

SUPPORTING INFORMATION

Energy framework and solubility: a new physicochemical approach in the evaluation of the structure-property of pharmaceutical solid forms

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COMPLEMENTARY FIGURES AND TABLES

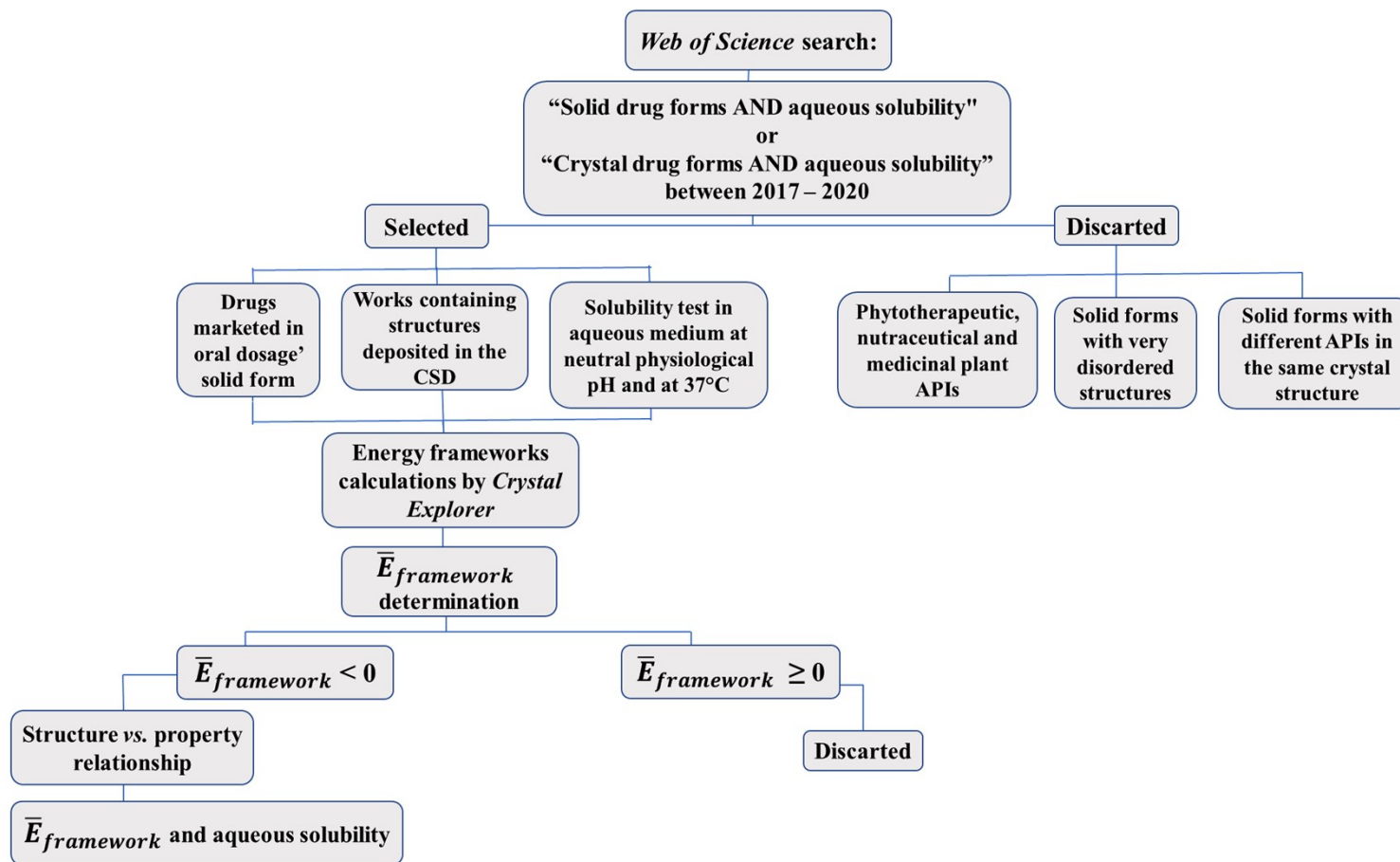


Figure S1. Steps of the methodology utilized in this work.

Table S1. Different solid forms selected for the analysis of structure-property correlation.

Drug	Solid form	Type of solid	Aqueous solubility (mg mL ⁻¹)*	HF $E_{framework}$ (KJ mol ⁻¹)	B3LYP $E_{framework}$ (KJ mol ⁻¹)	References
Acetazolamide (ACA)	ACA-form I	P	0.74	-27.44	-24.05	1-2
	ACA-proline	C	2.64	-15.5	-11.92	
	ACA-theophylline	C	2.18	-15.71	-16.33	
Albendazole (ABZ)	ABZ-form I	P	0.0033	-22.03	-22.04	3
	ABZ- hydrochloride	S	0.1	-11.09	-11.19	
Apatinib (APA)	APA- saccharinate	S	0.255	-9.58	-14.31	4
	APA-sebacylate	S	1.667	-14.84	-12.49	
	APA-succinate hydrated	S	0.989	-9.30	-13.94	
Apremilast (AP)	AP	P	0.004	-29.42	-25.07	5
Cilostazol (CLZ)	CLZ	P	0.0074	-28.56	-27.3	6
	CLZ-4-hydroxybenzoic acid	C	0.0703	-23.62	-23.08	
	CLZ-2,4-dihydroxybenzoic acid	C	0.1073	-23.79	-21.17	
	CLZ-2,5-dihydroxybenzoic acid	C	0.25382	-28.37	-26.37	
Chlorothiazide (CTZ)	CTZ-form 1	P	0.018	-28.07	-22.14	7
	CTZ-1,2-di(4-pyridyl)ethylene	C	0.028	-18.38	-12.27	
Doxazosin (DXZ)	DXZ - free-base	P	0.00006	-30.3	-28.36	8
	DXZ- mesylate polymorph A	S	0.0367	-12.14	-12.86	
	DXZ- mesylate hydrate	S	0.07434	-16.44	-14.08	
Desloratadine (DES)	DES	P	0.32	-20.83	-20.21	9

Dipfluzine (DF)	DF-fumarate	S	0.01911	-17.67	-20.25	10
Entacapone (ETP)	ETP	P	0.08	-31.73	-29.73	11
Epalrestat (EPR)	EPR	P	0.002956	-22.47	-21.27	12-13
	EPR-caffeine	C	0.004557	-15.78	-19.05	
Febuxostat (FEB)	FEB- isonicotinamide	C	0.0131	-18.78	-18.73	14-16
	HFEB-di-2-pyridylamine	S	0.1419	-7.52	-8.78	
	FEB	P	0.0108	-20.89	-20.00	
Furosemide (FSM)	FSM-form 1	P	0.081	-10.83	-9.07	17-18
	FSM-5-fluorocytosine monohydrated	C	0.22	-22.96	-21.92	
	FSM-triethanolamine	S	4.3	-7.75	-10.18	
Glibenclamide (GCM)	GCM	P	0.0101	-27.99	-25.72	19
Glimepiride (GLIM)	GLIM- form I	P	0.0464	-35.63	-32.73	20
	GLIM- form II	P	0.133	-37.47	-33.67	
Indomethacin (INC)	INC-DL-proline hydrate	C	0.00377	-24.81	-25.32	21-22
	INC- polymorph γ	P	0.00186	-26.56	-24.99	
	(INC) ₂ -(DPEN) ₂ -Dihydrate	C	0.3598	-1.83	-4.28	
Ketoconazole (KTZ)	KTZ	P	0.012	-31.78	-30.82	23
	KTZ-2,4 dihydroxybenzoic acid	C	0.1793	-17.11	-16.784	
	KTZ-3,4-dihydroxybenzoic acid	C	0.4858	-24.12	-21.83	
	KTZ- <i>trans</i> -4-hydroxycinnamic acid	C	0.0815	-26.80	-24.55	
Leflunomide (LEF)	LEF-form I	P	0.03165	-26.74	-24.85	24
	LEF-3-hydroxybenzoic acid	C	0.03658	-20.46	-18.08	
	LEF-pyrogallol	C	0.04463	-10.93	-11.27	

Lenalidomide (LEN)	LEN-hemihydrate	P/H	0.65	-11.89	-10.85	25
	LEN-acesulfame	S	1.35	-18.62	-17.39	
Riluzole (RZ)	RZ	P	0.35	-13.39	-12.86	26
	RZ-glutaric acid	C	0.62	-18.87	-17.38	
	RZ-sorbic acid	C	1.98	-10.64	-10.82	
	RZ-malonic acid	S	3.21	-10.39	-10.83	
Sulfametoxazole (SMZ)	SMZ-form I	P	0.705	-27.84	-23.42	27
	SMZ-form II	P	1.447	-23.91	-20.74	
	SMZ-benzamidine	S	3.289	-6.56	-7.11	
	SMZ-1,2-di(4-pyridyl)ethylene	C	0.334	-12.43	-10.50	
Teriflunomide (TFM)	TFM	P	0.637	-9.93	-8.66	28
	TFM-diethanolamine	S	1.785	-7.16	-7.62	
	TFM-monoethanolamine	S	4.202	-11.68	-9.97	
Telmisartan (TEL)	TEL	P	0.07	-39.72	-40.75	29
	TEL-gentisic acid	C	0.26	-21.62	-23.71	
Tinidazole (TNZ)	TMZ-salicylic acid	C	3.807	-20.11	-17.16	30
Zonisamide (ZNS)	ZNS	P	0.8466	-24.42	-20.12	31
	ZNS-caffeine	C	0.7418	-7.00	-9.4	

For better visualization, the data were grouped by drug and in alphabetical order.

P=pure; C=cocrystal; S=salt and H=hydrate

* Solubility data were standardized in mg mL⁻¹.

Calculation:

$$\bar{E}_{framework} = \frac{\sum (E_{total} \times n^{\circ} \text{interactions' type})}{N^{\circ} \text{Total interactions}}$$

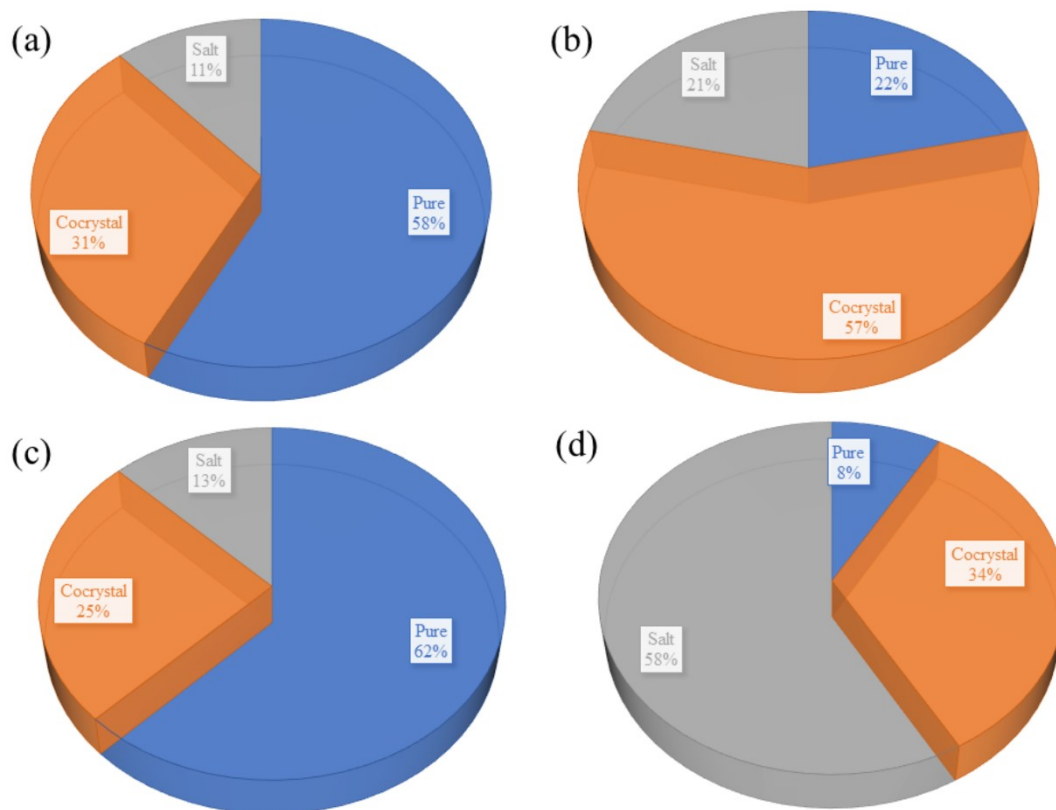


Figure S2. Distribution of the 60 solid forms selected in the four solubility ranges: (a) less than 0.10 mg mL⁻¹; (b) from 0.10 to 0.50 mg mL⁻¹; (c) from 0.50 to 1.00 mg mL⁻¹ and (d) from 1.00 to 5.00 mg mL⁻¹.

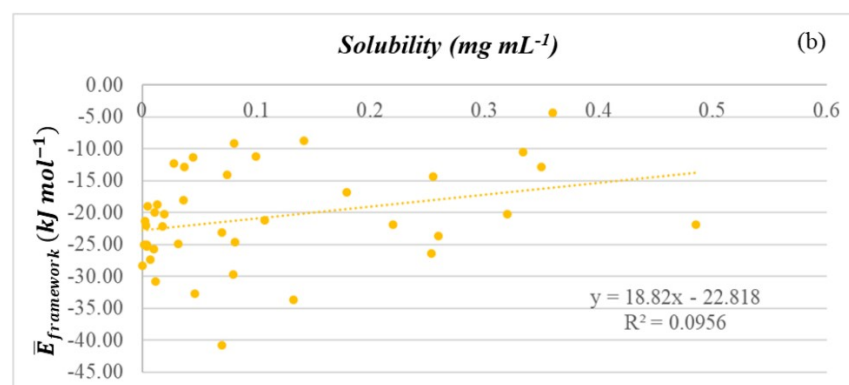
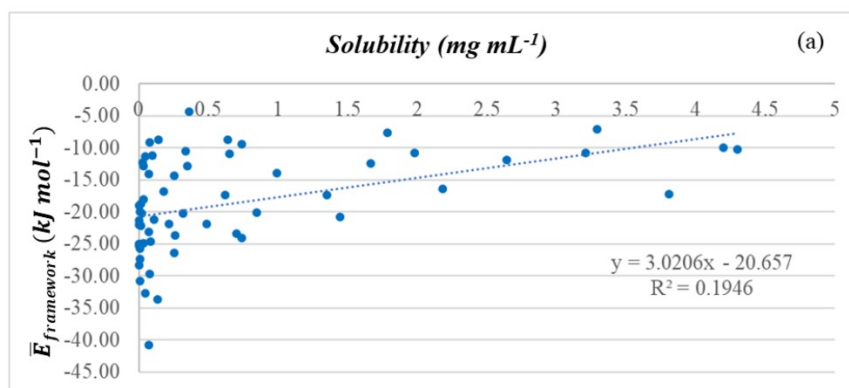


Figure S3. Linear representation of solubility and average of energy framework ($E_{framework}$) for 60 solid forms investigated a) showing the dispersion of points a) within the entire range of solubility studied ($0.00006 - 4.3 \text{ mg mL}^{-1}$) and b) only in the range of greatest variability between the points ($\leq 0.5 \text{ mg mL}^{-1}$). The energy values were obtained at the B3LYP/6-31G(d,p) level.

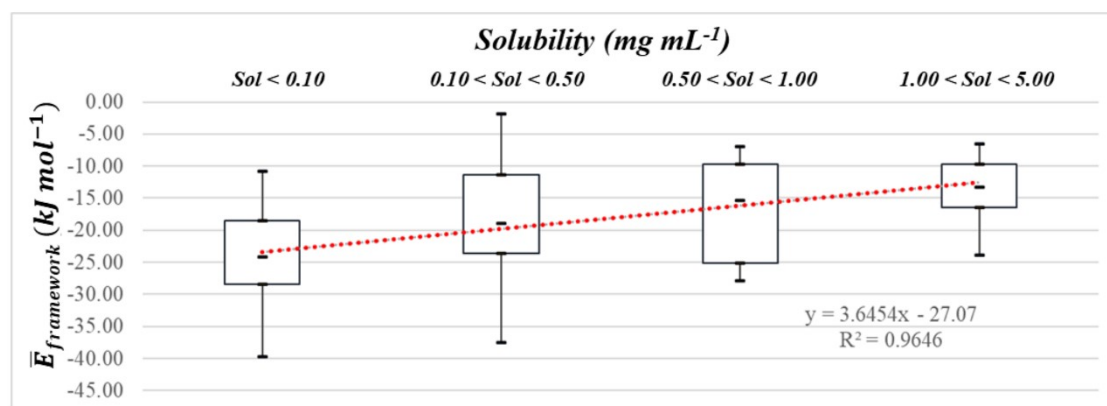


Figure S4. Box plot representation of solubility and average of energy framework ($E_{framework}$) for 60 solid forms investigated. The energy values were obtained at the HF/3-21G level.

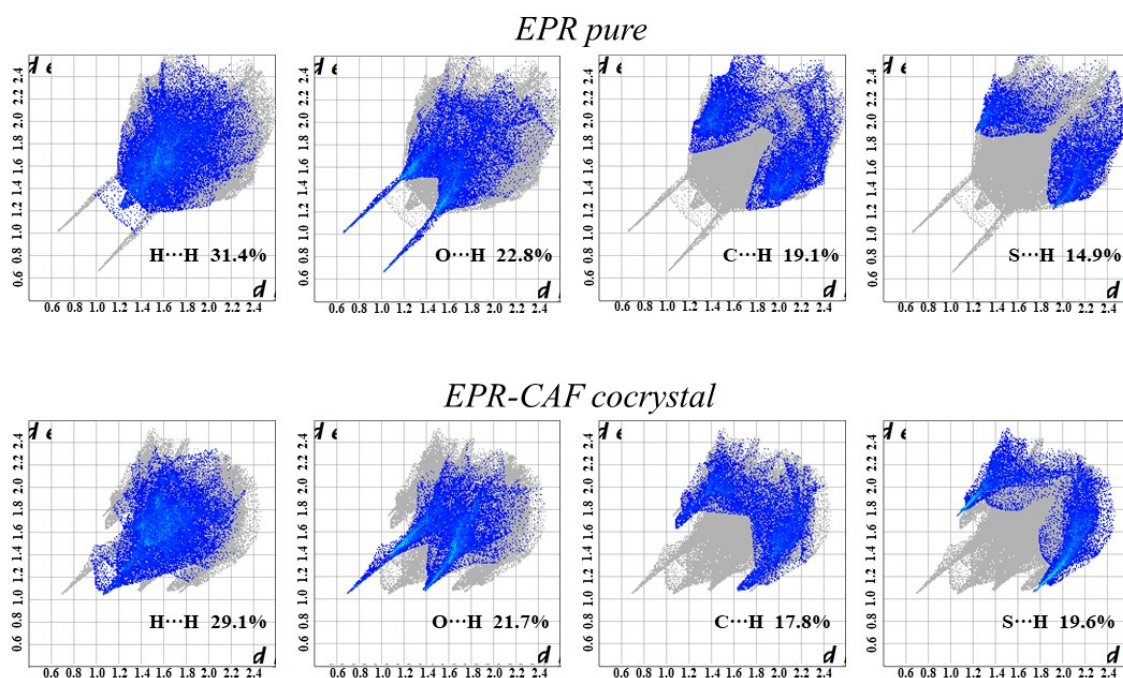


Figure S5. Partial fingerprint plots showing the percentual contributions of the main intermolecular contacts to the Hirshfeld surface area of the forms EPR (pure) and EPR-CAF cocrystal.

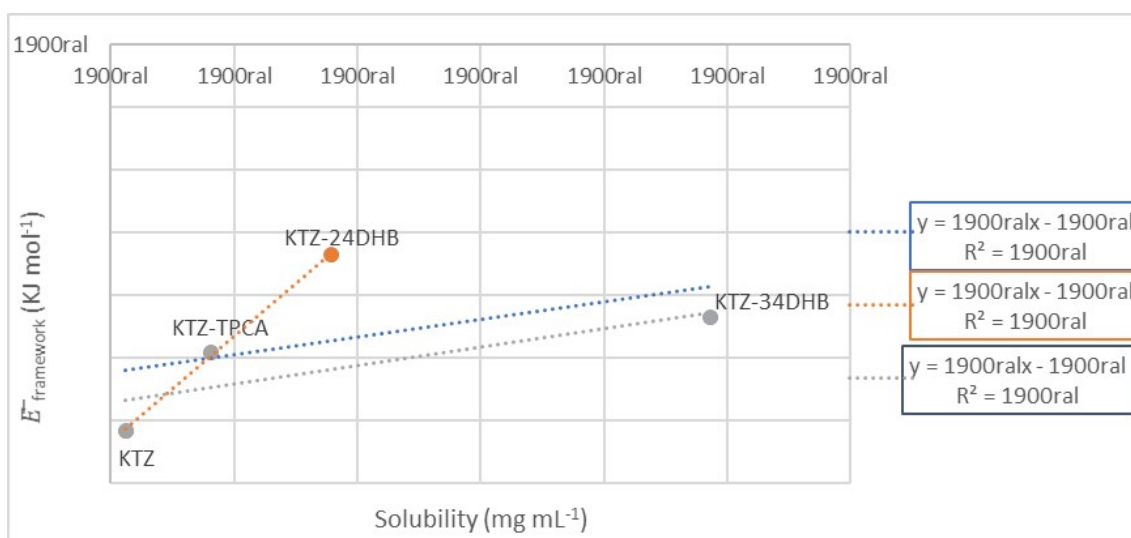


Figure S6. Correlation between the $E_{framework}$ and aqueous solubility for the four solid forms of ketoconazole studied. The blue dotted line corresponds to the linear trend between the 4 points; the orange dotted line represents the linear trend for the first three points (KTZ; KTZ-TPCA and KTZ-24DHB) and the gray dotted line corresponds to the linear trend between KTZ; KTZ-TPCA and KTZ-34DHB).

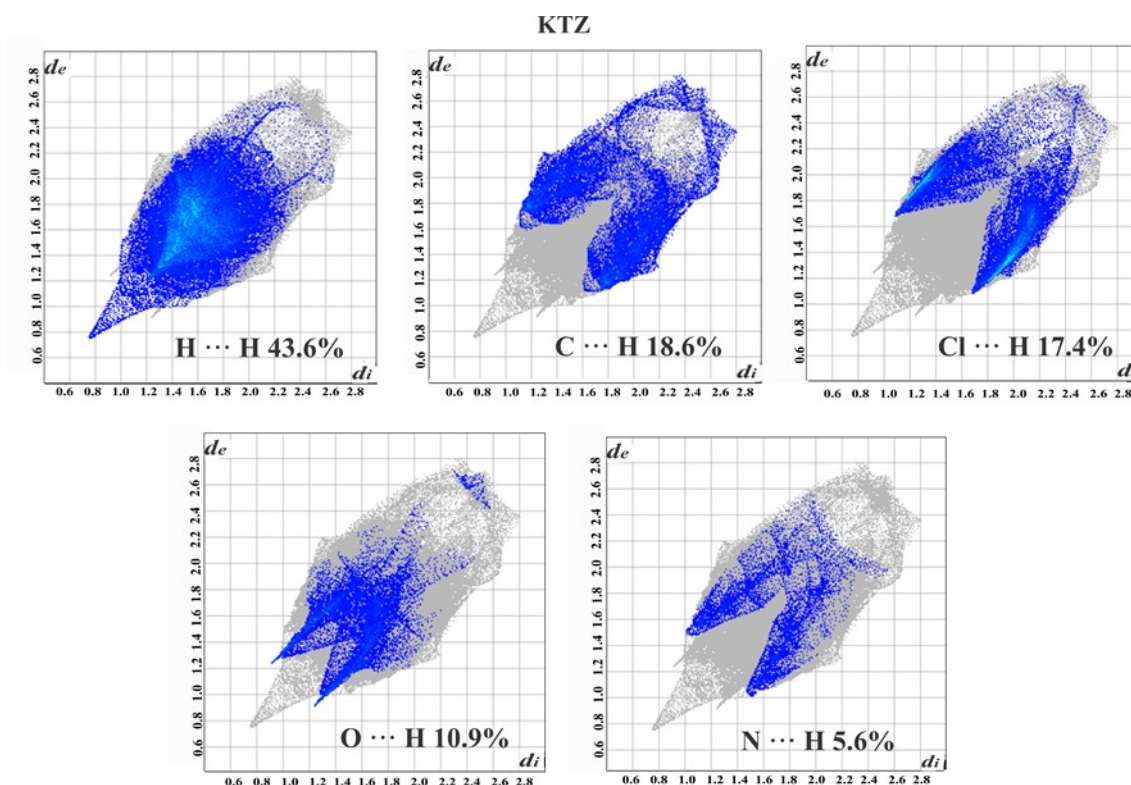


Figure S7. 2-D fingerprint plots of the KTZ structure showing the main contributions from specific pairs of atom-types.

KTZ- TPCA

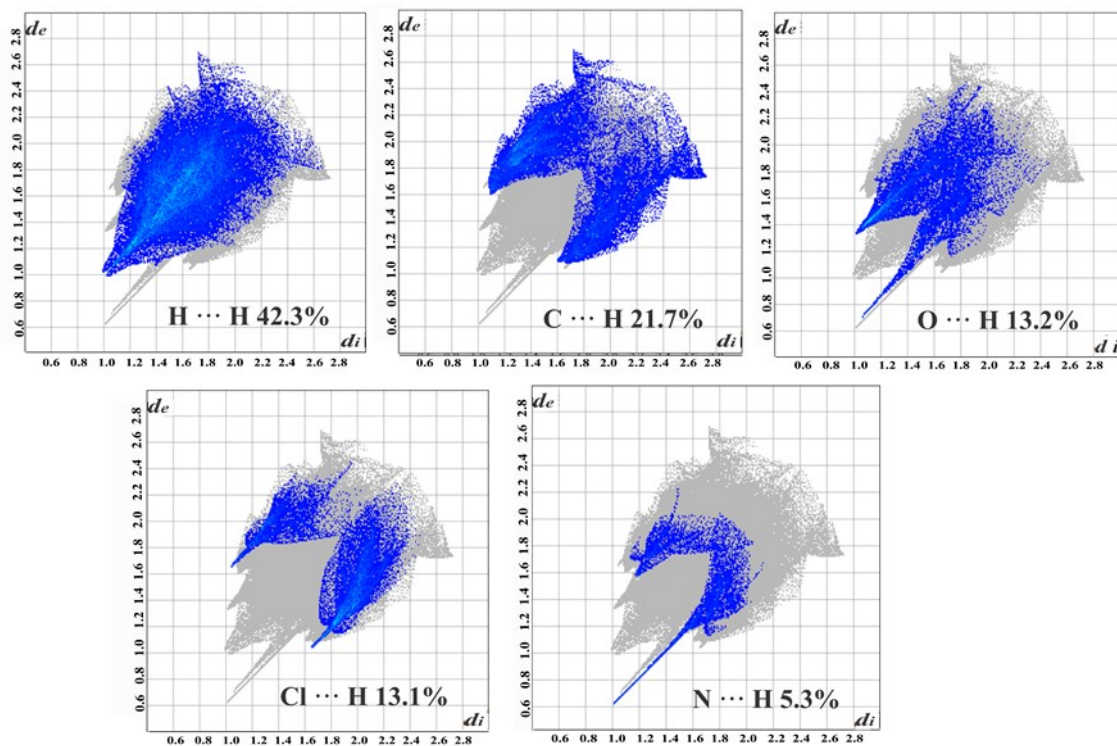


Figure S8. 2-D fingerprint plots of the KTZ-TPCA structure showing the main contributions from specific pairs of atom-types.

KTZ- 24DHB

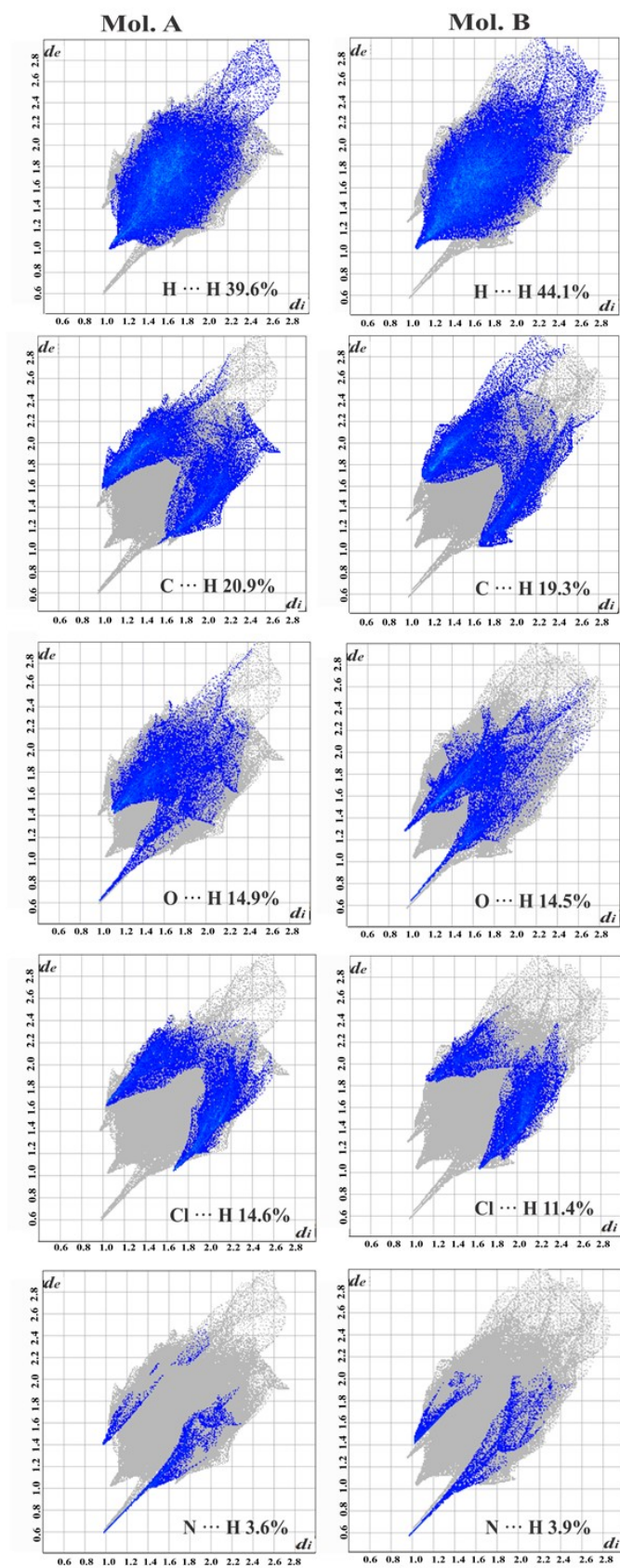


Figure S9. 2-D fingerprint plots for a) molecule A and b) molecule B of the KTZ-24DHB structure showing the main contributions from specific pairs of atom-types.

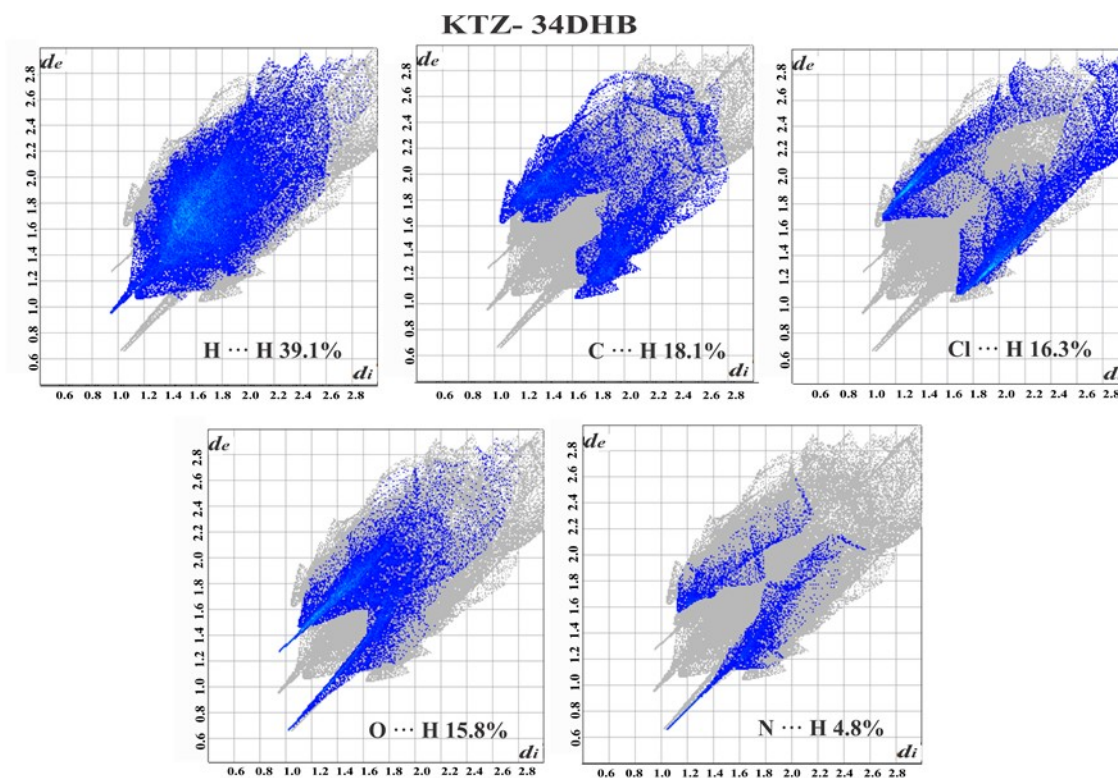


Figure S10. 2-D fingerprint plots of the KTZ-34DHB structure showing the main contributions from specific pairs of atom-types.

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