

Supporting Information

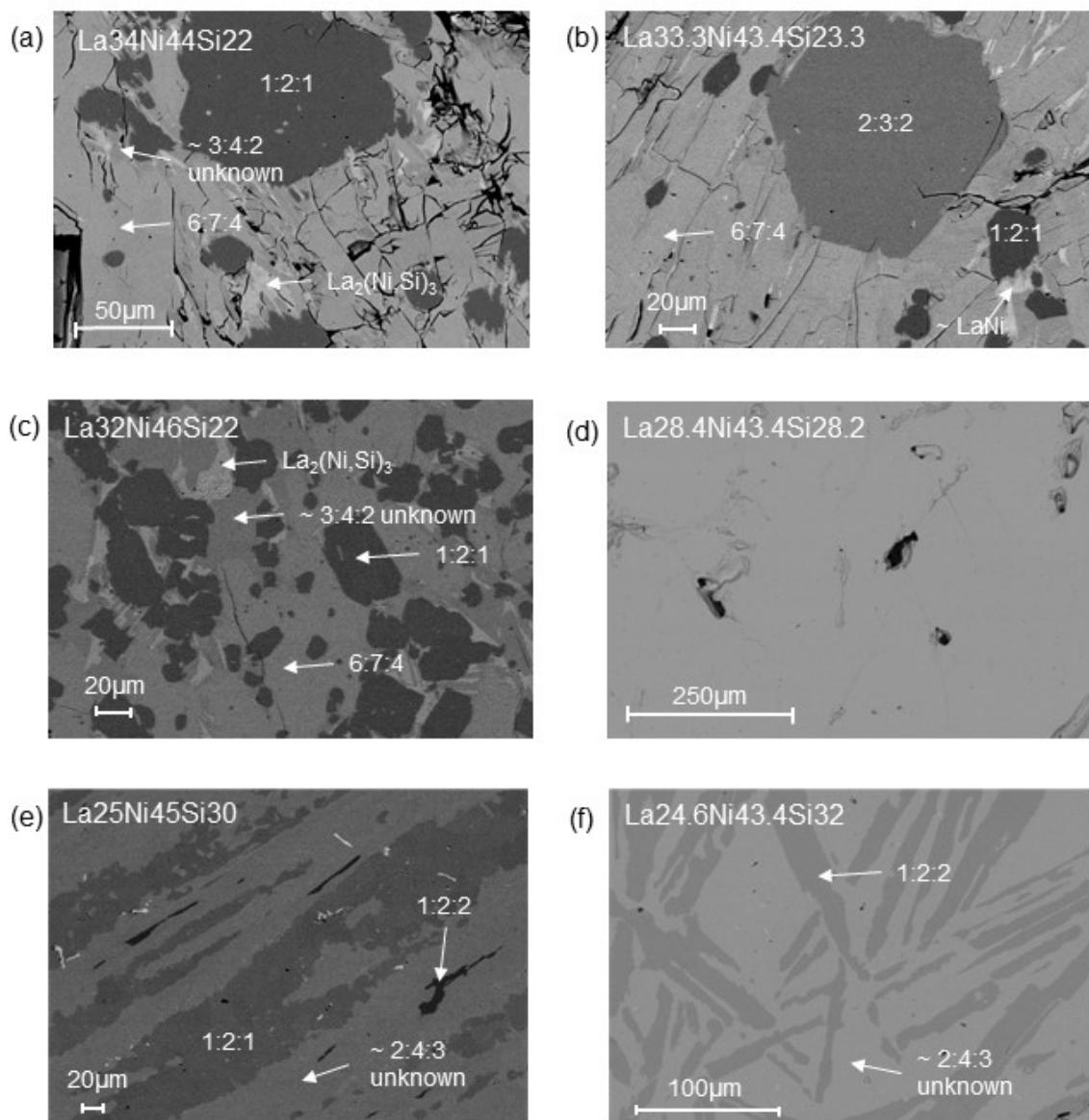


Figure S1. SEM images [backscattered electron (BSE) mode] showing the complex microstructure and multiphase appearance of samples. According to the numbering of Table 1, the nominal compositions are La₃₄Ni₄₄Si₂₂ [sample 3 (a)], La_{33.3}Ni_{43.4}Si_{23.3} [sample 4 (b)], La₃₂Ni₄₆Si₂₂ [sample 5 (c)], La_{28.4}Ni_{43.4}Si_{28.2} [sample 6 (d)], La₂₅Ni₄₅Si₃₀ [sample 8 (e)] and La_{24.6}Ni_{43.4}Si₃₂ [sample 9 (f)]. In each sample, the composition (La:Ni:Si at. %) of the different phases present is highlighted.

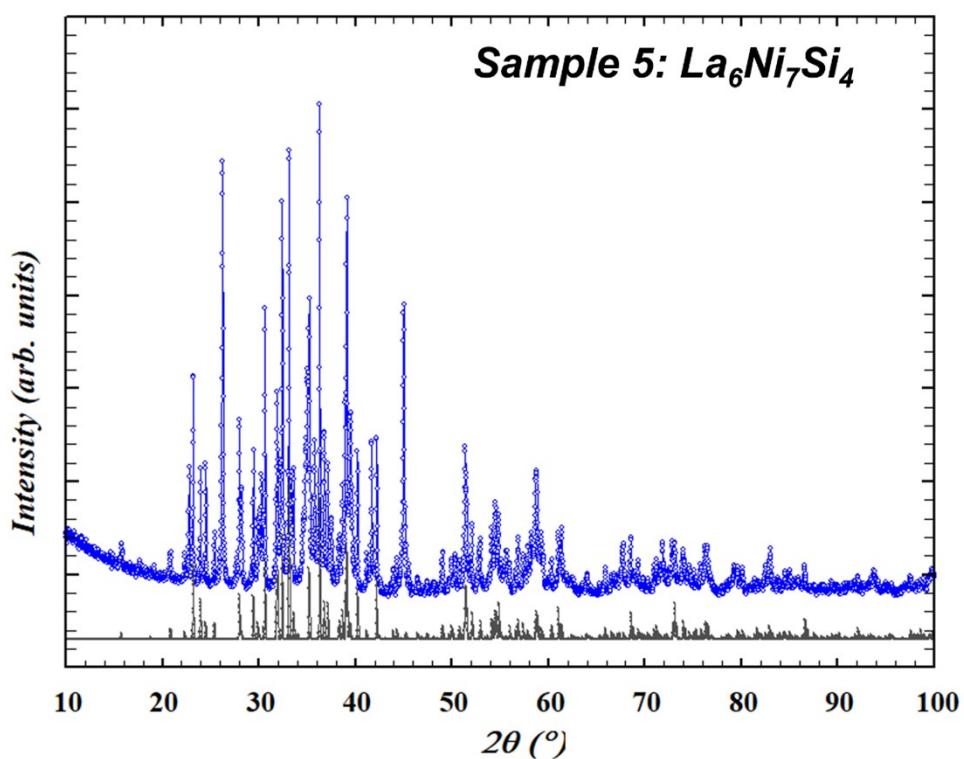
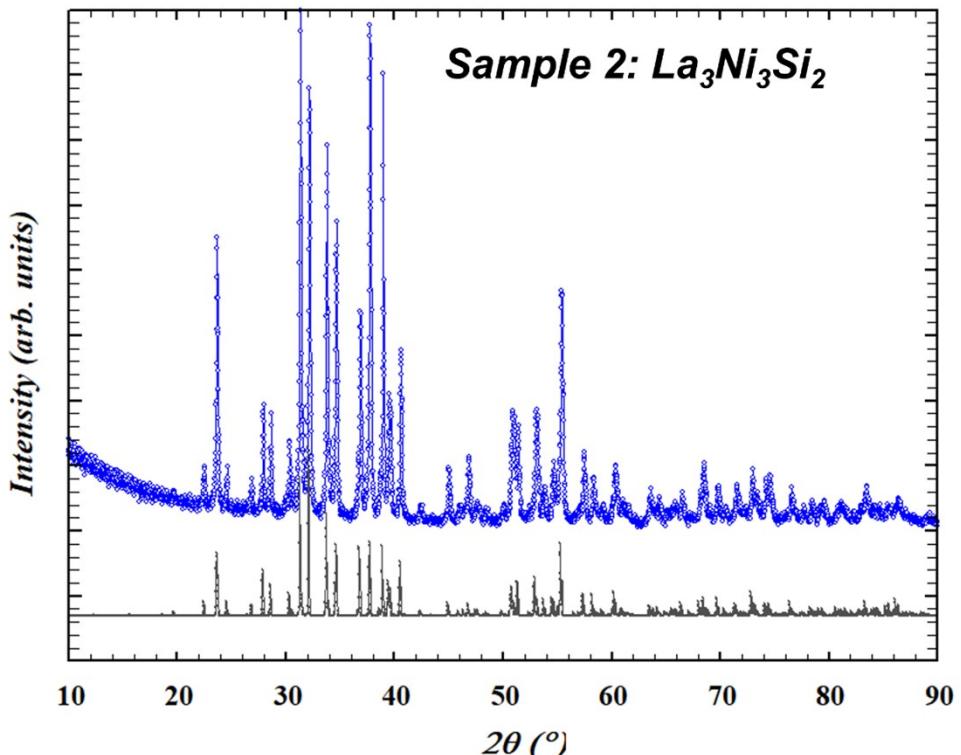


Figure S2. Top: powder pattern of sample 2 (nominal composition at. % La 37.5, Ni 37.5, Si 25.0), showing in blue and grey, respectively, the experimental and the calculated curves of the $\text{La}_3\text{Ni}_3\text{Si}_2$ compound. No additional phases are present. Bottom: powder pattern of sample 5, having nominal composition $\text{La}_{32}\text{Ni}_{46}\text{Si}_{22}$ and containing the $\text{La}_6\text{Ni}_7\text{Si}_4$ compound as main phase; $\text{LaNi}_2\text{Si}_1 \approx \text{La}_3\text{Ni}_4\text{Si}_2$ (yet unknown) and $\text{La}_2(\text{Ni},\text{Si})_3$ are present as secondary phases. The simulated pattern of the main phase is reported for comparison.

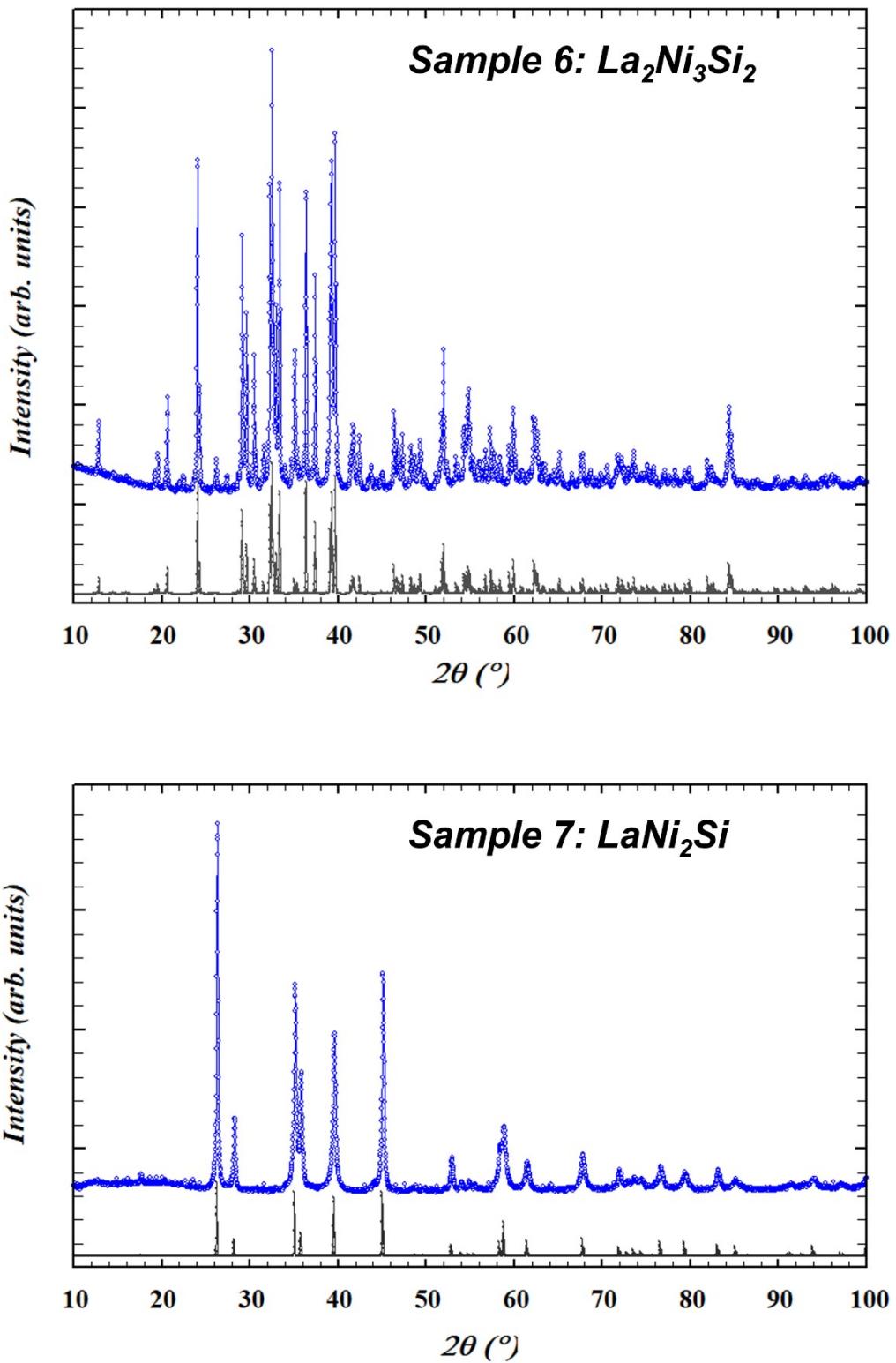


Figure S3. Top: powder pattern of sample 6 (nominal composition at. % La 28.4, Ni 43.4, Si 28.2), showing in blue and grey, respectively, the experimental and the calculated curves of the $\text{La}_2\text{Ni}_3\text{Si}_2$ compound. A few additional weak peaks are due to an impurity phase. Bottom: powder pattern of sample 7 (nominal composition at. % La 25, Ni 50, Si 25.0), almost single phase of the LaNi_2Si compound; the simulated pattern is represented in grey.

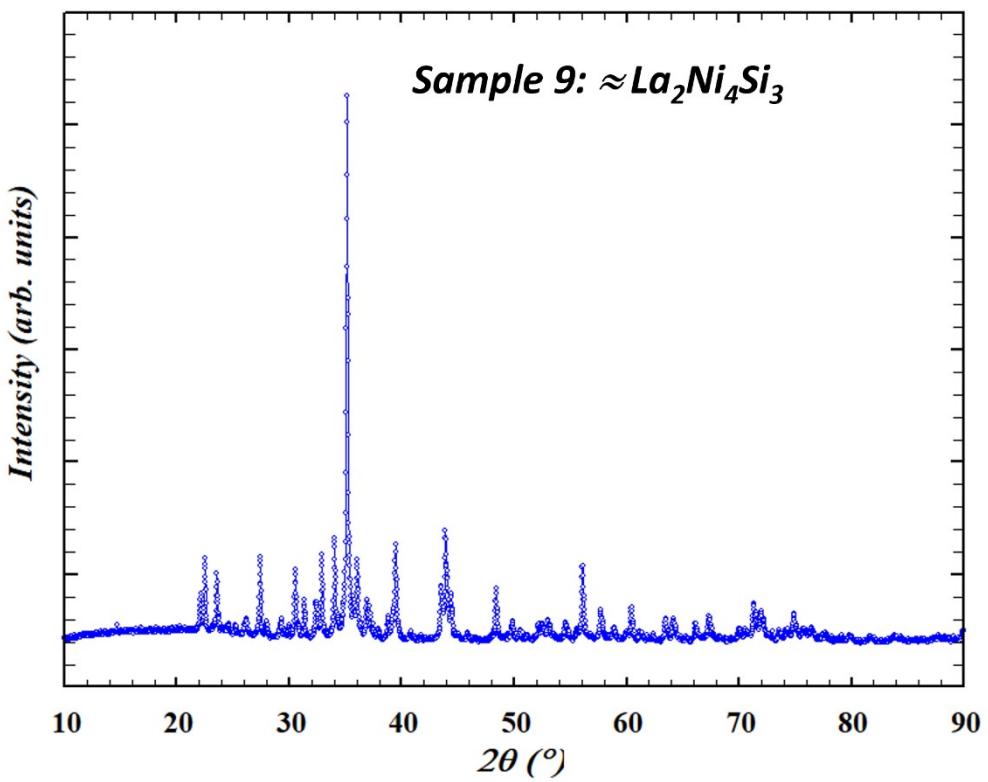


Figure S4. Powder pattern of sample **9** (nominal composition $\text{La}_{24.6}\text{Ni}_{43.4}\text{Si}_{32}$), containing the compound with approximate composition $\text{La}_2\text{Ni}_4\text{Si}_3$ (unknown structure), as predominant phase.

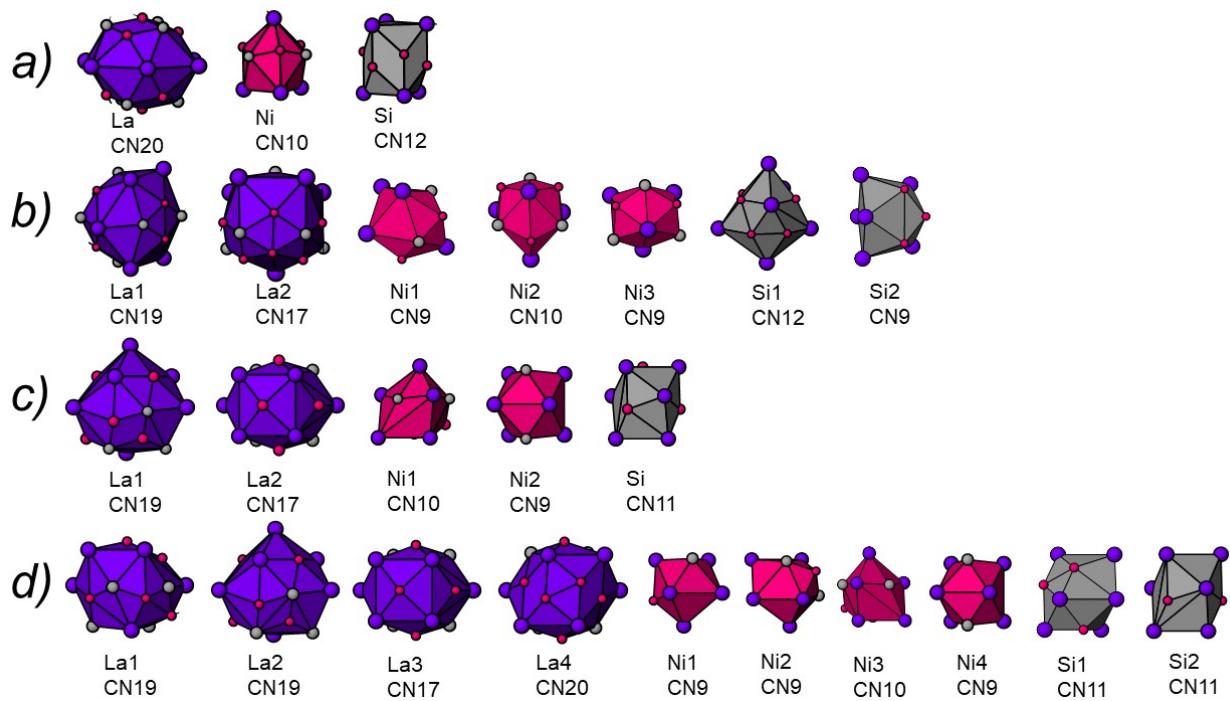


Figure S5. Coordination polyhedra in the crystal structures of LaNi_2Si (a), $\text{La}_2\text{Ni}_3\text{Si}_2$ (b), $\text{La}_3\text{Ni}_3\text{Si}_2$ (c) and $\text{La}_6\text{Ni}_7\text{Si}_4$ (d). La, Ni and Si atoms are represented in violet, pink and grey, respectively.

Table S1. Details of the single crystal X-ray data collection and structure refinements of LaNi₂Si, La₂Ni₃Si₂, La₆Ni₇Si₄ and La₃Ni₃Si₂ (T = 295 K).

Compound	LaNi ₂ Si	La ₂ Ni ₃ Si ₂	La ₆ Ni ₇ Si ₄	La ₃ Ni ₃ Si ₂
La:Ni:Si [composition in at.%]	25 : 50 : 25	28.6 : 42.8 : 28.6	35.3 : 41.2 : 23.5	37.5 : 37.5 : 25
Isotypic crystal	SrCu ₂ Ga	Own structure	Pr ₆ Ni ₇ Si ₄	Ce ₃ Rh ₃ Si ₂
Formula weight [g/mol]	284.42	510.13	1357.79	649.04
Crystal system	Rhombohedral	Monoclinic	Orthorhombic	Orthorhombic
Pearson symbol	hR12	mP28	oP68	oP32
Space group	$\bar{R}\bar{3}m$ (No. 166)	P2 ₁ /c (No. 14)	Pbcm (No. 57)	Pnma (No. 62)
Unit formula per cell, Z	3	4	4	4
Lattice parameters [Å, °]	$a = 4.0263(3)$ $c = 15.066(2)$	$a = 6.8789(7)$ $b = 6.2167(3)$ $c = 12.214(1)$ $\beta = 90.92(1)$	$a = 6.066(1)$ $b = 7.488(1)$ $c = 29.682(5)$	$a = 7.501(2)$ $b = 14.316(4)$ $c = 6.1485(16)$
Unit cell volume [Å ³]	211.48(4)	522.24(9)	1348.2(4)	660.3(3)
Calc. density, ρ [g/cm ³]	6.70	6.49	6.68	6.53
Absorption coefficient, μ [mm ⁻¹]	28.15	26.91	28.37	27.58
F(000)	381	904	2376	1132
Crystal shape	Elongated prism	Platelet	Block	Block
Crystal size [μm]	50 × 80 × 160	50 × 100 × 120	70 × 70 × 80	60 × 70 × 80
Scan mode	ω-θ	ω-θ	ω-θ	ω-θ
Theta range [°]	2 ≤ θ ≤ 32	2 ≤ θ ≤ 30	2.5 ≤ θ ≤ 33.5	3.6 ≤ θ ≤ 31.4
Ranges of h, k, l	-6 ≤ h ≤ 6 -6 ≤ k ≤ 6 -22 ≤ l ≤ 22	0 ≤ h ≤ 9 -8 ≤ k ≤ 8 -17 ≤ l ≤ 17	-9 ≤ h ≤ 9 -11 ≤ k ≤ 11 -44 ≤ l ≤ 44	-10 ≤ h ≤ 10 -20 ≤ k ≤ 19 -8 ≤ l ≤ 8
Number of collected reflections	970	3300	26857	6347
Number of independent reflections	116	1512	2503	1054
Reflections with F ₀ > 4σ (F ₀)	103	1080	1995	1044
Absorption correction method	ψ-scans, sphere	ψ-scans, sphere	multiscan	multiscan
Solution method	Patterson	Direct Methods	Intrinsic Phasing	Intrinsic Phasing
Number of refined parameters	9	65	81	41
R ₁ ^a	0.012	0.032	0.029	0.022
wR ₂ ^b (F ₀ ²), all data	0.034	0.064	0.053	0.055
R _{int} ^c (F ₀ ²)	0.059	0.044	0.046	0.040
Goodness-of-fit on F ₀ ²	1.273	0.920	0.923	1.249
Secondary extinction correction x	0.026(2)	0.0078(2)	—	0.0052(3)

^a $R_1(F) = \left[\sum (|F_o| - |F_c|) \right] / \sum |F_o|$, ^b $wR_2(F^2) = \left[\sum w(F_o^2 - F_c^2)^2 / \left[\sum w(F_o^2)^2 \right] \right]^{\frac{1}{2}}$, w is a weight function different for each compound;

^c $R_{int}(F^2) = \left[\sum |F_o^2 - \bar{F}_o^2| \right] / n$, is an internal agreement index quantifying the deviation of n equivalent intensities from the average value.

Table S2. Structure types adopted by lanthanide or alkaline earth ternary intermetallics with 1:2:1 stoichiometry: crystal data, number and representatives of each family.²

Structure Type	Pearson Code – Space Group	Wyckoff letters	Lattice Parameters [Å]	Number	Representatives
BaLi ₂ Si	<i>oP8</i> – <i>Pmmn</i>	4e, 2b, 2a	$a = 4.74$ $b = 6.75$ $c = 6.25$	3	KLi ₂ As BaNi ₂ Ge
PrCo ₂ Ga	<i>oP8</i> – <i>Pmma</i>	2 x 2f, 2e, 2a	$a = 5.021$ $b = 4.043$ $c = 6.860$	15	RCo ₂ Ga, R=La,Pr,Nd RCo ₂ In, R=Pr-Sm,Gd-Ho,Y RNi ₂ In, R=La,Pr,Nd LaCo _{2.17} Zn _{0.83}
GdPt ₂ Sn (or ZrPt ₂ Al)	<i>hP8</i> – <i>P6₃/mmc</i>	4f, 2c, 2a	$a = 4.532$ $c = 9.065$	29	RNi ₂ Sb, RPt ₂ Al, R=Zr,Hf RPd ₂ In, R=La, Ce RPt ₂ In, R=Ce,Gd,Tb-Ho,U RPt ₂ Sn, R=Gd,Tb,Er-Lu, Y,U UT ₂ Sn, T=Cu,Au RLi ₂ Ge, R=La,Ce CaNi ₂ Si, RNi ₂ Ge, R=Sr,Pr,Nd LaCu ₂ Mg, MgAu ₂ Ga
CeLi ₂ Ge (or AlCr ₂ C or LiCu ₂ Sn)	<i>hP8</i> – <i>P6₃/mmc</i>	4f, 2c, 2a	$a = 4.537$ $c = 7.568$	4	RLi ₂ Ge, R=La-Nd
SrCu ₂ Ga	<i>hR12</i> – $\bar{R}\bar{3}m$	6c, 3b, 3a	$a = 4.271$ $c = 15.925$	3	BaCu ₂ Ga LaNi ₂ Si
CePt ₂ B	<i>hP12</i> – <i>P6₂22</i>	6i,3d,3c	$a = 5.4811$ $c = 7.8830$	7	RPt ₂ B, R=La-Nd,Tm,Lu,Y
YPd ₂ Si (Fe ₃ C der.)	<i>oP16</i> – <i>Pnma</i>	8d, 2 x 4c	$a = 7.300$ $b = 6.927$ $c = 5.499$	62	RPd ₂ Ga, R=La-Sm,Gd-Dy,Y RPd ₂ Si, RPd ₂ Ge, R=La-Lu,Y RNi ₂ Si, R=Tb,Dy,Er,Tm,Y RPt ₂ Si, R=Gd-Lu,Y MgNi ₂ P, UAu ₂ Al RPd ₂ Al, R=Ce-Pr CaCd ₂ X, X=Pd,Pt CaMg ₂ X, X=Rh,Pd,Pt
EuAl ₂ Ge	<i>oP16</i> – <i>Pnma</i>	4 x 4c	$a = 7.294$ $b = 4.306$ $c = 11.237$	1	
CeRh ₂ Si (CeNiSi ₂ -type occupation variant)	<i>oS16</i> – <i>Cmcm</i>	4 x 4c	$a = 4.0413$ $b = 17.730$ $c = 4.0675$	3	RPt ₂ Si, R=Ce,Eu
YPd ₂ Sn (MnCu ₂ Al Heusler phase)	<i>cF16</i> – <i>Fm-3m</i>	8c, 4b, 4a	$a = 6.7144$	~200	RMg ₂ Cu, R=Ce,Tb RMg ₂ Ag, R=La-Gd RMg ₂ Zn, R=Pr,Nd LiCa ₂ X, X=Ti,Si,Ge MgNi ₂ X, X=Sn,Sb MgPd ₂ X, X=Ga,Sb MgM ₂ In, M=Li,Ni,Pd MgLi ₂ X, X=Ga,In,Tl,Ge,Sn,Sb,Bi,Ag,Au,Zn,Cd,Hg ScM ₂ Ga, M=Ni,Pd,Cu ScM ₂ Al, M=Cu,Ag,Au,Ni,Pd,Pt ScM ₂ In, M=Ni,Pt ScM ₂ Sn, M=Pt,Rh ScCo ₂ X, X=Ga,Ge,Sn RPd ₂ In, R=Gd,Dy,Er-Lu,Sc,Y RM ₂ In, R= La-Lu,Sc,Y; M=Cu,Ag,Au RPd ₂ Tl, R=Sm-Er,Y RNi ₂ Sn, R=Yb,Lu,Sc RPd ₂ Sn, R=Gd-Lu,Sc,Y RPd ₂ Pb, R=Pr,Sm-Lu,Sc,Y RPd ₂ Sb, R=Gd-Er,Yb,Y RPd ₂ Bi, R=Pr,Sm-Er,Y RZn ₂ Mg, R=Ce,Tb
YRh ₂ Si (CeNi ₃ der.)	<i>hP24</i> – <i>P6₃/mmc</i>	12k, 4f, 2d, 2c, 2b, 2a	$a = 5.495$ $c = 15.030$	14	RCo ₂ Zn, R=La,Ce,Sm RNi ₂ Zn, R=La,Ce RRh ₂ Si, R=Er,Y; RRh ₂ Ge, R=Sm,Er,Y SmCo ₂ Ga, LuNi ₂ Ga, TbFe ₂ Ga
DyB ₂ C	<i>tP32</i> – <i>P4₂/mbc</i>	8g, 3 x 8h	$a = 6.791$ $c = 7.522$	9	RB ₂ C, R=Tb-Lu,Sc,Y
CeIr ₂ Si	<i>tI32</i> – <i>I4₁/amd</i>	3 x 8e, 4b, 4a	$a = 4.0698$ $c = 35.408$	4	RRh ₂ Si, RIr ₂ Si, R=La,Ce

Table S3. Atomic coordinates for the VASP optimizations based on the CaNi₂Si structure (*hP8*).

Atom	Site	x	y	z
Ca/La	2 <i>c</i>	1/3	2/3	1/4
Ni	4 <i>f</i>	1/3	2/3	0.96051
Si	2 <i>a</i>	0	0	0

$a = 4.004$, $c = 9.886$ Å.