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1 Supplementary Information

2 Calix[6]arene Functionalized Lanthanide Metal-

- 3 Organic Frameworks with Boosted Performance on
- 4 Identifing Anti-epidemic Pharmaceutical
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16 1. Supplementary Experimental Section

17 **1.1 Materials and Reagents.** Carbon Mesh was purchased from Toray Industries, Inc (Japan), 18 Tb(NO₃)₃·6H₂O, dimethylformamide (DMF), and all the other interfering compounds for *Prednis* detection 19 experiments were purchased from Aladdin Reagent (Shanghai, China). All suspensions were prepared using 20 ultrapure water (18.2 M Ω cm) from a Millipore Direct-Q system.

21 **1.2 Apparatus.** The UV-vis spectra were collected on an UV-2450 spectrophotometer (Shimadzu, Japan). 22 Quanta 200 scanning electron microscopy (SEM, U.S.A.) was used to characterize the sizes and morphology. 23 Fourier transform infrared (FT-IR) spectra were obtained using a Nicolet 5700 FT-IR spectrometer. 24 Fluorescence spectra were performed on an F-97 Pro fluorescence spectrometer. The excitation wavelength, 25 slit widths (including excitation and emission), and the photomultiplier tube (PMT) voltage were set at 290 nm, 26 10 nm and 650 V, respectively. The X-ray photoelectron spectroscopy (XPS) spectra were recorded on powders 27 with a Thermo ESCALAB 250 spectrometer using an AI Ka monochromator source (hv=1486.6 eV) and a 28 multidetection analyzer.

29 1.3 Synthesis of Tb-BTC MOF and structurally modification of TB-Cx[6]. Tb-BTC was synthesized in 30 a relatively mild way according to the method in the literature with a slight adjustment [1]. 1 g of $Tb(NO_3)_3 \cdot 6H_2O$ 31 powder was mixed with 0.5 g of BTC and dissolved in 54 ml DMF mixed with 4 ml water. After stirring for 1 h, 32 the mixture was treated at 100 °C for 12 h to obtain Tb-BTC suspension in DMF with underreacted ligands BTC 33 and Tb metal ions The mixture suspension above was centrifuged and washed by DMF, water and ethanol, 34 respectively. Finally, the white powder of Tb-BTC was got from drying in vacuum oven at room temperature. 35 The structure modification route of Tb-BTC was according to a literature procedure in which 100 mg of the Tb-36 BTC powder and 50 mg Cx[6] were mixed in 10 ml methylbenzene within a glass bottle and took sonication for 37 1 h and left standing overnight. The solutions were filtered to get rid of parent solids. The solutions were dried 38 at 150°C for 24 h under vacuum to obtain the final product TB-Cx[6] powder.

1.4 Quantification determination of *Prednis* with TB-Cx[6] fluorescence probe. Firstly, the TB-Cx[6] stock suspension was prepared by mixing 5 mg TB-Cx[6] to 5 mL ultrapure water under sonication for 3 min. Subsequently, the TB-Cx[6] suspension (50 μ L) was respectively mixed with an aliquot of 200 μ L different concentrations of *Prednis* methanol solutions. The mixture above was diluted with ultrapure water to control the total volume 2000 μ L. The fluorescence spectra were measured with the excitation wavelength at 290 nm. The Tb³⁺ fluorescence peak at 545 nm was measured for quantification. **1.5 Selectivity of the assay to** *Prednis.* To assess the sensing selectivity of the TB-Cx[6] system for *Prednis*, potential interfering metal ions and were used to investigate the specificity of the *Prednis* assay under identical conditions. The concentration of *Prednis* is set at 100 ng/mL, and the level of other analytes is 5 folds than *Prednis*. The interference of other analytes is quite tiny, indicating that this fluorescence probe has high sensitivity and selectivity. It is worth noting that bare Tb-BTC for *Prednis* detection were disturbed by lots of analytes. This is due to the selective and efficient enrichment of *Prednis* into the solid material by Cx[6].

1.6 Analysis of *Prednis* **in real water samples.** To test the general applicability of TB-Cx[6] probe, standard addition method was performed on the water samples (river water and sea water) for our assay. Real samples of the Pearl River water were filtered through the 0.22 μ m cellulose acetate filters to remove insoluble impurities that may exist in the water. Subsequently, 350 μ L of the obtained water samples and 100 μ L of TB-Cx[6] probe were diluted to 700 μ L with deionized water. Subsequently, the mixture was used for performing fluorescent and L-MS measurements. The concentrations of *Prednis* in the tested water samples are presented in Table S1.

1.7 Preparation of TF. Firstly, the 20 mg solid TB-Cx[6] was added into 10 mL hexamethylene and then sonicated to around 30 mins. Then 40 mg PDMS was added into the above mixture. The mixed solution was strongly stirred and sonicated to obtain the suspension of TB-Cx[6]. Subsequently, a cleaned carbon mesh was fixed on the coating instrument (Figure S10) and the obtained TB-Cx[6] suspension was gradually dropped onto the carbon mesh surface. A round push rod rolls on the TF film to control the thickness and uniformity of the coating. Finally, the TF was obtained and dried for 30min at 85 °C under vacuum.

1.8 Design of the TF equipped drone water sampler. A drone of with automatic return function was purchased from Dajiang (Mavic Mini-DJI, Shenzhen, China). This drone was tested to have an additional lift capacity of 200 g and a flight range of 3000 m. The TB-Cx[6] TFs were fixed to the lower part of the buoyancy stents by clamps. The TB-Cx[6] TFs were all immersed into the water to a depth about 10 cm, when the drone was floating on water.

1.9 Analysis of *Prednis* by TF. As the universality and particularity of Prednis, as well as the uncertainty of nocuity in real samples, TFME drone water sampler was fabricated and used to analysis Pearl River Water. Two TB-Cx[6] membranes were equipped onto the drone sampler. The drone was then lifted off from a stent placed on the deck and flown toward real-water. After touching down on the surface of the water the membranes were all immersed into the water by gravity. The drone was then allowed to sit on the surface of the water for 10 min of sampling. After sampling, the TF is taken out and dried, and then the fluorescence test is performed. The concentrations of *Prednis* in the tested real water is presented in Table S2.



Figure S1. SEM image of bare Tb-BTC MOF







88 Figure S3. (A) FTIR and (B)XRD spectra of Cx[6], Tb-BTC and TB-Cx[6] adding different amounts of Cx[6],

respectively

89



92Figure S4. (A) TGA curves and (B) BET pattern of Tb-BTC and TB-Cx[6]. The TGA was performed at93temperature ranged 30 to 800 °C, heating rate was 10°C/min, and the atmosphere is nitrogen.





Figure S5. Photos of Tb-BTC under (A) sunlight and (B) UV light (254 nm). Photos of TB- Cx[6] under (C)

sunlight and (D) UV light (254 nm)





Figure S7. XRD pattern of TB- Cx[6] (A) and Tb-BTC (B) before and after soaking in water for 15 days.



109

110 Figure S8. Fluorescence spectra of Tb-BTC, TB- Cx[6], Tb-BTC and TB- Cx[6] with 1000 ng/mL Prednis

111 respectively.





114 Figure S9 Ultraviolet absorption spectrum of *Prednis* and fluorescence emission spectrum of TB-Cx[6].



117 Figure S10. (A) Effects of Cx[6] contents for *Prednis* analysis. (B) Effects of different pH before and after addition

118 of 1000 ng/mL *Prednis*.



121 Figure S11 Fluorescence intensity curves of the Tb-BTC MOF fluorescent probe with different concentrations of





- 125 Figure S12. Selectivity assay of Tb-BTC to *Prednis* in the presence of various interfering species (tetracycline,
- 126 tetrodotoxin, adenosine, 2-deoxyadenosine(EPN), malathion, norfloxacin, Cu^{2+} and UO_2^{2+}). The concentration
- 127 of *Prednis* was 100 ng/mL, the concentration of each interfering species was 5 times of *Prednis*.



- 130 Figure S13. Selectivity assay of TB-Cx[6] to *Prednis* in the presence of various interfering species (tetracycline,
- 131 tetrodotoxin, adenosine, 2-deoxyadenosine(EPN), malathion, norfloxacin, Cu^{2+} and UO_2^{2+}). The concentration
- 132 of *Prednis* was 100 ng/mL, the concentration of each interfering species was 5 times of *Prednis*.
- 133



Figure S14. Coating machine for making TF.



- 138 Figure S15. The spectra of TB-Cx[6] thin film after immersed into simulated sea water with different
- 139 concentration of *Prednis*

| methods | Spiked | Detected | Recovery | RSD (n=5) | LC-MS | Consistency |
|----------|---------|----------|----------|-----------|---------|-------------|
| | (ng/mL) | (ng/mL) | (%) | | (ng/mL) | (%) |
| | 0 | ND | - | 6.7% | - | - |
| TB-Cx[6] | 10.0 | 11.4 | 114.0 | 8.1% | 10.7 | 106.5 |
| | 50.0 | 51.7 | 103.4 | 5.9% | 50.9 | 101.5 |
| | 0 | ND | - | 6.5% | - | - |
| TF | 10.0 | 11.8 | 118.0 | 7.8% | 10.7 | 110.3 |
| | 50.0 | 52.7 | 105.4 | 5.3% | 50.9 | 103.5 |

141 **Table S1.** Detection *Prednis* in Pearl river water samples by using TB-Cx[6], TF and LC-MS.

142 ND indicated not detected. Consistency =C _{Fluorescence} / C_(LC-MS) 100%

143

145 Table S2. Detection *Prednis* in simulated sea water^a by using TF.

| methods | Spiked | Detected | Recovery | RSD (n=3) | LC-MS | Consistency |
|---------|---------|----------|----------|-----------|---------|-------------|
| | (ng/mL) | (ng/mL) | (%) | | (ng/mL) | (%) |
| TF | 5000.0 | 5397 | 108.0 | 7.4% | 5242 | 103.0 |

146 a: Simulated seawater was prepared in accord with the main constituents of real seawater as specified in

147 Table S3 (China Petroleum Processing and Petrochemical Technology, 2022, 24, 91-100).

| Main component | Percentage salt (%) | Concentration (g/L) | |
|-------------------|---------------------|---------------------|--|
| NaCl | 77.76 | 27.2 | |
| MgCl ₂ | 10.88 | 3.80 | |
| MgSO ₄ | 4.38 | 1.70 | |
| CaSO ₄ | 3.60 | 1.20 | |
| K_2SO_4 | 2.47 | 0.90 | |
| CaCO ₃ | 0.35 | 0.10 | |

149 Table S3. Main constituents of simulated seawater as used in the study