

Supporting information

Multicomponent solid forms of antibiotic Cephalexin towards improved chemical stability

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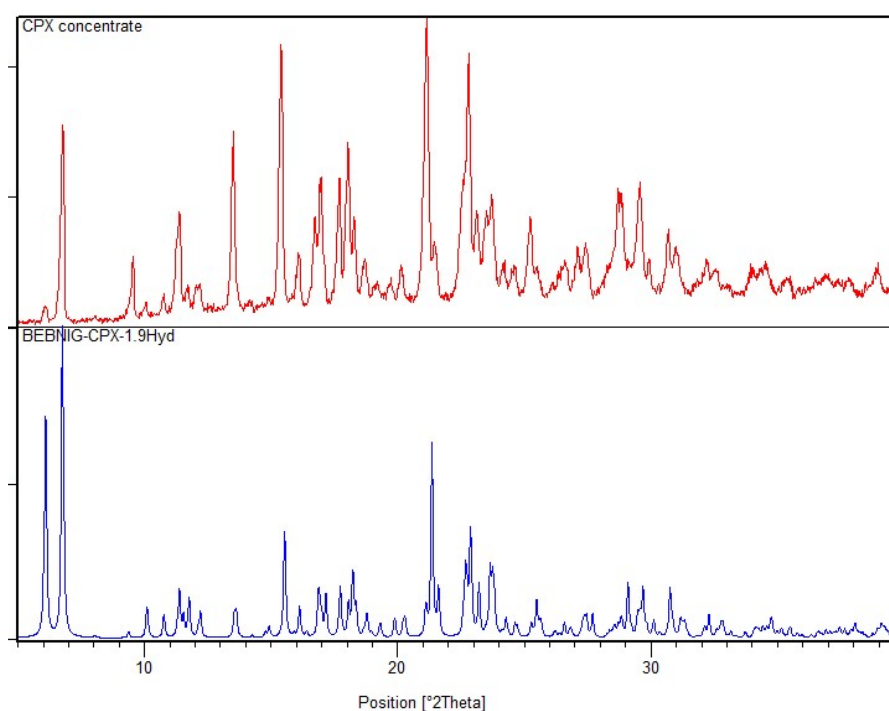
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Table S1. Hydrogen bond geometry ($\text{\AA}/^\circ$)

CPX binary solids	D-H \cdots A	D-H/ \AA	H \cdots A/ \AA	D \cdots A/ \AA	D-H \cdots A/ $^\circ$	Symmetry code
CPX-SER	N2-H2 \cdots O4	0.86	2.24	2.9792(4)	144	x,y,-1+z
	N3-H3A \cdots O1	0.89	1.73	2.5844(4)	159	1/2+x,1/2-y,-z
	N3-H3B \cdots O5	0.89	2.08	2.8598(4)	145	1-x,1/2+y,3/2-z
	N3-H3C \cdots O5	0.89	2.02	2.6942(4)	132	1-x,1/2+y,1/2-z
	N4-H4A \cdots O6	0.89	1.92	2.6659(4)	141	1/2-x,-y,-1/2+z
	N4-H4B \cdots O2	0.89	2.14	2.9803(4)	158	-
	N4-H4C \cdots O2	0.89	2.03	2.8657(4)	156	x,y,1+z
	O7-H7 \cdots O3	0.82	2.31	2.9770(4)	140	x,y,1+z
	C1-H1 \cdots O3	0.98	2.49	3.1797(5)	128	x,y,1+z
	C5-H5 \cdots O3	0.98	2.54	3.2274(5)	127	x,y,1+z
	C10-H10 \cdots O4	0.98	2.25	3.1225(5)	148	x,y,-1+z
	C16-H16 \cdots O5	0.93	2.16	3.0851(4)	174	1-x,1/2+y,3/2-z
	C18-H18 \cdots O6	0.98	1.86	2.8106(4)	163	x,y,-1+z
CPX-DHB .H ₂ O	O2-H2 \cdots O5	0.89	1.70	2.5699(1)	167	2-x,-1/2+y,1-z
	N2-H2A \cdots O9	0.83	2.32	3.1048(2)	156	1+x,y,z
	N3-H3A \cdots O2	0.89	2.30	3.0520(2)	143	-1+x,y,z
	N3-H3B \cdots O7	0.95	1.74	2.6720(1)	167	-
	N3-H3C \cdots O5	0.89	2.55	3.1982(2)	130	-
	N3-H3C \cdots O1	0.89	1.97	2.7153(1)	141	2-x,1/2+y,1-z
	O5-H5A \cdots O3	0.81	2.14	2.8914(2)	155	-1+x,y,z
	O5-H5B \cdots O6	0.93	1.77	2.6591(1)	159	-
	C1-H1 \cdots O9	0.98	2.36	3.2431(2)	149	1+x,-1+y,z
	C10-H10 \cdots O1	0.98	2.60	3.0112(2)	105	2-x,1/2+y,1-z

**Fig. S1.** PXRD comparison of CPX (extracted from commercialized tablets) with the calculated XRD pattern of CPX-1.9 hydrate from its crystal structure (CCDC refcode: BEBNIG). Peak to peak match confirms its bulk phase purity.

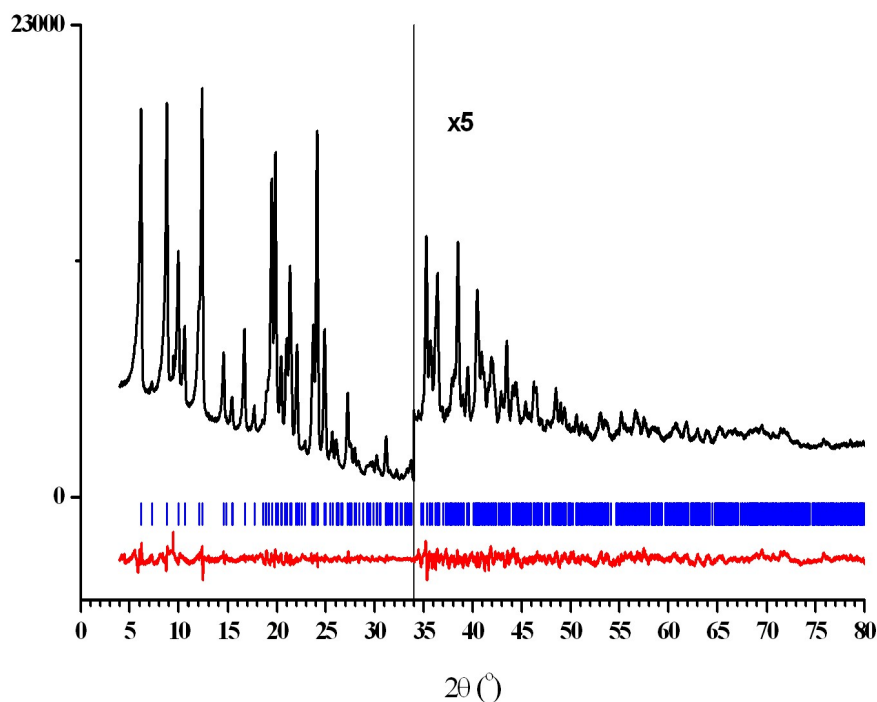


Fig. S2. The Rietveld plot after the final bond-restrained refinement for CPX-SER, showing the experimental and difference diffraction profiles as black (top) and red (bottom) curves, respectively. The vertical blue bars correspond to the calculated positions of the Bragg peaks.

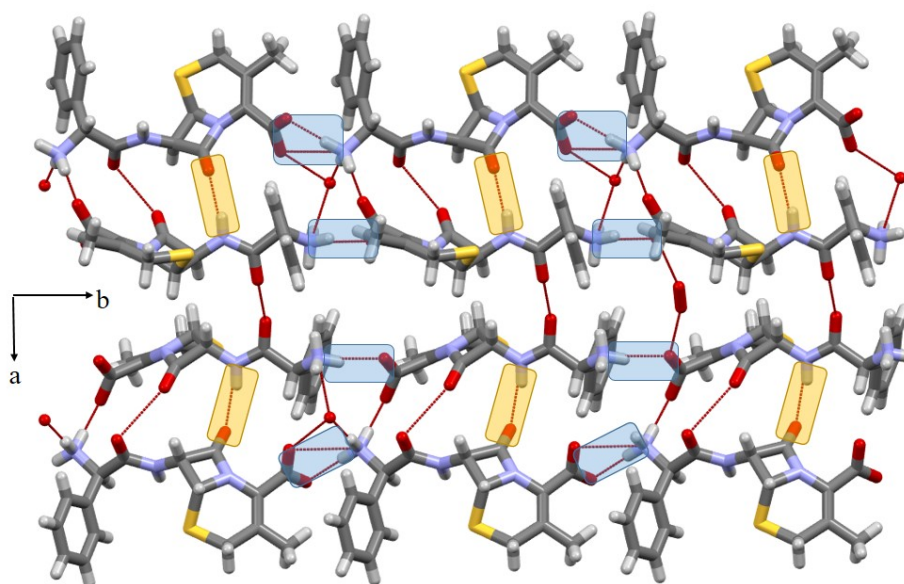


Fig. S3. CPX molecules form $N^+ \cdots H \cdots O^-$ hydrogen bonded 1D chain between carboxylate and ammonium ions along the b axis, whereas $N-H \cdots O$ hydrogen bond (amide NH/ and lactam CO) and $O \cdots O$ short contacts along the a axis resulted 2D sheet structure in CPX hydrate.

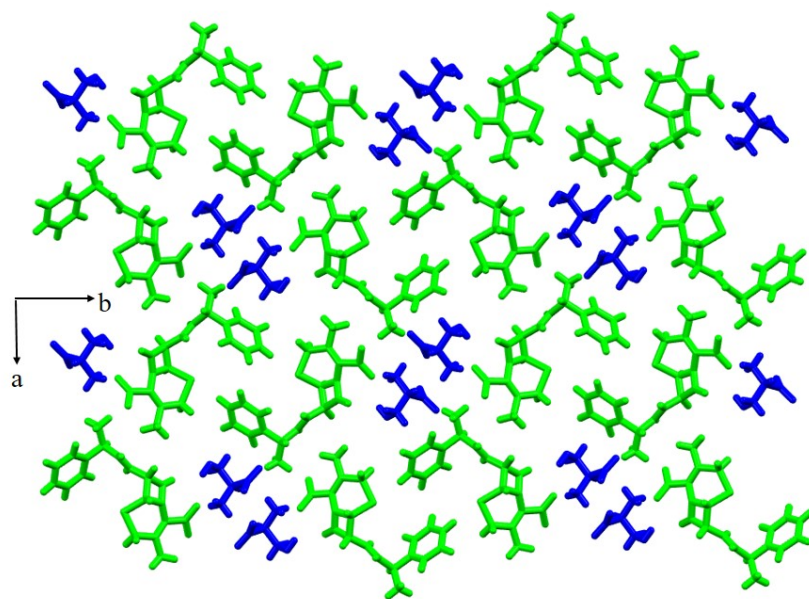
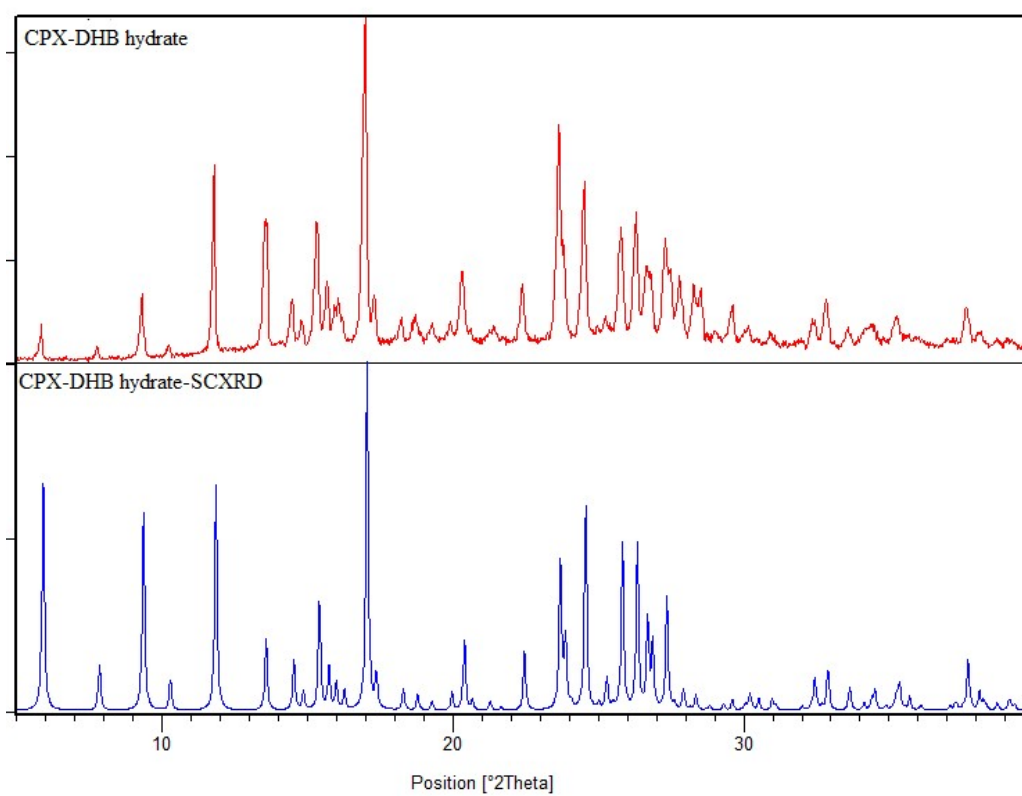
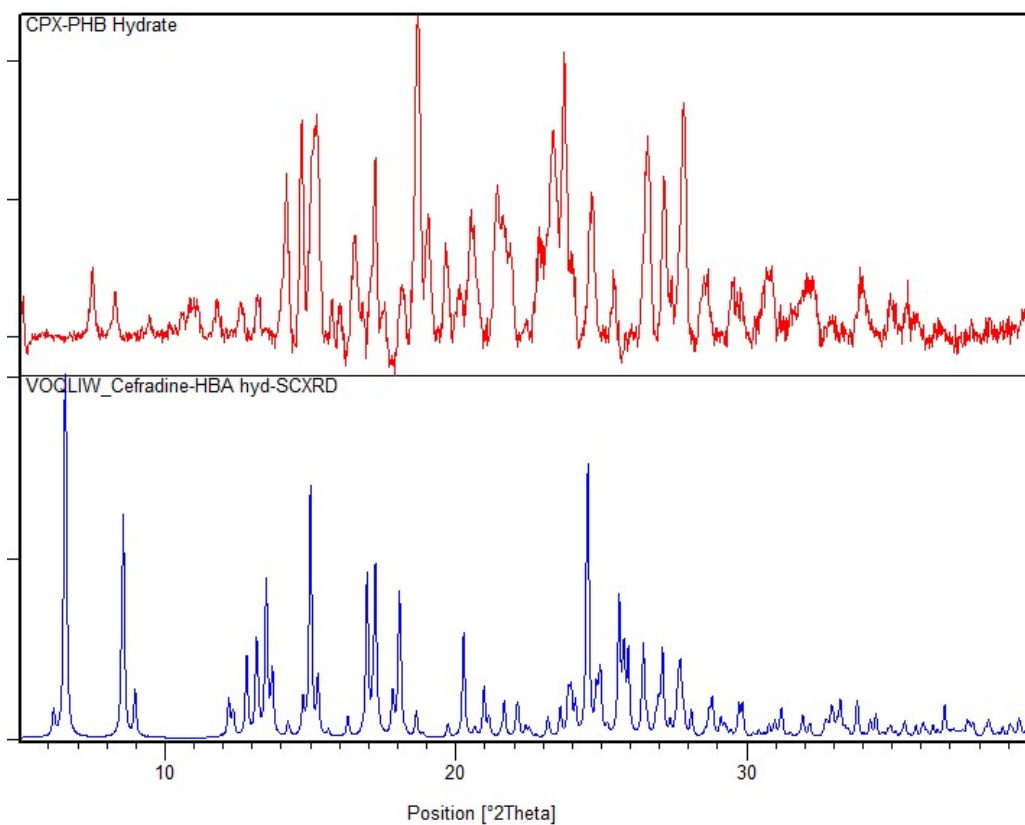


Fig. S4. Molecular packing of CPX–SER cocrystal viewed down the c axis represent host (CPX, green trace)-guest (SER, blue trace) assembly.

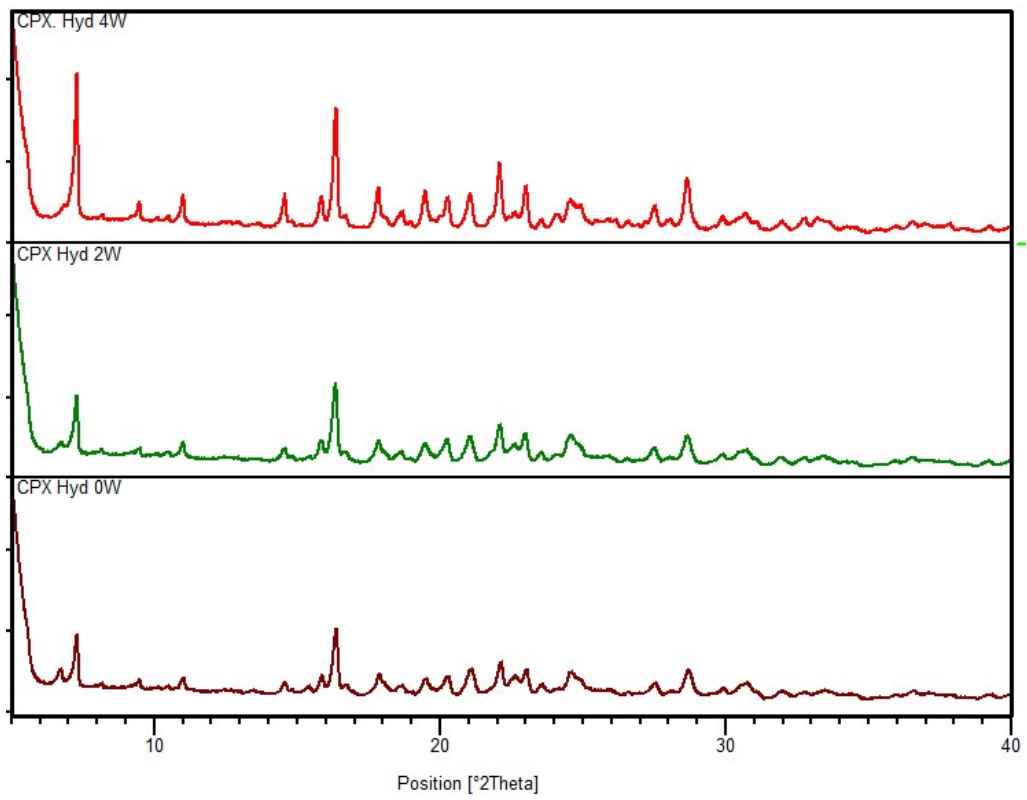


(a)

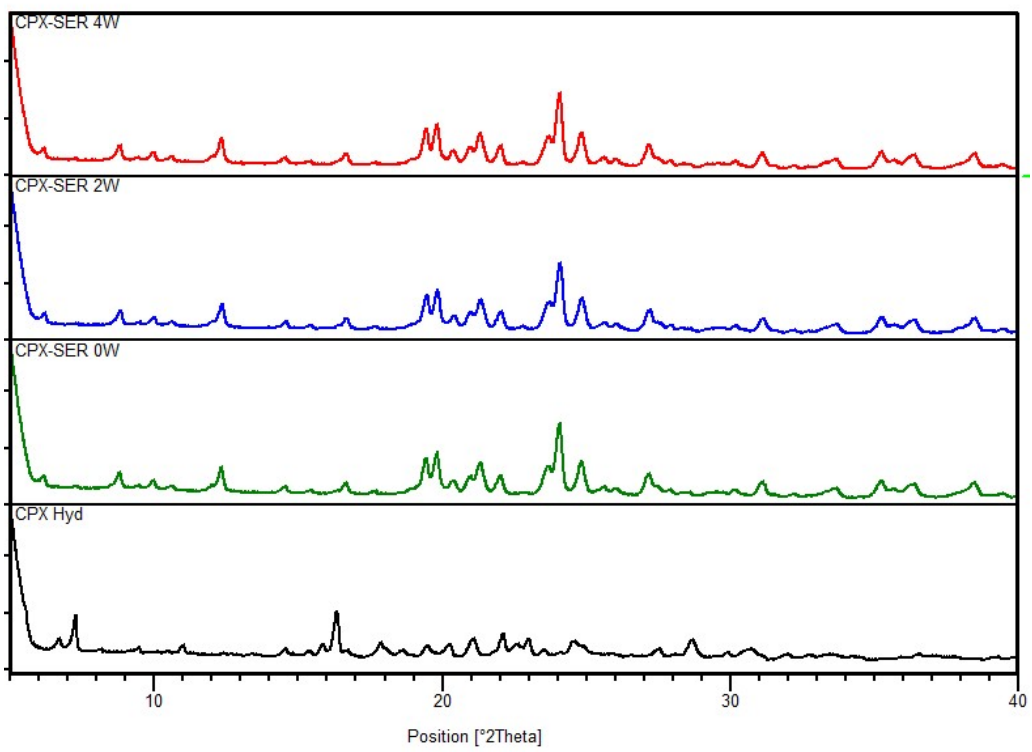


(b)

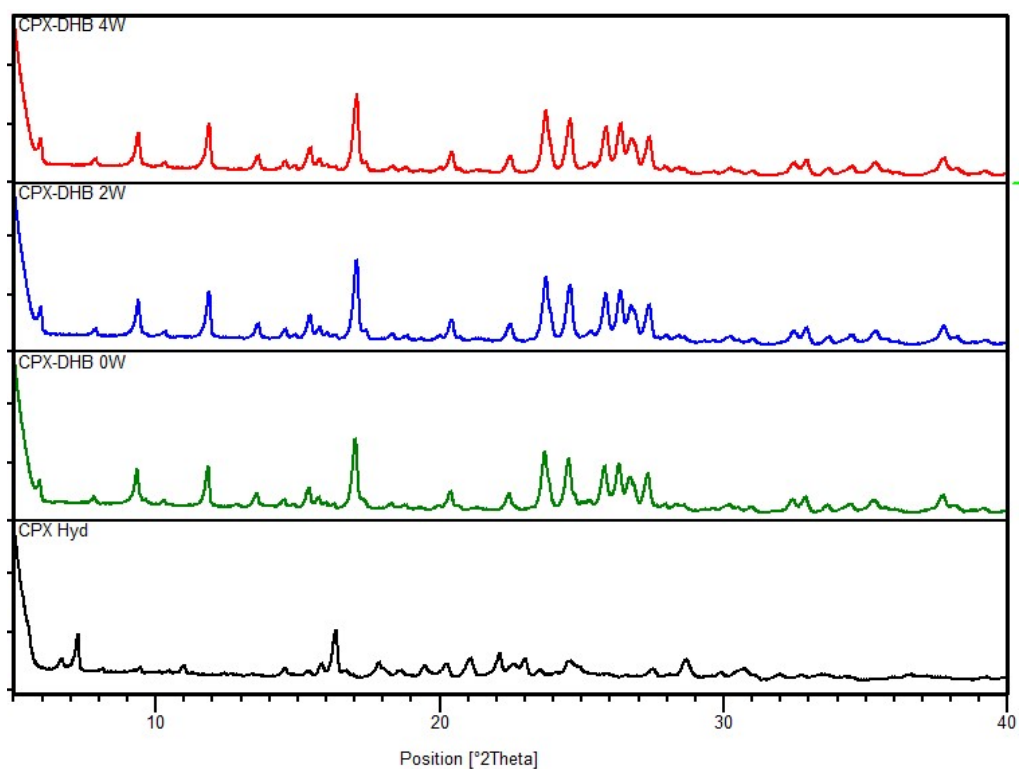
Fig. S5. a) Overlay of PXRD (red trace) of CPX–DHB salt hydrate with its calculated X-ray patterns (blue trace) from crystal structure confirms the bulk phase purity. b) Overlay of PXRD pattern of CPX–PHB hydrate with the simulated XRD pattern of VOQLIW indicated partial match that suggested possibility of multiple phases may be present in the sample.



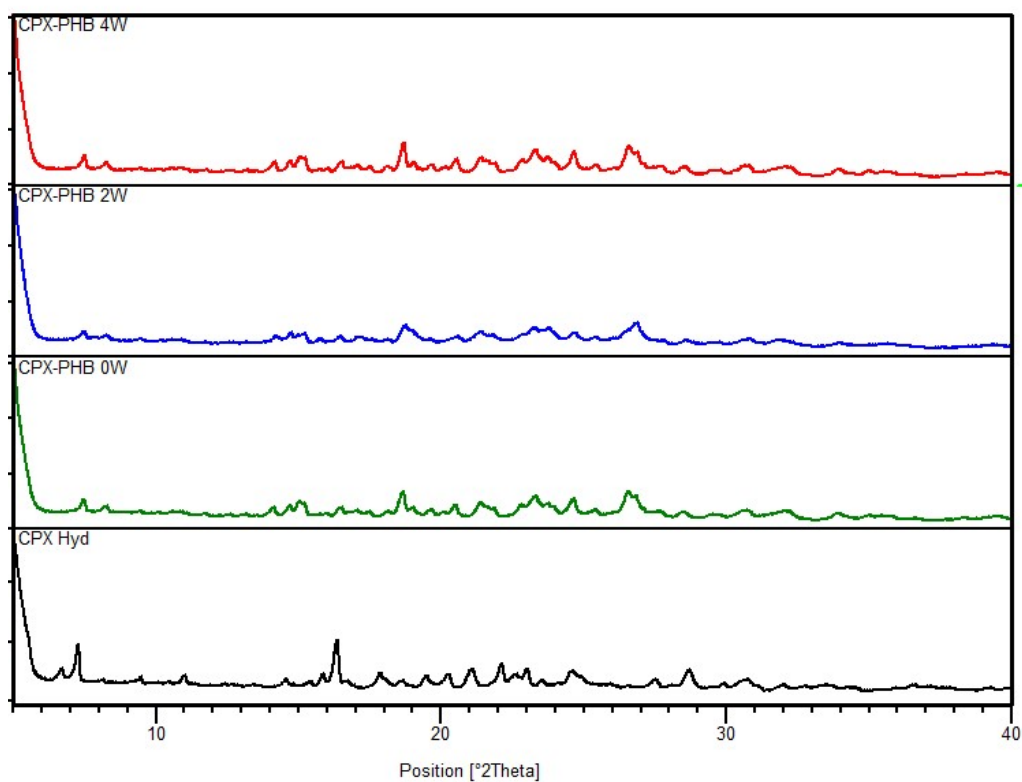
(a)



(b)



(c)



(d)

Fig. S6. PXRD comparison of a) CPX Hyd, b) CPX-SER, c) CPX-DHB, and d) CPX-PHB upto 4 weeks in 35 ± 5 °C and $75\pm 5\%$ relative humidity that confirmed the physical stability of all the multicomponent solids as CPX hydrate.

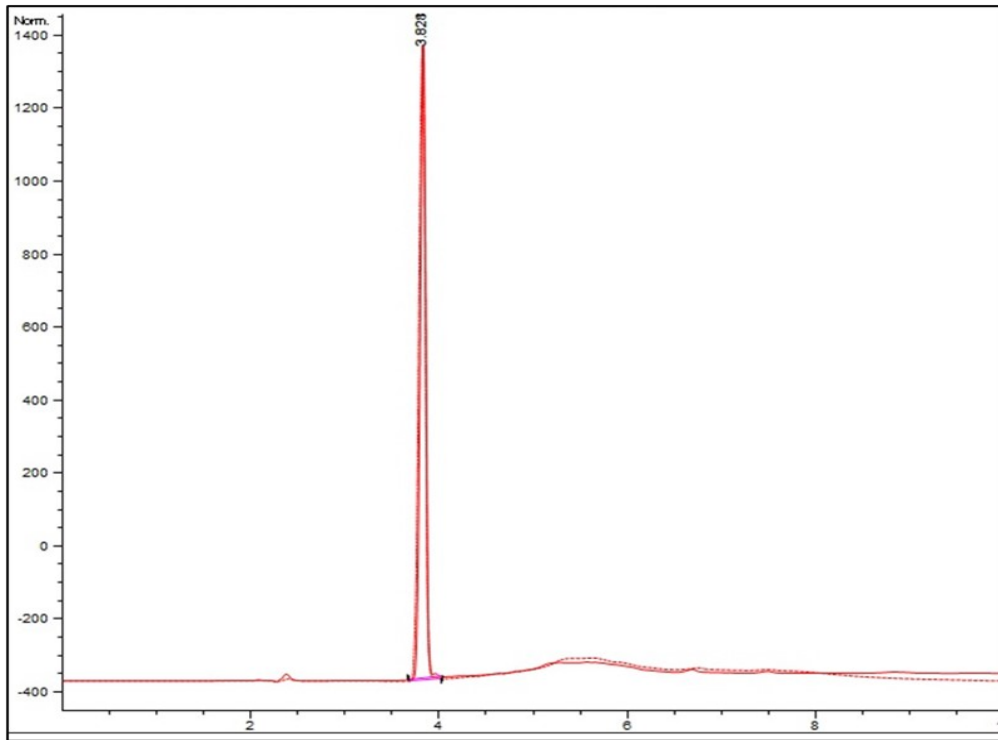
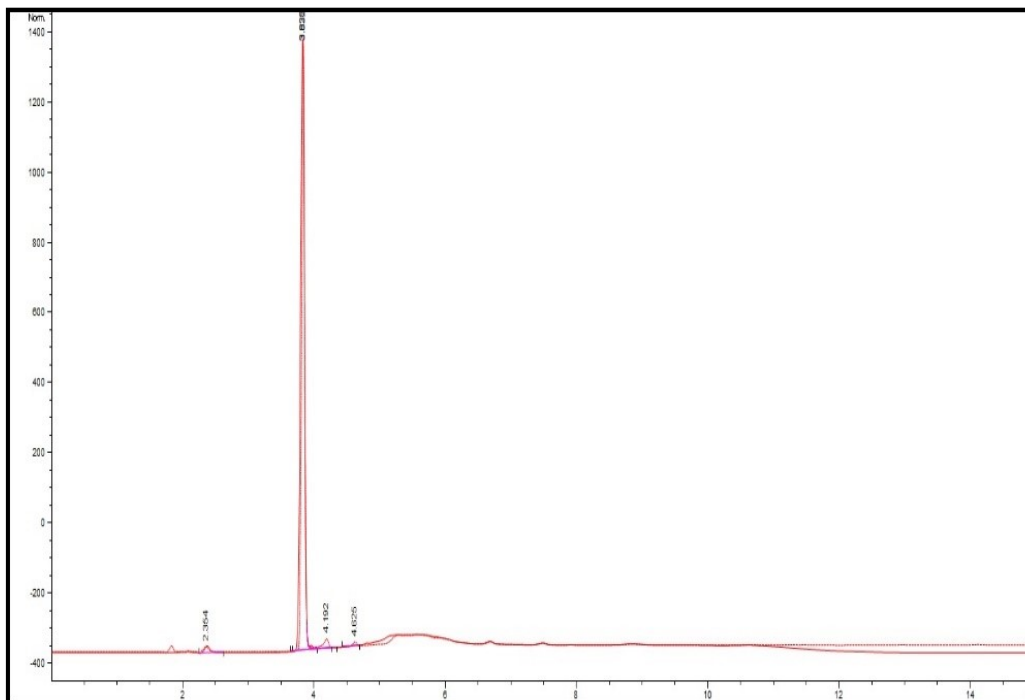
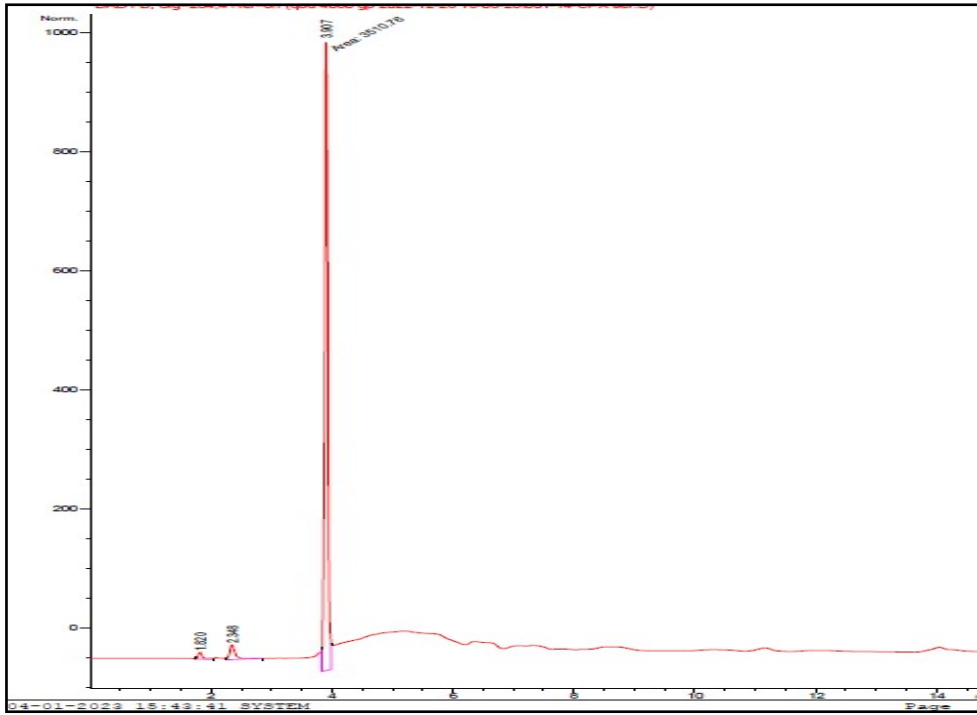


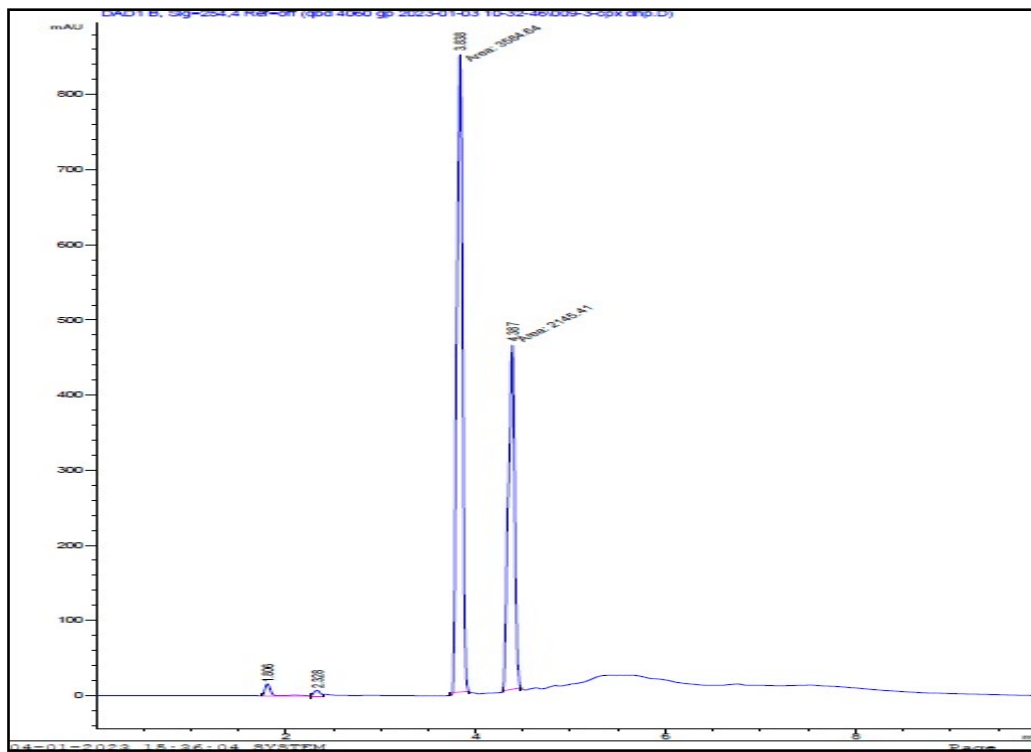
Fig. S7. HPLC overlay of CPX standard and CPX Sample used for cocrystallization



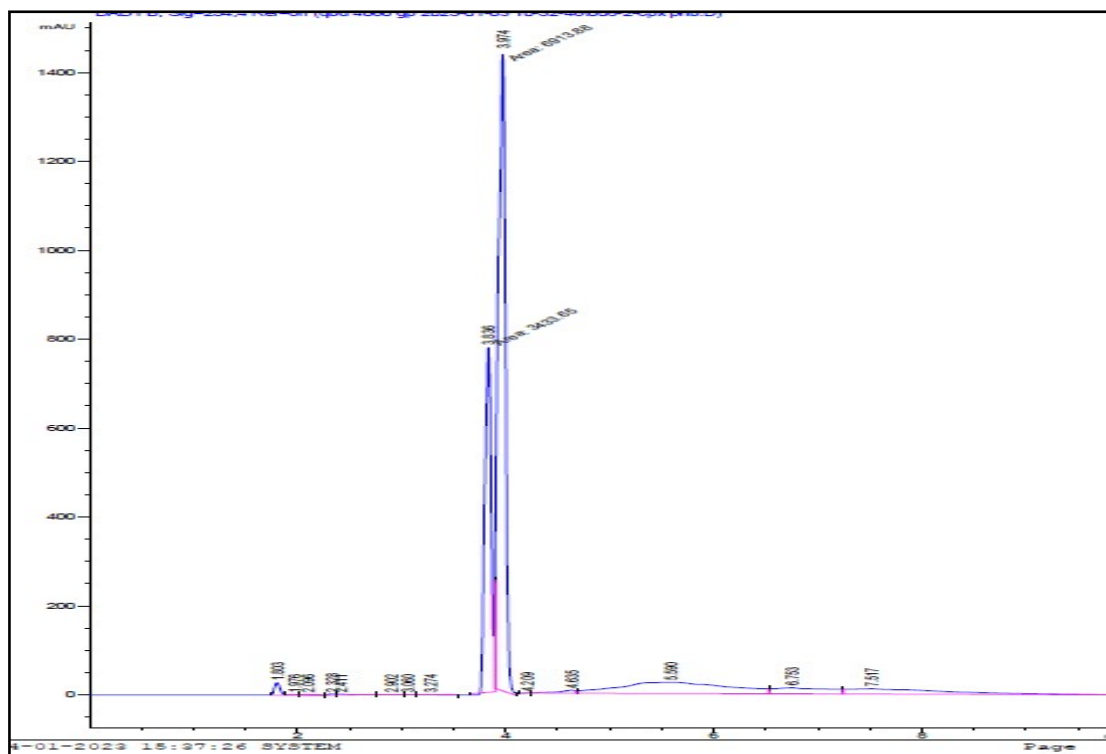
(a)



(b)



(c)



(d)

Fig. S8. HPLC chromatogram of a) CPX, b) CPX-SER, c) CPX-DHB and d) CPX-HBA following acid hydrolysis.