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Supplementary Information

2 **Introducing competitive adsorption species in sulfonimide salt-based electrolytes for** 3 **inhibiting aluminium corrosion**

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1 Experimental Section

2 1. Materials and Electrode Preparation

3 Ethylene carbonate (EC), diethyl carbonate (DEC) and lithium bis(trifluoromethanesulfonyl)imide
4 (LiTFSI) were purchased from Dodo Chem Co., Ltd. 1-ethyl-3-methylimidazolium nitrate
5 ([EMIM]NO₃) was purchased from Adamas Reagent Co., Ltd. The no-additive electrolyte (1 M LiTFSI
6 in EC:DEC 1:1 in vol.) served as the baseline electrolyte to which various concentrations of [EMIM]NO₃
7 additive were added. The LiCoO₂ was purchased from Ningbo Shanshan Co., Ltd. In this study, a
8 cathode slurry comprising of LiCoO₂ powders, PVdF binder and conductive carbon black with a weight
9 ratio of 90:5:5 was cast on an aluminum foil with a blade and dried in a vacuum oven at 120 °C for 8
10 hours. Afterwards, the obtained cathodes were punched into small disks (Φ12 mm, with a mass loading
11 of ~ 4.6 mg cm⁻²). The lithium foils with a thickness of 200 μm were purchased from China Energy
12 Lithium Co., Ltd.

13 2. Electrochemical Tests

14 The LiCoO₂ || Li coin cells containing 100 μL of different electrolytes were assembled in an argon-
15 filled glovebox (H₂O < 0.1 ppm, O₂ < 0.1 ppm). These assembled coin cells were then kept at 25 °C for
16 8 hours to ensure adequate electrolyte infiltration before further tests. The electrochemical performance
17 of these cells was evaluated on Landt Battery Test System (Wuhan LAND Electronic Co., Ltd). Cycling
18 tests were conducted in the voltage range of 3 - 4.2 V and the cells were charged and discharged at 0.1C
19 in the first cycle and then cycled at 0.5C. A high-low temperature chamber (Guangzhou GWS Co., Ltd)
20 was used to provide constant testing temperatures. Electrochemical impedance spectroscopy (EIS) tests
21 were conducted in the charged state (4.2 V) by using an electrochemical workstation (Autolab, Metrohm)
22 in a frequency range of 200 kHz to 20 mHz. Homemade three-electrode coin cells were prepared for the
23 cyclic voltammetry (CV) measurements with details shown in Figure S4. The CV tests were carried out
24 at a scanning rate of 10 mV s⁻¹ in a potential range from 2.0 to 4.5 V vs. Li⁺/Li (calibrated).

25 3. Characterizations

26 After the cycling tests, LiCoO₂ || Li cells were disassembled in a glove box to obtain LiCoO₂ cath-
27 odes in the charged state (4.2 V). The LiCoO₂ cathodes were then washed three times utilizing dimethyl
28 carbonate (DMC) to remove residual electrolytes and dried in a vacuum chamber. The surface morphol-
29 ogy of cycled cathodes were characterized using scanning electron microscopy (SEM, S-8100, Hitachi).

1 The aluminum electrodes in the three-electrode coin cells were obtained with the same method after the
2 CV tests and their surface morphologies were also characterized by SEM. The Raman spectra of pristine
3 aluminium foil and aluminium foils after immersion in the no-additive electrolyte and the 0.1 M additive
4 electrolyte were obtained using a HORIBA LabRAM HR Evolution system equipped with a frequency-
5 doubled Nd:YAG laser (532 nm green).

6 4. Theoretical Calculations

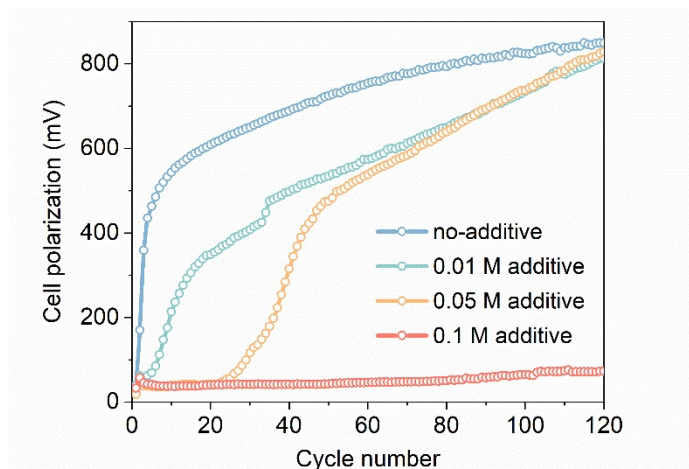
7 The adsorption configurations of various adsorbates on the (110) facet of Al_2O_3 were optimized by
8 density functional theory (DFT) method carried out in the Vienna Ab initio Simulation Package (VASP)
9 code. The generalized gradient approximation (GGA) of PBE exchange-correlation functional was used.
10 The plane-wave cutoff energy was set as 400 eV. The threshold values of energy convergence and force
11 were set as 10^{-5} eV and 0.03 eV \AA^{-1} , respectively. To eliminate the interaction between adjacent images,
12 a vacuum space larger than 15 \AA was constructed along the vertical direction. In the general models, the
13 k-point mesh was sampled with a separation of 0.03 \AA^{-1} . The 2×1 size (110) plane of $\alpha\text{-Al}_2\text{O}_3$ retained a
14 thickness of 5 molecular layers and the bottom two layers were fixed to simulate the adsorption surface.
15 In addition, the adsorption energy of different adsorbates is obtained by the following formula,

$$16 E_{ads} = E_T - E_s - E_a$$

17 where E_T , E_s and E_a are the energy of the total adsorption system, the Al_2O_3 substrate and the adsorbates
18 respectively. Besides, the DFT calculation results are presented in Table S2 and a list of atomic coordin-
19 ates of the optimized geometries (adsorbates and three layers of the Al_2O_3 substrate) are presented at the
20 end of the supplementary information.

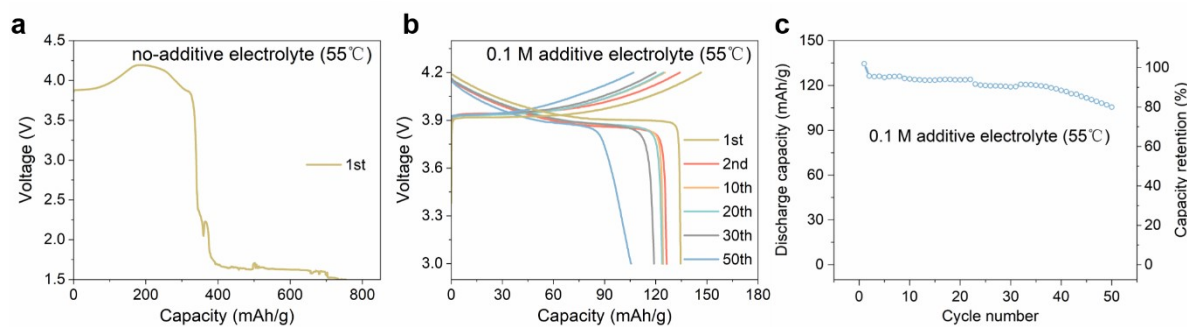
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1 Supplementary Figures



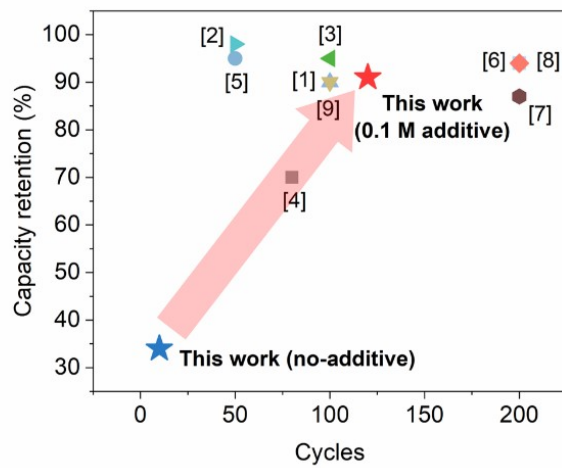
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3 **Figure S1.** The polarizations of $\text{LiCoO}_2 \parallel \text{Li}$ cells using different electrolytes. The cell polarization was
4 determined by subtracting the mid-voltage of discharge process from the mid-voltage of charge process
5 at each charge-discharge cycle.



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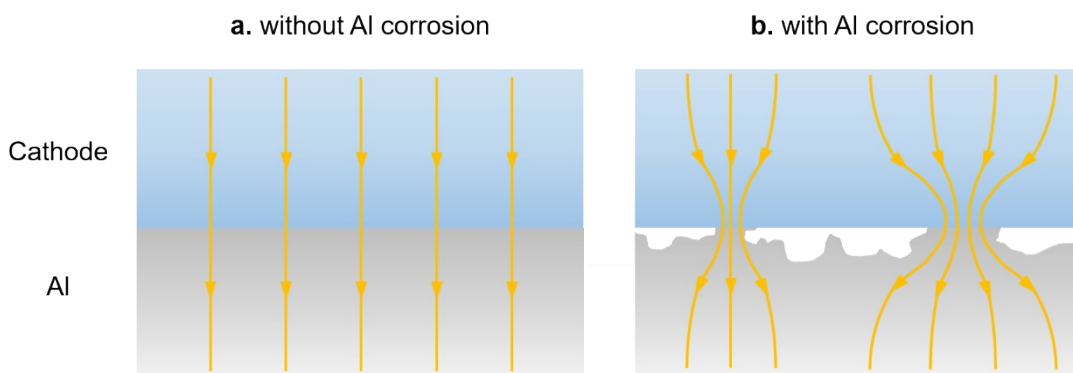
7 **Figure S2.** The charge-discharge curves of $\text{LiCoO}_2 \parallel \text{Li}$ cells using (a) no-additive electrolyte and (b)
8 0.1 M additive electrolyte at the elevated temperature of 55 °C. (c) The discharge capacity and capacity
9 retention of $\text{LiCoO}_2 \parallel \text{Li}$ cell with the 0.1 M additive electrolyte at 55 °C.



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2 **Figure S3.** Summary of the cycling stability of 4 V-class cathodes (with aluminium current collector) in

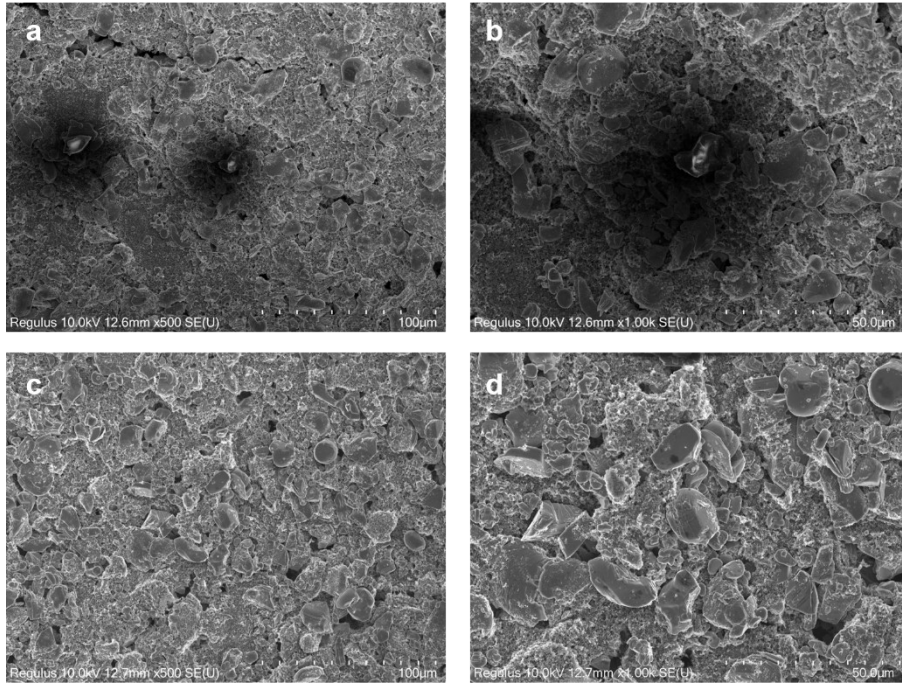
3 LiTFSI-based electrolytes achieved by various approaches.



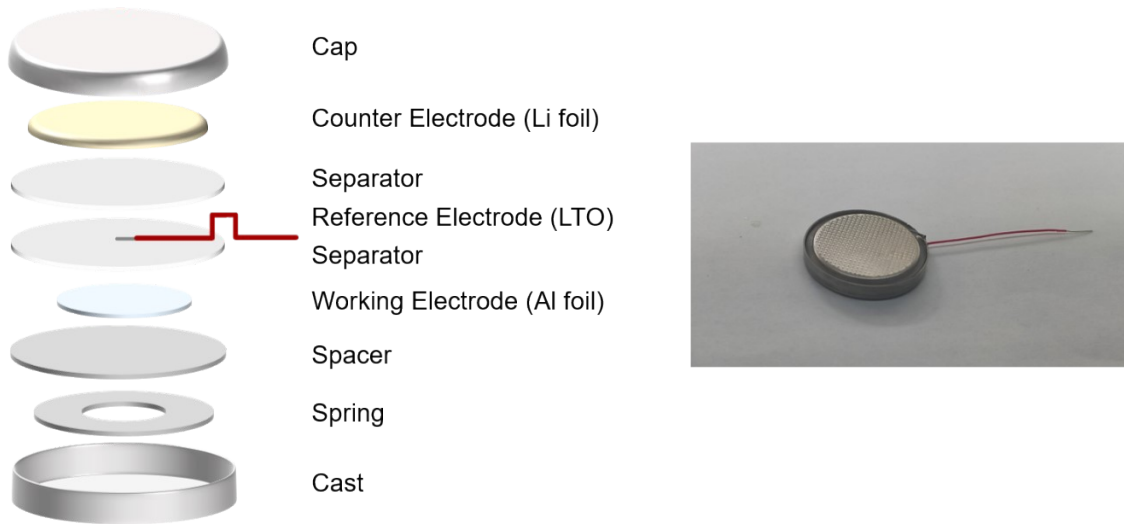
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5 **Figure S4.** Schematic illustrations of electron transfer process at the cathode/aluminum current collector

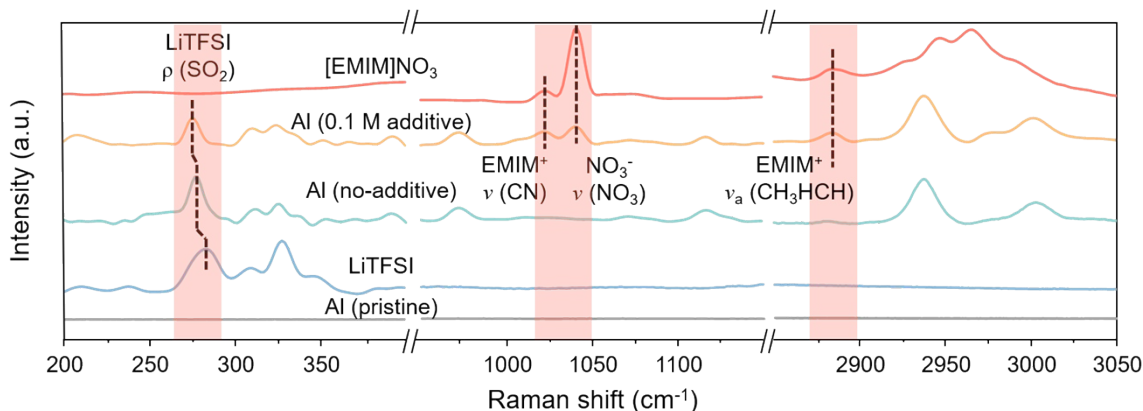
6 interface (a) without and (b) with aluminum corrosion.



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 2 **Figure S5.** SEM images of disassembled LiCoO₂ cathodes in the (a, b) no-additive electrolyte and (c, d)
 3 0.1 M additive electrolyte after 20 charge-discharge cycles.



4
 5 **Figure S6.** The inner structure and image of the homemade three-electrode coin cell. The Li₄Ti₅O₁₂ was
 6 casted on an enameled copper wire and served as the reference electrode (LTO). The aluminum foil was
 7 used as the working electrode and the lithium foil was used as the counter electrode.



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2 **Figure S7.** Raman spectra of pure [EMIM]NO₃, pure LiTFSI, pristine aluminium foil and aluminium
 3 foils after immersion in the no-additive electrolyte and 0.1 M additive electrolyte (vibrational modes: ν
 4 = stretching; ρ = rocking; ν_a = asymmetric stretching).

5 **Table S1.** Summary of the cycling stability of 4 V-class cathodes (with aluminium current collector)
 6 achieved in LiTFSI-based electrolytes by different methods

Ref.	Method	Electrolyte	Cell configuration	Voltage range	C-rate	Capacity retention (cycles)
[1]	Chromate conversion coating	1M LiTFSI EC/EMC 3:7 vol.	NMC333 Li	2.7 ~ 4.2 V	1C	90% (100)
[2]	Concentrated electrolytes	2.3M LiTFSI EC:DME 1:2 vol.	NMC622 Graphite	2.8 ~ 4.2 V	0.5C	98.7% (50)
[3]	Concentrated electrolytes	3.5M LiTFSI DMC/C ₂ [FSI] 1:1 vol.	NMC523 Li	2.8 ~ 4.5 V	C/6	95% (100)
[4]	Concentrated electrolytes	5M LiTFSI in EC:DEC 3:7 vol.	NMC811 Li	3.0 ~ 4.6 V	1C	70% (80)
[5]	Low solubility solvents	1M LiTFSI in F-C4	NMC111 Li	3.0 ~ 4.3 V	0.2C	95% (50)
[6]	Low solubility solvents	1M LiTFSI in MCP	NMC111 Graphite	2.8 ~ 4.2 V	1C	94.4% (200)
[7]	Partially hydrogenated anion	1M LiDFTFSI in EC:EMC 3:7 vol.	NMC111 Li	2.7 ~ 4.2 V	0.2C	87% (200)
[8]	Adding LiBOB	0.75M LiTFSI + 0.25M LiBOB in FEC/1,2-BC 1:1 mol.	NMC111 Graphite	3.0 ~ 4.2 V	0.2C	94.5% (200)
[9]	Adding LiDFOB and LiPF ₆	0.8M LiTFSI + 0.2M LiDFOB + 0.01M LiPF ₆ in EC:PC 1:1 vol.	LCO Li	3.0 ~ 4.2 V	0.5C	90% (100)
This work	Adding competitive adsorption species [EMIM]NO ₃	1M LiTFSI in EC:DEC 1:1 vol.	LCO Li	3.0 ~ 4.2 V	0.5C	90.2% (120)

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1 **Table S2.** The calculated energies of the total adsorption system, the Al₂O₃ substrate and the adsorbates
2 used for deducing the values of the adsorption energy of different adsorbates

Adsorbates	Energy (eV)			
	Adsorbates	Substrate	Total	Ads
NO ₃ ⁻	-28.071	-933.375	-966.007	-4.561
EMIM ⁺	-106.786	-933.375	-1045.27	-5.109
TFSI ⁻	-86.992	-933.375	-1022.551	-2.184

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22 **Cartesian coordinates of the optimized adsorption geometries**

23 Al₂O₃-NO₃⁻

24 N O Al

25 1 39 48

1	Cartesian		
2	9.863546584	6.924930020	7.339530140
3	3.232665593	5.419654616	3.229200169
4	9.840045775	6.981263906	4.180249989
5	3.264913933	2.873692575	3.269220024
6	9.735989778	1.277288833	4.251780137
7	3.315962958	16.567142818	3.444249868
8	9.821506759	4.123311587	4.164839968
9	7.609167018	2.647197122	3.244609848
10	1.126471701	4.225872491	4.252929986
11	7.635148722	0.081187121	3.246219978
12	1.009881556	6.764875320	4.280990049
13	7.606816859	5.462733237	3.435739964
14	1.105843222	1.350522885	3.981760159
15	11.956292821	8.149364079	3.240010113
16	5.557808475	1.465179020	4.238670155
17	11.979009984	5.591970089	3.238859922
18	5.364710257	4.072942422	4.280990049
19	11.910597022	2.720431174	3.388130084
20	5.396828055	6.888313118	4.078130111
21	3.242457467	13.699912208	3.240240082
22	9.861065688	15.271794773	4.240969852
23	3.260083073	11.135724673	3.262319908
24	9.796569397	9.518444161	4.274319902
25	3.306040153	8.283902741	3.387670144
26	9.759882020	12.424280160	4.122980013
27	7.631492747	10.904424162	3.280489907
28	1.138483190	12.514580291	4.231770039
29	7.627184641	8.359456859	3.285779893
30	1.063541728	15.076944758	4.275240123
31	7.615303889	13.786567190	3.458280072
32	0.994475468	9.670379987	4.048459917
33	11.963996206	16.433929485	3.253120095
34	5.465502492	9.738975141	4.238900125
35	11.973396062	13.892607330	3.257489860
36	5.341992706	12.327352397	4.216819957
37	11.938667410	11.045922282	3.462189898
38	5.462238728	15.163765653	4.154949903
39	9.784296433	8.069171310	6.469440252
40	7.615956019	6.599850143	5.648570031
41	10.007163067	5.810015337	6.638259888
42	4.597017706	4.121323347	2.551390074
43	4.515939567	0.128242508	4.895320132

1	8.459504221	4.121323347	2.551390074
2	8.293040616	0.248697691	5.058620155
3	1.931504538	4.121323347	2.551390074
4	1.797941792	0.028166960	5.148089990
5	11.125017584	4.121323347	2.551390074
6	11.205050602	16.456628952	5.014690131
7	8.948451615	1.359967088	2.551390074
8	9.078620054	5.571259109	5.047349930
9	12.812113988	1.359967088	2.551390074
10	12.758584066	5.323555430	4.930279970
11	6.284113721	1.359967088	2.551390074
12	6.368977794	5.535138958	4.862659991
13	2.420451737	1.359967088	2.551390074
14	2.496568278	5.676470671	4.950520039
15	0.244408304	6.882845396	2.551390074
16	0.753461738	2.634273437	5.343129933
17	4.108070312	6.882845396	2.551390074
18	3.776186666	3.011876385	5.145330012
19	10.636070190	6.882845396	2.551390074
20	10.856455150	2.645374548	4.885200098
21	6.772407817	6.882845396	2.551390074
22	6.756349697	2.769972008	4.955350086
23	4.597017706	12.405723580	2.551390074
24	4.343861651	8.335928814	4.878989890
25	8.459504221	12.405723580	2.551390074
26	8.334819745	8.240658206	5.120029926
27	1.931504538	12.405723580	2.551390074
28	0.750850542	8.314389750	5.357390106
29	11.125017584	12.405723580	2.551390074
30	11.070443320	8.186478227	5.014920101
31	8.948451615	9.644367197	2.551390074
32	8.910589158	13.908016501	4.939250156
33	12.812113988	9.644367197	2.551390074
34	12.665886650	13.996493563	5.064370081
35	6.284113721	9.644367197	2.551390074
36	6.313098103	13.650537287	4.911420062
37	2.420451737	9.644367197	2.551390074
38	2.464842109	13.711510318	4.986170128
39	0.244408304	15.167244888	2.551390074
40	0.211768311	11.130754196	4.965470120
41	4.108070312	15.167244888	2.551390074
42	4.095405914	11.122304484	4.998359889
43	10.636070190	15.167244888	2.551390074

1	10.655785064	11.000026596	4.950979978		
2	6.772407817	15.167244888	2.551390074		
3	6.981826408	10.591274220	5.103929996		
4					
5	Al ₂ O ₃ -EMIM ⁺				
6	H	C	N	O	Al
7	11	6	2	36	48
8	Cartesian				
9	4.618820875	6.996507164	7.309800088		
10	8.641113120	6.977784177	6.957900077		
11	7.958284641	9.369324906	7.831200063		
12	4.525470550	8.791239540	8.983500302		
13	4.582133497	9.989329971	7.672500014		
14	5.279062928	10.961917958	9.888600111		
15	6.591451827	11.086183839	8.703000247		
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18	5.806394949	4.995658982	7.782600224		
19	6.789119624	6.227217919	8.653799593		
20	5.419675955	7.511299679	6.760199815		
21	7.770800346	7.505831957	6.505499929		
22	7.323371384	8.617929576	7.366199791		
23	5.207124338	9.432783669	8.402100205		
24	5.991137263	10.377370873	9.290400445		
25	6.700469617	5.633226137	7.732799649		
26	6.041402643	8.525973278	7.608900368		
27	6.571345519	6.539871109	6.537600160		
28	3.236712807	5.402422871	3.247200027		
29	9.873600127	6.873401193	4.298399985		
30	3.263477768	2.843371726	3.253800049		
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37	0.998261694	6.822534535	4.253400117		
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7	3.277969764	8.267831400	3.468900025
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15	11.968957219	16.436083391	3.261299953
16	5.583006533	9.669717323	4.287300110
17	11.959034414	13.901720695	3.257400021
18	5.413800947	12.284274270	4.244100004
19	12.005644597	11.032667018	3.389699981
20	5.421112119	15.180996904	4.131900072
21	4.597017706	4.121323347	2.551200092
22	4.567380415	0.079364554	4.947000146
23	8.459504221	4.121323347	2.551200092
24	8.427647898	0.105377568	4.993199855
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26	1.978897977	16.432604156	4.943700135
27	11.125017584	4.121323347	2.551200092
28	11.310673891	16.319936384	5.082000196
29	8.948451615	1.359967088	2.551200092
30	10.133414056	5.290749096	5.424000174
31	12.812113988	1.359967088	2.551200092
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33	6.284113721	1.359967088	2.551200092
34	6.346260242	5.402754203	4.817099869
35	2.420451737	1.359967088	2.551200092
36	2.637703671	5.329023153	5.072700083
37	0.244408304	6.882845396	2.551200092
38	0.234224632	2.777427971	4.960500151
39	4.108070312	6.882845396	2.551200092
40	4.042790375	2.869881761	5.028300136
41	10.636070190	6.882845396	2.551200092
42	10.678371489	2.622012663	4.968300015
43	6.772407817	6.882845396	2.551200092

1	6.975298103	2.589371994	5.090999901
2	4.597017706	12.405723580	2.551200092
3	4.613990209	8.281748835	4.994700104
4	8.459504221	12.405723580	2.551200092
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7	1.972369867	8.224586865	4.941000044
8	11.125017584	12.405723580	2.551200092
9	11.443453186	8.443460196	5.063100010
10	8.948451615	9.644367197	2.551200092
11	8.988924927	13.787063942	5.050799847
12	12.812113988	9.644367197	2.551200092
13	12.710276633	13.870405602	5.025900006
14	6.284113721	9.644367197	2.551200092
15	6.328895916	13.746470559	4.950300157
16	2.420451737	9.644367197	2.551200092
17	2.509232676	13.602984446	5.037900209
18	0.244408304	15.167244888	2.551200092
19	0.009530880	11.104906332	4.884899855
20	4.108070312	15.167244888	2.551200092
21	3.883246120	11.200674679	5.091299862
22	10.636070190	15.167244888	2.551200092
23	9.989798351	11.167205679	5.423399806
24	6.772407817	15.167244888	2.551200092
25	6.778152476	10.991079144	4.988700002

26

27 Al₂O₃-TFSL-

28	C	N	O	F	Al	S
29	2	1	40	6	48	2

30 Cartesian

31	5.888778126	10.148390181	9.138300419
32	6.875680947	6.544013009	8.793599904
33	5.271882104	8.214810836	7.064699829
34	3.245982704	5.401428874	3.242399991
35	9.801530411	6.978116003	4.242900163
36	3.292723236	2.834921767	3.250799999
37	9.762623612	1.242328592	4.250999987
38	3.266872269	16.538147544	3.459299952
39	9.753093018	4.142034327	4.009800106
40	7.617001139	2.612568460	3.267600015
41	1.143313856	4.208475327	4.228499830
42	7.631101313	0.067103642	3.237899914
43	1.024504290	6.790888110	4.274999946

1	7.583577525	5.499350633	3.444600105
2	1.119943687	1.351682671	4.126499891
3	11.964518377	8.155329047	3.246900067
4	5.503756771	1.460042630	4.258499891
5	11.964779073	5.621296695	3.264899924
6	5.353873847	4.071451180	4.240500033
7	12.030581181	2.757876654	3.410099968
8	5.440696256	6.869258799	4.137900174
9	3.237887887	13.680195225	3.262799978
10	9.854015990	15.279912660	4.286099821
11	3.272747276	11.115345025	3.246299922
12	9.823333969	9.496573271	4.214999825
13	3.235276642	8.224752284	3.423300013
14	9.854015990	12.405557667	3.990899920
15	7.670399545	10.901773504	3.314400092
16	1.129213390	12.493041227	4.247999936
17	7.606556163	8.368072485	3.255300075
18	0.959877067	15.092189003	4.258499891
19	7.569999523	13.774140504	3.384599984
20	1.081689603	9.634260330	4.114800096
21	11.947937307	16.461765342	3.224399909
22	5.353873847	9.694404783	4.265699834
23	11.977835683	13.902880111	3.245100081
24	5.347345931	12.353200262	4.215299785
25	12.002641530	11.028359205	3.443400040
26	5.448791074	15.158298424	4.153799862
27	4.827716905	10.632199429	6.803399920
28	3.400826642	9.398486093	8.466299772
29	4.202203893	6.488839032	8.717699647
30	5.464197064	5.787316363	6.683400124
31	5.737981208	11.479858870	9.251699746
32	7.155732700	9.861417954	8.801399767
33	5.580917849	9.568647660	10.314299762
34	7.919508580	7.135188219	8.204399943
35	7.099199712	5.224473761	8.918099999
36	6.687674481	7.078025756	10.020300150
37	4.597017706	4.121323347	2.551200092
38	4.572733252	0.036451360	4.920900017
39	8.459504221	4.121323347	2.551200092
40	8.427647898	16.477506339	4.951499999
41	1.931504538	4.121323347	2.551200092
42	1.952133016	16.431278827	4.908899814
43	11.125017584	4.121323347	2.551200092

1	11.226723423	0.108359951	4.953300208
2	8.948451615	1.359967088	2.551200092
3	8.927300965	5.511611407	4.979399890
4	12.812113988	1.359967088	2.551200092
5	12.680117561	5.636871285	5.053199977
6	6.284113721	1.359967088	2.551200092
7	6.337251819	5.400931629	4.956900179
8	2.420451737	1.359967088	2.551200092
9	2.498526614	5.306158266	5.023499876
10	0.244408304	6.882845396	2.551200092
11	0.068805120	2.831607950	4.889700115
12	4.108070312	6.882845396	2.551200092
13	4.110681556	2.867562188	5.008800030
14	10.636070190	6.882845396	2.551200092
15	10.081712122	2.821335170	5.379599780
16	6.772407817	6.882845396	2.551200092
17	7.048412161	2.502054594	5.133900046
18	4.597017706	12.405723580	2.551200092
19	4.546099028	8.250268322	4.941000044
20	8.459504221	12.405723580	2.551200092
21	8.393571376	8.352995140	4.988100082
22	1.931504538	12.405723580	2.551200092
23	1.992084351	8.202550062	4.906800091
24	11.125017584	12.405723580	2.551200092
25	11.207531498	8.139588544	4.999199957
26	8.948451615	9.644367197	2.551200092
27	9.574747883	13.702728779	5.350800008
28	12.812113988	9.644367197	2.551200092
29	12.729730032	13.907850588	5.012100041
30	6.284113721	9.644367197	2.551200092
31	6.508285393	13.599174372	4.849800020
32	2.420451737	9.644367197	2.551200092
33	2.662771187	13.346002971	5.087399930
34	0.244408304	15.167244888	2.551200092
35	0.135782398	11.083864019	4.904999882
36	4.108070312	15.167244888	2.551200092
37	4.265656231	11.071603246	4.898999780
38	10.636070190	15.167244888	2.551200092
39	10.523266466	11.007481818	5.127300024
40	6.772407817	15.167244888	2.551200092
41	7.694031091	11.096622532	5.351399928
42	4.656552983	9.557546302	7.800900042
43	5.256998284	6.740353279	7.785900235