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Supplemental Information

Ruthenium(II)-Catalyzed Deoxygenation of Ketones

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I. General Methods

All reagents and solvents were purchased from commercial sources (Sigma-Aldrich) and used without further purification unless otherwise stated. Catalysts Ru-2, Ru-3, and Ru-5 were synthesized by mixing [(Ph₃P)₃RuCl₂] with one equiv of ligand. The procedure was referred to that of Ru-PNX synthesis¹ but optimized to 75°C temperature and overnight reaction time. All reactions were monitored by thin-layer chromatography (TLC). All reactions were carried out under a nitrogen atmosphere unless otherwise stated. Column chromatography was performed on silica gel (200-300 mesh) and visualized with ultraviolet light. Ethyl acetate and hexane were used as eluents.¹H, and ¹³C NMR spectra were taken on Bruker AV300, Bruker AV400, Varian Mercury 400, and Varian/Agilent QANUC 500 with TMS as an internal standard and CDCl₃ as solvent unless otherwise stated. GC-MS analyses were performed with a Thermo TRACE 1300 ISQ LT spectrometer.

II. Mechanism

Scheme 1. Proposed mechanism for the Ru-catalyzed ketone deoxygenation



III. Optimization of Reaction Conditions

1) Table S1. Screening ligands^a

0 	N₂H₄∙H₂O [Ru(p-cymene)Cl₂]₂ Ligand	н_н
Et 1aa	DMSO, KO ^f Bu H ₂ O, 75 °C, overnight	Ph Et 2aa
entry	ligand	yield(%) ^b
1	PPh ₃	6
2	1,2-Bis(dimethylphosphino)ethane	3
3	1,2-Bis(diphenylphosphino)ethane	10
4	1,3-Bis(diphenylphosphino)propane	12
5	1,4-Bis(diphenylphosphino)butane	10
6	1,6-Bis(diphenylphosphino)hexane	13
7	1,4-Bis(dicyclohexylphosphino)butane	14
	O Et 1aa entry 1 2 3 4 5 6 7	$\begin{array}{c} & \begin{array}{c} N_2H_4 \cdot H_2O \\ [Ru(p-cymene)Cl_2]_2 \\ Ligand \\ \end{array} \\ \hline \begin{array}{c} DMSO, KO^{f}Bu \\ H_2O, 75 \ ^{\circ}C, \ overnight \end{array} \\ \hline \begin{array}{c} entry \\ \end{array} \\ \begin{array}{c} ligand \\ \end{array} \\ \hline \begin{array}{c} 1 \\ 2 \\ 1,2 \cdot Bis(dimethylphosphino)ethane \\ 3 \\ 1,2 \cdot Bis(diphenylphosphino)ethane \\ 4 \\ 1,3 \cdot Bis(diphenylphosphino)propane \\ 5 \\ 1,4 \cdot Bis(diphenylphosphino)butane \\ 6 \\ 1,6 \cdot Bis(dicyclohexylphosphino)butane \\ 7 \\ \end{array} \\ \begin{array}{c} NSO, KO^{f}Bu \\ H_2O, 75 \ ^{\circ}C, \ overnight \\ \end{array} \\ \hline \end{array} \\ \end{array}$

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N₂H₄·H₂O (12 μL, 0.24 mmol, 1.2 equiv), H₂O (0.2 mL), [Ru(p-cymene)Cl₂]₂ (1.6 mg, 0.003 mmol, 1.5 mol%), ligand (0.006 mmol, 3 mol%), KO^tBu (11.2 mg, 0.1 mmol, 0.5 equiv), additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), 75°C, overnight, under N₂. ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

2) Table S2. Screening bases^a

	o ↓ -	N ₂ H ₄ •H ₂ O [Ru(p-cymene)Cl ₂] ₂ 1,6-Bis(diphenylphosphino)hexane	H H
Ph	Et 1aa	DMSO, base H ₂ O, 75 °C, overnight	Ph´ Et 2aa
_	entry	base	yield(%) ^b
	1	КОН	4
	2	DBU	5
	3	K ₃ PO ₄	4
	4	TMP	4
	5	1,8-Bis(dimethylamino)naphthalene	0
	6	Nal	0
	7	Cs ₂ CO ₃	8
	8	KO ^t Bu	19

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (12 μL, 0.24 mmol, 1.2 equiv), H_2O (0.2 mL), $[Ru(p-cymene)Cl_2]_2$ (1.6 mg, 0.003 mmol, 1.5 mol%), 1,6-bis(diphenylphosphino)hexane (2.7 mg, 0.006 mmol, 3 mol%), base (0.4 mmol, 2,0 equiv), additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), 75°C, overnight, under N_2 . ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

3) Table S3. Screening amount of bases^a

	0	N ₂ H ₄ •H ₂ O [Ru(p-cymene)Cl ₂] ₂ 1,6-Bis(diphenylphosphino)hexane	н_н
Ph	Et 1aa	DMSO, KO ^t Bu H ₂ O, 75 °C, overnight	Ph Et 2aa
-	entry	equivalent of KO ^f Bu	yield(%) ^b
_	1	0.5	13
	2	1.0	11
	3	2.0	19
	4	3.0	14

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (12 μL, 0.24 mmol, 1.2 equiv), H_2O (0.2 mL), $[Ru(p-cymene)Cl_2]_2$ (1.6 mg, 0.003 mmol, 1.5 mol%), 1,6-bis(diphenylphosphino)hexane (2.7 mg, 0.006 mmol, 3 mol%), KO¹Bu, additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), 75°C, overnight, under N_2 . ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

4) Table S4. Screening concentration^a



^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N₂H₄·H₂O (12 μL, 0.24 mmol, 1.2 equiv), H₂O, [Ru(p-cymene)Cl₂]₂ (1.6 mg, 0.003 mmol, 1.5 mol%), 1,6-bis(diphenylphosphino)hexane (2.7 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), 75°C, overnight, under N₂. ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

5) Table S5. Screening temperature^a

	O	N₂H₄•H₂O [Ru(p-cymene)Cl₂]₂ 1,6-Bis(diphenylphosphino)hexane	н
Pŀ	Et 1aa	DMSO, KO ^t Bu H ₂ O, 75 °C, overnight	Ph Et 2aa
	entry	temperature (°C)	yield(%) ^b
	1 2	75 100	19 7

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (12 μL, 0.24 mmol, 1.2 equiv), H_2O (0.2 mL), $[Ru(p-cymene)Cl_2]_2$ (1.6 mg, 0.003 mmol, 1.5 mol%), 1,6-bis(diphenylphosphino)hexane (2.7 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), overnight, under N_2 .^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

6) Table S6. Screening amount of additive^a

	0	N ₂ H ₄ •H ₂ O [Ru(p-cymene)Cl ₂] ₂ 1,6-Bis(diphenylphosphino)hexane	н_н
Ph	Et 1aa	DMSO, KO ^t Bu H ₂ O, 75 °C, overnight	Ph Et 2aa
-	entry	equivalent of DMSO	yield(%) ^b
	1	0	3
	2	0.2	19
	3	0.3	17

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (12 μL, 0.24 mmol, 1.2 equiv), H_2O (0.2 mL), $[Ru(p-cymene)Cl_2]_2$ (1.6 mg, 0.003 mmol, 1.5 mol%), 1,6-bis(diphenylphosphino)hexane (2.7 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO, 75°C, overnight, under N_2 .^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

7) Table S7. Screening additive^a

N O F		N ₂ H ₄ •H ₂ O Ru-PNN	н
Ph	Et 1aa	additive, KO ^t Bu H ₂ O, 75 °C, overnight	Ph Et 2aa
_	entry	additive	yield(%) ^b
	1	THF	14
	2	acetone	11
	3	ACN	17
	4	γ–Valerolactone	14
	5	HMPA	20
	6	DMSO	27

^a **1aa** (26.5 μL, 0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (12 μL, 0.24 mmol, 1.2 equiv), H_2O (0.2 mL), Ru-PNN (4.5 mg, 0.006 mmol, 3.0 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive (0.04 mmol, 20 mol%), 75°C, overnight, under N_2 . ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard.

IV. Other Substrates



Scheme 2. Substrate scope in methanol conditions^{a, b}

^a 1 (0.2 mmol, 1.0 equiv), N₂H₄·H₂O (50 μ L, 1.0 mmol, 5.0 equiv), methanol (0.2 mL), Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KOⁱBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO (2.6 μ L, 0.04 mmol, 20 mol%), 75°C, overnight, under N₂. ^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard. Isolated yields were given in the parentheses.

Scheme 3. Substrates not tolerated^a



^a **1** (0.2 mmol, 1.0 equiv), N_2H_4 · H_2O (50 µL, 1.0 mmol, 5.0 equiv), methanol (0.2 mL), Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO (2.6 µL, 0.04 mmol, 20 mol%), 75°C, overnight, under N_2 .

Scheme 4. Other substrates with poor yields^a



^a **1** (0.2 mmol, 1.0 equiv), N₂H₄·H₂O (50 μL, 1.0 mmol, 5.0 equiv), methanol or water (0.2 mL), Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv), additive: DMSO (2.6 μL, 0.04 mmol, 20 mol%), 75°C, overnight, under N₂.^b Yields were determined by crude ¹H NMR using mesitylene as an internal standard. ^c Yields were determined by crude GC using mesitylene as an internal standard. Isolated yields were given in the parentheses.

V. General Procedure for the Deoxygenation of Ketones General procedure with liquid ketones

A 10 mL V-shape microwave vial with a magnetic stir-bar was transferred to glovebox and charged with Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%) and KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv). The tube was capped with a rubber septum stopper and taken out from the glovebox. To the microwave vial, hydrazine monohydrate (50 μ L, 1.0 mmol, 5.0 equiv), ketone (0.2 mmol, 1.0 equiv), and DMSO (2.6 μ L, 0.04 mmol, 20 mol%) were charged via Hamilton microliter syringes. Solvent (0.2 mL) was added through a 1 mL plastic syringe. The tube was placed in a preheated oil bath at 75 °C and the mixture was stirred under an argon atmosphere overnight. The reaction mixture was cooled to room temperature and charged with mesitylene (9.2 μ L, 0.067 mmol, 0.33 equiv). The solution was diluted with

diethyl ether filtered through anhydrous MgSO₄ followed by silica gel, concentrated, and then purified by column chromatography on silica gel eluting with pentane: Et_2O (100:1-10:1). The solvent was frozen in an ice-water bath and gently evacuated by vacuum to afford the products.

General procedure with solid ketones

A 10 mL V-shape microwave vial with a magnetic stir-bar was transferred to glovebox and charged with Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KO^tBu (44.8 mg, 0.4 mmol, 2.0 equiv) and ketone (0.2 mmol, 1.0 equiv). The tube was capped with a rubber septum stopper and taken out from the glovebox. To the microwave vial, hydrazine monohydrate (50 μ L, 1.0 mmol, 5.0 equiv), and DMSO (2.6 μ L, 0.04 mmol, 20 mol%) were charged via Hamilton microliter syringes. Solvent (0.2 mL) was added through a 1 mL plastic syringe. The tube was placed in a preheated oil bath at 75 °C and the mixture was stirred under an argon atmosphere overnight. The reaction mixture was cooled to room temperature and charged with mesitylene (9.2 μ L, 0.067 mmol, 0.33 equiv). The solution was diluted with diethyl ether filtered through anhydrous MgSO₄ followed by silica gel, concentrated, and then purified by column chromatography on silica gel eluting with pentane: Et₂O (100:1-10:1). The solvent was frozen in an ice-water bath and gently evacuated by vacuum to afford the products.

General procedure with hydroxyl, amino and carboxylic acid substituted ketones

A 10 mL V-shape microwave vial with a magnetic stir-bar was transferred to glovebox and charged with Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KO^tBu (67.2 mg, 0.6 mmol, 3.0 equiv) and ketone (0.2 mmol, 1.0 equiv). The tube was capped with a rubber septum stopper and taken out from the glovebox. To the microwave vial, hydrazine monohydrate (50 μ L, 1.0 mmol, 5.0 equiv), and DMSO (2.6 μ L, 0.04 mmol, 20 mol%) were charged via Hamilton microliter syringes. Water (0.2 mL) was added through a 1 mL plastic syringe. The tube was placed in a preheated oil bath at 75 °C and the mixture was stirred under an argon atmosphere overnight. The reaction mixture was cooled to room temperature and charged with mesitylene (9.2 μ L, 0.067 mmol, 0.33 equiv). The solution was diluted with diethyl ether, filtered through anhydrous MgSO₄ followed by silica gel, concentrated, and then purified by column chromatography on silica gel eluting with pentane: Et₂O (4:1-1:1). The solvent was frozen in an ice-water bath and gently evacuated by vacuum to afford the products.

General procedure with sodium carboxylate substituted ketones

A 10 mL V-shape microwave vial with a magnetic stir-bar was transferred to glovebox and charged with Ru-PNN (4.5 mg, 0.006 mmol, 3 mol%), KO^tBu (67.2 mg, 0.6 mmol, 3.0 equiv) and ketone (0.2 mmol, 1.0 equiv). The tube was capped with a rubber septum stopper and taken out from the glovebox. To the microwave vial, hydrazine monohydrate (50 μ L, 1.0 mmol, 5.0 equiv), and DMSO (2.6 μ L, 0.04 mmol, 20 mol%) were charged via Hamilton microliter syringes. Water (0.2 mL) was added through a 1 mL plastic syringe. The tube was placed in a preheated oil bath at 75 °C and the mixture was stirred under an argon atmosphere overnight. The reaction mixture was cooled to room temperature and charged

with mesitylene (9.2 μ L, 0.067 mmol, 0.33 equiv). Concentrate hydrochloric acid was added dropwise until the solution became acidic. Centrifuge the slurry to isolate the products, and extract the aqueous solution with diethyl ether. Dry the organic layer, pass through a layer of silica and evacuate the solvent to afford the remaining products.

VI. Characterization of Products

Hexylbenzene² (2ab)

Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.31 – 7.24 (m, 2H), 7.18 (d, *J* = 7.5 Hz, 3H), 2.64 – 2.57 (m, 2H), 1.66 – 1.58 (m, 2H), 1.39 – 1.25 (m, 6H), 0.92 – 0.85 (m, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 143.0, 128.4, 128.2, 125.5, 36.0, 31.7, 31.5, 29.0, 22.6, 14.1.



1,2-diphenylethane³ (2b)

White solid. ¹H NMR (400 MHz, CDCl₃) δ 7.28 (ddd, J = 9.1, 6.3, 0.9 Hz, 4H), 7.25 – 7.13 (m, 6H), 2.93 (s, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 141.8, 128.4, 128.3, 125.9, 37.9.

1,2,3,4-tetrahydronaphthalene⁴ (2c)

Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.09 (h, J = 4.6, 4.2 Hz, 4H), 2.80 (h, J = 3.1 Hz, 4H), 1.87 – 1.78 (m, J = 4.3 Hz, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 137.1, 129.1, 125.4, 29.4, 23.2.

(4-ethylphenyl)(methyl)sulfane (2db)

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.24 – 7.18 (m, 2H), 7.16 – 7.10 (m, 2H), 2.61 (q, J = 7.6 Hz, 2H), 2.47 (s, 3H), 1.22 (t, J = 7.6 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 141.5, 134.9, 128.4, 127.3, 28.3, 16.5, 15.6. IR wavenumber (cm⁻¹) 2963.7, 2921.6, 2872.2, 1493.6, 1121.9, 816.8. HRMS calc. for C₉H₁₂OSNa [M+Na+O]+: 191.0501; found, 191.0493.



1-ethyl-4-(methylsulfonyl)benzene

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.89 – 7.81 (m, 2H), 7.42 – 7.35 (m, 2H), 3.04 (s, 3H), 2.75 (q, *J* = 7.6 Hz, 2H), 1.27 (t, *J* = 7.6 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 150.8, 137.9, 128.8, 127.5, 44.6, 28.9, 15.1. IR wavenumber (cm⁻¹) 2968.5, 2930.1, 2875.3, 1300.0, 1144.5, 547.3, 517.8. HRMS calc. for C₉H₁₂O₂SNa [M+Na]+: 207.0450; found, 207.0451.

1-methoxy-4-propylbenzene⁵ MeO



Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.13 – 7.05 (m, 2H), 6.87 – 6.78 (m, 2H), 3.79 (s, 3H), 2.53 (dd, J = 8.5, 6.7 Hz, 2H), 1.68 – 1.54 (m, 2H), 0.93 (t, J = 7.3 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 157.61, 134.81, 129.29, 113.61, 55.24, 37.13, 24.78, 13.77.



Diphenylmethane⁶ (2fa)

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.33 – 7.25 (m, 4H), 7.24 – 7.13 (m, 6H), 3.99 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 141.1, 128.9, 128.4, 126.1, 41.9.

1-benzyl-4-methylbenzene⁷ (2fb)

Brown liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.34 – 7.25 (m, 2H), 7.20 (ddt, *J* = 7.4, 3.1, 1.9 Hz, 3H), 7.15 – 7.06 (m, 4H), 3.96 (s, 2H), 2.33 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 141.4, 138.1, 135.5, 129.1, 128.9, 128.8, 128.4, 126.0, 41.5, 21.0.



Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.32 – 7.24 (m, 2H), 7.28 – 7.07 (m, 7H), 4.00 (s, 2H), 2.25 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 140.4, 138.9, 136.6, 130.3, 129.9, 128.7, 128.4, 126.4, 126.0, 125.9, 39.4, 19.7.



1-benzyl-3,5-dibutoxybenzene (2fd)

Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.29 – 7.13 (m, 5H), 6.97 (d, J = 8.2 Hz, 1H), 6.45 (d, J = 2.4 Hz, 1H), 6.40 (dd, J = 8.3, 2.4 Hz, 1H), 3.97 – 3.89 (m, 6H), 1.75 (dddd, J = 15.3, 12.2, 7.6, 6.3 Hz, 4H), 1.55 – 1.38 (m, 4H), 0.97 (dt, J = 18.4, 7.4 Hz, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 158.9, 157.7, 141.7, 130.4, 128.8, 128.1, 125.5, 122.1, 104.4, 99.7, 67.7, 67.6, 35.5, 31.4, 31.3, 19.3, 13.9, 13.8. IR wavenumber (cm⁻¹) 3060.7, 3026.8, 2957.2, 2932.0, 2870.4, 1300.0, 1173.4. HRMS calc. for C₂₁H₂₈O₂Na [M+Na]+: 335.1982; found, 335.1978.

va)

1-methoxy-3-(4-methoxybenzyl)benzene (2ga)

Clear yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.20 (td, J = 7.7, 0.8 Hz, 1H), 7.15 – 7.07 (m, 2H), 6.87 – 6.80 (m, 2H), 6.80 – 6.74 (m, 1H), 6.77 – 6.69 (m, 2H), 3.90 (s, 2H), 3.78 (s, 3H), 3.77 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.7, 158.0, 143.2, 133.0, 129.8, 129.4, 121.2, 114.6, 113.9, 111.2, 55.3, 55.1, 41.0. IR wavenumber (cm⁻¹) 3029.4, 2998.3, 2954.8, 2933.2, 2834.1.4, 1280.3. HRMS calc. for C₁₅H₁₆NaO₂ [M+Na]+: 251.1043; found, 251.1034.

1-methoxy-4-(4-methylbenzyl)benzene⁹ (2gb)

Clear yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.14 – 7.03 (m, 6H), 6.90 – 6.79 (m, 2H), 3.89 (s, 2H), 3.78 (s, 3H), 2.32 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 157.9, 138.5, 135.4, 133.5, 129.8, 129.1, 128.7, 113.8, 55.2, 40.6, 21.0.



(E)-1-benzyl-4-styrylbenzene (2h)

White solid. ¹H NMR (400 MHz, CDCl₃) δ 7.54 – 7.47 (m, 2H), 7.47 – 7.41 (m, 2H), 7.35 (dd, J = 8.4, 6.9 Hz, 2H), 7.35 – 7.26 (m, 2H), 7.29 – 7.19 (m, 2H), 7.20 (td, J = 6.1, 1.8 Hz, 4H), 7.08 (d, J = 1.6 Hz, 2H), 4.00 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 141.0, 140.7, 137.4, 135.3, 129.3, 128.9, 128.6, 128.5, 128.1, 127.5, 126.6, 126.4, 126.1, 41.7. IR wavenumber (cm⁻¹) 3079.0, 3053.5, 3022.3, 2921.0, 2852.5, 1491.5, 1448.5, 1416.8. HRMS calc. for C₂₁H₁₉ [M+H]+: 271.14813; found, 271.14840.



1-benzyl-4-(dec-1-yn-1-yl)benzene (2i)

Clear yellow liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.33 – 7.26 (m, 4H), 7.25 – 7.13 (m, 3H), 7.13 – 7.07 (m, 2H), 3.96 (s, 2H), 2.38 (t, *J* = 7.1 Hz, 2H), 1.43 (p, *J* = 7.0 Hz, 2H), 1.37 – 1.21 (m, 10H), 0.97 – 0.77 (m, 3H).

¹³C NMR (126 MHz, CDCl₃) δ 140.7, 140.5, 131.6, 128.9, 128.8, 128.5, 126.1, 121.8, 90.1, 80.4, 41.7, 31.8, 29.2, 29.1, 28.9, 28.8, 22.7, 19.4, 14.1. IR wavenumber (cm⁻¹) 3027.0, 2953.9, 2924.0, 2854.0, 1508.1, 1494.4, 696.6. HRMS calc. for $C_{23}H_{29}$ [M+H]+: 305.22638; found, 305.22638.



1-benzyl-3,5-difluorobenzene (2j)

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.36 – 7.13 (m, 5H), 6.71 (t, *J* = 1.9 Hz, 1H), 6.69 (t, *J* = 2.0 Hz, 1H), 6.64 (tt, *J* = 9.0, 2.4 Hz, 1H), 3.95 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 163.0 (dd, *J* = 12.8, 248.0 Hz), 145.0 (t, *J* = 9.0 Hz), 139.4, 128.9, 128.7, 126.6, 111.6 (dd, *J* = 6.3, 12.6 Hz), 101.6 (t, *J* = 25.5 Hz), 41.6 (t, *J* = 2.6 Hz). ¹⁹F NMR (376 MHz, CDCl₃) δ -110.4 – -110.5 (m, *J*=1.9 Hz, 7.5 Hz). IR wavenumber (cm⁻¹) 3087.8, 3063.5, 3029.2, 2919.3, 1737.5, 1623.2, 1115.5, 971.9, 700.1. HRMS calc. for C₁₃H₉F₂ [M-H]+: 203.06778; found, 203.06705.



2-(4-methoxybenzyl)benzofuran (2k)

Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.50 – 7.44 (m, 1H), 7.41 (dq, J = 8.1, 0.9 Hz, 1H), 7.28 – 7.14 (m, 4H), 6.91 – 6.84 (m, 2H), 6.35 (q, J = 1.0 Hz, 1H), 4.06 (d, J = 1.0 Hz, 2H), 3.81 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 158.5, 158.3, 154.9, 129.9, 129.2, 128.8, 123.3, 122.5, 120.4, 114.0, 110.9, 103.1, 55.3, 34.1. IR wavenumber (cm⁻¹) 3033.4, 2997.9, 2954.4, 2933.2, 2905.5, 2834.4, 1510.2, 1300.5. HRMS calc. for C₁₆H₁₄O₂Na [M+Na]+: 261.08860; found, 261.08815.



Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 8.55 (dt, J = 4.6, 1.7 Hz, 1H), 7.58 (td, J = 7.7, 1.9 Hz, 1H), 7.34 – 7.25 (m, 4H), 7.25 – 7.18 (m, 1H), 7.11 (dd, J = 7.5, 4.6 Hz, 2H), 4.17 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 161.0, 149.3, 139.5, 136.6, 129.1, 128.6, 126.4, 123.1, 121.2, 44.7.



White solid. ¹H NMR (400 MHz, CDCl₃) δ 7.31 – 7.24 (m, 4H), 7.19 (tt, *J* = 6.4, 1.2 Hz, 6H), 7.10 (s, 4H), 3.95 (s, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 141.2, 138.8, 129.0, 128.9, 128.4, 126.0, 41.5.



1,3-dibenzylbenzene

1-benzyl-4-fluorobenzene¹³ (2m) F

White cloudy oil. ¹H NMR (400 MHz, CDCl₃) δ 7.28 (dd, J = 8.0, 6.6 Hz, 4H), 7.24 – 7.15 (m, 7H), 7.11 – 6.99 (m, 3H), 3.95 (s, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 141.2, 141.1, 129.6, 128.9, 128.6, 128.4, 126.7, 126.0, 41.9. IR wavenumber (cm⁻¹) 3059.4, 3023.8, 2960.7, 2912.2, 2899.5, 2833.3, 722.5, 696.0. HRMS calc. for C₂₀H₁₇ [M-H]+: 257.13248; found, 257.13182.

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.33 – 7.27 (m, 2H), 7.23 – 7.20 (m, 1H), 7.20 – 7.12 (m, 4H), 7.01 – 6.91 (m, 2H), 3.95 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 161.4 (d, *J* = 263.9 Hz) 140.9 (d, *J* = 1.0 Hz), 136.7 (d, *J* = 3.2 Hz) 130.3 (d, *J* = 7.8 Hz) 128.8, 128.5, 126.2, 115.27 (d, *J* = 21.2 Hz), 41.1. ¹⁹F NMR (471 MHz, CDCl₃) δ -117.4 (t, *J* = 6.8 Hz).



Clear colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.15 – 7.08 (m, 4H), 7.02 – 6.93 (m, 4H), 3.92 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 161.5 (d, J = 244 Hz), 136.6 (d, J = 3.8 Hz), 130.2 (d, J = 7.6 Hz), 115.3 (d, J = 21.4 Hz), 40.2. ¹⁹F NMR (471 MHz, CDCl₃) δ -110.5 (d, J = 8.4 Hz), -117.2 (dt, J = 8.5, 5.1 Hz).



4-benzyl-1,2-difluorobenzene (20) ^F

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.35 – 7.27 (m, 2H), 7.26 – 7.20 (m, 1H), 7.19 – 7.12 (m, 2H), 7.06 (dt, *J* = 10.3, 8.3 Hz, 1H), 6.96 (ddd, *J* = 11.3, 7.6, 2.2 Hz, 1H), 6.92 – 6.86 (m, 1H), 3.93 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 150.2 (dd, *J* = 75.6, 248.0 Hz), 149.5 (dd, *J* = 75.6, 248.0 Hz), 140.1, 138.1, 128.8, 128.7, 126.5, 125.1 – 124.4 (m), 117.6 (d, *J* = 16.6 Hz), 117.0 (d, *J* = 17.2 Hz), 41.0. ¹⁹F NMR (376 MHz, CDCl₃) δ -138.2 (dddd, *J* = 21.2, 11.4, 8.2, 1.4 Hz), -141.9 – -142.1 (m, *J* = 3.8, 11.3 Hz). IR wavenumber (cm⁻¹) 3066.3, 3029.7, 2911.0, 2837.1, 751.3, 715.9. HRMS calc. for C₁₃H₉F₂ [M-H]+: 203.06778; found, 203.06690.



bis(4-chlorophenyl)methane¹⁵ (2p) CI

Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.34 – 7.23 (m, 4H), 7.15 – 7.07 (m, 4H), 3.92 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 139.0, 132.1, 130.2, 128.8, 40.5.

1-methoxy-4-(4-(trifluoromethyl)benzyl)benzene (2q) F₃C

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.53 (d, J = 8.1 Hz, 2H), 7.36 – 7.23 (m, 2H), 7.14 – 7.04 (m, 2H), 6.92 – 6.80 (m, 2H), 3.98 (s, 2H), 3.79 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 158.2, 145.7, 132.1, 129.9, 129.0, 128.5, 125.4 (q, J = 4.2, 3.6 Hz), 114.1, 55.3, 40.8. ¹⁹F NMR (376 MHz, CDCl₃) δ -62.4 (s). IR wavenumber (cm⁻¹) 3006.1, 2934.3, 2837.5, 1510.8, 1323.0, 1113.4, 1065.3. HRMS calc. for C₁₅H₁₂OF₃ [M-H]+: 265.08457; found, 265.08389.

bis(4-bromophenyl)methane (2r) Br

Pale yellow oil, solidify at room temperature.¹H NMR (500 MHz, CDCl₃) δ 7.51 – 7.34 (m, 4H), 7.09 – 6.94 (m, 4H), 3.88 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 139.4, 131.6, 130.6, 120.2, 40.6. IR wavenumber (cm⁻¹) 2937.2, 2919.1, 2851.0, 1483.6, 1066.5, 1009.3, 826.7, 805.2. HRMS calc. for C₁₃H₉Br₂ [M-H]+: 322.90655; found, 322.90871.



5-benzyl-1,2,3-trimethoxybenzene

Clear colorless liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.44 – 7.15 (m, 5H), 6.40 (s, 2H), 3.93 (s, 2H), 3.82 (s, 3H), 3.81 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 206.9, 153.2, 128.8, 128.5, 126.2, 105.9, 60.9, 56.0, 42.2, 30.9. IR wavenumber (cm⁻¹) 2925.4, 1127.3, 1008.4. HRMS calc. for C₁₆H₁₉O₃ [M+H]+: 259.13287; found, 259.13261.



4-propylbenzoic acid¹⁶ (2s) HOOC

White solid. ¹H NMR (400 MHz, CDCl₃) δ 8.06 – 8.00 (m, 2H), 7.33 – 7.24 (m, 2H), 2.73 (q, *J* = 7.6 Hz, 2H), 1.27 (t, *J* = 7.6 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 171.1, 150.8, 130.4, 128.0, 126.6, 29.0, 15.2.

4-benzylphenol¹⁷ (2ta) HO

White solid. ¹H NMR (400 MHz, CDCl₃) δ 7.33 – 7.24 (m, 2H), 7.26 – 7.14 (m, 3H), 7.10 – 7.02 (m, 2H), 6.80 – 6.71 (m, 2H), 4.59 (s, 1H), 3.92 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 153.8, 141.5, 133.4, 130.1, 128.8, 128.4, 126.0, 115.3, 41.0.



4-benzylaniline¹⁸ (2tb) H₂N

Clear colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.31 – 7.23 (m, 2H), 7.18 (td, J = 5.3, 2.8 Hz, 3H), 7.02 – 6.94 (m, 2H), 6.67 – 6.59 (m, 2H), 3.88 (s, 2H), 3.57 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 144.5, 141.9, 131.2, 129.8, 128.8, 128.3, 125.8, 115.3, 41.1.



4,4'-methylenediphenol¹⁹ HO

White solid. ¹H NMR (400 MHz, CD₃OD) δ 7.08 – 6.82 (m, 4H), 6.82 – 6.41 (m, 4H), 3.75 (s, 2H). ¹³C NMR (126 MHz, CD₃OD) δ 156.5, 134.2, 130.7, 116.1, 41.1, 30.7.



3-benzylbenzoic acid (2u)

White solid. ¹H NMR (400 MHz, CDCl₃) δ 7.94 (dt, J = 9.1, 1.7 Hz, 2H), 7.46 – 7.34 (m, 2H), 7.34 – 7.26 (m, 2H), 7.26 – 7.15 (m, 3H), 4.04 (s, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 171.4, 141.6, 140.3, 134.3, 130.6, 129.6, 128.9, 128.7, 128.6, 128.1, 126.3, 41.7. IR wavenumber (cm⁻¹) 3021.6, 299.9, 2916.2, 2849.6, 2671.3, 2561.5, 1683.0. HRMS calc. for C₁₄H₁₂NaO₂ [M+Na]+: 235.0730; found, 235.0725.

VII. References

- 1. W. Xu and R. Langer, *Dalton Trans.*, 2015, 44, 16785-16790.
- T. C. Fessard, H. Motoyoshi and E. M. Carreira, Angew. Chem. Int. Ed., 2007, 46, 2078-2081.
- 3. Y. Yuan and Y. Bian, Appl. Organomet. Chem., 2008, 22, 15-18.
- 4. a) F. Nador, Y. Moglie, C. Vitale, M. Yus, F. Alonso and G. Radivoy, *Tetrahedron*, 2010, **66**, 4318-4325. b) X. Wu and A. E. V. Gorden, *Eur. J. Org. Chem.*, 2009, **2009**, 503-509.
- 5. D. R. Crist, G. J. Jordan, D. W. Moore, J. A. Hashmall, A. P. Borsetti and S. A. Turujman, *J. Am. Chem. Soc.*, 1983, **105**, 4136-4142.
- M. Peña-López, M. Ayán-Varela, L. A. Sarandeses and J. Pérez Sestelo, *Eur. J. Chem.*, 2010, 16, 9905-9909.
- 7. G. Sun and Z. Wang, *Tetrahedron Lett.*, 2008, **49**, 4929-4932.
- 8. O. Masato and K. Ryoichi, Chem. Lett., 2008, 37, 796-797.
- 9. C. C. Kofink and P. Knochel, Org. Lett., 2006, 8, 4121-4124.
- 10. T. Niwa, H. Yorimitsu and K. Oshima, Angew. Chem. Int. Ed., 2007, 46, 2643-2645.
- 11. H. S. Park, H. Y. Lee and Y. H. Kim, *Org. Lett.*, 2005, **7**, 3187-3190.
- 12. A. Bramborg and T. Linker, Adv. Synth. Catal., 2010, 352, 2195-2199.
- 13. M. J. Burns, I. J. S. Fairlamb, A. R. Kapdi, P. Sehnal and R. J. K. Taylor, *Org. Lett.*, 2007, **9**, 5397-5400.
- 14. Y. Chang, H. Lee and C. Bae, in Organic Syntheses, pp. 245-252.
- 15. J. R. Coats, J. W. Williams, C. Chang, A.-H. Lee and R. L. Metcalf, *Environ. Toxicol. Chem.*, 1989, **8**, 45-52.
- 16. K. Nemoto, H. Yoshida, N. Egusa, N. Morohashi and T. Hattori, *J. Org. Chem.*,2010, **75**, 7855-7862.
- 17. C.-R. Chen, S. Zhou, D. B. Biradar and H.-M. Gau, Adv. Synth. Catal., 2010, 352, 1718-1727.

- 18. G. Manolikakes, C. Muñoz Hernandez, M. A. Schade, A. Metzger and P. Knochel, J. Org. Chem., 2008, 73, 8422-8436.
- 19. T. H. Fisher, P. Chao, C. G. Upton and A. J. Day, *Magnetic Resonance in Chemistry*, 1995, **33**, 717-723.

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m/z= 257.12449-257.13669													
m/z	Intensity	Relative	Resolution	Charge	Theo. Mass	Delta	RDB	Composition					
						(ppm)	equiv.						





























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 Z20427-06APCI HRMS-Li-Ruchua Gui-RG-03-73C#738-757 RT: 3.83-3.87 AV: 20

 T:FTMS - p APCI corona sid=30.00 Full ms [200.0000-500.0000]

 m/z= 322.8980-322.92982

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220427-05ESI HRMS-Li-Ruchua Gui-RG-03-73A#405-412 RT: 0.96-0.98 AV: 8 SB:15 0.01-0.04 T:FTMS + p ESI Full ms [150.0000-500.0000] m/z= 258 12/242-558 13792

m/z= 259.12429-259.13/92												
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 SB:15 0.01-0.04

 T: FTMS + p ESI Full ms [150.0000-500.0000]

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