Supplementary materials

## Zn(II) and Co(II) coordination polymers based on semi-rigid bis-pyridyl-bis-amide and angular dicarboxylate ligands: synthesis, structures and properties

Venkatesan Lakshmanan,<sup>a</sup> Chia-Yi Lee,<sup>a</sup> Yu-Wen Tseng,<sup>a</sup>

Yu-Hsiang Liu,<sup>a</sup> Chia-Her Lin<sup>b,\*</sup> and Jhy-Der Chen<sup>a,\*</sup>

<sup>1</sup> Department of Chemistry, Chung-Yuan Christian University, Taoyuan City, 320, Taiwan R. O. C;

<sup>2</sup> Department of Chemistry, National Taiwan Normal University, Taipei, Taiwan, R. O. C.

## Scheme S1 Structures of selected pesticides.







NH



C 



Nitrofen

1,3-Dichlorobenzene

C C DH

2,4-Dichlorophenol



3,5-Dinitrobenzoic acid



5-Nitroisophthalic acid

Fig. S1 FT-IR spectra of complexes (a) 1 (b) 2 (c) 3 (d) 4 (e) 5 and (f) 6.











Fig. S2 (a) Simulated and (b) as synthesized PXRD patterns of complex 1.



Fig. S3 (a) Simulated and (b) as synthesized PXRD patterns of complex 2.



Fig. S4 (a) Simulated and (b) as synthesized PXRD patterns of complex 3.



Fig. S5 (a) Simulated and (b) as synthesized PXRD patterns of complex 4.









Fig. S8 TGA curves for complexes (a) 1 (b) 2 (c) 3 (d) 4 (e) 5 and (f) 6.



**Fig. S9** IR spectra for the structural transformations from **6** to **5**. (a) CP **6**, (b) CP **6** in hydrothermal reaction. The IR spectrum was measured using the purple crystals collected, and (c) CP**5**.





## Fig. S10 PXRD patterns for transformation from 5 to 6.

Fig. S11 IR spectra for the structural transformations from 5 to 6. (a) CP 5, (b) CP 5 in water at room temperature and (c) CP 6.





Fig. S12 PXRD patterns of the dehydrated and rehydrated (a) 1 and (b) 2, and 1 and 2 immersed in solutions with various pH values.





Fig. S13 PXRD patterns of (a) 1 and (b) 2 immersed in various solvents for 48 h.







Fig. S14 (a)  $CO_2$  sorption isotherm of complex 1 at 273 K and (b) pore size distributions.



Fig. S15 (a)  $CO_2$  sorption isotherm of complex 2 at 273 K and (b) pore size distributions.



Fig. S16  $CO_2$  sorption isotherms of complexes (a) 1 and (b) 2 at 298 K and PXRD patterns of (c) 1 and (d) 2 measured after experiments.





(c)



(d)



Fig. S17 Excitation and emission spectra for (a)  $L^1$  and  $L^2$  and (b)  $H_2FIPBB$ ,  $H_2OBA$  and 1 and 2.



Fig. S18 Emission spectra of solvent-free (a) 1 and (b) 2 dispersed in DCM with various pesticides.

Fig. S19 PXRD patterns before and after treatments with 2,6-DCNA and 4-NA for (a) 1 and (b) 2.



(a)



**Fig. S20** Emission spectra of the activated samples **1** dispersed in DCM of (a) 2,6-DCNA and (b) 4-NA at various concentration.



(a)





**Fig. S21** Emission spectra of the activated samples **2** dispersed in DCM of (a) 2,6-DCNA and (b) 4-NA at various concentration.







Fig. S22 (a) Bar diagrams showing the relative emission intensities of **1** and **2** dispersed in various pesticides and the Stern–Volmer (SV) plot of **1** in (b) 2,6-DCNA and (c) 4-NA and **2** in (d) 2,6-DCNA and (e) 4-NA, showing  $I_0/I$  versus analyte concentration.









Fig. S24 The IR spectra of complex (a) 1 and (b) 2 before and after treated with 4-NA and 2,6-DCNA.

Wavenumbers (cm<sup>-1</sup>)

Fig. S25 UV-vis absorption spectrum of 4-NA and 2,6-DCNA in DCM solution and the excitation and emission spectra of (a) 1 and (b) 2.





Fig. S26 Water contact angle (WCA) images of samples (a) 1 (b) 2 (c) 3 (d) 4 (e) 5 and (f) 6













CP-2

(b)



CP-**2-**C-18













CP-4



CP-4-C-18





CP-5





(f)

Fig. S27 FT-IR analysis for 2 (black curve) and 2-C18 (red curve).



Fig. S28 TGA curves for activated 2a (black) and 2a-C18 (red).



Fig. S29 (a)  $CO_2$  uptake at 273 K for 2a and 2a-C18. (b) SEM images of 2 before and after post-synthesis modification.







PSM

2

**2-**C-18

Complex	Analytes	Stern-Volmer	Adjusted	Limit of	Reference
		constant	<b>R</b> <sup>2</sup>	detection (µM)	
		(K <sub>sv</sub> , M <sup>-1</sup> )			
${[Zn(L^1)_{0.5}(FIPBB) \cdot H_2O]}_n$ , 1	2,6-DCNA,	1.136 x 10 <sup>4</sup>	0.981	3.62	this work
	4-NA	1.313 x 10 <sup>4</sup>	0.992	4.14	
$\{[Zn(L^2)(OBA) \cdot CH_3OH]\}_n, 2$	2,6-DCNA,	2.025 x 10 <sup>5</sup>	0.988	2.86	this work
	4-NA	2.623 x 10 <sup>5</sup>	0.992	1.94	
Mg-APDA	2,6-DCNA	$7.50  imes 10^4$	0.9815	1.50	Inorg. Chem. 2018, 57, 21, 13330–13340.
		( 401 - 104	0.005	2.07	Cham Asian I 2022 17 202101204
$[\{Zn(L)_2(DMF)_4\} \cdot 2BF_4]_{\alpha}$	2,0-DNA,	6.401 x 10 <sup>4</sup>	0.995	2.97	Chem Asian. J. 2022, 17, 202101204.
$[{Cd(L)_2(Cl)_2} \cdot 2H_2O]_{\alpha}$	4-NA	7.130 x 10 <sup>4</sup>	0.919	0.90	
$(H_3O)[Zn_2L(H_2O)]\cdot 3NMP\cdot 6H_2O$	2,6-DCNA	6.072 x 10 <sup>3</sup>	0.922	5.4	RSC Adv., 2019, 9, 38469-38476.
[Eu <sub>2</sub> (dtztp)(OH) <sub>2</sub> (DMF)(H <sub>2</sub> O) <sub>2.5</sub> ]·2H <sub>2</sub> O	2,6-DCNA	6.25×10 <sup>4</sup>	0.995	2.91	Sensors and Actuators: B. Chemical, 2021,
					<i>331</i> , 129377.
[Ag(3-dpyb)(H <sub>3</sub> odpa)]·H <sub>2</sub> O	2,6-DCNA	$2.028 \times 10^{5}$	0.988	1.15	RSC Adv., 2020, 10, 44712–44718.
${[Cd(tptc)_{0.5}(bpz)(H_2O)] \cdot_{0.5}H_2O}_n$	2,6-DCNA	4.71 x 10 <sup>4</sup>	0.989	1.14	Molecular and Biomolecular Spectroscopy,
[Cd(tptc) <sub>0.5</sub> (bpy)] <sub>n</sub>	2,6-DCNA	8.27 x 10 <sup>3</sup>	0.994	6.38	2020, <i>239</i> , 118467.
[Zn <sub>2</sub> (bpdc) <sub>2</sub> (BPyTPE)]	2,6-DCNA	8.04 x 10 <sup>3</sup>	0.996	1.30	Chem. Commun., 2017, 53, 9975-9978.
$[Cd_3(L^2)_2(H2O)4] \cdot 5H_2O(4)$	4-NA	2.15 x 10 <sup>4</sup>	0.998	1.14	Cryst. Growth Des. 2018, 18, 431–440.
[Cd(ppvppa)(1,4-NDC)] <sub>n</sub> (10)	4-NA	1.3 x 10 <sup>3</sup>	0.985	1.20	Cryst. Growth Des. 2015, 15, 2753–2760.
[Cd(tptc) <sub>0.5</sub> (phen)]	2,6-DCNA	5.18 ×10 <sup>4</sup>		1.02	New J. Chem., 2019, 43, 13349-13356.
[Zn <sub>3</sub> (DDB)(DPE)]	2,6-DCNA	$3.3  imes 10^4$	0.990	3.3	Dalton Trans., 2019, 48, 16776-16785.
LNU-45	2,6-DCNA	9.09 ×10 <sup>4</sup>	0.998	1.5	Molecules, 2022, 27, 126.
LNU-47	2,6-DCNA	1.62 ×10 <sup>3</sup>	0.992	1.9	

Table S1. Sensing properties of reported compounds towards 2,6-DCNA, and 4-NA pesticides.