Monotrifluoroacetoxyborane-amines: Chemoselective reagents for challenging reductive aminations

P. Veeraraghavan Ramachandran,* Shivani Choudhary

Department of Chemistry, 560 Oval Drive, Purdue University West Lafayette, Indiana 47907-2084; Fax: 765 494 0239; Tel: 765 494 5303; E-mail: chandran@purdue.edu

Supporting Information

Table of Contents

Numbering Legend for carbonyls	S2
Numbering Legend for amines	S2
Table SI-1: Synthesis of TFAB-amines: Optimization of reaction conditions	S3
General Experimental Procedures	S4
Additional comparison of reductive aminating reagents	S8
Green matrix calculation (Process Mass Intensity)	
X-Ray crystallography Data	S11
Experimental NMR Data	S14

Numbering Legend for carbonyls

Aldehydes

Ketones



5g: R = H **5h**: R = *p*-OMe **5i**: R = *p*-CF₃ **5j**

5k

Numbering Legend for Amines



$\begin{array}{ccc} R_{3}N-BH_{3} & \xrightarrow{CF_{3}COOH} & \xrightarrow{R_{3}N} & \xrightarrow{CF_{3}} \\ \textbf{1a-j} & & \textbf{2a-j: TFAB-amine complex} \end{array}$									
	R ₃ N●BH ₃ R ₂ N	Eq. of TE∆	Solvent	Tempe	Time (h)	TI	TFAB-amine complex		
	1311	1 ea	THF	RT	1				
		1 eq.	DCM	RT	2				
19	Ammonia	1 eq.	Toluene	RT	1	2.9	$CF_{2}COO-BH_{2} \bullet NH_{2}$		
1	1 minionia	2 eq.	THF	RT	1				
		<u> </u>	THF	RT	1				
		1 ea.	THF	RT	18	_			
1c	Benzylamine	1 eq.	Toluene	RT	2	2c	CF ₃ COO-BH ₂ •NH ₂ CH ₂ Ph		
11		1 eq.	THF	RT	20				
lb	Methylamine	1 eq.	Toluene	RT	2	26	$CF_3COO-BH_2 \bullet NH_2Me$		
1£	Dingriding	1 eq.	THF	reflux	6	26			
11	Piperidine	1 eq.	Toluene	RT	4	21	CF ₃ COO-BH ₂ •piperidine		
1.1	Dimethylemine	1 eq.	THF	reflux	4	24	CE COO DIL MIIMA		
10	Dimetriyianine	1 eq.	Toluene	RT	4	2u	$CF_3COO-B\Pi_2$ •NILIVIE ₂		
10	Diisonronylamine	1 eq.	THF	reflux	4	20	$CF_{i}COO BH_{i}NH(i Pr)$		
10	Disopropylainine	1 eq.	Toluene	RT	4	20	$C\Gamma_3COO-D\Pi_2 \bullet N\Pi(l-\Gamma_1)_2$		
1σ	Trimethylamine	1.5 eq.	THF	reflux	12	20	NR		
Ig		1.5 eq.	Toluene	RT	4	2g	CF ₃ COO-BH ₂ •NEt ₃		
1h	Triethylomine	1.5 eq.	THF	reflux	12	2h	NR		
111		1.5 eq.	Toluene	RT	4	211	CF ₃ COO-BH ₂ •NMe ₃		
1i	Pyridine	1 eq.	THF	RT	-	2i	CF.COO_BH. Pyridine		
		1 eq.	Toluene	RT	2	21			
11	2-Picoline	1 eq.	THF	RT	-	21	CE_COO_BH_•2_Picoline		
-J	2-1 10011110	1 eq.	Toluene	RT	4	<u> </u>			

 Table SI-1. Optimization of the synthesis of TFAB-amine complexes

General Experimental Procedure

General Information: ¹¹B, ¹⁹F, ¹³C, and ¹H NMR spectra were recorded at room temperature, on a Varian INOVA 300 MHz NMR spectrophotometer. Chemical shifts (δ values) are reported in parts per million relative to BF₃.Et₂O f{Peterson, 2002 #173} or ¹¹B NMR spectra. PMR spectral data are reported as: δ value, multiplicity (s=singlet, d=doublet, t=triplet, q=quartet, p=pentet, h=hextet, m=multiplet, br=broad) with integration. All solvents for routine isolation of products were reagent-grade. All amines and carbonyls were purchased from commercial sources and used without further purification.

Synthesis of TFAB-ammonia complex: Anhydrous THF (3 mL) was added to ammonia borane (3 mmol, 1 equiv.) in a 15mL round bottom flask. Trifluoroacetic acid (3 mmol, 1 equiv.) was added to the above solution at 0 °C and the resulting reaction mixture was stirred for 1 hour at room temperature. The completion of reaction was confirmed by ¹¹B-NMR spectroscopy as indicated by the disappearance of the δ - 22 ppm peak due to ammonia borane and the appearance of δ -5 ppm peak due to TFAB-NH₃.

Preparation of Crystal for X-ray crystallography: 1.5 mmol of 18-crown-6 was added to 1M THF solution containing 3 mmol of TFAB-NH₃. The mixture was stirred for 5 min and filtered. The clear solution was kept in a small vial inside a larger vial containing hexane at RT for a week to get colorless transparent block crystals. Details of crystal structure are described in X-Ray crystallography data (vide infra).

Synthesis of TFAB-amine complex: Anhydrous toluene (3 mL) was added to the respective amine boranes (3 mmol, 1 equiv.) in a 15mL round bottom flask. Trifluoroacetic acid (4.5 mmol, 1.5 equiv.) was added to the above solution at 0 °C and the resulting reaction mixture was stirred at room temperature for 4-6 h. The completion of reaction was confirmed by ¹¹B-NMR spectroscopy as indicated by the disappearance of the amine borane peak (δ -13 to -19 ppm) and appearance of the TFAB-amine peak (δ 0 to -5 ppm). Excess trifluoroacetic acid was then removed in vacuo to obtain the TFAB-amine complex.

Reductive amination of ketones with TFAB-ammonia:

Procedure A: To a 1M solution of TFAB-ammonia (3.3 mmol, 1.1 equiv.) in THF, ketone (6 mmol, 2 equiv.) and the amine (6 mmol, 2 equiv.) were added. The reaction mixture was stirred for 12-24 hrs at room temperature. Upon completion of the reaction, as revealed by TLC, the resulting mixture was quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄

and concentrated in vacuo to obtain the crude product, which was purified by silica gel flash chromatography to obtain the desired product.

Procedure B: To a 1M solution of TFAB-ammonia (3.3 mmol, 1.1 equiv.) in THF, ketone (9 mmol, 3 equiv.) and the amine (6 mmol, 2 equiv.) were added. The reaction mixture was refluxed for 6-12 hrs. Upon completion of the reaction, as revealed by TLC, the resulting mixture was cooled to RT and quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the crude product, which was purified by silica gel flash chromatography to obtain the desired product.

Reductive amination of aldehydes with TFAB-NEt₃

Procedure C: To a 1M solution of TFAB-NEt₃ (3.3 mmol, 1.1 equiv.) in Toluene, aldehyde (3 mmol, 1 equiv.) and the amine (3.3 mmol, 1.1 equiv.) were added. The reaction mixture was stirred at 60 °C for 12 h. Upon completion of the reaction, as revealed by TLC, the resulting mixture was cooled to RT and quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the crude product, which was purified by silica gel flash chromatography to obtain the desired product.

Reductive amination of aromatic ketones with TFAB-NEt₃

Procedure D: To a 1M solution of TFAB-NEt₃ (3.3 mmol, 1.1 equiv.) in Toluene, ketone (3 mmol, 1 equiv.) and the amine (3 mmol, 1 equiv.) were added. The reaction mixture was stirred at 80 °C for 12-18 h. Upon completion of the reaction, as revealed by TLC, the resulting mixture was cooled to RT and quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the crude product, which was purified by silica gel flash chromatography to obtain the desired product.

Reduction of Imines with TFAB-NH₃ and TFAB-Et₃N

Reduction of N-benzylidenebenzylamine with TFAB-NH₃ (2a):

N-benzylidenebenzylamine (3 mmol) was synthesised by stirring benzaldehyde (3 mmol) and benzylamine (1 equiv.) for 12h in presence of molecular sieves. A 1M solution of TFAB-ammonia (3 mmol, 1 equiv.) in THF was added to this mixture and the reaction was stirred for 6 hr at room temperature. The reaction was quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the product. ¹HNMR analysis shows a complete reduction of starting imine to corresponding amine (dibenzyl amine). This shows the efficacy of TFAB-NH₃ (**2a**) as a reagent for imine reduction.

Reduction of N-benzylidenebenzylamine with TFAB-Et₃N (2h):

N-benzylidenebenzylamine (3 mmol) was synthesised by stirring benzaldehyde (3 mmol) and benzylamine (3 mmol) for 12h in presence of molecular sieves. A 1M solution of TFAB-NEt₃ (3 mmol, 1 equiv.) in toluene was added to this mixture and the reaction was stirred at 60 °C for 6 h. Upon completion the resulting mixture was cooled to RT and quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the product. ¹HNMR analysis shows a complete reduction of starting imine to corresponding amine (dibenzyl amine). This shows the capability of TFAB-Et₃N (**2h**) as an efficient reagent for imine reduction.

Hydride analysis* of TFAB-NH₃ + Benzylamine polymer

A equimolar mixture of TFAB-NH₃ and benzylamine in 1M THF is prepared in a 3-dram vial with a septum inlet fitted with a connecting tube. The connecting tube was attached to an analytical gas burette filled with CuSO4 solution. 3 M HCl was syringed into the vial, dropwise, till no further gas evolution was observed. The gas evolved was measured using the analytical gas burette. The temperature of the reaction was maintained at 25 °C.

*(H. C. Brown, Organic Syntheses Via Boranes, Wiley, New York, 1975, ch. 9.)

- 6 -

Competitive reduction of ketone vs imine with TFAB-NH₃ (2a):



N-benzylidenebenzylamine was as mentioned in the above procedure. N-benzylidenebenzylamine (3 mmol, 1 equiv.) and cyclohexanone (3 mmol, 1 equiv.) were added to a 1M solution of TFAB-NH₃ (1.5 mmol, 0.5 equiv.) in THF. The reaction mixture was stirred for 12 hrs at room temperature. Upon completion of the reaction, the mixture was quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the crude product. ¹HNMR analysis shows that N-benzylidenebenzylamine (δ 4.84 ppm) is selectively reduced to dibenzyl amine (δ 3.83 ppm) and cyclohexanone remains unreacted with no reduction to cyclohexanol observed. Hence, TFAB-NH₃ exclusively reduces imines in presence of ketones.



SI Figure 1: ¹HNMR analysis of competitive reduction of ketone vs imine with 2a.

Reductive amination with polyaminoboranes: To a 1M solution of TFAB-amine (3mmol) in toluene, amine (6 mmol, 2 equiv.) was added. The reaction was stirred for 15 mins to obtain the corresponding aminoborane polymer. To this 1 equiv. of benzylamine and cyclohexanone were added and stirred for 24h at 60 °C. Upon completion the resulting mixture was cooled to RT and quenched with NaOH (5 mL, 3 M), diluted with water (10 mL), and extracted with diethyl ether (3 × 15 mL). The combined organic extracts were washed with brine (1 × 45 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo to obtain the product.

Comparison of reductive amination of acetophenone with benzylamine with reagents from literature.

No.	Reagent	Reaction condition	Yield
1	TFAB-NEt ₃	24h, 80 °C	99%
2	pyridine-BH ₃	-	10% ²⁹
3	2-picoline-BH ₃	72h, RT	87% ³⁰
4	5-ethyl-2-methylpyridine-BH ₃	4 h, 50 °C	70% ²⁹
5	$PhMe_2SiH + B(C_6F_5)_3$ catalyst	48h, 100 °C	75% ³¹
6	Hydrosilatrane	16h, 70 °C	50% ³²
7	1,2,3-triazole-boranes	5h, 60 °C	87% ³³

Green Matrix calculation - Process mass intensity (PMI)

(A) Reductive amination with TFAB-NH $_3$

N-benzylcyclohexanamine (6aa)

H N

		Equivalents	mmol	MW (g/	~~~~	
		used	тто	mol)	У	
	Cyclohexanone	2	6	98.14	0.589	
Reactants	Benzylamine	2	6	107.15	0.643	
	TFAB-NH ₃	1.1	3.3	142.87	0.471	
	THF				2.66	
Auvilion	NaOH			38	7.2	
Auxiliary	H ₂ O			18	10	
	diethyl ether				35	
	Na ₂ SO ₄				10	
Draduat	N-	0.90	F 24	189.30	1.010	
FIOUUCI	benzylcyclohexanamine	0.09	5.54			
	Total mass in process				67.573	
Coloulation	Mass of product				1.010	
Calculation	%Yield				89	
	РМІ				66.9	

(B) Reductive amination with $\mathsf{TFAB}-\mathsf{NEt}_3$

N-benzyl-1-phenylethan-1-amine (6ga)



		Equivalents	mmol	MW (g/	g
		used	mmoi	mol)	
Reactants	Acetophenone	1	3	120.15	0.360
	Benzylamine	1	3	107.15	0.643
	TFAB-NH ₃	1.1	3.3	227.03	0.749
	Toluene				2.59
Auxiliany	NaOH			38	7.2
Auxilialy	H ₂ O			18	10
	diethyl ether				35
	Na_2SO_4				10
Droduct	N-benzyl-1-	0.00	2.07	211.31	0 6 2 9
FIOUUCI	phenylethan-1-amine	0.99	2.97		0.020
Calculation	Total mass in process				67.17
	Mass of product				0.628
	%Yield				99
	РМІ				106.96

X-Ray crystallography:

Data were collected, reflections were indexed and processed, and the files scaled and corrected for absorption using APEX3 [1] and SADABS [2]. The space groups were assigned using XPREP within the SHELXTL suite of programs [3,4] and solved by direct methods using ShelXS [4] or dual methods using ShelXT [5] and refined by full matrix least squares against F² with all reflections using Shelxl2018 [6] using the graphical interface Shelxle [7]. If not specified otherwise H atoms attached to carbon, boron and nitrogen atoms as well as hydroxyl hydrogens were positioned geometrically and constrained to ride on their parent atoms. C-H bond distances were constrained to 0.95 Å for aromatic and alkene C-H and CH₂ and alkyne C-H moieties, and to 1.00, 0.99 and 0.98 Å for aliphatic C-H, CH₂ and CH₃ moieties, respectively. B-H bond distances were constrained to 1.00 Å for pyramidal (sp³ hybridized) R₃B-H moieties. N-H bond distances were constrained to 0.88 Å for planar (sp² hybridized) N-H, N-H⁺ and NH₂ groups. N-H bond distances were constrained to 0.91 Å for pyramidal (sp³ hybridized) ammonium NH₂⁺ and NH₃⁺ groups. O-H distances of alcohols were constrained to 0.84 Å. Methyl CH₃, ammonium NH₃⁺ and hydroxyl H atoms were allowed to rotate but not to tip to best fit the experimental electron density. H atoms of pyramidalized R₂NH and RNH₂ units were refined and N-H distances were restrained to 0.88(2) Å. Water H atom positions were refined and O-H distances were restrained to 0.84(2) Å. Where necessary, water H...H distances were restrained to 1.36(2) Å, and H atom positions were further restrained based on hydrogen bonding considerations. U_{iso}(H) values were set to a multiple of U_{ea}(C) with 1.5 for CH₃, NH₃⁺ and OH, and 1.2 for C-H, CH₂, B-H, N-H and NH₂ units, respectively.

Complete crystallographic data, in CIF format, have been deposited with the Cambridge Crystallographic Data Centre. CCDC-2175383 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

[2] Krause, L., Herbst-Irmer, R., Sheldrick, G.M. & Stalke, D. (2015). J. Appl. Cryst. 48, 3-10.

^[1] Bruker (2019). Apex3 v2019.1-0, SAINT V8.40A, Bruker AXS Inc.: Madison (WI), USA.

^[3] SHELXTL suite of programs, Version 6.14, 2000-2003, Bruker Advanced X-ray Solutions, Bruker AXS Inc., Madison, Wisconsin: USA

^[4] Sheldrick, G.M. A short history of SHELX. Acta Crystallogr A. 2008, 64(1), 112–122.

^[5] Sheldrick, G. M., "SHELXT--Integrated space-group and crystal-structure determination", Acta Crystallogr A. 2015, A71, 3-8.

^[6] a) Sheldrick, G.M. University of Göttingen, Germany, 2018. b) Sheldrick, G.M. Crystal structure refinement with SHELXL. Acta Crystallogr Sect C Struct Chem. 2015, 71(1), 3–8.

^[7] Hübschle, C.B., Sheldrick, G.M. & Dittrich, B. ShelXle: a Qt graphical user interface for SHELXL. J. Appl. Crystallogr. 2011, 44(6), 1281–1284.

Table SI-2. Experimental details

For all structures: monoclinic, $P2_1/n$, Z = 4. Experiments were carried out with Mo Ka radiation using a CCD area detector.

	(1957962)	(I)
Crystal data		
Chemical formula	C ₈ H ₁₉ BClNO ₅	C ₈ H ₁₈ BCl ₂ NO ₅
M _r	255.50	289.94
Temperature (K)	103	153
<i>a</i> , <i>b</i> , <i>c</i> (Å)	10.312 (2), 12.187 (2), 10.6044 (19)	8.6163 (11), 11.9140 (15), 13.6933 (16)
β (°)	100.686 (6)	99.807 (3)
$V(Å^3)$	1309.5 (4)	1385.1 (3)
μ (mm ⁻¹)	0.30	0.48
Crystal size (mm)	$0.42 \times 0.40 \times 0.38$	$0.36 \times 0.24 \times 0.22$
Data collection		
Absorption correction	Multi-scan SADABS V2014/4 (Bruker AXS Inc.)	Multi-scan <i>SADABS</i> V2014/5 (Bruker AXS Inc.)
T_{\min}, T_{\max}	0.53, 0.90	0.78, 0.90
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	14073, 2370, 1779	13711, 3186, 1698
R _{int}	0.134	0.070
(sin θ/λ) _{max} (Å-1)	0.600	0.650
Refinement		
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.075, 0.203, 1.02	0.056, 0.148, 0.91
No. of reflections	2370	3186
No. of parameters	145	313
No. of restraints	0	702
H-atom treatment	H-atom parameters constrained	H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{\text{max}}$, $\Delta \rho_{\text{min}}$ (e Å ⁻³)	0.52, -0.53	0.45, -0.46

Computer programs: *APEX2* (Bruker AXS Inc., 2014), *SAINT* V8.34A (Bruker AXS Inc., 2013), *SHELXT* (Bruker AXS Inc., 2014), *SHELXL2018*/3 (Sheldrick, 2018), *SHELXL2014*/7 (Sheldrick, 2014), *XP* (Bruker AXS Inc., 2014).

B1—O1	1.505 (4)	C4—O3	1.428 (4)
B1—N1	1.579 (4)	C4—C5	1.497 (5)
C1—C2	1.504 (5)	C5—O4	1.431 (4)
C1—Cl1	1.778 (3)	C6—O4	1.433 (4)
C2—O2	1.217 (4)	C6—C7	1.492 (5)
C2—O1	1.303 (4)	C7—O5	1.430 (4)
C3—O3	1.426 (4)	C8—O5	1.430 (4)
C3—C8 ⁱ	1.490 (4)		
O1—B1—N1	102.7 (2)	O4—C6—C7	109.6 (3)
C2—C1—C11	117.6 (2)	O5—C7—C6	107.8 (2)
O2—C2—O1	125.5 (3)	O5—C8—C3 ⁱ	109.0 (2)
O2—C2—C1	116.9 (3)	C2—O1—B1	118.2 (2)
O1—C2—C1	117.5 (3)	C3—O3—C4	110.8 (2)
O3—C3—C8 ⁱ	109.0 (3)	C5—O4—C6	110.3 (2)
O3—C4—C5	108.9 (2)	C8—O5—C7	110.8 (2)
O4—C5—C4	108.2 (3)		

Table SI-3. Selected geometric parameters: bond lengths (Å), bond angles (°)



SI Figure 2: Crystal structure of TFAB-NH₃ (2a) co-crystalized with 18-crown ether

Experimental: NMR Data

N H

N-benzylcyclohexanamine (6aa)

The product was prepared using *procedure A* and isolated in 89% yield (1.0 g, 5.34 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature¹.

¹H NMR (300 MHz, $CDCI_3$) δ 7.37 – 7.05 (m, 5H), 3.77 (d, J = 1.2 Hz, 2H), 2.45 (tt, J = 10.0, 3.6 Hz, 1H), 1.96 – 1.83 (m, 2H), 1.80 – 1.55 (m, 3H), 1.35 – 1.00 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 139.4, 132.3, 129.3, 128.4, 56.2, 50.3, 33.7, 26.3, 25.1.



N-benzylcyclopentanamine (6ba)

The product was prepared using *procedure A* and isolated in 83% yield (0.871 g, 4.98 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature¹.

¹H NMR (300 MHz, Chloroform-*d*) δ 7.25 (d, *J* = 8.9 Hz, 5H), 3.72 (d, *J* = 9.4 Hz, 2H), 3.06 (p, *J* = 6.6 Hz, 1H), 1.95 – 1.20 (m, 8H).

¹³C NMR (75 MHz, CDCl₃) δ 138.9, 132.4, 129.4, 128.4, 59.2, 52.0, 33.2, 24.2.

N-benzyl-3,3-dimethylbutan-2-amine (6ca)

The product was prepared using *procedure A* and isolated in 93% yield (1.148 g, 5.58 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature³.

¹H NMR (300 MHz, CDCl₃) δ 7.32 (d, *J* = 4.3 Hz, 4H), 7.28 – 7.17 (m, 1H), 4.18 – 3.40 (m, 2H), 2.89 – 2.29 (m, 1H), 1.56 – 1.38 (m, 1H), 1.42 – 1.18 (m, 7H), 1.08 (d, *J* = 6.2 Hz, 3H), 0.94 – 0.81 (t, *J* = 7.0 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 140.8, 128.3, 128.0, 126.7, 52.6, 51.5, 37.2, 32.2, 25.8, 22.8, 20.5, 14.2.

Ph N

N-benzylpropan-2-amine (6da)

The product was prepared using *procedure A* and isolated in 88% yield (0.787 g, 5.3 mmol) as a yellow liquid after purification via column chromatography on silica gel using hexane as an eluent. The spectral data matched with those reported in the literature⁵.

¹H NMR (300 MHz, Chloroform-*d*) δ 7.38 – 7.20 (m, 5H), 3.79 (s, 2H), 2.86 (p, *J* = 6.2 Hz, 1H), 1.69 (s, 1H), 1.11 (d, *J* = 6.2 Hz, 6H). ¹³C{1H} NMR (75 MHz, CDCl₃) δ 134.9, 129.3, 128.6, 128.0, 49.3, 48.1, 20.6.

N-benzyl-2-methylcyclohexan-1-amine (6ea)

The product was prepared using *procedure B* and isolated in 91% yield as a trans/cis (91:9) mixture (1.108 g, 5.5 mmol) as a yellow liquid after purification via column chromatography on silica gel using hexane as an eluent. The spectral data matched with those reported in the literature.⁴

¹H NMR (300 MHz, CDCl₃) δ 7.65 – 6.98 (m, 5H), 3.96 – 3.64 (m, 2H), 2.65 (dt, *J* = 8.0, 4.0 Hz, 1H), 1.96 (dq, *J* = 6.7, 3.4 Hz, 1H), 1.81 – 1.17 (m, 8H), 0.95 (t, *J* = 6.8 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 141.1, 128.2, 128.0, 126.6, 57.7, 51.2 33.0, 31.0, 28.4, 23.6, 22.5, 14.1.

N H H

N-benzyl-3,3-dimethylbutan-2-amine (6fa)

The product was prepared using *procedure A* and isolated in 82% yield (0.935 g, 4.92 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature².

¹H NMR (300 MHz, CDCl₃) δ 7.40 – 7.18 (m, 5H), 3.94 (d, *J* = 13.2 Hz, 1H), 3.66 (d, *J* = 13.2 Hz, 1H), 2.36 – 2.24 (m, 1H), 1.09 (d, *J* = 6.4 Hz, 3H), 0.87 (s, 9H).

¹³C NMR (75 MHz, CDCl₃) δ 141.2, 128.2, 128.1, 126.6, 61.2, 52.7, 34.5, 26.6, 14.8.

N-(2-methoxybenzyl)cyclohexanamine (6ab)

The product was prepared using *procedure A* and isolated in 75% yield (0.977 g, 4.5 mmol) as a orangeyellow liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature¹⁵.

¹H NMR (300 MHz, CDCl₃) δ 7.29 – 7.16 (m, 2H), 6.96 – 6.81 (m, 2H), 3.82 (d, *J* = 6.6 Hz, 5H), 2.43 (tt, *J* = 9.7, 3.6 Hz, 1H), 1.98 – 1.84 (m, 2H), 1.80 – 1.54 (m, 4H), 1.35 – 1.03 (m, 5H).

 ^{13}C NMR (75 MHz, CDCl_3) δ 157.5, 129.6, 128.8, 127.9, 120.3, 110.1, 55.9, 55.3, 46.3, 33.6, 26.3, 25.1.

N-(3-methoxybenzyl)cyclohexanamine (6ac)

The product was prepared using *procedure A* and isolated in 86% yield (1.130 g, 5.2 mmol) as a yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature⁶.

 1 H NMR (300 MHz, CDCl₃) δ 7.26 – 7.67 (m, 4H), 3.8 (m, 5H), 2.4-2.5 (m, 1H), 1.96 – 1.83 (m, 2H), 1.80 – 1.55 (m, 3H), 1.35 – 1.00 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 159.6, 142.6, 129.3, 120.3, 56.2, 55.2, 51.1, 33.7, 26.3, 25.1.

MeO

N-(4-methoxybenzyl)cyclohexanamine (6ad)

The product was prepared using *procedure A* and isolated in 92% yield (1.314 g, 5.5 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature⁶.

¹H NMR (300 MHz, CDCl₃) δ 7.30 – 7.18 (m, 2H), 6.94 – 6.77 (m, 2H), 3.79 (s, 3H), 3.74 (s, 2H), 2.47 (tt, *J* = 10.0, 3.7 Hz, 1H), 1.98 – 1.84 (m, 2H), 1.80 – 1.67 (m, 2H), 1.68 – 1.54 (m, 1H), 1.33 – 0.92 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 158.3, 133.0, 129.2, 113.7, 56.1, 55.3, 50.5, 33.6, 26.3, 25.1.



N-(4-fluoro)cyclohexanamine (6ae)

The product was prepared using *procedure A* and isolated in 79% yield (0.973 g, 4.7 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature⁶.

¹H NMR (300 MHz, CDCl₃) δ 7.34 – 7.20 (m, 2H), 7.08 – 6.91 (m, 2H), 3.76 (s, 2H), 2.60 – 2.38 (m, 1H), 1.96 – 1.53 (m, 5H), 1.28 – 1.11 (m, 5H).

 ^{13}C NMR (75 MHz, CDCl_3) δ 163.3, 160.0, 136.6, 129.5, 129.4, 115.2, 114.9, 56.2, 50.4, 33.7, 26.3, 25.1.

N-(4-chloro)cyclohexanamine (6af)

The product was prepared using *procedure A* and isolated in 84% yield (1.124 g, 5 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature⁷.

¹H NMR (300 MHz, Chloroform-*d*) δ 7.52 – 7.09 (m, 4H), 3.77 (s, 2H), 2.45 (tt, *J* = 10.1, 3.7 Hz, 1H), 1.89 (dd, *J* = 11.8, 3.8 Hz, 2H), 1.81 – 1.69 (m, 2H), 1.66 – 1.54 (m, 1H), 1.38 – 0.87 (m, 5H). ¹³C{1H} NMR (75 MHz, CDCl₃) δ 139.4, 132.3, 129.3, 128.3, 56.1, 50.3, 33.6, 26.2, 25.1.

N-phenethylcyclohexanamine (6ag)

The product was prepared using *procedure A* and isolated in 81% yield (0.987 g, 4.9 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as an eluent. The spectral data matched with those reported in the literature⁸.

¹H NMR (300 MHz, CDCl₃) δ 7.43 – 7.10 (m, 5H), 3.03 – 2.70 (m, 4H), 2.42 (tt, *J* = 10.3, 3.7 Hz, 1H), 2.01 – 1.50 (m, 5H), 1.47 – 0.90 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 140.1, 128.6, 128.4, 126.0, 56.8, 48.4, 36.8, 33.7, 26.3, 25.2.

N-hexylcyclohexanamine (6ah)

The product was prepared using *procedure B* and isolated in 86% yield (0.952 g, 5.2 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5) as an eluent. The spectral data matched with those reported in the literature⁹.

¹H NMR (300 MHz, CDCl₃) δ 2.57 (t, *J* = 7.2 Hz, 2H), 2.36 (tt, *J* = 10.3, 3.7 Hz, 1H), 1.92 – 1.52 (m, 2H), 1.42 (q, *J* = 7.5 Hz, 2H), 1.36 – 1.20 (m, 3H), 1.25 – 0.90 (m, 12H), 0.90 – 0.77 (m, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 57.0, 47.2, 33.8, 31.9, 30.6, 27.3, 26.3, 25.2, 22.7, 14.2.

N-hexylcyclohexanamine (6ai)

The product was prepared using *procedure B* and isolated in 71% yield (0.771 g, 4.26 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹¹.

¹H NMR (300 MHz, CDCl₃) δ 2.53 (ttd, *J* = 9.9, 3.8, 1.7 Hz, 2H), 1.92 – 1.52 (m, 10H), 1.34 – 0.91 (m, 10H).

¹³C NMR (75 MHz, CDCl₃) δ 53.1, 34.4, 26.3, 25.4.



N-hexylcyclohexanamine (6aj)

The product was prepared using *procedure B* and isolated in 78% yield (0.721 g, 4.69 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹¹.

¹H NMR (300 MHz, CDCl₃) δ 2.42 (tt, *J* = 10.6, 3.9 Hz, 1H), 2.07 – 1.07 (m, 24H). ¹³C NMR (75 MHz, CDCl₃) δ 60.3, 36.9, 32.1, 29.3, 27.9, 25.8, 24.3.



N-hexylcyclohexanamine (6ak)

The product was prepared using *procedure D* for 18h and isolated in 72% yield (0.856 g, 4.32 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹².

¹H NMR (300 MHz, CDCl₃) δ 2.61 – 2.43 (m, 4H), 2.22 (dt, *J* = 5.8, 2.6 Hz, 1H), 1.90 – 0.81 (m, 16H). ¹³C NMR (75 MHz, CDCl₃) δ 64.4, 53.1, 50.1, 34.5, 28.9, 26.6, 26.6, 26.3, 25.4, 25.0.



N-benzyl-N-methylcyclohexanamine (6al)

The product was prepared using *procedure B* and isolated in 82% yield (0.998 g, 4.92 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹¹.

¹H NMR (300 MHz, CDCl₃) δ 7.65 – 6.93 (m, 5H), 3.57 (s, 2H), 2.09 – 1.75 (m, 1H), 2.19 (s, 3H), 1.64 (dtd, *J* = 11.6, 3.5, 2.1 Hz, 4H), 1.43 – 1.02 (m, 6H).

¹³C NMR (75 MHz, CDCl₃) δ 140.3, 128.7, 128.1, 126.6, 62.6, 57.9, 37.8, 28.8, 26.6, 26.2.

N-(heptan-2-yl)aniline (6cm)

The product was prepared using *procedure A* and isolated in 91% yield (1.050 g, 5.46 mmol) as a brown oil after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹³.

¹H NMR (300 MHz, CDCl₃) δ 7.29 – 7.10 (m, 2H), 6.73 – 6.53 (m, 3H), 3.47 (m, 2H), 1.64 – 1.38 (m, 1H), 1.44 – 1.30 (m, 3H), 1.35 – 1.23 (m, 4H), 1.19 (dd, *J* = 6.2, 1.4 Hz, 3H), 0.96 – 0.85 (m, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 147.6, 129.2, 116.7, 113.0, 48.5, 37.3, 32.0, 26.0, 22.8, 20.9, 14.2.



N-cyclohexylaniline (6am)

The product was prepared using *procedure A* and isolated in 89% yield (0.934 g, 5.34 mmol) as a yellow oil after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹¹.

¹H NMR (300 MHz, CDCl₃) δ 7.29 – 7.10 (m, 2H), 6.75 – 6.62 (m, 1H), 6.67 – 6.52 (m, 2H), 3.56 – 3.50 (m, 1H), 3.28 (tt, J = 10.1, 3.8 Hz, 1H), 2.16 – 2.01 (m, 2H), 1.87 – 1.60 (m, 3H), 1.50 – 1.08 (m, 5H). ¹³C NMR (75 MHz, CDCl₃) δ 147.3, 129.2, 116.8, 113.1, 51.7, 33.6, 26.1, 25.2.



N-cyclohexyl-4-nitroaniline (6ao)

The product was prepared using *procedure B* and isolated in 71% yield (0.937 g, 4.26 mmol) as a yellow solid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as the eluent. The spectral data matched with those reported in the literature.¹⁵

¹H NMR (300 MHz, CDCl₃) δ 8.06 (d, *J* = 9.2 Hz, 2H), 6.49 (d, *J* = 9.2 Hz, 2H), 4.43 (s, 2H), 2.12 – 1.98 (m, 3H), 1.86 – 1.60 (m, 5H), 1.50 – 1.13 (m, 9H).

¹³C NMR (75 MHz, CDCl₃) δ 152.4, 137.2, 126.5, 111.1, 51.6, 32.9, 25.7, 24.9.

N-cyclohexyl-4-cyanoaniline (6an)

The product was prepared using *procedure B* and isolated in 78% yield (0.936 g, 4.68 mmol) as a yellow solid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as the eluent. The spectral data matched with those reported in the literature.¹⁵

¹H NMR (300 MHz, CDCl₃) δ 7.17 (tdd, *J* = 7.5, 1.2, 0.6 Hz, 1H), 6.92 – 6.82 (m, 1H), 6.80 – 6.69 (m, 2H), 3.84 (s, 1H), 3.21 (tt, *J* = 10.1, 3.8 Hz, 1H), 2.09 – 1.94 (m, 2H), 1.88 – 1.59 (m, 3H), 1.47 – 1.06 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 147.5, 129.8, 119.9, 119.6, 117.4, 114.9, 112.8, 51.5, 33.1, 25.9, 25.0.



N-cyclohexyl-2-nitroaniline (6ap)

The product was prepared using *procedure C* for 18h and isolated in 82% yield (0.541 g, 2.46 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc (1:1) as an eluent. The spectral data matched with those reported in the literature.¹⁵

¹H NMR (300 MHz, CDCl₃) δ 8.15 (td, *J* = 7.1, 3.3 Hz, 2H), 7.39 (ddd, *J* = 8.3, 6.6, 1.4 Hz, 1H), 6.59 (ddt, *J* = 8.8, 6.9, 1.0 Hz, 1H), 3.51 (d, *J* = 8.9 Hz, 1H), 2.12 – 2.01 (m, 2H), 1.71 – 1.61 (m, 1H), 1.53 – 1.22 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 144.7, 135.9, 131.4, 127.0, 114.7, 114.1, 51.0, 32.8, 25.7, 24.7.



N-(prop-2-yn-1-yl)cyclohexanamine (6as)

The product was prepared using *procedure A* and isolated in 92% yield (0.756 g, 5.52 mmol) as a brown liquid after purification via column chromatography on silica gel using hexane/EtOAc 95:5 as the eluent. The spectral data matched with those reported in the literature.¹⁵

¹H NMR (300 MHz, CDCl₃) δ 3.41 (dd, *J* = 2.4, 0.8 Hz, 2H), 2.69 – 2.43 (m, 1H), 2.16 (td, *J* = 2.4, 0.8 Hz, 1H), 1.94 – 1.49 (m, 4H), 1.35 – 0.92 (m, 5H).

¹³C NMR (75 MHz, CDCl₃) δ 82.5, 70.9, 54.9, 35.2, 33.1, 26.2, 24.9.

N-benzyl-1-phenylethan-1-amine (6ga)

The product was prepared using *procedure D* and isolated in 99% yield (0.626 g, 2.97 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature³.

¹H NMR (300 MHz, CDCl₃) δ 7.48 – 7.21 (m, 10H), 3.86 (q, *J* = 6.6 Hz, 1H), 3.76 – 3.58 (m, 2H), 1.61 (s, 1H), 1.41 (dd, *J* = 6.6, 0.5 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 145.5, 140.6, 128.4, 128.3, 128.1, 126.9, 126.8, 126.7, 57.6, 51.8, 24.7.

1-phenyl-N-(4-(trifluoromethyl)benzyl)ethan-1-amine (6ia)

The product was prepared using *procedure D* and isolated in 82% yield (0.686 g, 2.46 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature²³.

¹H NMR (300 MHz, CDCl₃) δ 7.62 (d, *J* = 8.1 Hz, 2H), 7.51 (d, *J* = 8.1 Hz, 2H), 7.42 - 7.20 (m, 5H), 3.90 (q, *J* = 6.5 Hz, 1H), 3.72 - 3.55 (m, 2H), 1.64 (s, 1H), 1.38 (dd, *J* = 6.6, 1.4 Hz, 3H).

 ^{13}C NMR (75 MHz, CDCl_3) δ 149.7, 140.2, 128.4, 128.0, 127.0, 126.9, 125.4, 125.4, 57.3, 51.8, 24.7.



N-(4-methoxybenzyl)-1-phenylethan-1-amine (6ha)

The product was prepared using *procedure D* and isolated in 99% yield (0.715 g, 2.97 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature²⁴.

¹H NMR (300 MHz, CDCl₃) δ 7.61 – 7.12 (m, 7H), 7.02 – 6.87 (m, 2H), 3.93 – 3.76 (m, 2H), 3.84 (s, 3H), 3.75 – 3.56 (m, 2H), 1.40 (d, *J* = 6.6 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 158.5, 140.7, 137.6, 128.3, 128.1, 127.7, 126.8, 113.8, 56.9, 55.3, 51.7, 24.7.



N-benzyl-2,2,2-trifluoro-1-phenylethan-1-amine (6ja)

The product was prepared using *procedure D* and isolated in 72% yield (0.552 g, 2.16 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc (1:1) as an eluent. The spectral data matched with those reported in the literature.⁶

¹H NMR (300 MHz, CDCl₃) δ 7.49 – 7.35 (m, 1H), 4.99 (q, J = 6.8 Hz, 0H), 3.43 (s, 0H), 3.14 (s, 0H). ¹³C NMR (75 MHz, CDCl₃) δ 193.10, 166.10, 134.45, 129.21, 128.61, 128.55, 128.40, 127.44, 73.12, 72.70, 72.28, 71.86, 50.30.



N-(1-phenylethyl)hexan-1-amine (6gh)

The product was prepared using *procedure D* and isolated in 87% yield (0.535 g, 2.61 mmol) as a pale blue liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature⁹.

¹H NMR (300 MHz, CDCl₃) δ 7.58 – 7.01 (m, 5H), 3.75 (q, *J* = 6.6 Hz, 1H), 2.59 – 2.34 (m, 2H), 1.57 – 1.17 (m, 14H), 0.88 (dt, *J* = 9.2, 4.5 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 145.8, 128.3, 126.7, 126.5, 58.5, 48.0, 31.9, 30.4, 27.2, 24.5, 22.7, 14.2.



N-benzyl-1,1-diphenylmethanamine (6ka)

The product was prepared using *procedure D* and isolated in 90% yield (0.737 g, 2.7 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature²⁵.

¹H NMR (300 MHz, CDCl₃) δ 7.83 – 6.87 (m, 15H), 4.94 (s, 1H), 3.83 (s, 2H), 1.94 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 143.9, 140.4, 128.5, 128.4, 128.2, 127.4, 127.0, 126.9, 66.6, 52.0.



N-benzyl-N-methyl-1-phenylethan-1-amine (6gl)

The product was prepared using *procedure D* and isolated in 96% yield (0.648 g, 2.88 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature²⁶.

¹H NMR (300 MHz, CDCl₃) δ 7.51 – 7.20 (m, 10H), 3.82 – 3.51 (m, 2H), 3.35 (d, *J* = 13.3 Hz, 1H), 2.19 (s, 3H), 1.47 (dt, *J* = 6.7, 0.8 Hz, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 144.1, 140.0, 128.7, 128.1, 127.6, 126.8, 126.7, 63.3, 59.0, 38.5, 18.6.

N-(1-phenylethyl)aniline (6gm)

The product was prepared using *procedure D* and isolated in 95% yield (0.561 g, 2.85 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (9:1) as an eluent. The spectral data matched with those reported in the literature²⁷.

¹H NMR (300 MHz, CDCl₃) δ 7.47 – 7.33 (m, 4H), 7.33 – 7.23 (m, 1H), 7.15 (t, J = 7.7 Hz, 2H), 6.70 (t, J = 7.3 Hz, 1H), 6.57 (d, J = 7.9 Hz, 2H), 4.54 (q, J = 6.7 Hz, 1H), 4.07 (s, 1H), 1.56 (d, J = 6.9 Hz, 3H) ¹³C NMR (75 MHz, CDCl₃) δ 147.2, 145.2, 129.1, 128.6, 126.8, 125.8, 117.2, 113.3, 53.5, 25.2.

1-(1-phenylethyl)piperidine (6gk)

The product was prepared using *procedure D* and isolated in 86% yield (0.487 g, 2.58 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc (8:2) as an eluent. The spectral data matched with those reported in the literature²⁸.

¹H NMR (300 MHz, CDCl₃) δ 7.38 – 7.17 (m, 5H), 3.47 – 3.34 (m, 1H), 2.39 (tp, *J* = 11.5, 5.2 Hz, 4H), 1.57 (dq, *J* = 10.7, 5.5 Hz, 6H), 1.49 – 1.35 (m, 3H).

 ^{13}C NMR (75 MHz, CDCl_3) δ 143.8, 127.9, 127.7, 126.6, 65.3, 51.6, 26.4, 24.8, 19.6.

N-benzyl-N-methylhexan-1-amine (7bl)

The product was prepared using *procedure C* and isolated in 88% yield (0.541 g, 2.64 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹⁶.

¹H NMR (300 MHz, CDCl₃) δ 7.38 – 7.18 (m, 5H), 3.48 (s, 2H), 2.42 – 2.31 (m, 2H), 2.19 (s, 3H), 1.51 (qd, *J* = 6.7, 3.6 Hz, 2H), 1.41 – 1.20 (m, 6H), 0.95 – 0.84 (m, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 139.2, 129.0, 128.1, 126.7, 62.4, 57.7, 42.4, 31.9, 27.5, 27.3, 22.8, 14.2.

N,N-dibutyIhexan-1-amine (7br)

The product was prepared using *procedure C* and isolated in 76% yield (0.485 g, 2.28 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. HRMS (CI) m/z: [M+H]+ calcd. for C₁₄H₃₂N 214.2535, found 214.2535.

¹H NMR (300 MHz, CDCl₃) δ 2.36 (tt, *J* = 7.9, 2.3 Hz, 6H), 1.68 – 1.11 (m, 16H), 0.89 (t, *J* = 7.1 Hz, 9H).

¹³C NMR (75 MHz, CDCl₃) δ 54.3, 54.0, 32.0, 29.3, 27.4, 27.1, 22.8, 20.9, 14.2.

Benzylpiperidine (7ak)

The product was prepared using *procedure C* and isolated in 97% yield (0.509 g, 2.91 mmol) as a clear yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc (95:5) as an eluent. The spectral data matched with those reported in the literature¹⁸.

¹H NMR (300 MHz, Chloroform-*d*) δ 7.40 – 7.23 (m, 5H), 3.51 (s, 2H), 2.42 (t, *J* = 5.3 Hz, 4H), 1.71 – 1.58 (m, 4H), 1.48 (td, *J* = 6.1, 3.2 Hz, 2H).

¹³C{1H} NMR (75 MHz, CDCl₃) δ 138.6, 129.2, 128.0, 126.8, 64.0, 54.6, 26.2, 24.6.

N-benzyl-N-methyl-1-phenylmethanamine (7al)

The product was prepared using *procedure C* and isolated in 98% yield (0.620 g, 2.94 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹⁹.

 ^{1}H NMR (300 MHz, CDCl_3) δ 7.67 – 7.00 (m, 10H), 3.58 (s, 4H), 2.27 – 2.21 (m, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 139.3, 128.9, 128.2, 126.9, 61.9, 42.4.

N-benzyl-N-butylbutan-1-amine (7ar)

The product was prepared using *procedure C* and isolated in 99% yield (0.620 g, 2.94 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹⁷.

¹H NMR (300 MHz, CDCl₃) δ 7.67 – 7.00 (m, 5H), 3.58 (s, 2H), 2.27 – 2.21 (m, 4H), 1.5 – 1.3 (m, 8H), 0.9 (t, *J* = 6.5 Hz, 6H).

¹³C NMR (75 MHz, CDCl₃) δ 139.3, 128.7, 127.9, 126.5, 58.7, 53.6, 29.4, 20.8, 14.2.

Dibenzyl amine (7aa)

The product was prepared using *procedure C* and isolated in 54% yield (0.319 g, 1.54 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (95:5) as an eluent. The spectral data matched with those reported in the literature⁷.

¹H NMR (300 MHz, Chloroform-*d*) δ 7.42 – 7.25 (m, 10H), 3.85 (s, 4H), 2.09 (s, 1H). ¹³C{1H} NMR (75 MHz, CDCl₃) δ 140.0, 128.4, 128.2, 127.0, 53.1.

N-benzylhexan-1-amine (7ca)

The product was prepared using *procedure C* and isolated in 70% yield (0.401 g, 2.1 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (95:5) as an eluent. The spectral data matched with those reported in the literature²⁰.

¹H NMR (300 MHz, CDCl₃) δ 7.33 – 7.17 (m, 5H), 3.8 (s, 2H), 2.63 (t, *J* = 7.2 Hz, 2H), 1.62 – 1.39 (m, 2H), 1.36 – 1.28 (m, 6H), 0.92 – 0.83 (m, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 140.5, 128.3, 128.0, 126.8, 54.2, 49.6, 31.9, 30.2, 27.2, 22.8, 14.2.

N-benzylhexan-1-amine (7ch)

The product was prepared using *procedure C* and isolated in 77% yield (0.427 g, 2.31 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (95:5) as an eluent. The spectral data matched with those reported in the literature²¹.

¹H NMR (300 MHz, CDCl₃) δ 2.55 (t, *J* = 6.9 Hz, 4H), 1.53 – 1.36 (m, 4H), 1.28 (ddt, *J* = 10.0, 3.9, 2.0 Hz, 12H), 0.95 – 0.78 (m, 6H).

¹³C NMR (75 MHz, CDCl₃) δ 50.3, 31.9, 30.3, 27.2, 22.7, 14.1.

`N′

N-phenethylhexan-1-amine (7cg)

The product was prepared using *procedure C* and isolated in 72% yield (0.442 g, 2.16 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane/EtOAc (95:5) as an eluent. The spectral data matched with those reported in the literature¹⁷.

¹H NMR (300 MHz, CDCl₃) δ 7.88 – 6.98 (m, 5H), 2.97 – 2.73 (m, 4H), 2.69 – 2.55 (m, 2H), 1.37 – 1.18 (m, 8H), 0.95 – 0.82 (m, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 140.1, 128.6, 128.4, 126.0, 50.0, 36.5, 31.9, 30.2, 27.2, 22.7, 14.2, 1.2.

N-benzyl-1-cyclohexylmethanamine (7ba)

The product was prepared using *procedure C* and isolated in 84% yield (0.511 g, 2.5 mmol) as a pale yellow liquid after purification via column chromatography on silica gel using hexane as an eluent. The spectral data matched with those reported in the literature²².

¹H NMR (300 MHz, Chloroform-*d*) δ 7.31 (dd, *J* = 23.7, 4.5 Hz, 5H), 3.80 (s, 2H), 2.50 (d, *J* = 6.7 Hz, 2H), 1.89 – 1.63 (m, 4H), 1.62 – 1.46 (m, 1H), 1.42 – 1.09 (m, 4H), 0.97 (td, *J* = 11.7, 2.9 Hz, 2H). ¹³C{1H} NMR (75 MHz, CDCl₃) δ 140.6, 128.3, 128.1, 128.0, 126.8, 56.3, 54.3, 38.1, 31.9, 31.6, 26.9, 26.3.

N-(4-chlorobenzyl)-4-methoxyaniline (7dq)

The product was prepared using *procedure C* and isolated in 96% yield (0.714 g, 2.88 mmol) as a yellow liquid after purification via column chromatography on silica gel using hexane/EtOAc (90:10) as an eluent. The spectral data matched with those reported in the literature¹⁷.

¹H NMR (300 MHz, CDCl₃) δ 7.31 (s, 4H), 6.79 (d, *J* = 8.9 Hz, 2H), 6.59 (d, *J* = 8.9 Hz, 2H), 4.27 (s, 2H), 3.75 (s, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 152.2, 142.0, 138.2, 132.7, 128.7, 128.6, 114.9, 114.1, 55.9, 48.6.

1-(4-methoxybenzyl)piperidine (7ek)

The product was prepared using *procedure C* and isolated in 94% yield (0.578 g, 2.82 mmol) as a colourless liquid after purification via column chromatography on silica gel using hexane the eluent. The spectral data matched with those reported in the literature¹⁴.

¹H NMR (300 MHz, CDCl₃) δ 7.29 – 7.17 (m, 2H), 3.79 (d, *J* = 0.5 Hz, 3H), 3.42 (s, 2H), 2.36 (t, *J* = 5.3 Hz, 4H), 1.57 (p, *J* = 5.5 Hz, 4H), 1.42 (hept, *J* = 4.3 Hz, 2H).

 ^{13}C NMR (75 MHz, CDCl_3) δ 158.4, 130.3, 113.4, 63.2, 55.3, 54.4, 26.0, 24.5.

References:

- 1. O.-Y. Lee, K.-L. Law, D. Yang, Org. Lett. 2009, 11, 3302-3305.
- M. R. Adams, C. H. Tien, B. S. Huchenski, M. J. Ferguson, A. W. Speed, *Angew. Chem. Int. Ed.* 2017, 56, 6268-6271.
- 3. C. Wang, A. Pettman, J. Bacsa, J. Xiao, Angew. Chem. Int. Ed. 2010, 49, 7548-7552.
- 4. P. V. Ramachandran, P. D. Gagare, K. Sakavuyi, P. Clark, *Tetrahedron Lett.* **2010**, *51*, 3167-3169.
- 5. J. L. G. Ruano, A. Parra, J. Aleman, F. Yuste, V. M. Mastranzo, *ChemComm* **2009**, 404-406.
- Fleury-Brégeot, N., Raushel, J., Sandrock, D.L., Dreher, S.D. and Molander, G.A., *Chem. Eur. J.*, 2012 18, 9564-9570.
- 7. P.V. Ramachandran, S. Choudhary, A. Singh, J. Org. Chem. 2021, 86, 4274-4280.
- 8. K.-I. Fujita, Y. Enoki, R. Yamaguchi, *Tetrahedron*, 2008, 64, 1943-1954
- 9. D. Hollmann, A. Tillack, D. Michalik, R. Jackstell, M. Beller, Chem. Asian J. 2007, 2, 403-410.
- 10. S. B. Amin, S. Seo, T. J. Marks, Organometallics 2008, 27, 2411-2420.
- 11. Q. Zou, F. Liu, T. Zhao, X. Hu, Chem. Commun. 2021, 57, 8588-8591.
- 12. P. Spies, S. Schwendemann, S. Lange, G. Kehr, R. Fröhlich, G. Erker, *Angew. Chem. Int. Ed.* **2008**, *47*, 7543-7546.
- 13. P. Yin, T.-P. Loh, Org. Lett. 2009, 11, 3791-3793.
- 14. G. A. Molander, D. L. Sandrock, Org. Lett. 2007, 9, 1597-1600.
- 15. V. Airoldi, O. Piccolo, G. Roda, R. Appiani, F. Bavo, R. Tassini, S. Paganelli, S. Arnoldi, M. Pallavicini, C. Bolchi, *Eur. J. Org. Chem.* **2020**, *2020*, 162-168.
- 16. N. A. Sitte, M. Bursch, S. Grimme, J. Paradies, J. Am. Chem. Soc. 2018, 141, 159-162.
- 17. A. B. Enyong, B. Moasser, J. Org. Chem. 2014, 79, 7553-7563.
- 18. Molander, G. A.; Gormisky, P. E.; Sandrock, D. L., J.Org.Chem 2008, 73, 2052-2057
- 19. Zhao, Y., Foo, S.W. and Saito, S., Angew. Chem. Int. Ed., 2011, 50, 3006-3009
- M. S. Kwon, S. Kim, S. Park, W. Bosco, R. K. Chidrala, J. Park, *J. Org. Chem.* 2009, 74, 2877-2879.
- 21. L. L. Lorentz-Petersen, P. Jensen, R. Madsen, Synthesis 2009, 2009, 4110-4112.
- 22. Byun, E.; Hong, B.; De Castro, K. A.; Lim, M.; Rhee, H. J. Org. Chem. 2007, 72, 9815–9817.
- 23. S. Shirai, H. Nara, Y. Kayaki, T. Ikariya, Organometallics 2009, 28, 802-809.
- 24. K. Anil Kumar, T. Sreelekha, K. Shivakumara, K. Prakasha, D. C. Gowda, Synth. Commun. 2009, 39, 1332-1341.
- 25. Yang, X., Zhao, L., Fox, T., Wang, Z.-X. and Berke, H., *Angew. Chem. Int. Ed*, **2010**, 49: 2058-2062
- 26. X. Chen, Y. Ai, P. Liu, C. Yang, J. Yang, F. Li, J. Catal. 2021, 402, 325-334.
- 27. Z. Luo, S. Wan, Y. Pan, Z. Yao, X. Zhang, B. Li, J. Li, L. Xu, Q. H. Fan, *Asian J. Org. Chem.* **2022**, e202100707.
- P. Spies, S. Schwendemann, S. Lange, G. Kehr, R. Fröhlich, G. Erker, *Angew. Chem. Int. Ed.* 2008, 47, 7543-7546.

- 29. E. R. hardt and B. M. Coleridge, *Tetrahedron Lett*, 2008, **49**, 5152-5155.
- 30. S. Sato, T. Sakamoto, E. Miyazawa and Y. Kikugawa, *Tetrahedron*, 2004, **60**, 7899-7906.
- 31. V. Fasano, M. J. Ingleson, *Chem. Eur. J.* **2017**, 23, 2217.
- 32. S. E. Varjosaari, V. Skrypai, P. Suating, J. J. M. Hurley, A. M. D. Lio, T. M. Gilbert, M. J. Adler, *Adv. Synth. Catal.* **2017**, *359*, 1872.
- 33. W. Liao, Y. Chen, Y. Liu, H. Duan, J. L. Petersen and X. Shi, *Chem. Commun*, **2009**, 6436-6438.


































- 46 -





¹³C NMR (75 MHz, Chloroform-*d*) of 6ah







- 49 -





















- 57 -







¹³C NMR (75 MHz, Chloroform-d) of 7aa









¹³C NMR (75 MHz, Chloroform-*d*) of 7ba




























