of dodecamer **2b** in variable temperature, a) 298 K, b) 288 K, c) 278 K, d) 268 K, e) 258 K.

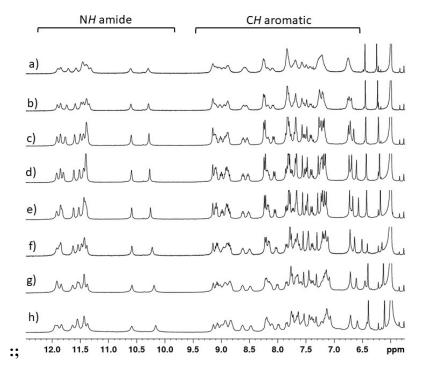


Figure S8. Part of ¹H NMR spectra (400 MHz, 0.5 mM, C₂D₂Cl₄) of dodecamer **2b** in variable temperature, a) 353 K, b) 343 K, c) 333 K, d) 323 K, e) 313 K, f) 298 K, g) 283 K, h) 273 K.

3.6 Gel permeation chromatography (GPC)

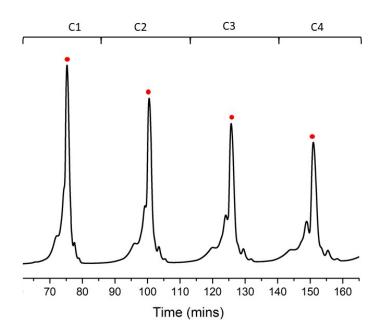


Figure S9. Chromatogram of recycling GPC of 1a purification. The peaks marked with red circle stand for the DMB functionalized macrocycle 1a.

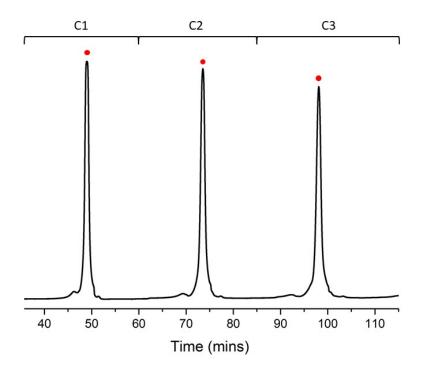
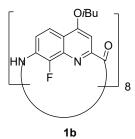
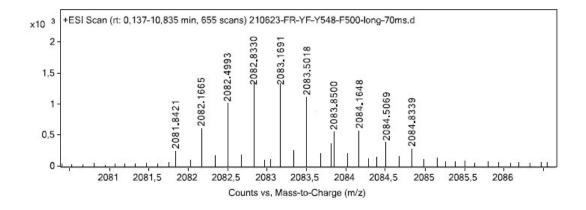


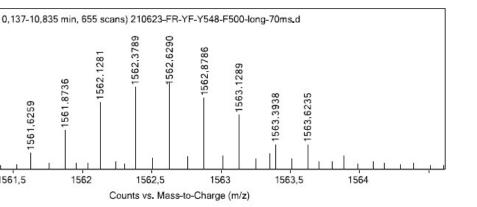
Figure S10. GPC Chromatogram of 2a purification. The peaks marked with red circle stand for 2a. The residual time can be calculated be the time between two cycles of same peaks.

3.7 Mass spectrometry of cyclo-octamer 1b

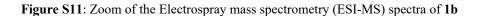




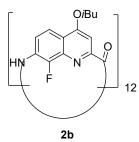
MS (ESI): m/z calcd for $C_{336}H_{316}F_{24}N_{48}O_{48}\,[3M\!+\!3H]^{3+}\,2082.4450$; found 2082.4993

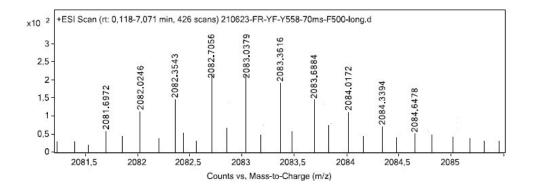


MS (ESI): m/z calcd for $C_{336}H_{316}F_{24}N_{48}O_{48}\,[3M\!+\!4H]^{4+}\,1562.3364$; found 1562.3789

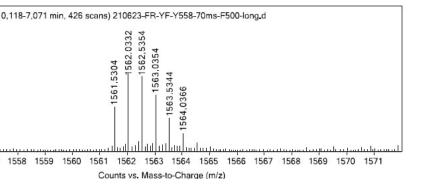


3.8 Mass spectrometry of cyclo-dodecamer 2b





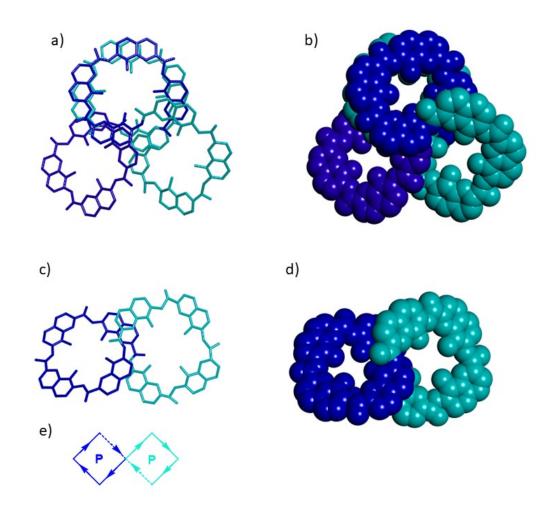
MS (ESI): m/z calcd for $C_{336}H_{316}F_{24}N_{48}O_{48}\,[2M\!+\!3H]^{3+}\,2082.7795$; found 2082.7056



MS (ESI): m/z calcd for $C_{168}H_{158}F_{12}N_{24}O_{24}\,[M+2H]^{2+}\,1562.0856$; found 1562.0332

Figure S12: Zoom of the Electrospray mass spectrometry (ESI-MS) of 2b

4. Solid state studies: X-ray crystallography



4.1 Crystal structure and X-ray data for 1b

Figure S13. X-ray structure of octamer macrocycle **1b**, a, b) top view of crystal packing showing a dimeric architecture $(1b)_2$; c, d) Top view of monomeric **1b** which is one strand from $(1b)_2$, both cycles have P helicity; e) Schematic representations of monomeric **1b**, straight line represent a Q^F unit, the solid line is in the front and dot line is in the back. Arrows represent handedness P or M. Side chains (O*i*Bu groups) and hydrogen atoms have been removed for clarity.

CCDC d-	2160708	Table S1. Crystal data
CCDC code		and structure
Empirical formula	$C_{232}H_{190}Cl_{3.5}F_{16}N_{32}O_{32}$	refinement for
Formula weight	4366.22	1b.
Temperature/K	130	
Crystal system	monoclinic	
Space group	P2 ₁ /c	
a/Å	36.4661(5)	
b/\AA	42.5825(5)	
c/Å	20.4587(3)	
$\alpha/^{\circ}$	90.0	
β^{\prime}	99.8128(14)	
$\gamma^{\prime \circ}$	90.0	
Volume/Å3	31303.9(8)	
Ζ	4	
pcalcg/cm3	0.926	
μ/mm-1	0.844	
<i>F(000)</i>	9062.0	
Crystal size/mm3	0.1 imes 0.01 imes 0.01	
Radiation	Cu Ka ($\lambda = 1.54178$)	
2 Θ range for data collection/°	4.15 to 88.588	
Index ranges	$-31 \le h \le 32, -37 \le k \le 38, -18 \le l \le 18$	
Reflections collected	126576	
Independent reflections	23865 [$R_{int} = 0.1958, R_{sigma} = 0.0964$]	
Data/restraints/parameters	23865/22844/2864	
Goodness-of-fit on F2	1.389	
Final R indexes [$I > = 2\sigma$ (I)]	$R_1 = 0.1390, wR_2 = 0.3808$	
Final R indexes [all data]	$R_1 = 0.1979, wR_2 = 0.4304$	

4.2 Crystal structure and X-ray data for 2b

g)

Figure S14. X-ray structure of dodecamer macrocycles $(2b)_2$, a) Front view of $(2b)_2$, opposite handedness cycles were marked as blue (P) and light blue (M), middle cycle was marked as white; b) front view of $(2b)_2$ in CPK style; c) top view of $(2b)_2$; d) top view of $(2b)_2$ in CPK style; e) top view of monomeric 2b which is one strand from crystal packing $(2b)_2$; f) front view of monomeric 2b in CPK style; g) schematic representation of 2b, straight line represent a Q^F unit, the solid line is in the front layer and dot line is in the back. Arrows denote the P or M handedness; side chains (O*i*Bu groups) and hydrogen atoms have been removed for clarity.

CCDC code	2161479
Empirical formula	$C_{348}H_{321}Cl_2F_{24}N_{48}O_{48}$
Formula weight	6470.41
Temperature/K	135
Crystal system	triclinic
Space group	P-1
a/Å	24.6529(8)
b/\AA	28.2336(8)
c/Å	32.8540(10)
$\alpha^{\prime \circ}$	76.852(2)
$\beta^{\prime \circ}$	72.247(3)
$\gamma^{\prime \circ}$	74.951(3)
Volume/Å ³	20757(1)
Z	2
$ ho_{calc}g/cm^3$	1.035
μ/mm^{-1}	0.760
<i>F(000)</i>	6758.0
Crystal size/mm ³	$0.05\times0.05\times0.05$
Radiation	Cu Ka ($\lambda = 1.54178$)
2 Θ range for data collection/ $^{\circ}$	3.844 to 87.974
Index ranges	$-22 \le h \le 22, -25 \le k \le 21, -29$
Reflections collected	101599
Independent reflections	31689 [$R_{int} = 0.0908$, $R_{sigma} =$
Data/restraints/parameters	31689/72118/3679
Goodness-of-fit on F^2	2.082
Final R indexes [$I \ge 2\sigma$ (I)]	$R_1 = 0.2225, wR_2 = 0.4996$
Final R indexes [all data]	$R_1 = 0.2792, wR_2 = 0.5454$

Table S2. Crystal data and structure refinement for 2b

4.3 Symmetry of 1b

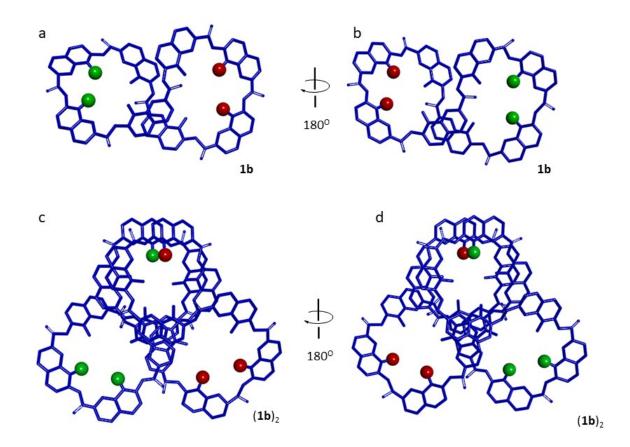


Figure S15. X-ray structure of **1b**, a) top view of monomeric **1b** which is one strand in the crystal packing, two side F atoms were marked as green and red balls; b) **1b**, rotated 180° horizontal from a); c) top view of $(1b)_2$, picked F atoms were marked as green and red; d) $(1b)_2$, rotated 180° horizontal from c). Side chains (O*i*Bu groups) and included solvent molecules have been removed for clarity.

4.4 Symmetry of 2b

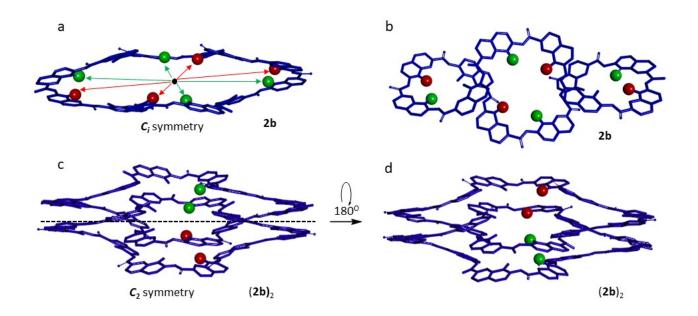
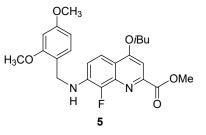
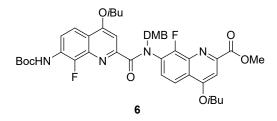


Figure S16. X-ray structure of macrocycle **2b**, a) front view of macrocycle monomeric **2b** only one strand is shown, fluorine atoms are marked as green and red balls, equal distance between a pair of antipodal fluorine atoms and inversion center were marked as different color arrow (green and red); b) top view of **2b**; c) side view of macrocycle dimeric (**2b**)₂; d) side view of (**2b**)₂, rotated 180° vertical from c). Side chains (O*i*Bu groups) and included solvent molecules have been removed for clarity.

5. Synthetic procedures

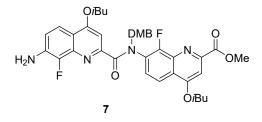


Compound 5: amine $3^{[5]}$ (610 mg, 1.98 mmol), 2,4-dimethoxybenzaldehyde (659 mg, 3.97mmol) were dissolved in 1,2-dichloroethane (5 ml), under inert atmosphere of N₂. The slurry was stirred at room temperature for 2 hours. Then sodium triacetoxyborohydride (841 mg, 3.97 mmol) was added to the slurry. The mixture was stirred at room temperature under a N₂ atmosphere for 18 h. The reaction mixture was quenched by adding aqueous saturated NaHCO₃, and the product was extracted with dichloromethane. The organic layer was washed with brine, and dried over anhydrous MgSO₄. After the solvent was removed in vacuo, the residue was purified by flash column chromatography (silica gel, ethyl acetate / cyclohexane = 1 / 6) to give **5** as yellow solid (731 mg, 83 %). ¹H NMR (300 MHz, CDCl₃, 298 K) δ 7.90 (dd, *J*(H, H) = 1.5, *J*(H, H) = 9.1, 1H), 7.37 (s, 1H), 7.26-7.18 (m, 2H), 6.50 (d, *J*(H, H) = 2.3, 1H), 6.45-6.41 (m, 1H), 4.91 (br, 1H), 4.49 (d, *J*(H, H) = 6.2, 2H), 4.07 (s, 3H), 4.03 (d, *J*(H, H) = 6.4, 2H), 3.87 (s, 3H), 3.81 (s, 3H), 2.34-2.20 (m, 1H), 1.15 (s, 3H), 1.12 (s, 3H) ¹³C NMR (100 MHz, CDCl₃) δ 166.4, 162.7, 160.4, 158.4, 149.6, 145.3, 142.8, 138.9, 136.9, 129.3, 118.9, 117.5, 116.0, 114.7, 103.8, 98.7, 98.3, 74.9, 55.3, 53.0, 42.9, 28.1, 19.2. HRMS (ESI): m/z calcd for C₂₄H₂₈FN₂O₅ [M+H]⁺ 443.1977, found 443.1989.

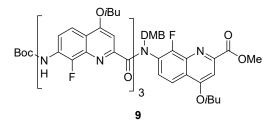


Compound 6: A solution of $4^{[1]}$ (645 mg, 1.7 mmol) in dry CHCl₃ (3 mL) was cooled to 0°C. 1-Chloro-N,N,2trimethyl-1-propenylamine (0.75 mL, 3.0 eq.) was added. The solution was stirred at room temperature for 3 h under inert atmosphere of N₂. The solvent was removed under high-vacuum for at least 2 hours to yield the corresponding acid chloride. The resulting acid chloride was dissolved in dry CHCl₃ (3 mL) and added to a solution of 5 (630 mg, 0.9 eq.) and dry DIPEA (0.73 mL, 4.0 eq.) in dry CHCl₃ (1 mL). The mixture was stirred

for 12 h. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane, washed with saturated aqueous NH₄Cl, saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica gel, ethyl acetate / cyclohexane = 1 / 4) to give **6** as yellow solid (1.1 g, 89 %). ¹H NMR (300 MHz, CDCl₃, 298 K) δ 8.19 (t, *J*(H, H) = 7.9, 1H), 7.82 (t, *J*(H, H) = 10.9, 2H), 7.58-7.49 (m, 3H), 7.25 (s, 1H), 6.75 (s, 1H), 6.46 (d, *J*(H, H) = 8.1, 1H), 6.30 (s, 1H), 5.21 (dd, *J*(H, H) = 14.5, 2H), 4.01 (br, 7H), 3.78 (s, 3H), 3.47 (s, 3H), 2.35-2.18 (m, 2H), 1.52 (s, 9H), 1.15-1.10 (m, 12H) ¹³C NMR (100 MHz, CDCl₃) δ 168.6, 165.9, 162.4, 162.0, 160.4 158.5, 154.6, 154.1, 152.1, 151.6, 149.3, 146.6, 144.1, 138.9, 136.9, 131.6, 131.3, 129.6, 126.6, 122.2, 118.9, 117.9, 117.1, 116.9, 116.0, 104.1, 101.6, 100.2, 98.1, 81.4, 75.2, 75.0, 55.2, 54.9, 53.2, 47.7, 38.6, 37.0, 35.5, 28.2, 28.1, 28.0, 19.1. HRMS (ESI): m/z calcd for C₄₃H₄₉F₂N₄O₉ [M+H]⁺ 803.3462, found 803.3486.

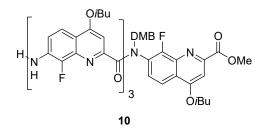


Synthesis of compound 7: **6** (200 mg, 0.24 mmol) was dissolved in dioxane (0.2 mL). HCl in dioxane (4M, 2mL) was added to the solution, and stirred at room temperature for 4 h. The mixture was diluted with DCM, the organic layer was washed with saturated aqueous NaHCO₃ and brine, dried over MgSO4, filtered and concentrated in vacuo to give **7** as yellow solid (169 mg, 98%).¹H NMR (400 MHz, CDCl₃, 298 K) δ 7.82 (d, *J*(H, H) = 8.9, 1H), 7.64 (d, *J*(H, H) = 8.8, 1H), 7.55 (t, *J*(H, H) = 7.8, 1H), 7.49-7.47 (m, 2H), 7.12 (s, 1H), 6.84 (t, *J*(H, H) = 8.1, 1H), 6.42 (d, *J*(H, H) = 8.1, 1H), 6.27 (s, 1H), 5.19 (dd, *J*(H, H) = 14, 2H), 3.98-3.94 (m, 7H), 3.75 (s, 3H), 3.44 (s, 3H), 2.30-2.16 (m, 2H), 1.12-1.06 (m, 12H). ¹³C NMR (100 MHz, CDCl₃) δ 168.9, 165.9, 162.4, 162.1, 160.3, 158.5, 154.3, 151.6, 149.2, 144.9, 142.5, 139.0, 138.9, 138.1, 138.0, 133.8, 131.6, 131.4, 129.8, 122.2, 118.2, 117.0, 116.0, 115.2, 104.1, 101.5, 98.8, 98.1, 75.2, 74.8, 67.0, 55.2, 54.9, 53.2, 47.7, 28.2, 19.1. HRMS (ESI): m/z calcd for C₃₈H₄₀F₂N₄O₇ [M+H]⁺702.7558, found 703.7089.

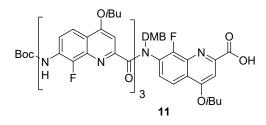


Compound 9: A solution of $8^{[1]}$ (845 mg, 1.3 mmol) in dry CHCl₃ (3 mL) was cooled to 0°C. 1-Chloro-N,N,2trimethyl-1-propenylamine (0.27 mL, 1.5 eq.) was added. The solution was stirred at room temperature for 3 h under inert atmosphere of N₂. The solvent was removed under high-vacuum for at least 2 hours to yield the

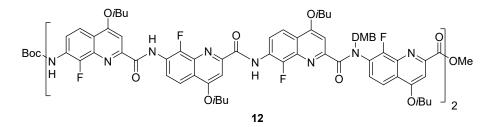
corresponding acid chloride. The resulting acid chloride was dissolved in dry CHCl₃ (3 mL) and added to a solution of 7 (929 mg, 1.0 eq.) and dry DIPEA (0.70 mL, 3.0 eq.) in dry CHCl₃ (2 mL). The mixture was stirred for 12 h. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane and washed with saturated aqueous NH₄Cl, saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica gel, ethyl acetate / cyclohexane = 1 / 4) to give **9** as yellow solid (1.3 g, 74 %). ¹H NMR (300 MHz, CDCl₃, 298 K) δ 10.96 (s, 1H), 10.59 (s, 1H), 8.87 (q, *J*(H, H) = 6.8, 1H), 8.60 (t, *J*(H, H) =7.9, 1H), 8.53 (q, *J*(H, H) =5.3, 1H), 8.12 (q, *J*(H, H) =3.5, 1H), 8.04 (d, *J*(H, H) =5.2, 1H), 7.91(d, *J*(H, H) =8.9, 2H), 7.74 (s, 1H), 7.69 (s, 1H), 7.60 (t, *J*(H, H) =7.9, 1H), 7.55 (s, 1H), 7.50 (s, 1H), 7.47 (s, 1H), 7.31 (s, 1H), 6.45 (q, *J*(H, H) =3.4, 1H), 6.31 (s, 1H), 5.47 (d, *J*(H, H) =7.3, 1H), 4.97(d, *J*(H, H) =7.3, 1H), 4.14-4.09 (m, 4H), 4.02-3.98 (m, 7H), 3.76 (s, 3H), 3.50 (s, 3H), 2.40-2.12 (m, 4H), 1.59 (s, 9H), 1.17-1.10 (m, 24H).¹³C NMR (100 MHz, CDCl₃) δ 168.6, 166.0, 163.5, 163.4, 162.7, 162.6, 162.4, 162.1, 160.5, 158.7, 154.7, 152.4, 151.4, 151.2, 149.4, 148.6, 147.1, 145.2, 143.8, 137.8, 137.6, 137.3, 131.6, 130.0, 127.8, 126.8, 126.0, 122.5, 120.9, 120.0, 119.1, 118.8, 117.8, 116.3, 104.2, 101.8, 100.7, 98.3, 97.9, 81.8, 75.6, 75.3, 55.4, 55.1, 53.3, 48.0, 30.4, 29.8, 28.1, 19.1. HRMS (ESI): m/z calcd for C₇₁H₇₄F₄N₈O₁₃ [M+H]⁺1323.4096, found 1324.5714.



Compound **10**: **9** (0.20 g, 0.15 mmol) was dissolved in dioxane (0.2 mL). HCl in dioxane (4M, 2mL) was added to the solution, and stirred at room temperature for 4 h. The mixture was diluted with DCM, the organic layer was washed with saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo to give **10** as yellow solid (180 mg, 98%).¹H NMR (300 MHz, CDCl₃, 298 K) δ 11.07 (s, 1H), 10.64 (s, 1H), 8.90 (q, *J*(H, H) =5.3, 1H), 8.61 (t, *J*(H, H) =7.8, 1H), 8.12 (d, *J*(H, H) =9.1, 1H), 7.98-7.89 (m, 3H), 7.70-7.64 (m, 3H), 7.55 (s, 1H), 7.50 (d, *J*(H, H) =8.3, 1H), 7.32 (s, 1H), 7.14 (t, *J*(H, H) =8.3, 1H), 6.47 (d, *J*(H, H) =8.2, 1H), 6.34 (s, 1H), 5.50 (d, *J*(H, H) =14.6, 1H), 4.98 (d, *J*(H, H) =14.6, 1H), 4.13-3.99 (m, 8H), 3.79 (s, 3H), 3.72 (s, 3H), 3.53 (s, 3H), 2.34-2.16 (m, 4H), 1.26 (s, 3H), 1.18-1.12 (m, 18H), 0.97 (d, *J*(H, H) =6.6, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 168.8, 166.0, 163.3, 162.9, 162.7, 162.3, 162.0, 160.6, 158.7, 154.6, 151.2, 150.7, 149.4, 148.2, 145.2, 144.9, 142.0, 139.1, 138.4, 137.4, 135.3, 131.5, 130.1, 127.0, 126.8, 125.9, 122.6, 120.3, 119.5, 119.2, 118.8, 117.5, 117.1, 116.5, 115.9, 104.3, 101.8, 100.5, 98.4, 98.1, 96.4, 75.5, 75.2, 67.2, 55.4, 55.2, 53.3, 48.1, 31.7, 29.8, 28.2, 22.7, 19.3, 19.1, 14.2. HRMS (ESI): m/z calcd for C₆₆H₆₆F₄N₈O₁₁ [M+H]⁺ 1223.2926, found 1224.5142.

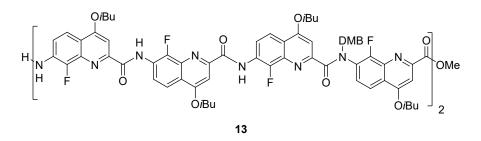


Compound **11**: **9** (750 mg, 0.56 mmol) was dissolved in a mixture of THF (10 mL) and H₂O (2 mL). To this solution was added LiOH (71 mg, 3.0 equiv.). The solution was stirred at room temperature for 3 h. Then solution was neutralized with 1N HCl to pH = 4~5, and concentrated under reduced pressure to remove THF. H₂O (30 mL) was added to the residue. The aqueous phase was extracted with CH₂Cl₂ (3×20 mL). The combined organic phases were dried over Na₂SO₄, filtered, then evaporated to give dimer acid **11** as a yellow solid (702 mg, 97%). ¹H NMR (300 MHz, CDCl₃, 298 K) δ 10.99 (s, 1H), 10.56 (s, 1H), 8.89 (q, *J*(H, H) =5.3, 1H), 8.61 (t, *J*(H, H) =7.6, 1H), 8.52 (t, *J*(H, H) =8.1, 1H), 8.12 (d, *J*(H, H) =9.5, 1H), 8.04 (d, *J*(H, H) =9.3, 1H), 7.91 (t, *J*(H, H) =7.8, 2H), 7.73 (s, 1H), 7.68 (s, 1H), 7.61 (s, 1H), 7.53 (t, *J*(H, H) =7.9, 1H), 7.45 (s, 1H), 7.36 (s, 1H), 6.47 (d, *J*(H, H) =7.7, 1H), 6.29 (s, 1H), 5.42 (d, *J*(H, H) =13.5, 1H), 5.08 (d, *J*(H, H) =14.8, 1H), 4.14-4.09 (m, 4H), 4.05-4.03 (m, 4H), 3.76 (s, 3H), 3.46 (s, 3H), 2.40-2.15 (m, 4H), 1.59 (s, 9H), 1.25 (s, 3H), 1.17-1.11 (m, 18H), 0.95 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 168.3, 164.4, 163.9, 162.8, 162.5, 160.6, 158.7, 154.5, 152.5, 147.7, 145.1, 143.8, 137.7, 131.7, 129.9, 127.9, 126.9, 122.7, 120.9, 120.2, 119.9, 119.1, 118.8, 117.7, 117.0, 104.2, 100.63, 99.7, 98.3, 97.9, 81.7, 75.7, 75.6, 75.3, 55.4, 55.0, 31.7, 30.4, 29.8, 28.4, 28.2, 28.0, 22.7, 19.3, 19.1, 14.2. HRMS (ESI): m/z calcd for C₇₀H₇₂F₄N₈O₁₃ [M+H]⁺ 1309.3823, found 1310.5554.

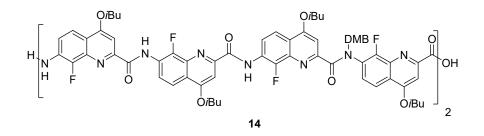


Compound 12: A solution of 11 (400 mg, 0.30 mmol) in dry CHCl₃ (3 mL) was cooled to 0°C. 1-Chloro-N,N,2trimethyl-1-propenylamine (0.10 mL, 2.2 eq.) was added. The solution was stirred at room temperature for 3 h under N₂. The solvent was removed under high-vacuum for at least 2 hours to yield the corresponding acid chloride. The resulting acid chloride was dissolved in dry CHCl₃ (3 mL) and added to a solution of 10 (377 mg, 1.0 eq.) and dry DIPEA (0.16 mL, 3.0 eq.) in dry CHCl₃ (2 mL). The mixture was stirred for 12 h. The organic layer was washed with saturated aqueous NH₄Cl, saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica gel, ethyl acetate / cyclohexane = 1 / 3) to give 12 as yellow solid (594 mg, 77 %.). ¹H NMR (400 MHz, CDCl₃, 298 K) δ 11.44-10.28 (m, 5H), 9.09-8.33 (m, 5H), 8.16-7.31 (m, 20H), 7.19-6.97 (m, 2H), 6.51-5.99 (m, 3H), 5.47-4.85 (m,

4H), 4.12-3.36 (m, 31H), 2.39-2.09 (m, 8H), 1.66-1.51 (m, 12H), 1.22-1.07 (m, 45H).¹³C NMR (100 MHz, CDCl₃) δ 167.6, 164.8, 162.1, 161.4, 160.9, 159.4, 159.2, 157.4, 153.5, 151.3, 150.2, 150.1, 149.9, 148.0, 146.8, 137.6, 136.4, 130.7, 130.2, 128.4, 128.1, 126.6, 125.8, 125.5, 124.8, 121.5, 120.8, 119.4, 118.9, 118.7, 118.1, 117.6, 116.4, 116.0, 115.3, 103.1, 99.7, 99.5, 98.2, 97.0, 76.2, 74.3, 74.0, 54.2, 54.1, 53.9, 52.0, 46.6. HRMS (ESI): m/z calcd for C₁₃₆H₁₃₆F₈N₁₆O₂₃ [M+H]⁺ 2514.6602, found 2515.0552.

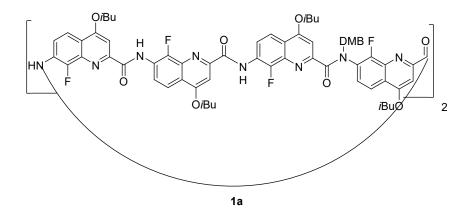


Compound **13**: **12** (100 mg, 0.039 mmol) was dissolved in dioxane (0.1 mL). HCl in dioxane (4M, 1.0 mL) was added to the solution, and stirred at room temperature for 4 h. The mixture was diluted with DCM and the organic layer was washed with saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo to give **13** as yellow solid (93 mg, 97%).¹H NMR (400 MHz, CDCl₃, 298 K) δ 11.30-10.37 (m, 5H), 9.13-8.32 (m, 4H), 8.22-7.37 (m, 20H), 7.22-6.92 (m, 3H), 6.51-6.18 (m, 3H), 5.47-4.94 (m, 4H), 4.35-3.91 (m, 25H), 3.77-3.36 (m, 6H), 2.42-2.11 (m, 8H), 1.72-1.62 (m, 6H), 1.24-0.95 (m, 42H). ¹³C NMR (100 MHz, CDCl₃) δ 168.6, 165.8, 163.1, 162.7, 162.4, 161.9, 160.5, 160.3, 160.0, 158.6, 158.4, 154.6, 151.1, 150.5, 138.8, 137.7, 137.1, 135.1, 131.7, 11.4, 129.5, 126.9, 125.9, 122.5, 121.9, 120.0, 119.1, 118.9, 117.4, 117.0, 116.3, 104.1, 101.6, 101.2, 100.7, 100.4, 99.2, 98.1, 96.1, 95.3, 72.2, 71.1, 61.5, 59.4, 53.1, 47.6, 42.8, 37.6, 35.6, 30.2, 29.6. HRMS (ESI): m/z calcd for C₁₃₁H₁₂₈F₈N₁₆O₂₁ [M+H]⁺2414.5432, found 2414.9951.

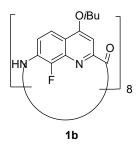


Compound 14: 13 (90 mg, 0.037 mmol) was dissolved in THF (1.0 mL). LiOH (7.45 mg, 0.186 mmol) was added to the solution, followed with H₂O (0.1 mL) and stirred at room temperature for 4 h. The organic layer was washed with citric acid (5%, 25 mL) and brine, dried over MgSO₄, filtered and concentrated in vacuo to give 14 as yellow solid (82 mg, 98%).¹H NMR (400 MHz, CDCl₃, 298 K) δ 11.27-10.34 (m, 5H), 9.10-8.32 (m, 4H), 8.07-7.30 (m, 20H), 7.19-6.90 (m, 3H), 6.64-6.04 (m, 3H), 5.44-4.87 (m, 4H), 4.42-3.79 (m, 18H), 3.76-3.30 (m, 10H), 2.40-2.07 (m, 8H), 1.23-0.94 (m, 48H). ¹³C NMR (100 MHz, CDCl₃) δ 167.2, 163.2, 162.6, 162.1, 161.3,

160.9, 160.4, 159.5, 159.0, 157.5, 153.3, 150.0, 149.4, 146.6, 144.1, 136.1, 134.0, 130.7, 130.4, 128.7, 125.9, 124.8, 124.4, 121.4, 118.9, 118.1, 117.8, 116.4, 115.2, 114.6, 103.1, 102.9, 102.7, 102.5, 99.4, 98.0, 97.0, 95.0, 94.2, 66.9, 66.0, 58.4, 50.0, 47.0, 36.0, 34.5, 28.6, 28.4, 24.5. HRMS (ESI): m/z calcd for $C_{130}H_{126}F_8N_{16}O_{21}$ [M+H]⁺2400.5162, found 2400.9805.

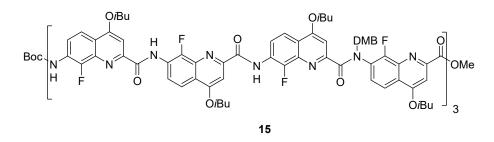


Compound **1a**: **14** (350 mg, 0.15 mmol) and triphenylphosphine (582 mg, 15 eq.) were mixed in dry chloroform (5 mL) and then trichloroacetonitrile (250 μ L, 17 eq.) and dry DIPEA (384 μ L, 15 eq.) were added. The mixture was stirred at room temperature for 48 h. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane and washed with saturated aqueous NH₄Cl, NaHCO₃ and extracted with chloroform. The organic layer was washed with brine, and dried over anhydrous MgSO₄. After the solvent was removed in vacuo, the residue was purified by column chromatography (silica gel, ethyl acetate / *n*-hexane = 1 / 2) and GPC to give **1a** as yellow solid (127 mg, 36%).¹H NMR (300 MHz, CDCl₃, 298 K) δ 11.15-10.80 (m, 2H), 10.68 (s, 2H), 10.44-10.23 (m, 2H), 8.97-8.46 (m, 4H), 8.37-8.26 (m, 2H), 8.17-8.02 (m, 3H), 7.95-7.40 (m, 11H), 7.30 (s, 4H), 6.59 (br, 1H), 6.40 (d, *J*(H, H) =7.9, 2H), 6.22-6.18 (m, 3H), 5.64 (d, *J*(H, H) =14.2, 2H), 5.03 (d, *J*(H, H) =14.2, 2H), 4.17-3.96 (m, 12H), 3.85 (s, 4H), 3.68 (s, 6H), 3.47-3.33 (m, 6H), 2.30-2.07 (m, 8H), 1.24-1.05 (m, 24H), 1.02-0.94 (m, 24H). HRMS (ESI): m/z calcd for C₁₃₀H₁₂₅F₈N₁₆O₂₀ [M+2H]²⁺ 1191.9615, found 1191.8117.

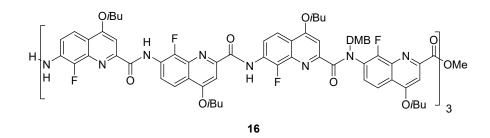


Compound **1b**: Trifluoroacetic acid (5 mL) was added to **1a** (100.0 mg, 0.04 mmol) and the mixture was stirred at 60°C for 2 h. The reaction mixture was quenched by adding sat. NaHCO₃ aq. and extracted with chloroform. The organic layer was washed with brine, and dried over MgSO₄. After the solvent was removed in vacuo, the residue

was purified by GPC to give **1b** as yellow solid (87.4 mg, quant.). ¹H NMR (300 MHz, CDCl₃, 298 K) δ 11.37-9.92 (m, 8H), 9.17-8.73 (m, 4H), 8.51-8.06 (m, 6H), 7.97-7.32 (m, 10H), 7.06-6.64 (m, 4H), 4.38-3.69 (m, 16H), 1.37-1.04 (m, 48H). HRMS (ESI): m/z calcd for C₁₁₂H₁₀₄F₈N₁₆O₁₆ [3M+3H]³⁺ 2082.4450, found 2082.4993.

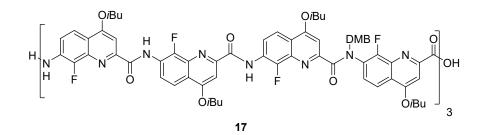


Compound **15**: A solution of **11** (107 mg, 0.08 mmol) in dry CHCl₃ (1 mL) was cooled to 0°C. 1-Chloro-N,N,2-trimethyl-1-propenylamine (0.04 mL, 3.0 eq.) was added. The solution was stirred at room temperature for 3 h. The solvent was removed under high-vacuum. The resulting acid chloride was dissolved in dry CHCl₃ (3 mL) and added to a solution of **13** (200 mg, 1.0 eq.) and dry DIPEA (0.05 mL, 3.0 eq.) in dry CHCl₃ (2 mL). The mixture was stirred for 12 h. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane and washed with saturated aqueous NH₄Cl, saturated aqueous NaHCO₃ and brine, dried over MgSO₄, filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica gel, ethyl acetate / cyclohexane = 1 / 3) to give **15** as yellow solid (162 mg, 54%.). ¹H NMR δ (400 MHz, CDCl₃, 298 K) δ 10.96-10.40 (m, 8H), 9.01-8.51 (m, 8H), 8.10-7.48 (m, 28H), 6.48-6.24 (m, 9H), 5.50-4.96 (m, 6H), 4.24-3.83 (m, 27H), 3.73-3.67 (m, 9H), 3.52-3.41 (m, 9H), 2.43-2.11 (m, 12H), 1.32-1.00 (m, 81H). ¹³C NMR (100 MHz, CDCl₃) δ 167.6, 167.1, 164.8, 162.0, 161.4, 160.8, 159.2, 157.4, 153.5, 151.3, 150.0, 148.1, 147.3, 146.7, 145.5, 145.0, 144.2, 137.7, 136.5, 130.5, 128.8, 125.7, 125.4, 124.5, 121.3, 120.1, 119.2, 118.7, 118.2, 117.4, 116.4, 116.0, 115.2, 103.0, 100.5, 99.6, 97.0, 96.6, 80.3, 79.9, 74.3, 74.0, 54.1, 53.8, 46.8, 27.1, 182. HRMS (ESI): m/z calcd for C₂₀₁H₁₉₉F₁₂N₂₄O₃₃ [M+H]⁺ 3706.4501, found 3706.4991.

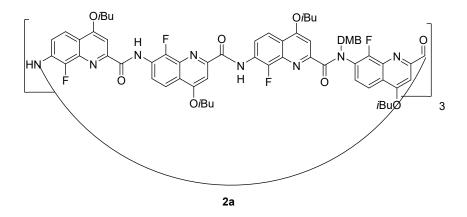


Compound 16: 15 (80 mg, 0.021 mmol) was dissolved in dioxane (0.1 mL). HCl in dioxane (4M, 1mL) was added to the solution, and stirred at room temperature for 4 h. The mixture was diluted with DCM and washed with saturated aqueous NaHCO₃ and brine, the organic layer was dried over MgSO₄, filtered and concentrated in vacuo to give 16 as yellow solid (75 mg, 98%). ¹H NMR δ (400 MHz, CDCl₃, 298 K) 11.07-10.55 (m, 8H), 8.93-

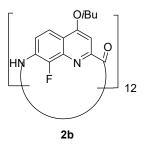
8.51 (m, 8H), 8.17-7.40 (m, 26H), 7.22-7.09 (m, 2H), 6.50-6.22 (m, 9H), 5.50-5.03 (m, 6H), 4.18-3.45 (m, 45H), 2.45-2.17 (m, 12H), 1.29-0.83 (m, 72H).



Compound **17**: **16** (80 mg, 0.021 mmol) was dissolved in THF (1.0 mL). LiOH (5.09 mg, 0.121 mL) was added to the solution, followed with H_2O (0.1 mL) and stirred at room temperature for 4 h. The organic layer was washed with citric acid (5%, 25 mL) and brine, dried over MgSO₄, filtered and concentrated in vacuo to give **17** as yellow solid (78 mg, 98%).¹H NMR (400 MHz, CDCl₃, 298 K) δ 10.73-10.48 (m, 8H), 8.70-8.39 (m, 8H), 8.00-7.39 (m, 26H), 6.32-6.13 (m, 9H), 5.37-5.03 (m, 6H), 4.21-3.23 (m, 42H), 2.42-2.19 (m, 12H), 1.03-0.78 (m, 72H).



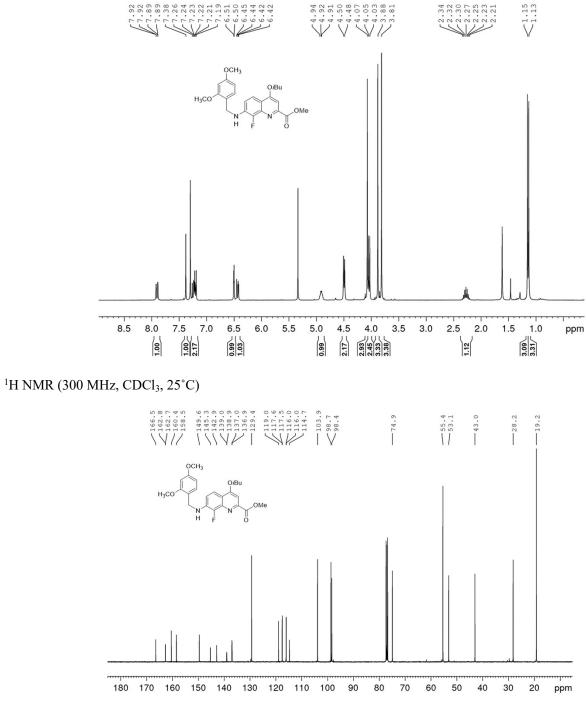
Compound **2a**: **17** (85 mg, 0.023 mmol) and triphenylphosphine (93.4 mg, 15 eq.) were mixed in dry chloroform (3 mL) and then trichloroacetonitrile (47 μ L, 17 eq.) and dry DIPEA (61 μ L, 15 eq.) were added. The mixture was stirred at room temperature for 48 h under argon. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane and washed with saturated aqueous NH₄Cl, NaHCO₃ and extracted with chloroform. The organic layer was washed with brine, and dried over anhydrous MgSO₄. After the solvent was removed in vacuo, the residue was purified by column chromatography (silica gel, ethyl acetate / n-hexane = 1 / 2) and GPC to give **2a** as yellow solid (52 mg, 61%).¹H NMR (400 MHz, CDCl₃, 298 K) δ 11.01-10.46 (m, 9H), 9.03-8.37 (m, 9H), 8.20-7.34 (m, 24H), 7.20-7.01 (m, 4H), 6.44-6.12 (m, 8H), 5.09 (br, 6H), 4.21-3.94 (m, 20H), 3.81-3.53 (m, 12H), 3.46-3.22 (m, 10H), 2.42-2.04 (m, 12H), 1.16-0.84 (m, 72H).



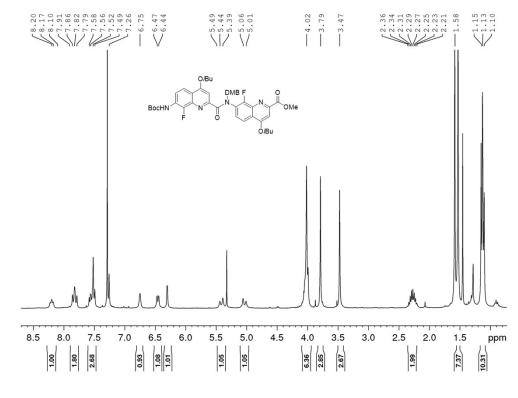
Compound **2b**: Trifluoroacetic acid (5 mL) was added to **2a** (56.0 mg, 0.015 mmol) and the mixture was stirred at 60°C for 2 h. The reaction mixture was quenched by adding sat. NaHCO₃ aq. and extracted with chloroform. The organic layer was combined and washed with brine, dried over MgSO₄. After the solvent was removed in vacuo, the residue was purified by GPC to give **2b** as yellow solid (48.2 mg, quant.). ¹H NMR (400 MHz, CDCl₃, 298 K) δ 12.01 (s, 1H) 11.87 (s, 1H), 11.74 (s, 1H), 11.57 (s, 1H), 11.50 (s, 1H), 11.48 (s, 1H), 11.44 (s, 1H), 11.36 (s, 2H), 10.50 (s, 1H), 10.37 (s, 1H), 9.24 (t, *J*(H, H) =7.2, 1H), 9.11 (s, 1H), 9.07 (t, *J*(H, H) =7.8, 1H), 9.00-8.89 (m, 3H), 8.81 (t, *J*(H, H) =7.6, 1H), 8.63 (t, *J*(H, H) =7.3, 1H), 8.46 (t, *J*(H, H) =7.2, 1H), 8.20 (q, *J*(H, H) =7.8, 2H), 8.14 (t, *J*(H, H) =7.2, 1H), 8.09 (s, 1H), 8.07-8.01 (m, 1H), 7.87 (d, *J*(H, H) =9.0, 1H), 7.80-7.71 (m, 4H), 7.65 (s, 1H), 7.51 (s, 1H), 7.47 (d, *J*(H, H) =4.6, 2H), 6.41 (s, 1H), 6.18 (s, 1H), 4.35 (br, 2H), 4.20-4.14 (m, 5H), 4.12-4.03 (m, 2H), 3.92-3.86 (m, 3H), 3.81-3.72 (m, 5H), 3.70-3.57 (m, 5H), 3.40 (t, *J*(H, H) =8.1, 1H), 3.03 (t, *J*(H, H) =8.1, 1H), 2.42-2.03 (m, 12H), 1.31-1.00 (m, 72H). HRMS (ESI): m/z calcd for C₁₆₈H₁₅₆F₁₂N₂₄O₂₄ [M+2H]²⁺1562.0856, found 1562.0332, [2M+3H]³⁺: 2082.7795, found 2082.7056 .

6. ¹H and ¹³C NMR spectra

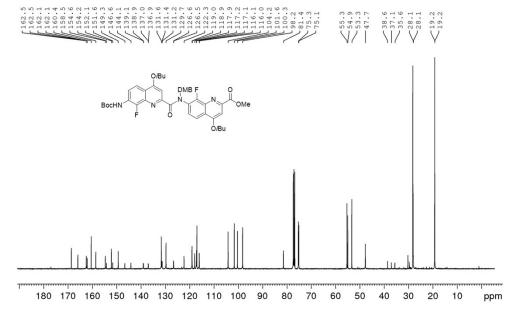
Compound 5



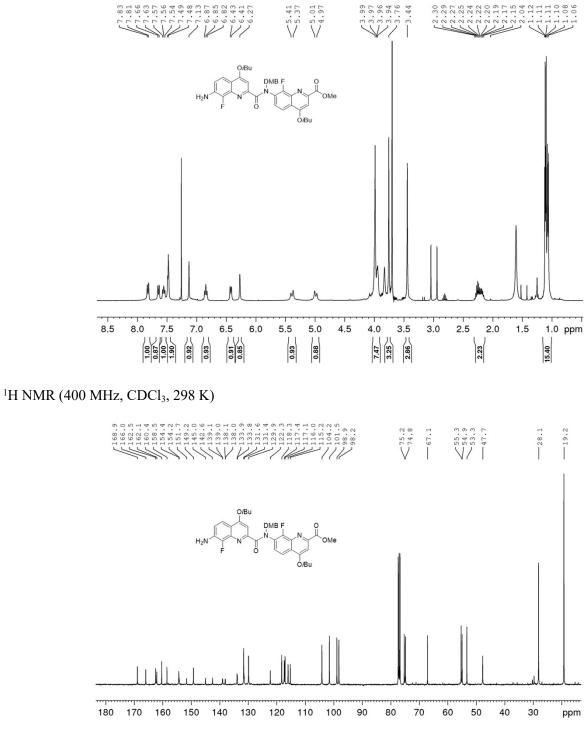
¹³C NMR (100 MHz, CDCl₃, 25°C)



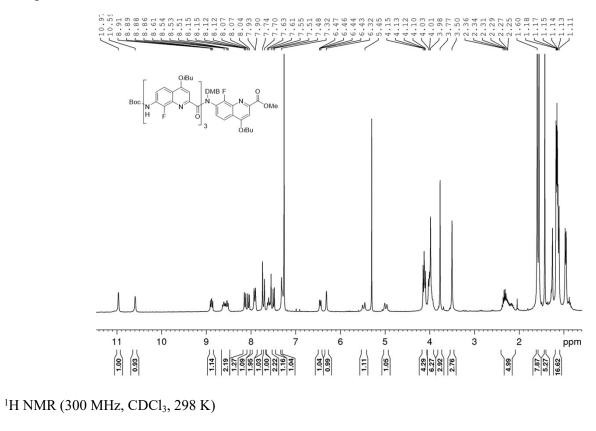
¹H NMR (300 MHz, CDCl₃, 25°C)

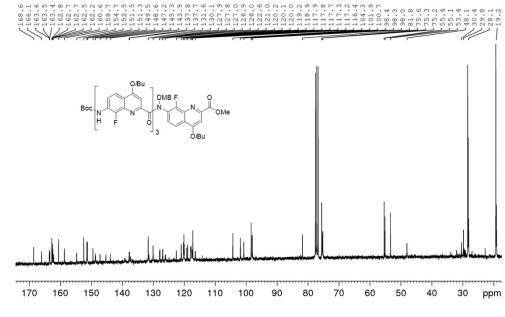


¹³C NMR (100 MHz, CDCl₃, 25°C)

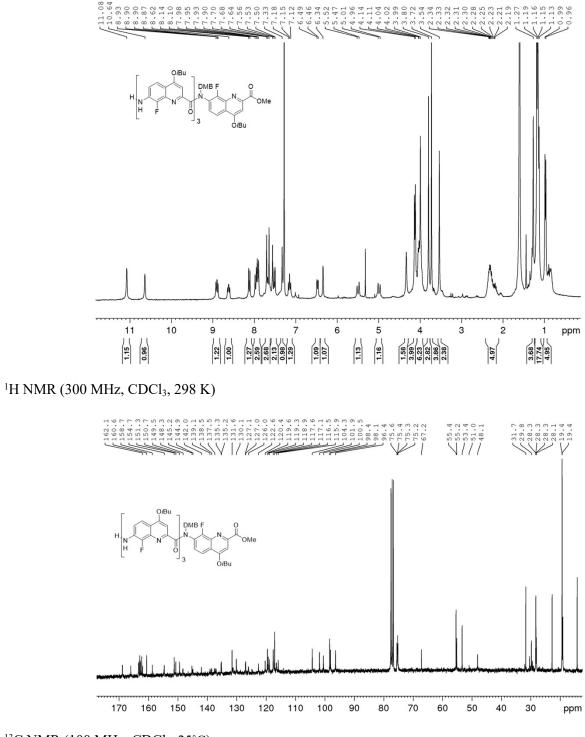


¹³C NMR (100 MHz, CDCl₃, 25°C)

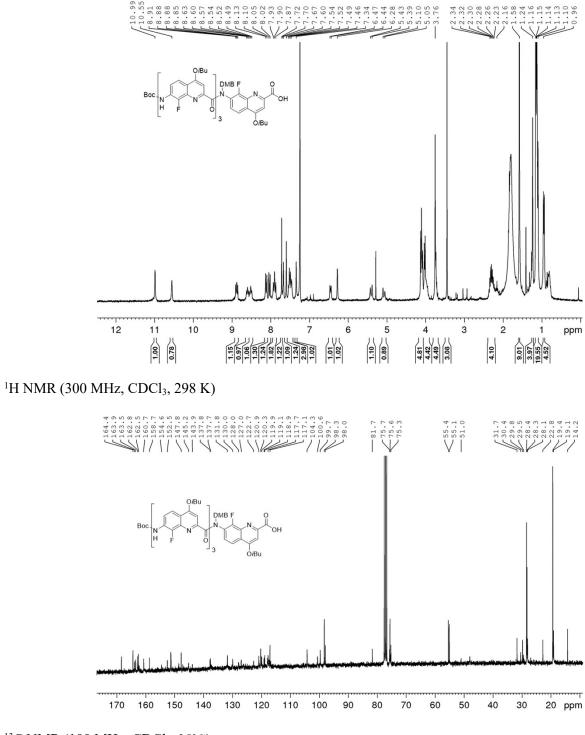




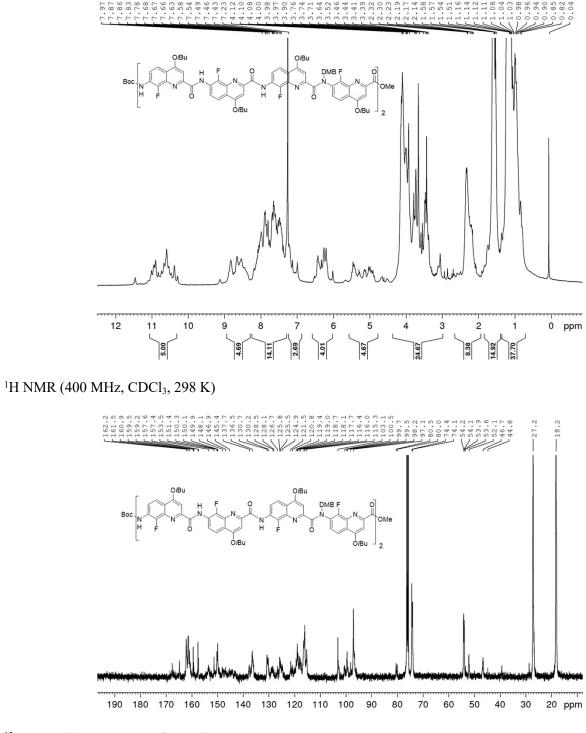
¹³C NMR (100 MHz, CDCl₃, 25°C)



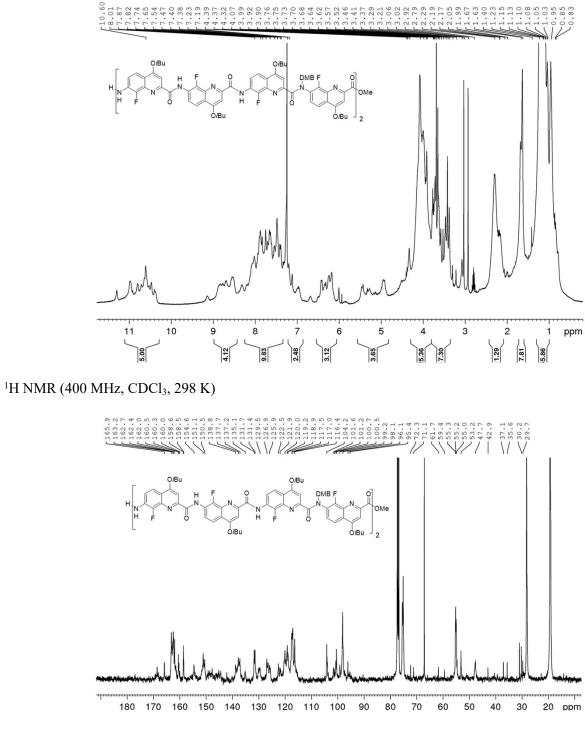
¹³C NMR (100 MHz, CDCl₃, 25°C)



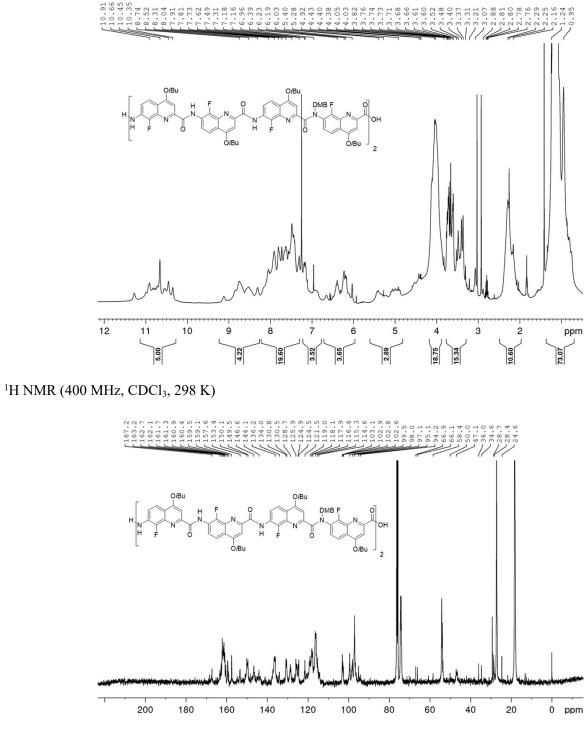
¹³C NMR (100 MHz, CDCl₃, 25°C)



¹³C NMR (100 MHz, CDCl₃, 25°C)

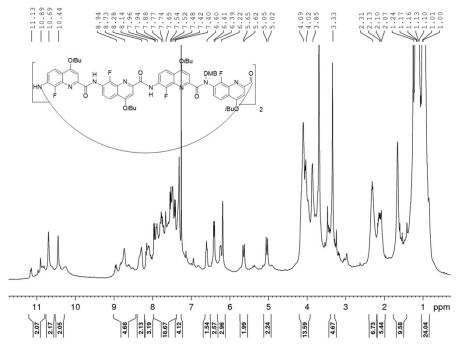


¹³C NMR (100 MHz, CDCl₃, 25°C)



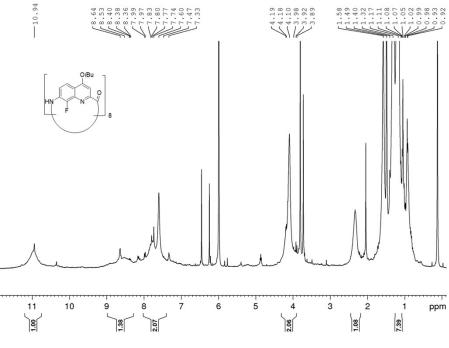
¹³C NMR (100 MHz, CDCl₃, 25°C)

Compound 1a

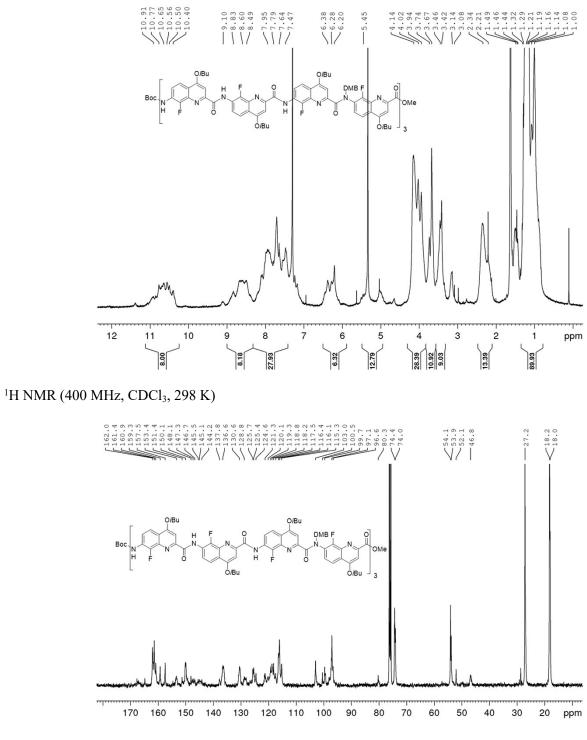


¹H NMR (300 MHz, CDCl₃, 298 K)

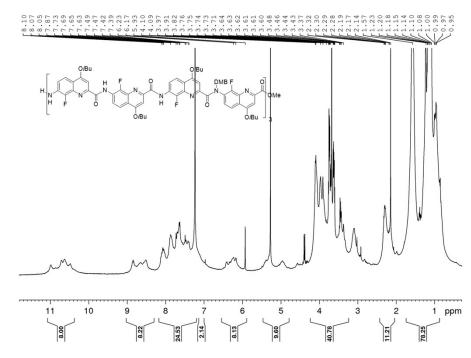
Compound 1b



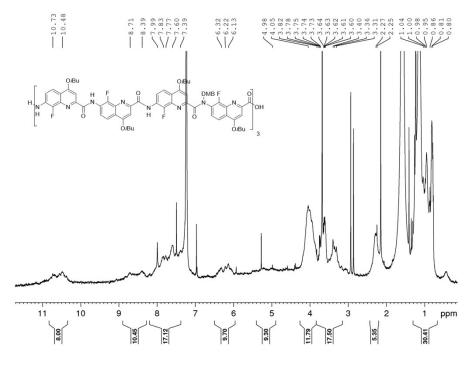
¹H NMR (300 MHz, C₂D₂Cl₄, 353 K)



¹³C NMR (100 MHz, CDCl₃, 25°C)

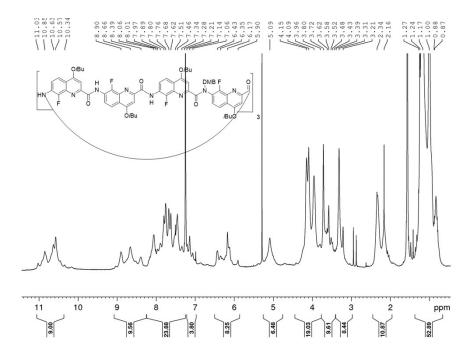


¹H NMR (400 MHz, CDCl₃, 298 K)



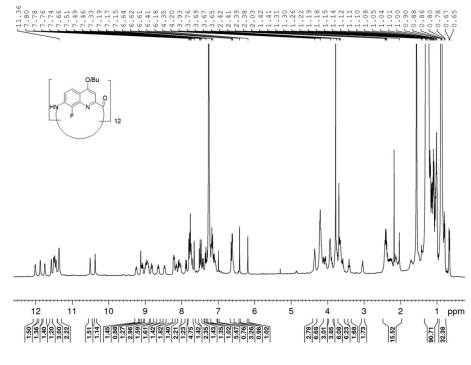
¹H NMR (400 MHz, CDCl₃, 298 K)

Compound 2a



¹H NMR (400 MHz, CDCl₃, 298 K)

Compound **2b**



¹H NMR (400 MHz, CDCl₃, 298 K)

7. Reference

- [1] CrysAlisPRO: CrysAlisPRO, Oxford Diffraction / Agilent Technologies UK Ltd, Yarnton, England.
- [2] Sheldrick, G. M. Acta Cryst. 2015, A71, 3-8.
- [3] OLEX2: O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. Puschmann. J. Appl. Cryst. 2009, 42, 339–341.
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- [5] Gan, Q.; Bao, C.; Kauffmann, B.; Grélard, A.; Xiang, J.; Liu, S.; Huc, I.; Jiang, H. Angew. Chem., Int. Ed. 2008, 47, 1715–1718.