

Combined crystallographic and computational investigation of the Solvent Disorder present in a new Tipiracil hydrochloride methanol solvate - hydrate

Rafael Barbas,¹ Mercè Font-Bardia,² Dafne de Sande,¹ Antonio Frontera^{3,*} and Rafel Prohens^{1,*}

¹ Unitat de Polimorfisme i Calorimetria, Centres Científics i Tecnològics, Universitat de Barcelona, Baldiri Reixac 10, 08028 Barcelona, Spain.

² Unitat de Difracció de Raigs X, Centres Científics i Tecnològics, Universitat de Barcelona, Spain.

³ Department of Chemistry, Universitat de les Illes Balears, Crta. de Valldemossa km 7.5, 07122 Palma, Spain

Electronic Supplementary Information

1. Crystal data and structure refinement	2
2. Characterization of Tipiracil hydrochloride methanol solvate–hydrate crystal form..	5
3. References	11

1. Crystal data and structure refinement

1.1 Tipiracil hydrochloride methanol solvate – hydrate (4:1:3, molar ratio)

Table S1. Crystal data and structure refinement for mo_023XB102A_0m_a.

Identification code	mo_023XB102A_0m_a		
Empirical formula	C ₃₇ H ₅₈ Cl ₈ N ₁₆ O ₁₂		
Formula weight	1202.59		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	Orthorhombic		
Space group	P 21 21 21		
Unit cell dimensions	a = 7.1291(2) Å	α = 90°.	
	b = 13.5358(5) Å	β = 90°.	
	c = 27.2056(11) Å	γ = 90°.	
Volume	2625.29(16) Å ³		
Z	2		
Density (calculated)	1.521 Mg/m ³		
Absorption coefficient	0.502 mm ⁻¹		
F(000)	1248		
Crystal size	0.290 x 0.073 x 0.062 mm ³		
Theta range for data collection	2.704 to 33.782°.		
Index ranges	-7 ≤ h ≤ 11, -19 ≤ k ≤ 21, -42 ≤ l ≤ 30		
Reflections collected	29776		
Independent reflections	10351 [R(int) = 0.0317]		
Completeness to theta = 25.242°	99.1 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7467 and 0.6558		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	10351 / 6 / 374		
Goodness-of-fit on F ²	1.048		
Final R indices [I > 2σ(I)]	R1 = 0.0498, wR2 = 0.1175		
R indices (all data)	R1 = 0.0593, wR2 = 0.1255		
Extinction coefficient	n/a		
Largest diff. peak and hole	1.507 and -0.872 e.Å ⁻³		

Table S2. Hydrogen bonds for Tipiracil hydrochloride methanol solvate - hydrate [Å and °]

Donor --- H...Acceptor	[ARU]	d(D – H)	d(H...A)	d(D...A)	<(D - H...A)
N1 --H1N ..Cl3	[1/2+x,1/2-y,1-z]	0.88	2.36	3.240(3)	175
N1 --H1N ..Cl3'	[1/2+x,1/2-y,1-z]	0.88	2.31	3.175(3)	167
N2 --H2N ..O3	[]	0.88	1.86	2.743(3)	176
O2W --H2W ..Cl3	[1+x,y,z]	0.84	1.81	2.578(7)	151
O2W --H2W ..Cl3'	[1+x,y,z]	0.84	2.49	3.236(7)	148
N4 --H4A ..O1	[1/2+x,1/2-y,1-z]	0.88	1.98	2.817(4)	159
N4 --H4B ..O2W	[]	0.88	1.86	2.722(6)	165
N9 --H9NA ..Cl3	[1+x,1+y,z]	0.88	2.22	3.017(5)	151

N9	--H9NA	..Cl3'	[1+x,1+y,z]	0.88	2.52	3.370(5)	162
N5	--H5N	..O1W	[]	0.88	1.90	2.766(3)	168
N9	--H9NB	..Cl4	[x,1+y,z]	0.88	2.30	3.155(3)	165
N6	--H6N	..Cl4	[2-x,1/2+y,3/2-z]	0.88	2.21	3.089(2)	174
O2W'	--H2A'	..Cl4	[]	0.79(4)	2.81(9)	3.089(6)	103(6)
O2W'	--H2A'	..N9	[x,-1+y,z]	0.79(4)	2.59(9)	3.047(8)	119(10)
O2W'	--H2B'	..Cl4	[]	0.80(6)	2.63(7)	3.089(6)	118(7)
O1W	--H1OA	..Cl4	[3/2-x,1-y,-1/2+z]	0.80(3)	2.44(3)	3.221(2)	165(3)
O1W	--H1OB	..O2	[]	0.796(17)	2.071(17)	2.855(3)	168(4)

1.2 Comparative of crystal data parameters of Tipiracil hydrochloride crystal forms

Table S3. Crystal data of Tipiracil hydrochloride crystal forms

Crystal data	US 9.527.833 B2		methanol solvate – hydrate [4:1:3]
	Crystal I	Crystal III	
Temperature (K)	291	- ¹	100
Crystal system	Monoclinic	Monoclinic	Orthorhombic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁	<i>P</i> 2 ₁ 2 ₁ 2 ₁
Unit cell dimensions	a (Å)	11.6006(9)	10.3221(14)
	b (Å)	10.3106(11)	9.8634(13)
	b (Å)	10.3036(10)	11.6643(16)
	α (°)	90	90
	β (°)	101.951(7)	100.317
	γ (°)	90	90
Volume (Å ³)	1205.7(2)	1169.5(3)	2625.29(16)

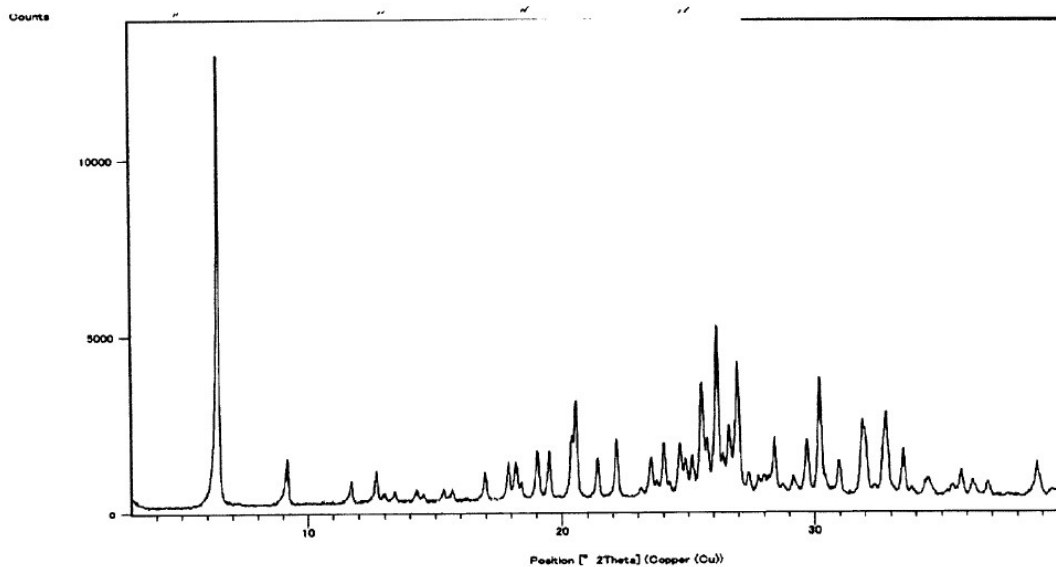
¹ “room temperature” is reported

1.3 Comparative XRPD of Tipiracil hydrochloride crystal II and methanol solvate – hydrate

Figure S1: (a) XRPD diffractogram of crystal II reported in patent US 9.527.833 B2 and (b) bulk powder of Tipiracil hydrochloride methanol solvate - hydrate

(a)

Fig. 2



(b)

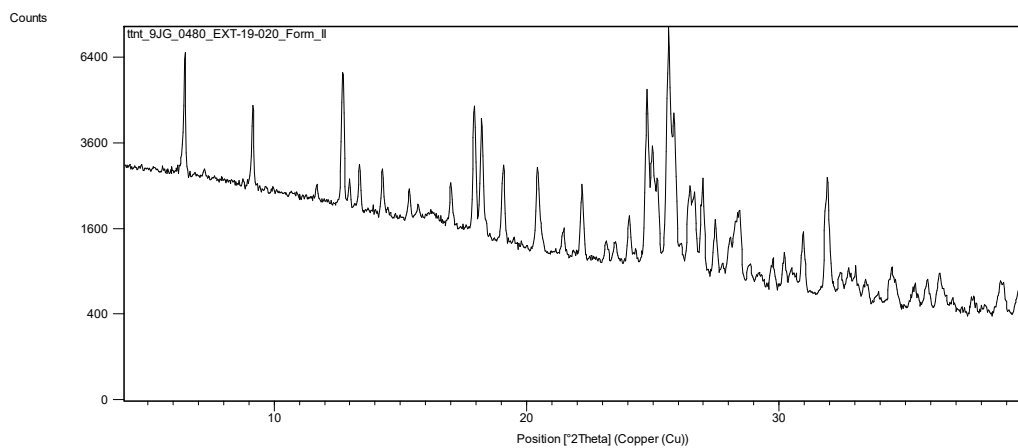


Table S4. 2 θ peaks of Tipiracil hydrochloride crystal II and methanol solvate – hydrate at 298K

Tipiracil hydrochloride	2 θ peaks					
Crystal II	6.5	20.6	25.5	26.1	27.0	30.2
methanol solvate – hydrate	6.5	20.6	25.6	26.1	27.0	30.2

2.- Characterization of the Tipiracil hydrochloride methanol solvate – hydrate crystal form

Powder X-ray Diffraction Analysis: the powder diffractogram was indexed and the lattice parameters were refined by means of LeBail fits by means of Dicvol04,¹ and the space groups was determined from the systematic absences.

Figure S2: The XRPD of Tipiracil hydrochloride methanol solvate - hydrate has been indexed with the following proposed orthorhombic cell: $a=27.27(7)$ Å, $b=13.613(2)$ Å, $c=7.1818(8)$ Å, $V=2666.5(9)$ Å³ (Figures of Merit: $M= 35$, $F= 84$), with number of impurities equal to zero. A $P 2_12_12_1$ space group is compatible with the cell and the cell volume is compatible with 4 molecules of Tipiracil hydrochloride, 1 molecule of methanol and 3 molecules of water. (R_{wp} : 3.17; R_{exp} : 2.40), $Z=2$

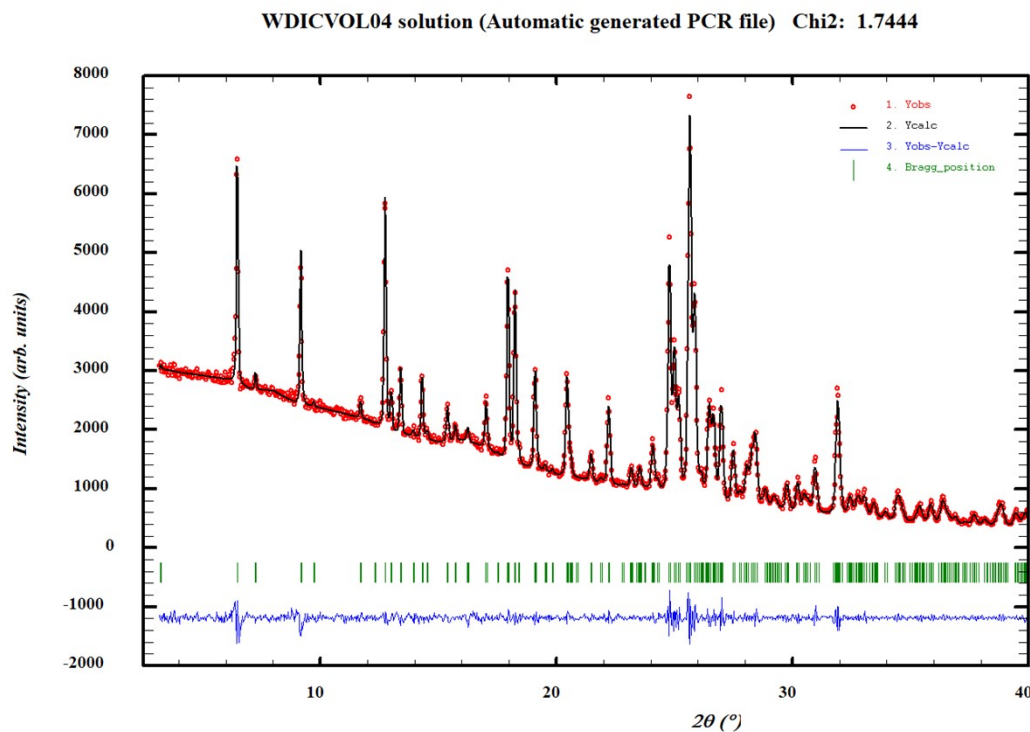


Figure S3: The structure was determined by single crystal X-ray diffraction at 100K showing a 4:1:3 (Tipiracil hydrochloride: methanol: water) molar ratio with the following orthorhombic cell: $a=7.1291(2)$ Å, $b=13.5358(5)$ Å, $c=27.2056(11)$ Å, $V=2625.29(12)$ Å³, $Z=2$ and $P 2_1 2_1 2_1$ space group. (R_{int} (%) = 3.17; R-Factor (%) = 5.10). Comparative XRPD diffractograms between bulk Tipiracil hydrochloride methanol solvate - hydrate and simulated from the cif is shown

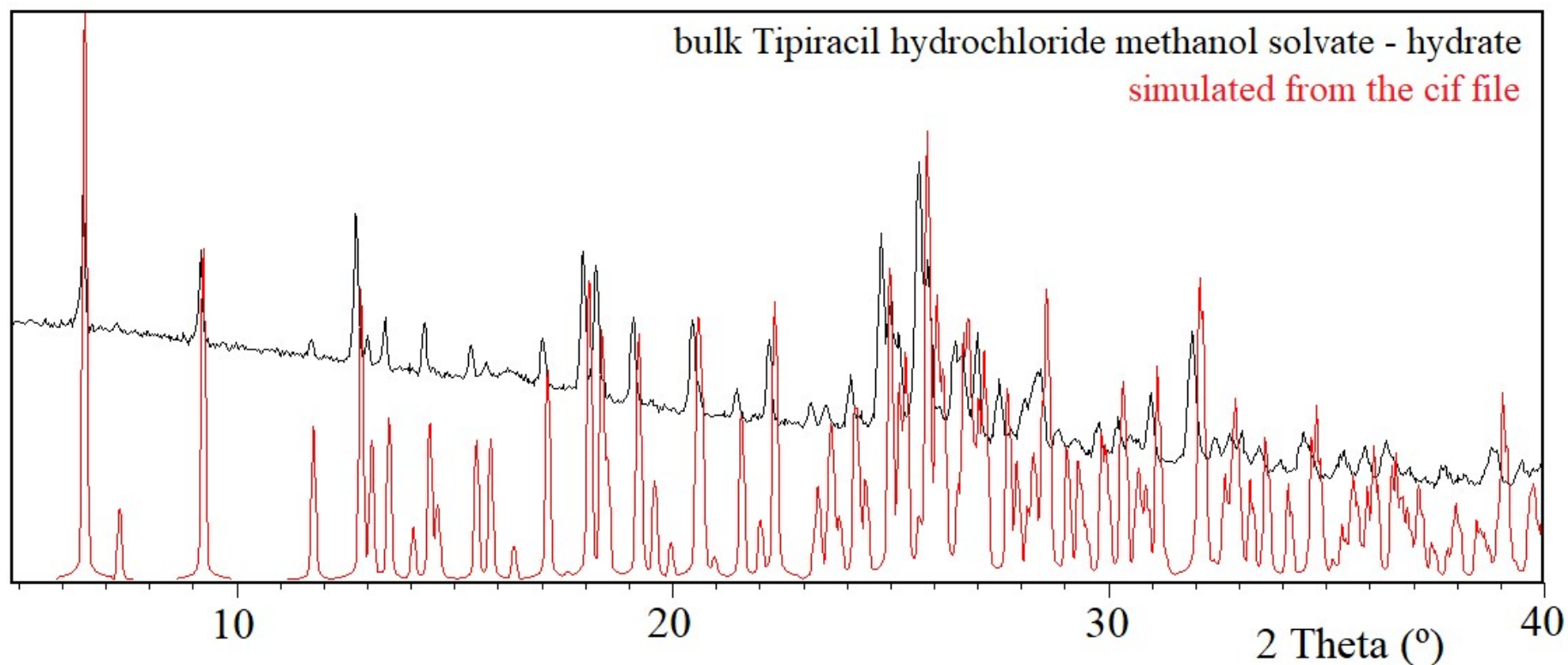


Figure S4: DSC of Tipiracil hydrochloride methanol solvate – hydrate. Differential scanning calorimetry analysis was carried out by means of a Mettler-Toledo DSC-822e calorimeter. Experimental conditions: aluminium crucibles of 40 μL volume, atmosphere of dry nitrogen with 50 mL/min flow rate, heating rate of 10 $^{\circ}\text{C}/\text{min}$. The calorimeter was calibrated with indium of 99.99% purity (mp: 156.6 $^{\circ}\text{C}$; ΔH : 28.31 J/g).

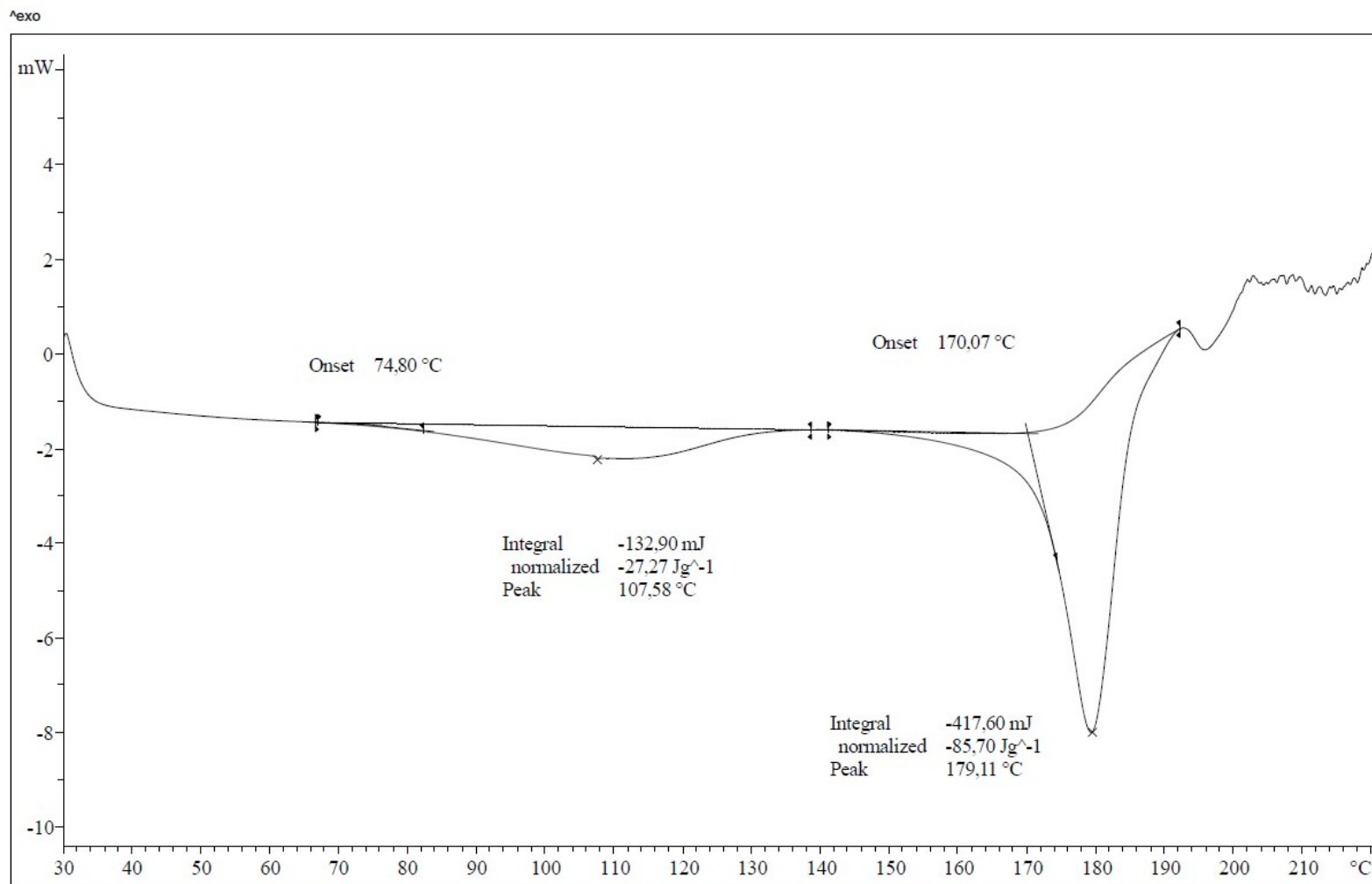


Figure S5: TGA of Tipiracil hydrochloride methanol solvate - hydrate

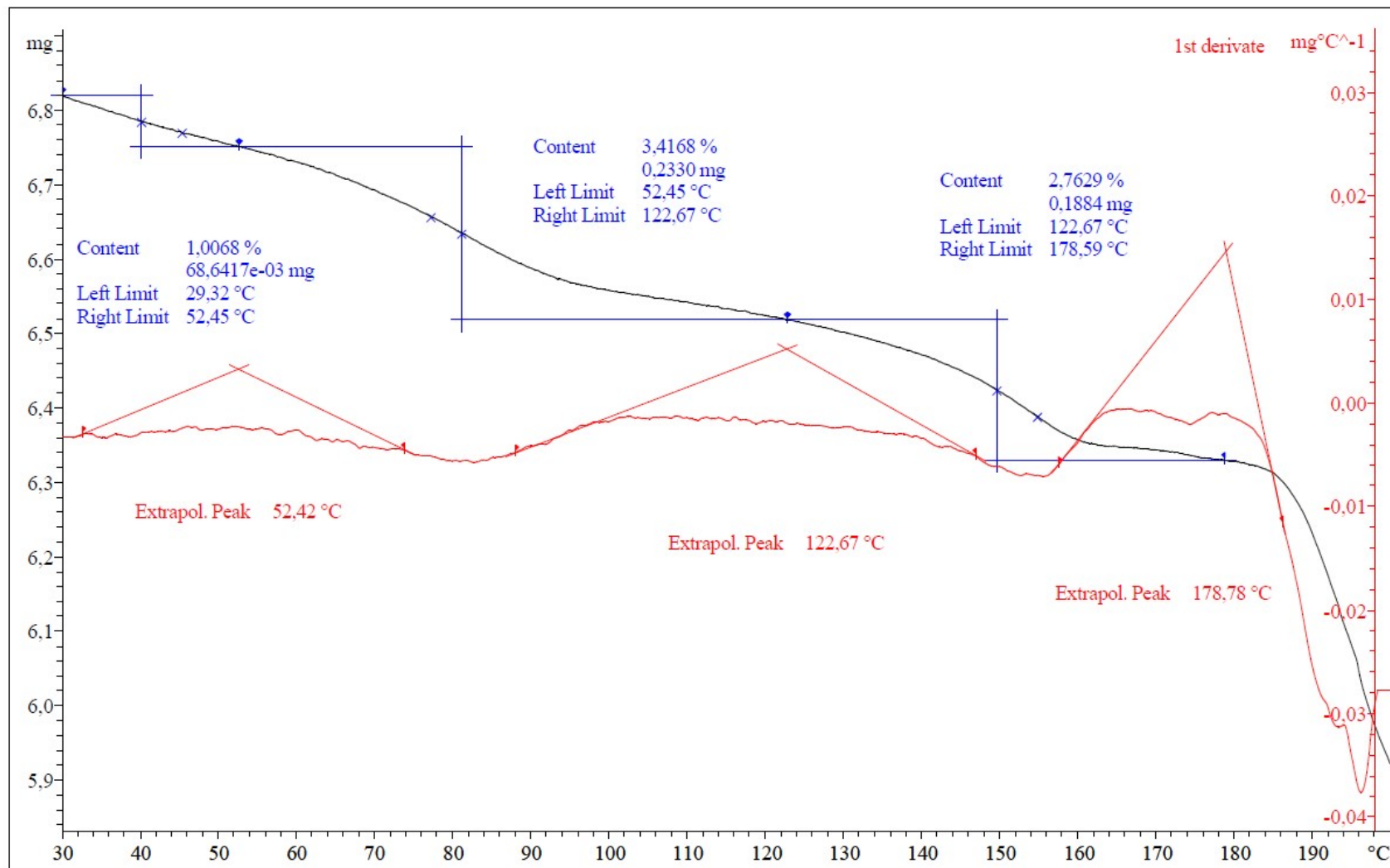
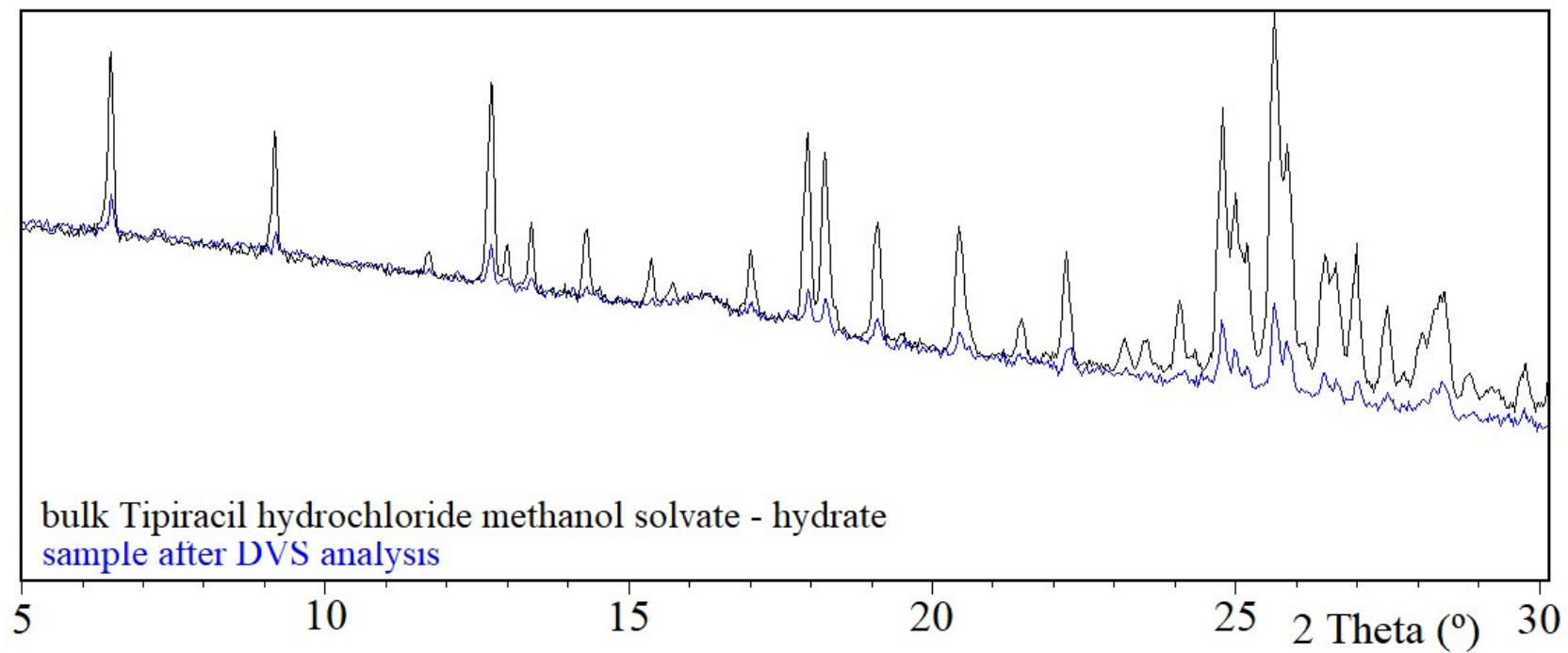


Figure S6: XRPD diffractograms of bulk Tipiracil hydrochloride methanol solvate - hydrate and the resulting solid after DVS analysis. Enlargement from 5 to 30° 2θ.



3. References

- [1] Boultif, A.; Louër, D. Indexing of powder diffraction patterns for low-symmetry lattices by the successive dichotomy method *J. Appl. Crystallogr.* 1991, **24**, 987-993.