

Supporting Information for

Aggregate emission behaviors and reversible mechanofluorochromic properties of α -cyanostilbene functionalized indolo[3,2-*b*]carbazole derivatives

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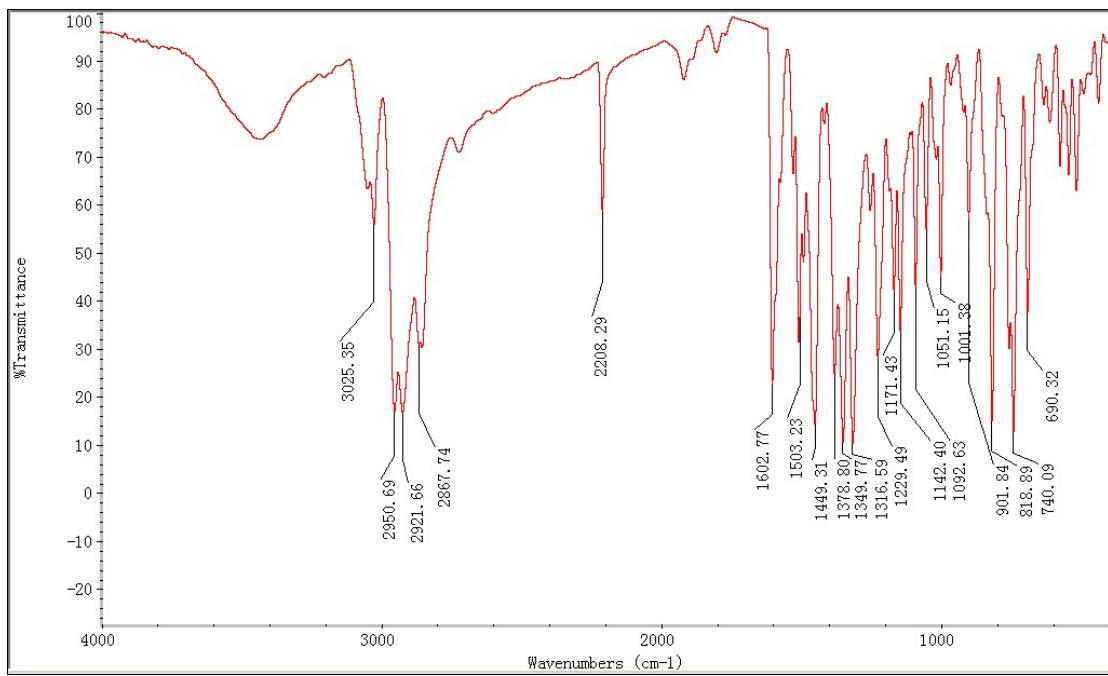


Fig. S1 FT-IR spectrum of ICZH.

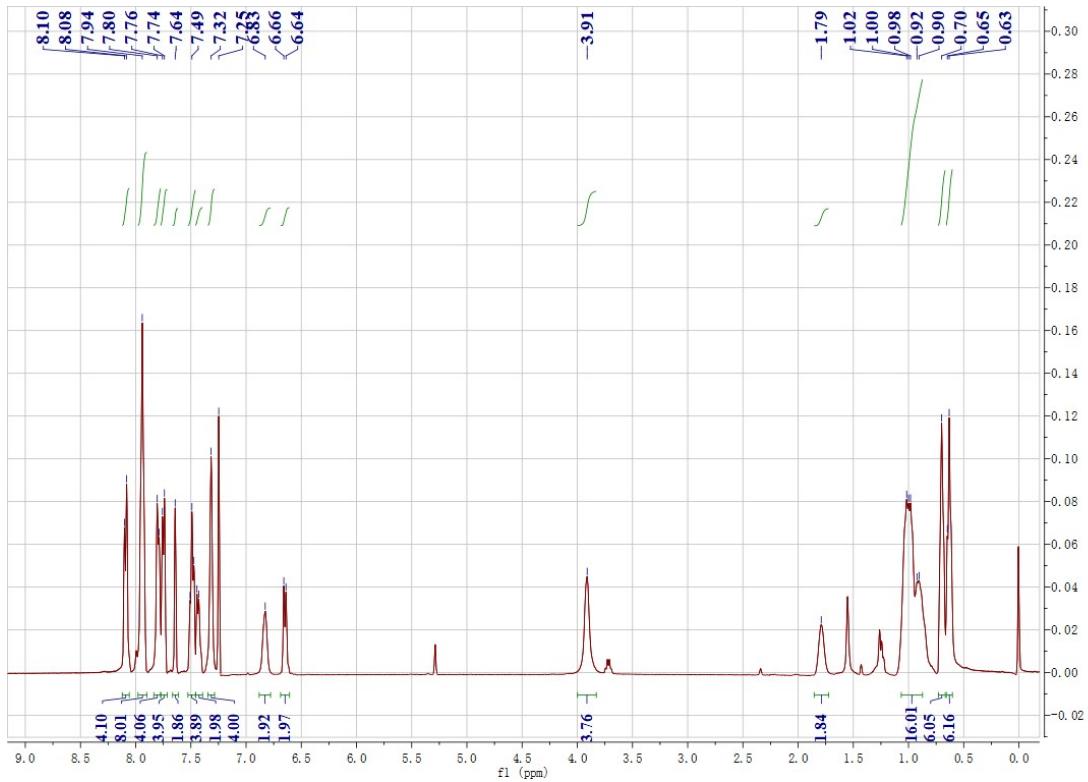


Fig. S2 ^1H NMR spectrum of **ICZH** in CDCl_3 .

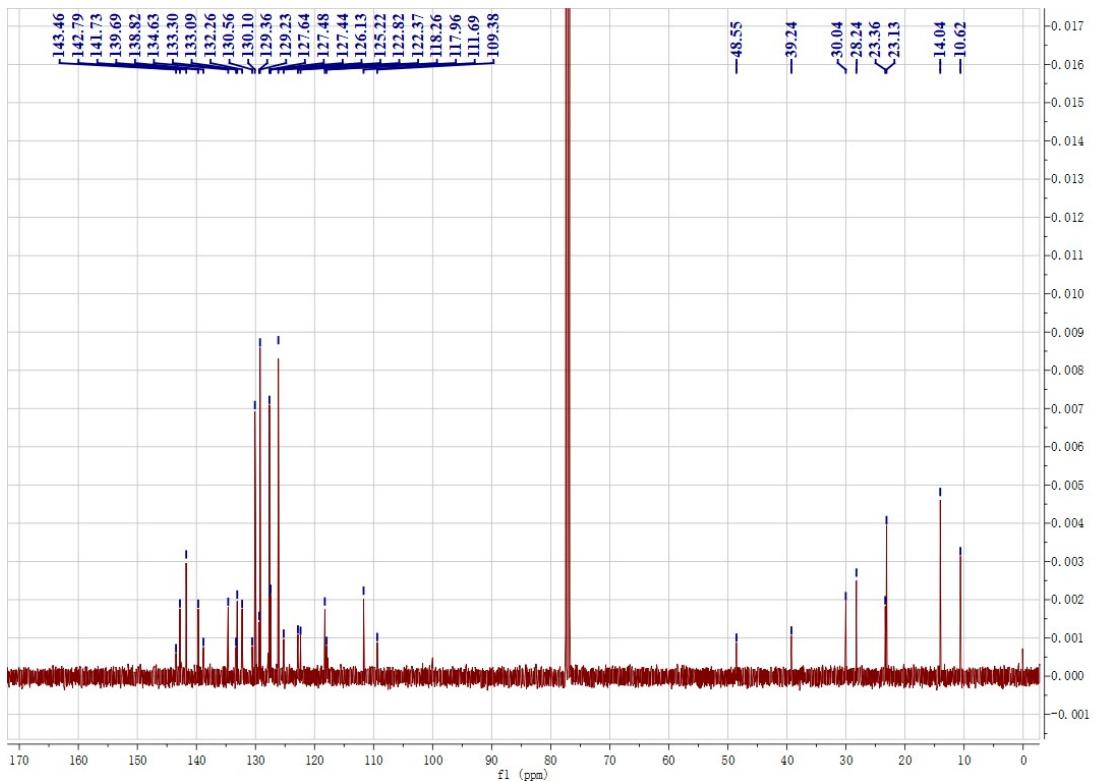


Fig. S3 ^{13}C NMR spectrum of **ICZH** in CDCl_3 .

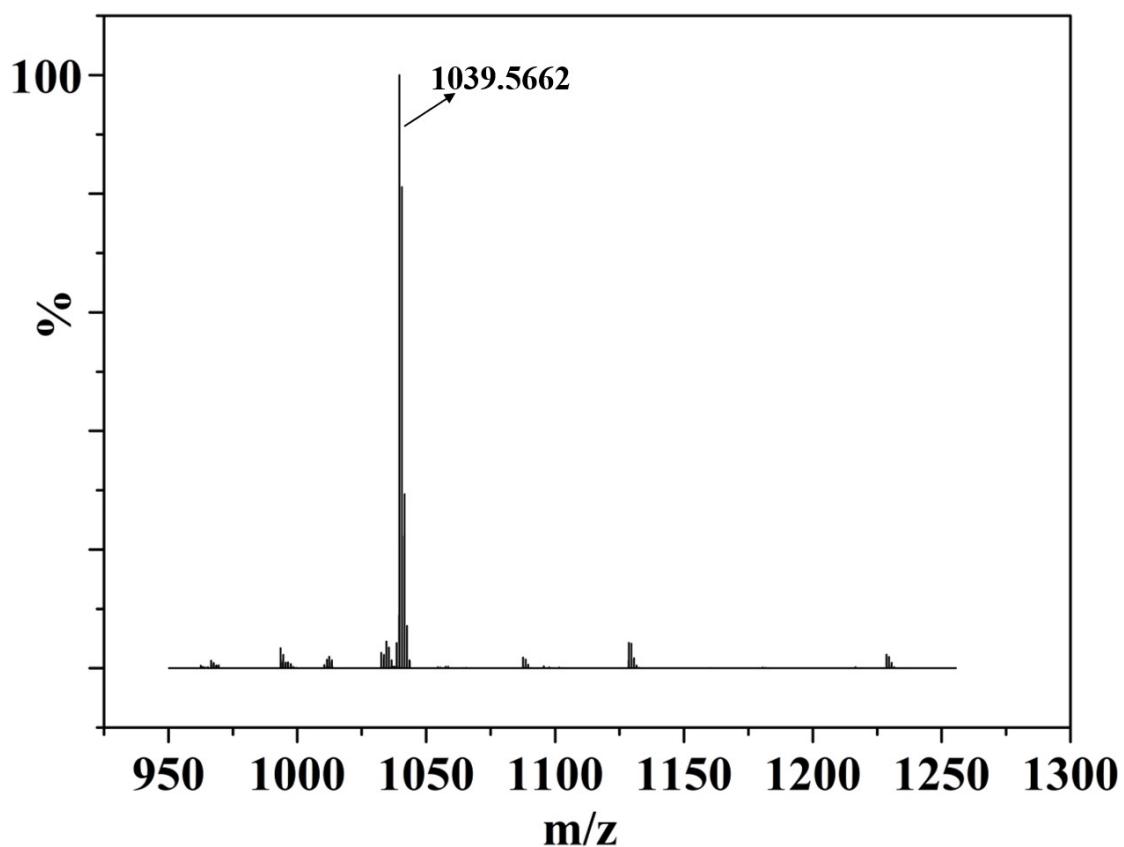


Fig. S4 HP-MS spectrum of **ICZH**.

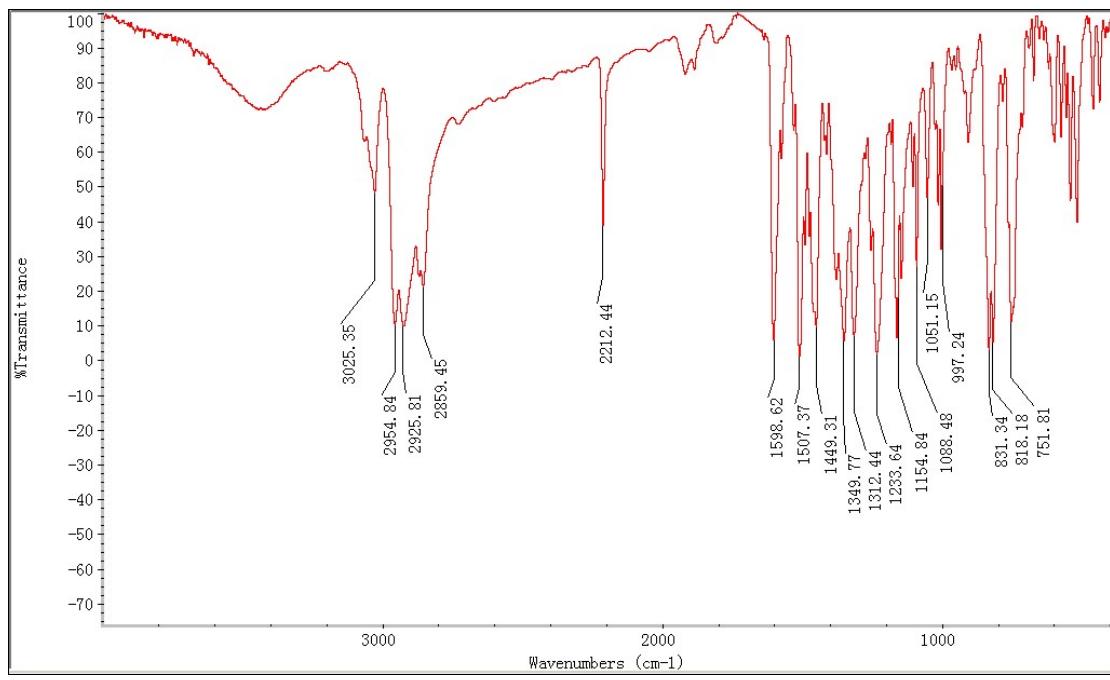


Fig. S5 FT-IR spectrum of **ICZF**.

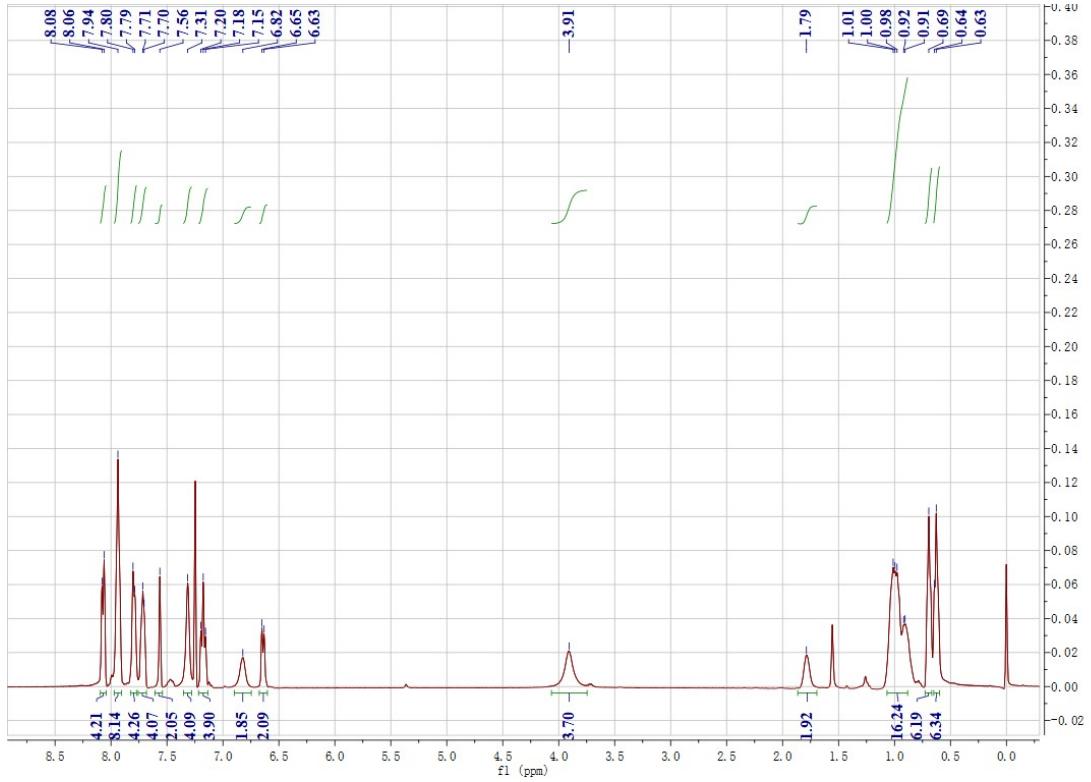


Fig. S6 ^1H NMR spectrum of **ICZF** in CDCl_3 .

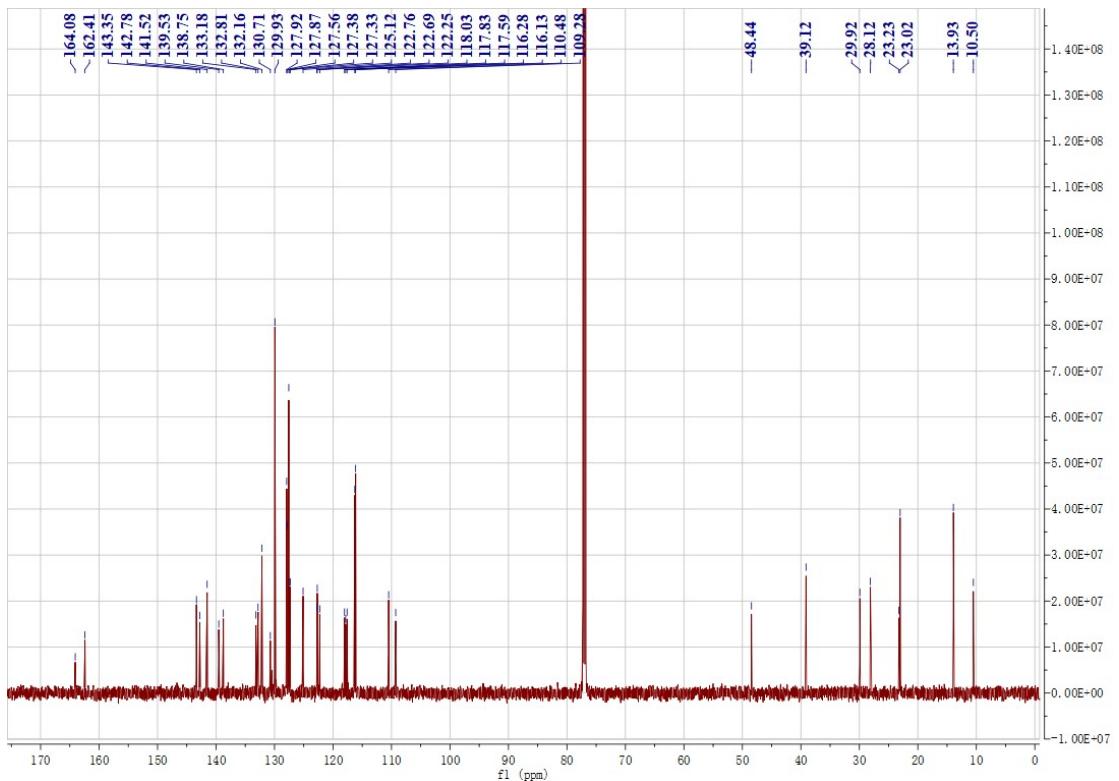


Fig. S7 ^{13}C NMR spectrum of ICZF in CDCl_3 .

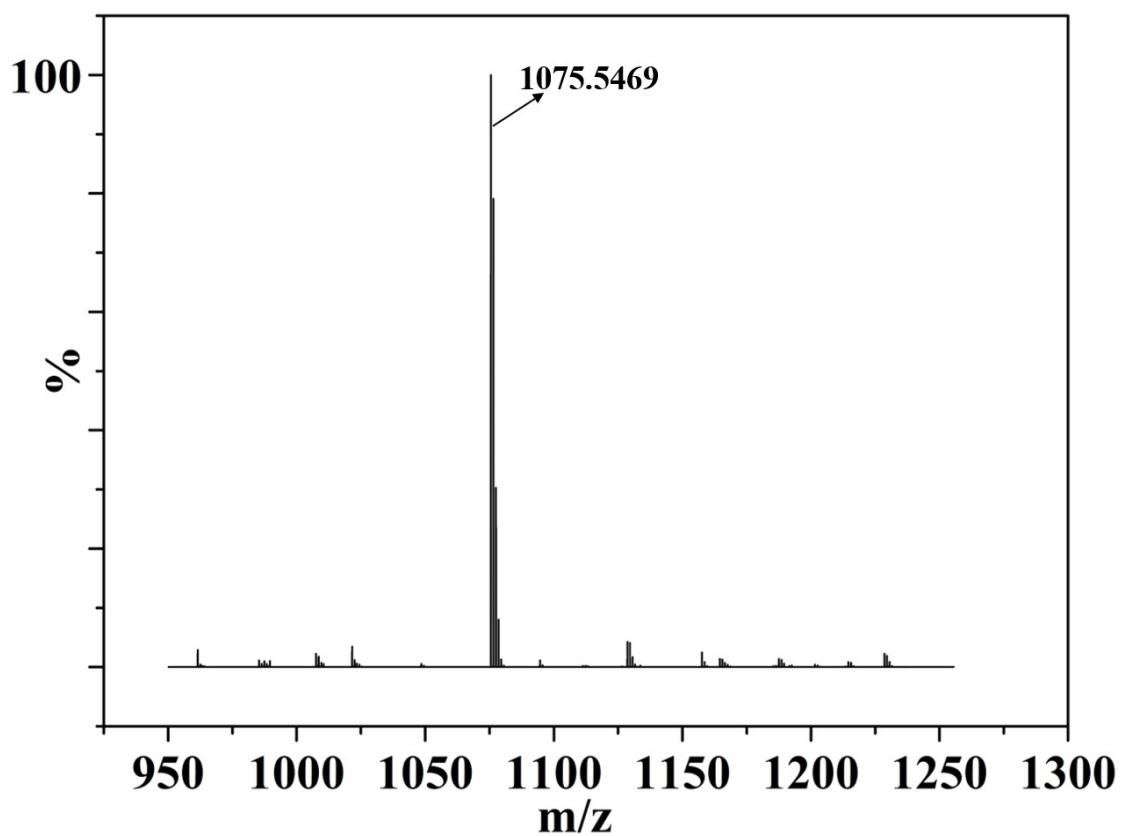


Fig. S8 HP-MS spectrum of ICZF.

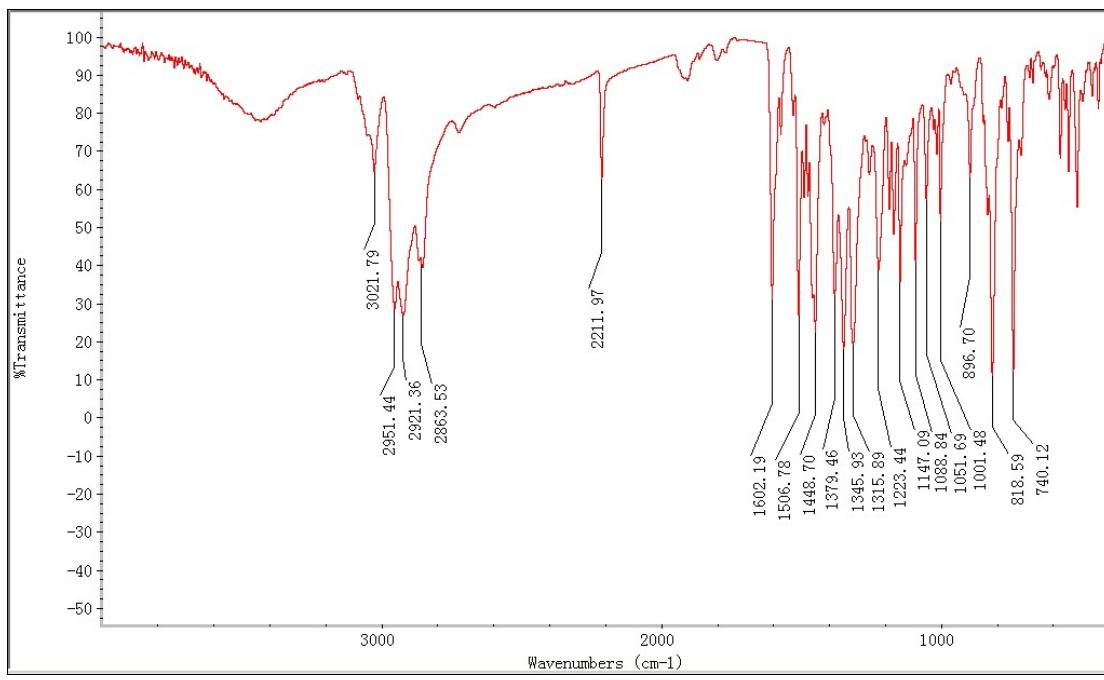


Fig. S9 FT-IR spectrum of **ICZCH₃**.

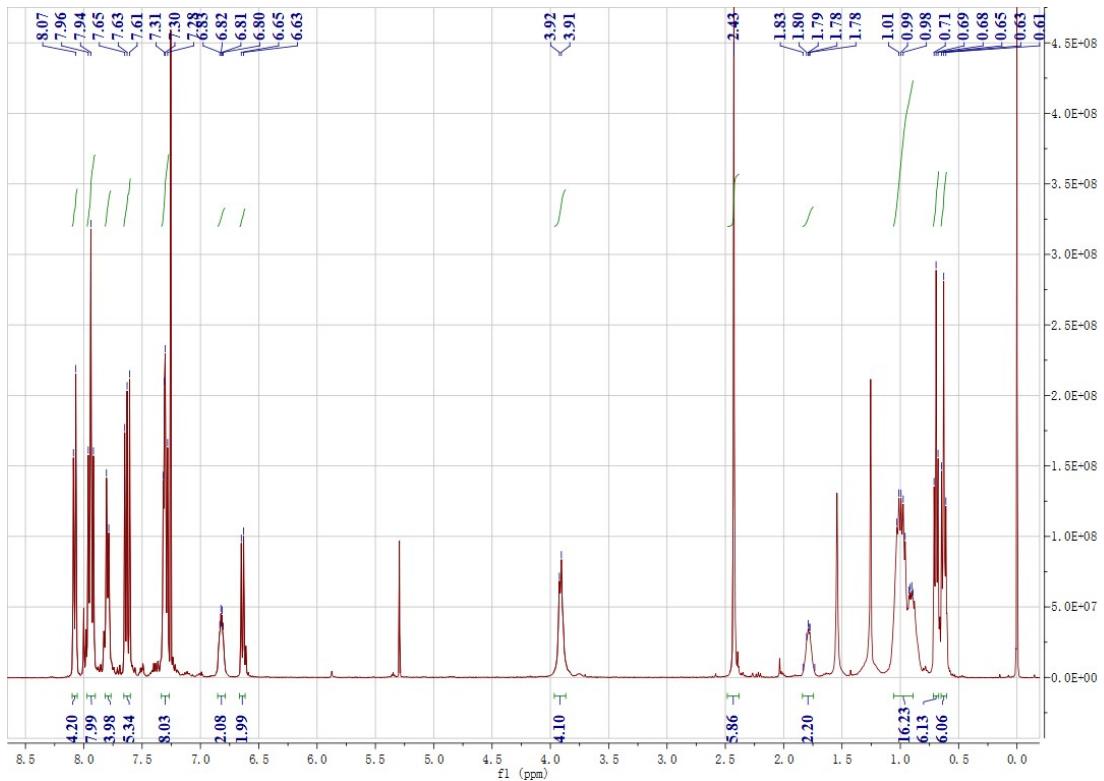


Fig. S10 ¹H NMR spectrum of **ICZCH₃** in CDCl_3 .

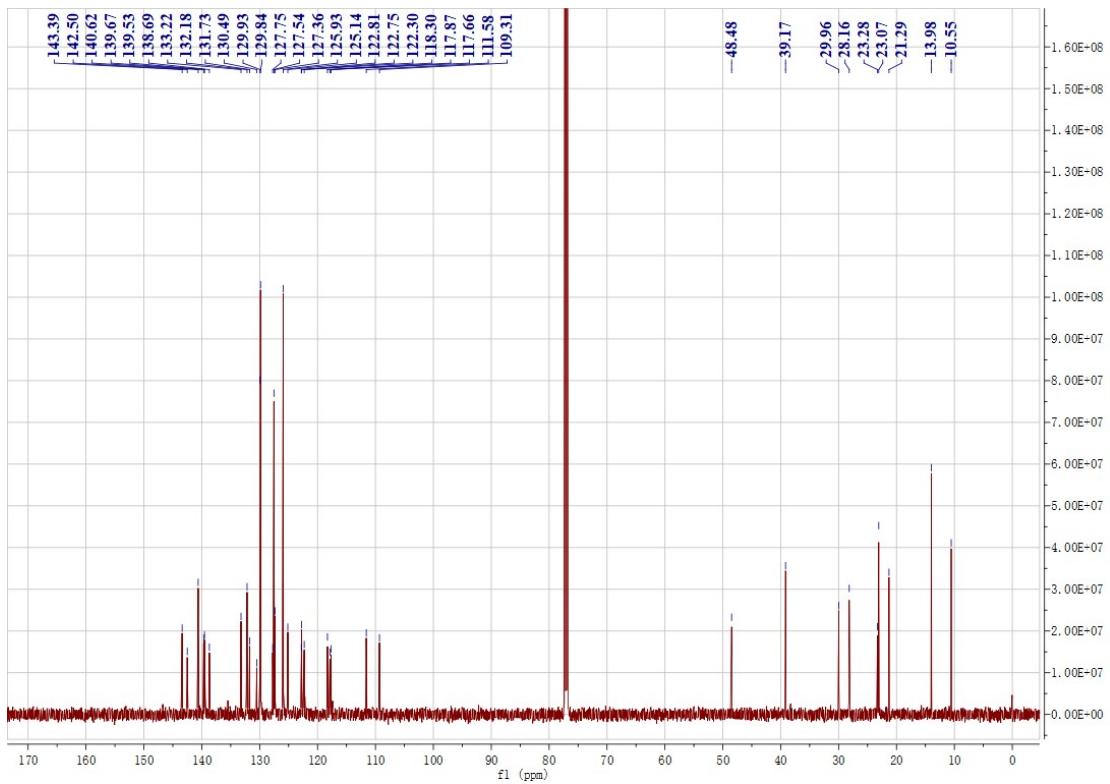


Fig. S11 ^{13}C NMR spectrum of **ICZCH₃** in CDCl_3 .

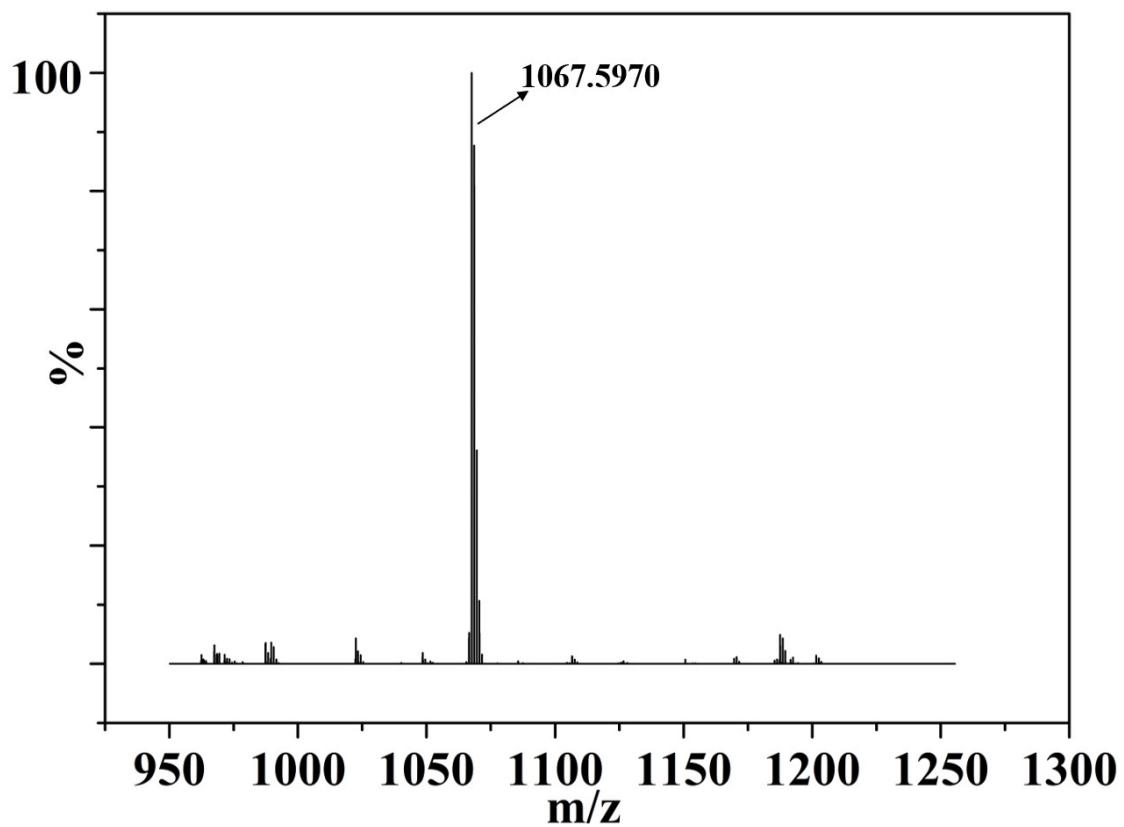


Fig. S12 HP-MS spectrum of **ICZCH₃**.

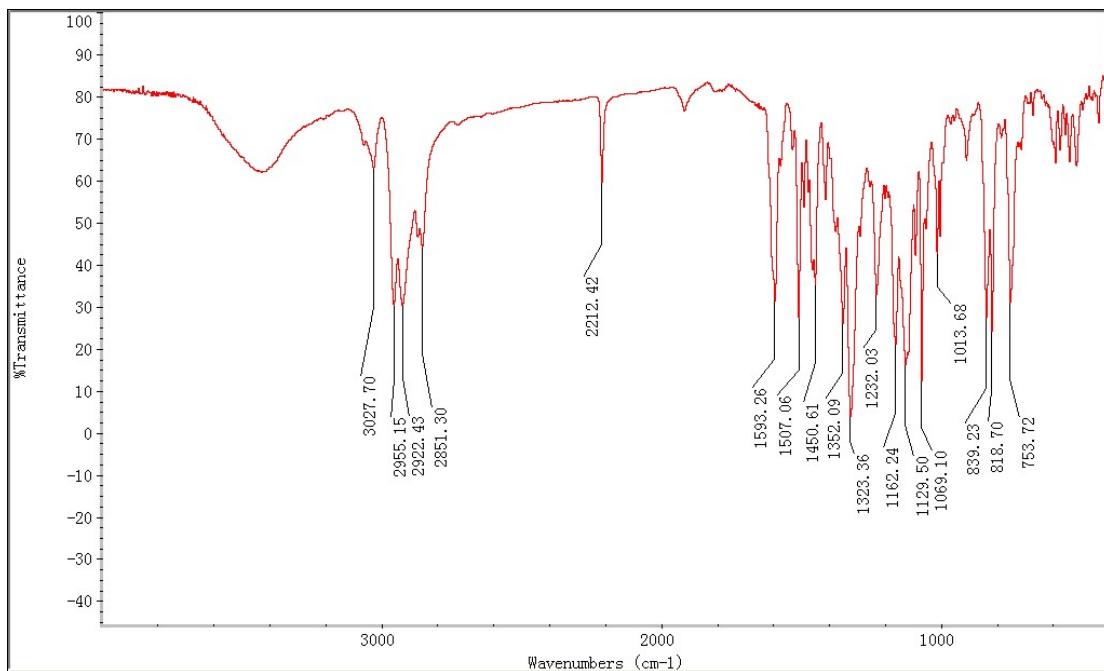


Fig. S13 FT-IR spectrum of ICZCF₃.

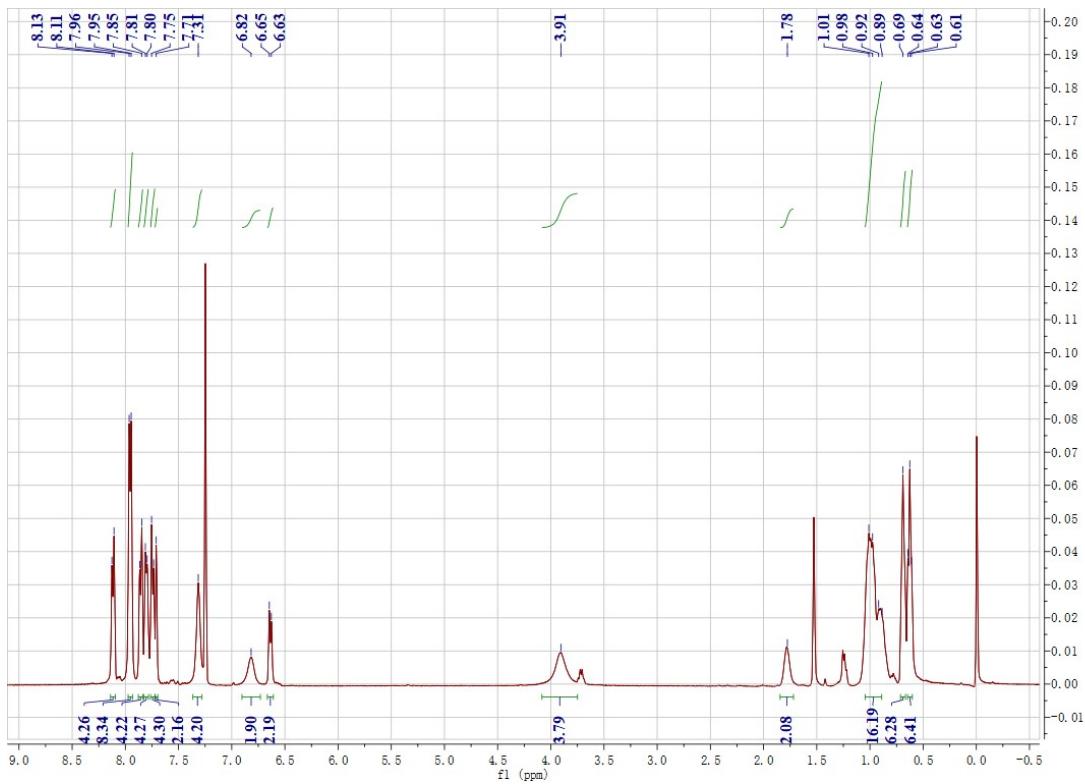


Fig. S14 ^1H NMR spectrum of **ICZCF₃** in CDCl₃.

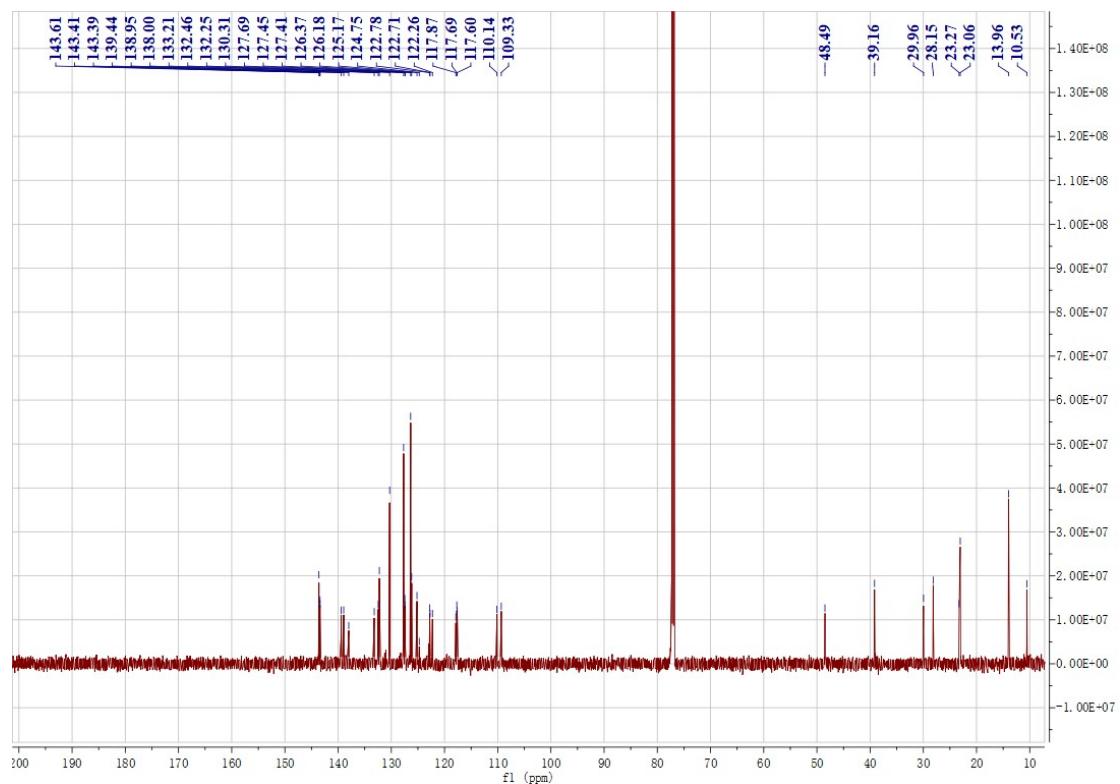


Fig. S15 ^{13}C NMR spectrum of ICZCF_3 in CDCl_3 .

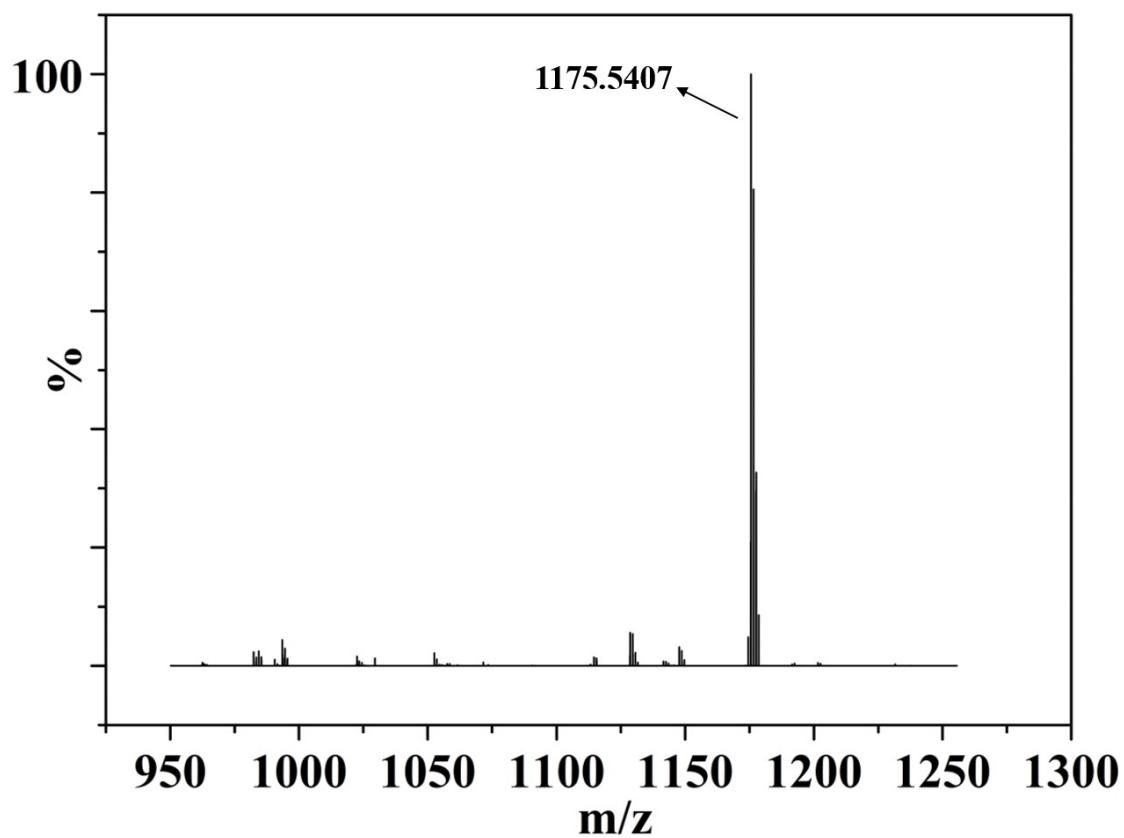


Fig. S16 HP-MS spectrum of ICZCF_3 .

Table S1 Summary of crystallographic data for compound **ICZCF₃**

Identification	ICZCF₃	Data/restraints/parameters	6759/6/400
Wavelength (Å)	1.54178	Final R indices [I>2σ(I)]	$R_I = 0.2572$ $wR_2 = 0.5680$
Crystal system	triclinic		
Space group	pī	Empirical formula	C ₇₈ H ₆₈ F ₆ N ₄
<i>a</i> (Å)	10.0055(2)	Formula weight	1175.36
<i>b</i> (Å)	13.0597(3)	<i>F</i> (000)	618
<i>c</i> (Å)	14.2217(3)	Volume	1696.84(6)
α (°)	101.9360(10)	Z, Calculated density	2, 1.150 g/cm ³
β (°)	104.1170(10)	<i>Goodness-of-fit on F²</i>	2.705
γ (°)	101.9760(10)	Largest diff. peak and hole	3.883 and -0.629 e.A ⁻³
Reflections collected / unique	14629/6759 [R(int) = 0.0342]	θ range for data collection	3.33 to 74.47 °
	-12 ≤ <i>h</i> ≤ 11		
Limiting indices	-16 ≤ <i>k</i> ≤ 15 -17 ≤ <i>l</i> ≤ 17	R indices (all data)	$R_I = 0.2756$ $wR_2 = 0.5934$

Table S2 Selected Bond distance (Å), bond angle (°) and torsion angle (°) for compound **ICZ-CF₃**

Bond distance (Å)		bond angle (°)		torsion angle °)	
F3-C1	1.326(9)	C31-N2-C24	108.0(4)	F2-C1-C2-C5	-121.24
N2-C32	1.463(6)	C31-N2-C32	123.0(5)	F2-C1-C2-C3	58.02
N2-C31	1.385(7)	C15-C14-C17	121.7(5)	C1-C2-C3-C4	179.08
N2-C24	1.403(7)	C13-C14-C17	119.9(5)	N1-C9-C8-C7	2.23
N1-C9	1.161(8)	C5-C2-C1	122.3(6)	C10-C8-C7-C6	169.46
C22-C21	1.382(7)	F2-C1-F3	105.7(7)	C11-C10-C8-C9	-7.76
C22-C17	1.407(8)	F2-C1-C2	114.4(6)	C13-C14-C17-C22	-142.14
C23-C20	1.500(6)	C6-C7-C8	120.9(5)	C21-C20-C23-C24	92.17
C14-C17	1.478(7)	N2-C32-C33	115.5(5)	C23-C24-N2-C32	2.70
C11-C10	1.464(7)	C34-C33-36	108.4(14)	C33-C32-N2-C31	92.63
C8-C7	1.483(7)	C27-C26-C25	135.4(5)	C20-C23-C25-C26	1.82
C31-C26	1.417(7)	C31-C26-C25	106.1(5)	C32-C33-C34-C35	162.91
C29-C28	1.374(9)	C32-C33-C36	105.3(8)	C33-C36-C37-C38	152.68
C29-C30	1.383(8)	C24-C25-C26	106.9(4)	F1-C1-C2-C3-C5	-114.94

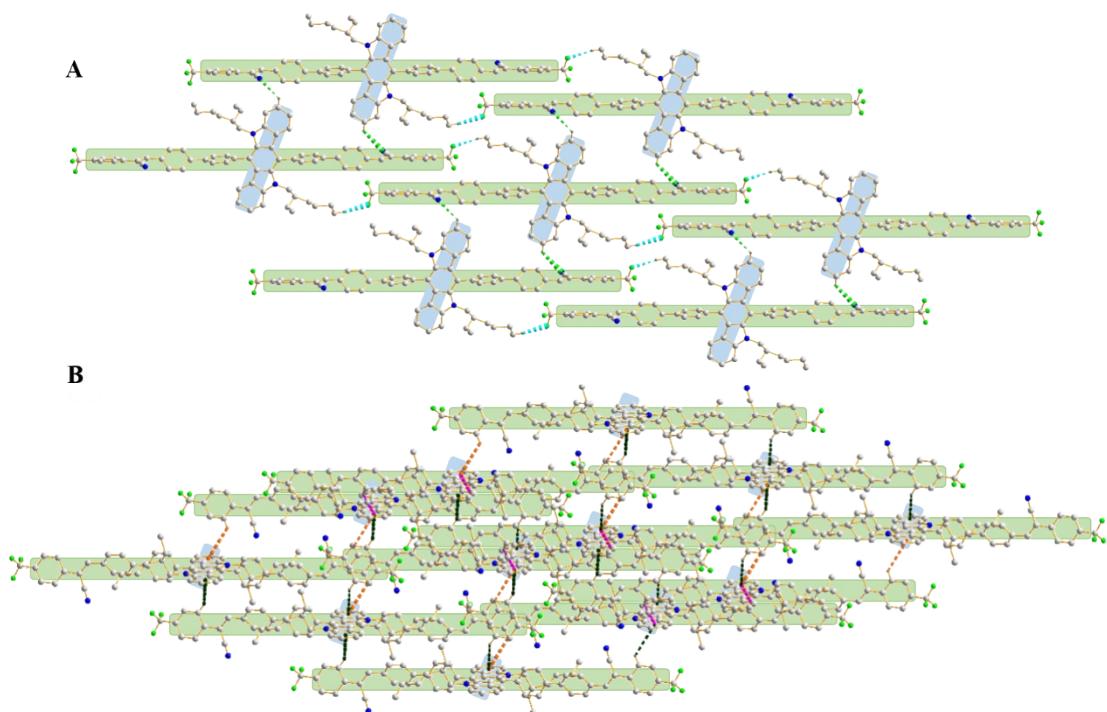


Fig. S17 (A) The molecular packing by hydrogen bond interaction and (B) C-H...πinteraction (hydrogen atoms are omitted for the sake of clarity).

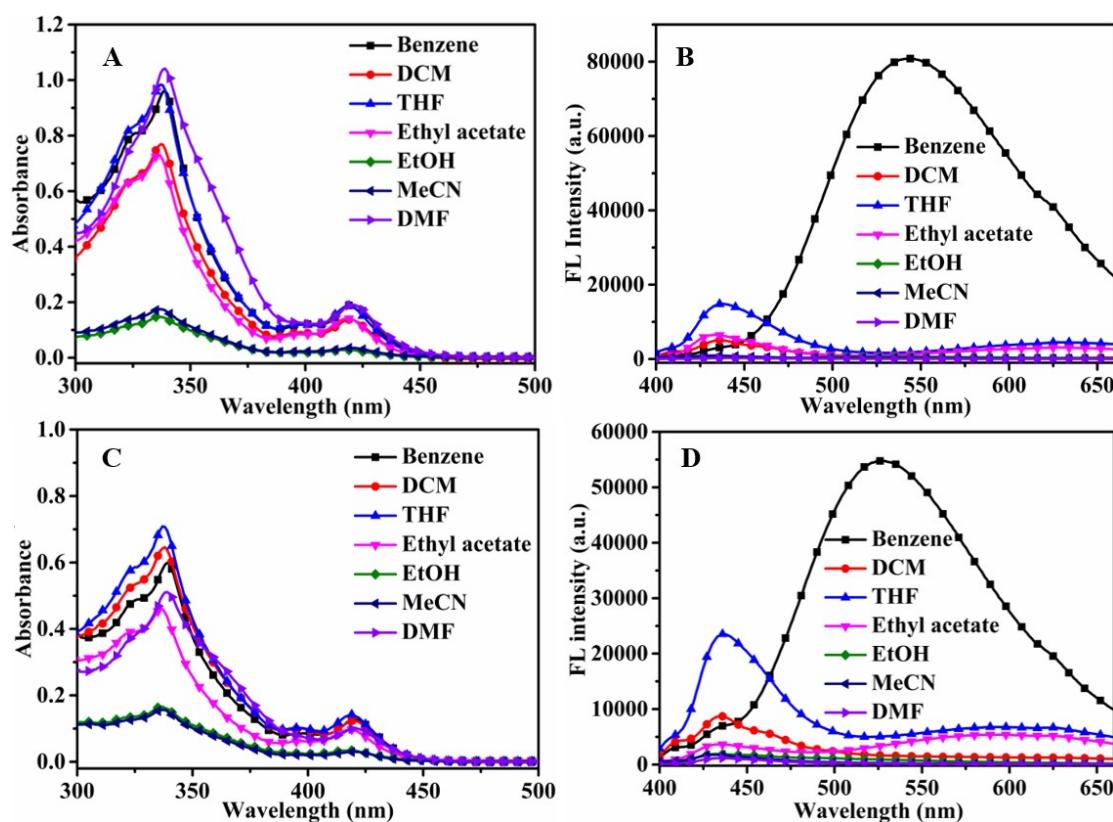


Fig. S18 (A, C) The UV-vis spectra and (B, D) fluorescence spectra of compounds ICZH and ICZCH₃ in different solvents, respectively.

Table S3 Photophysical properties of compounds **ICZ-H**, **ICZ-F**, **ICZ-CH₃** and **ICZ-CF₃** in different polar solvents.

Compounds	Solvents	λ [a] max	λ [b] max	ϵ ($\times 10^4$) ^[c]	$\Delta\nu$ [d]	Φ ^[e]
ICZH	Benzene	339	544	9.58	11116	
	DCM	338	436	7.68	6650	
	THF	337	437	9.85	6790	
	Ethyl acetate	336	434	7.28	6720	17.53%
	EtOH	336	/	1.44	/	
	MeCN	336	433	1.80	6667	
	DMF	339	/	10.42	/	
ICZF	Benzene	339	541	7.90	11014	
	DCM	338	436	9.03	6650	pristine:
	THF	337	438	10.38	6843	8.31%
	Ethyl acetate	336	438	9.64	6931	
	EtOH	334	451	1.40	7767	ground:
	MeCN	337	436	2.02	6738	11.73%
	DMF	338	443	7.08	7012	
ICZCH₃	Benzene	339	526	5.98	10487	
	DCM	337	434	6.44	6632	
	THF	338	436	7.07	6650	
	Ethyl acetate	336	433	4.57	6667	17.10%
	EtOH	336	432	1.66	6614	
	MeCN	337	432	1.56	6525	
	DMF	338	432	5.08	6438	
ICZCF₃	Benzene	338	581	1.56	12374	
	DCM	338	436	3.21	6650	pristine:
	THF	337	436	1.87	6738	5.80%
	Ethyl acetate	337	428	2.26	6309	
	EtOH	336	431	1.57	6560	ground:
	MeCN	337	432	2.39	6525	5.08%
	DMF	337	437	1.56	6790	

[a] Peak position of the longest absorption band. [b] Peak position of fluorescence emission, excited at the absorption maximum. [c] Molar absorptivity (L/cm/mol). [d] Stokes' shift in cm⁻¹.

[e] Quantum yields in the solid state.

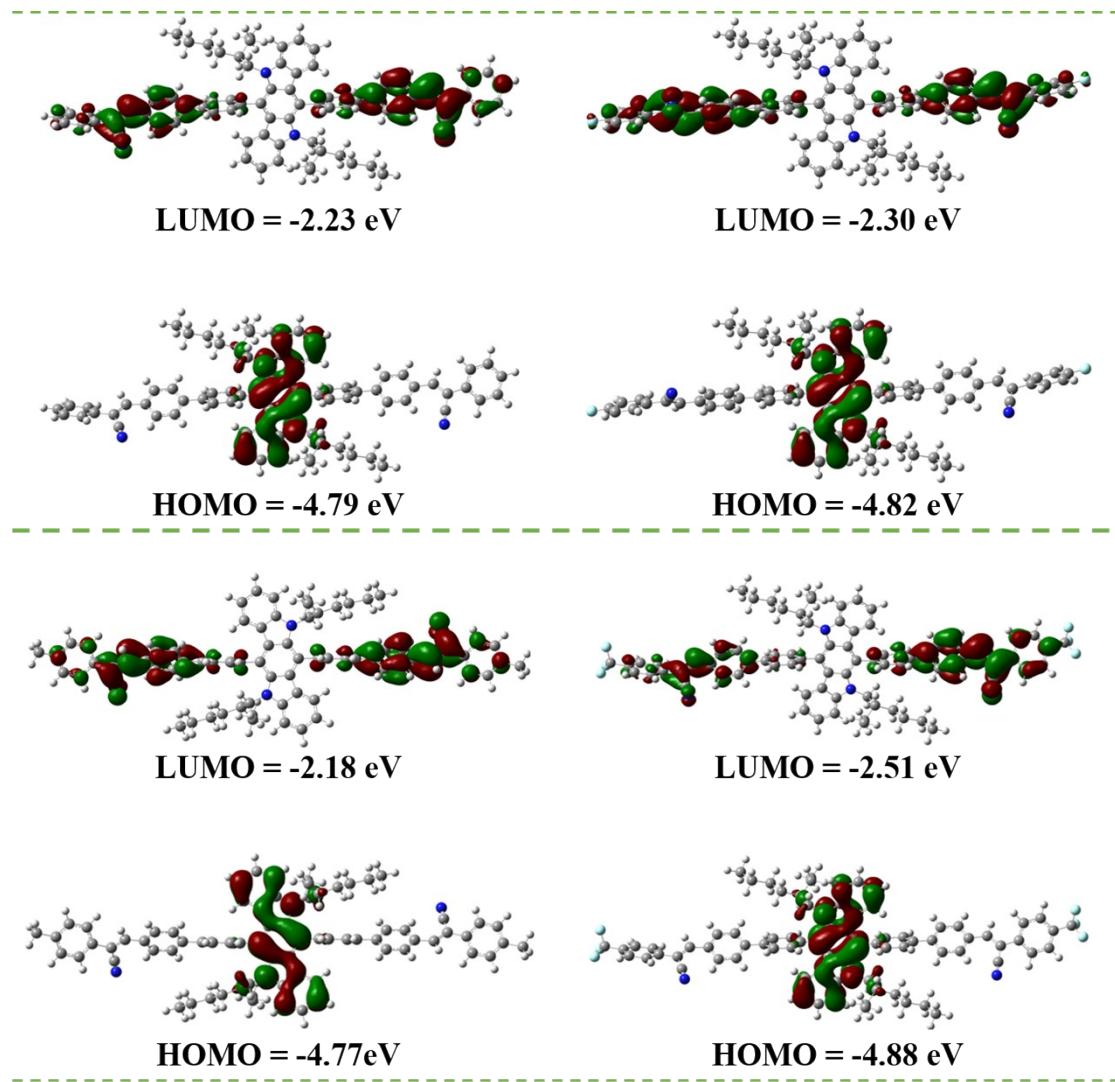


Fig. S19 Optimized molecular structure and molecular orbital amplitude plots of the HOMO and LUMO levels and electron cloud distribution of **ICZH**, **ICZF**, **ICZCH₃** and **ICZCF₃** calculated using the B3LYP/6-31G* basis set.

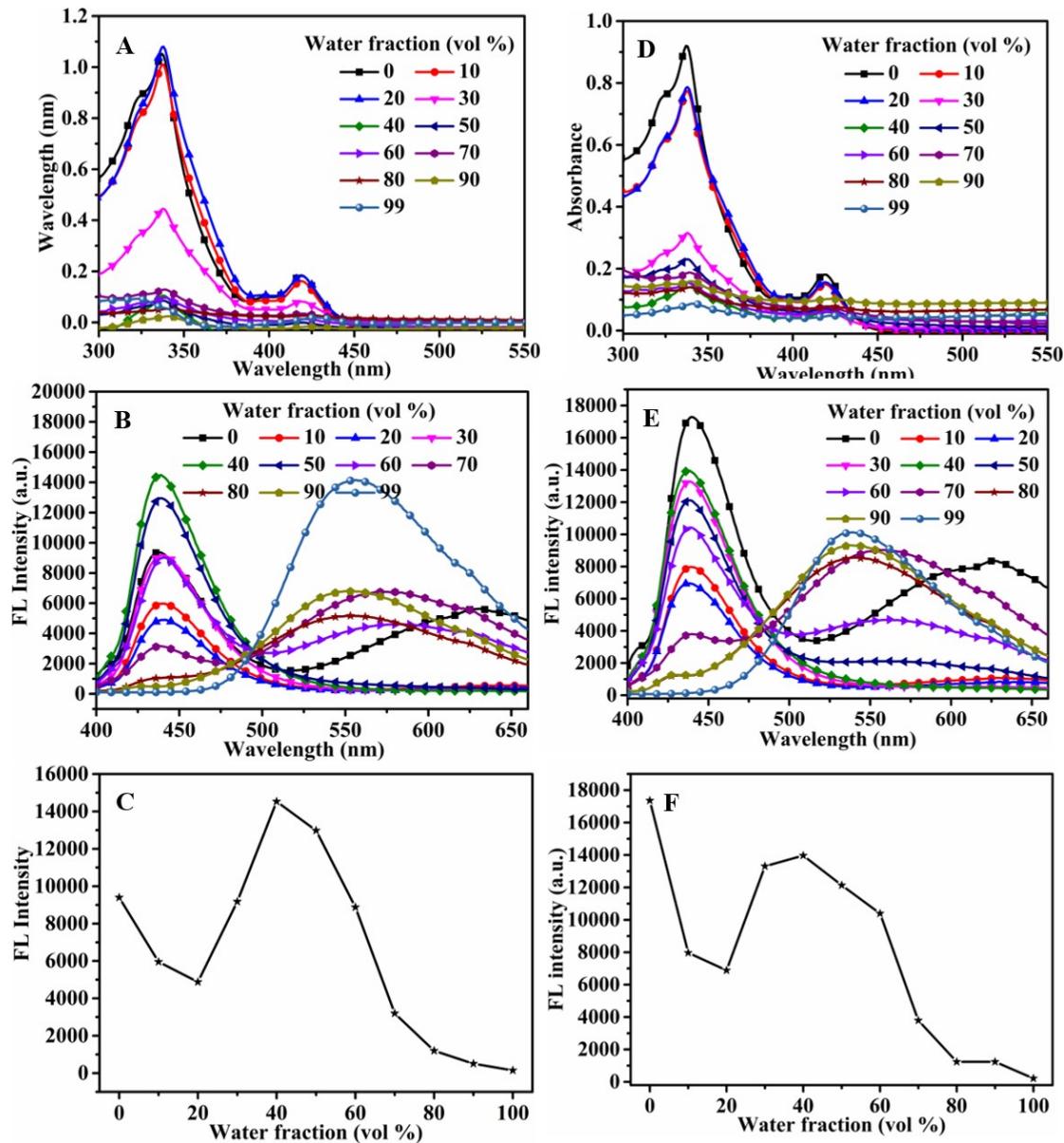


Fig. S20 (A, D) The UV-vis absorption and (B, E) fluorescence spectra of compounds **ICZH** and **ICZCH₃** in THF-water mixtures with different water volume fractions; (C, F) the effect of water volume fractions on the emission intensity of 440 nm.

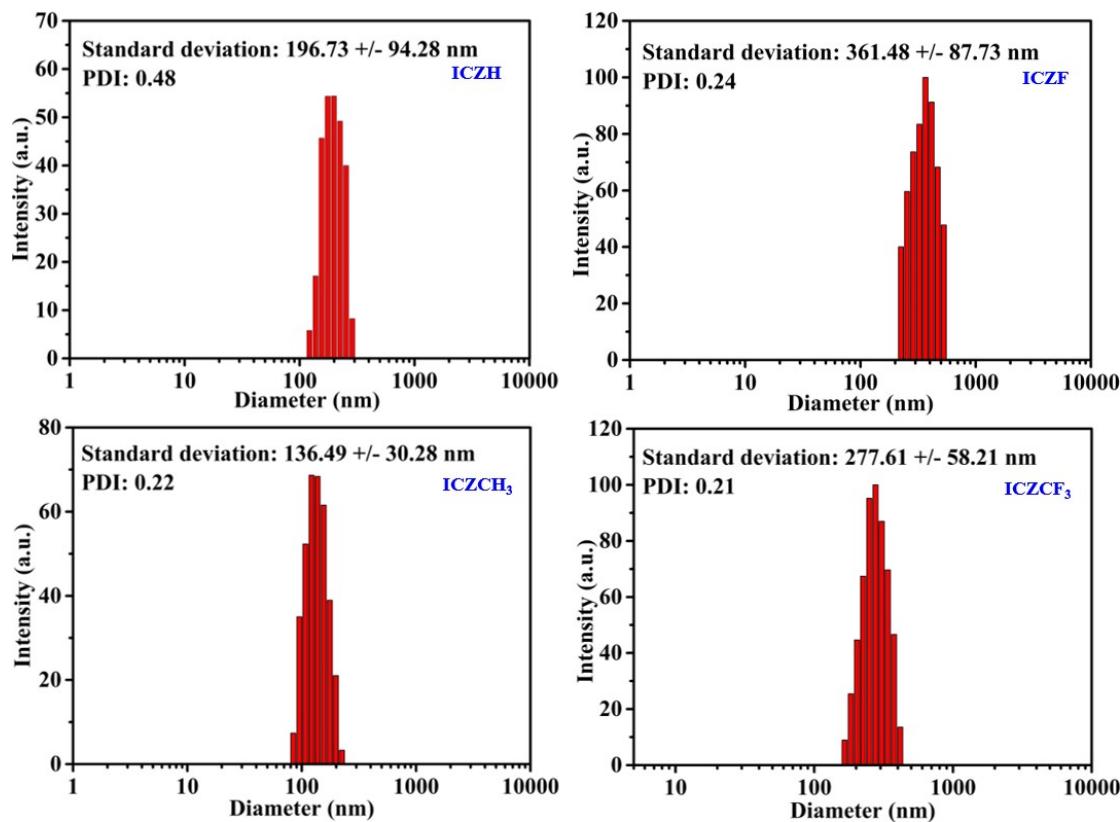


Fig. S21 Size distributions of ICZH, ICZF, ICZCH₃ and ICZCF₃ in THF-H₂O mixtures ($f_w = 50\%$).

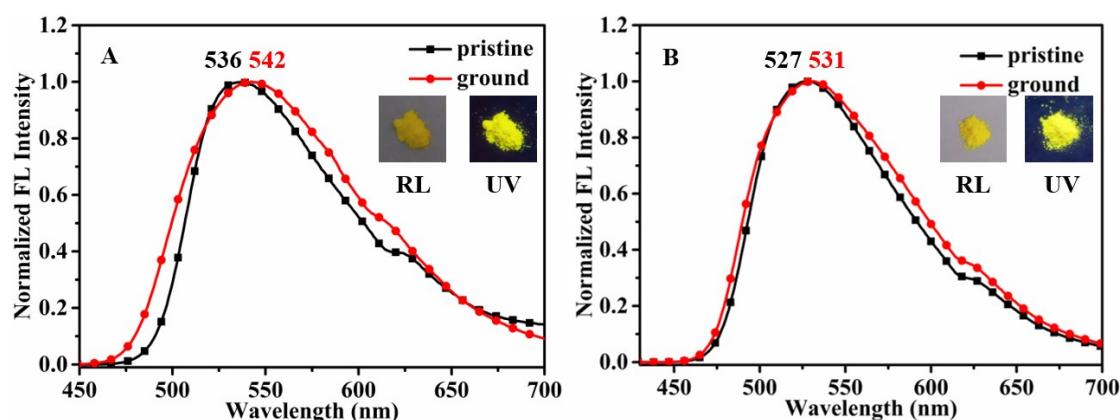


Fig. S22 Emission spectra of compound ICZH (A) and ICZCH₃ (B) of pristine and ground samples. Inset: Photographs under room light and 365 nm light.

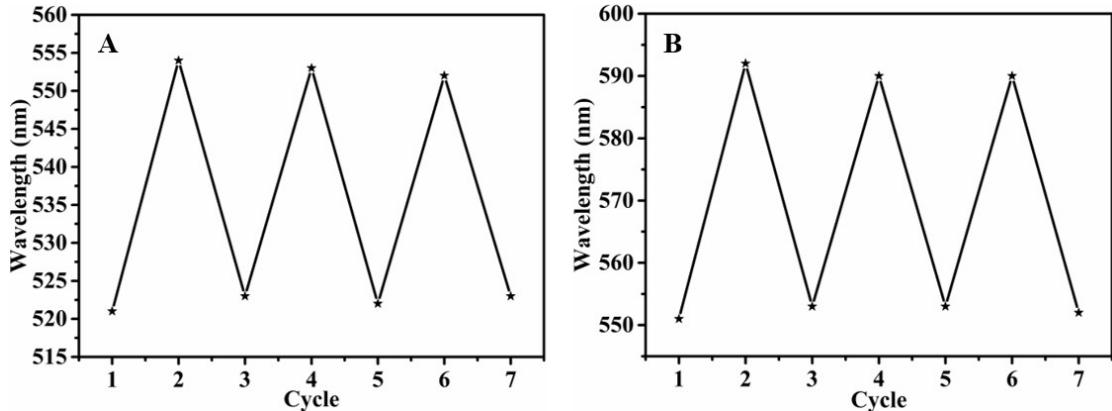


Fig S23 Repeated switching of compounds **ICZF** (A) and **ICZCF₃** (B) by ground-fumed cycles.

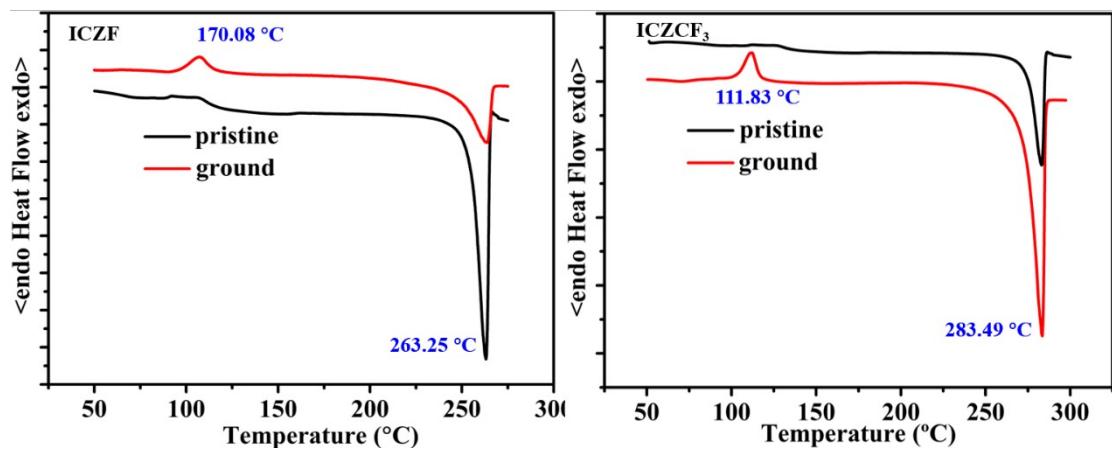


Fig. S24 DSC curve of compounds **ICZF** (A) and **ICZCF₃** (B) in the pristine and ground powders. (Scan rate: 10 °C min⁻¹).