

Two Cu(I) coordination polymers based on a new benzimidazolyl-tetrazolyl heterotopic ligand for visible-light-driven photocatalytic dye degradation

Ping Tang,^{a, ‡} Xue-Xian Xie,^{a, ‡} Zi-Yuan Huang,^a Zhi-Yang Kuang,^a Song-Liang Cai,^a Wei-Guang Zhang,^{a,b,*} Sheng-Run Zheng^{a,*}

^aSchool of Chemistry, Guangzhou Key Laboratory of Analytical Chemistry for Biomedicine, South China Normal University, Guangzhou, 510006, P. R. China.

^bGDMPA Key Laboratory for Process Control and Quality Evaluation of Chiral Pharmaceuticals, South China Normal University, Guangzhou 510006, China.

[‡]These authors are contributed equally to the work.

* Corresponding author: Prof. Sheng-Run, Zheng; Prof. Wei-Guang Zhang

E-mail address: zhengsr@scnu.edu.cn; wgzhang@scnu.edu.cn

Tel./Fax.: +86-20-39310187

Supporting Information

Table S1 Crystallographic data and structure refinement summary for CPs 1-2

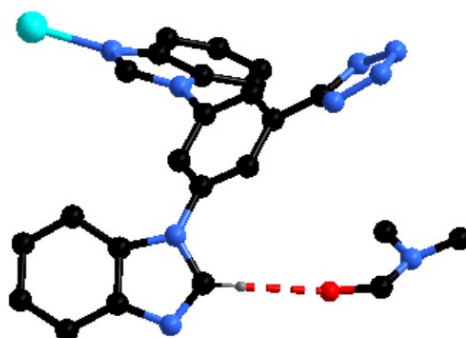
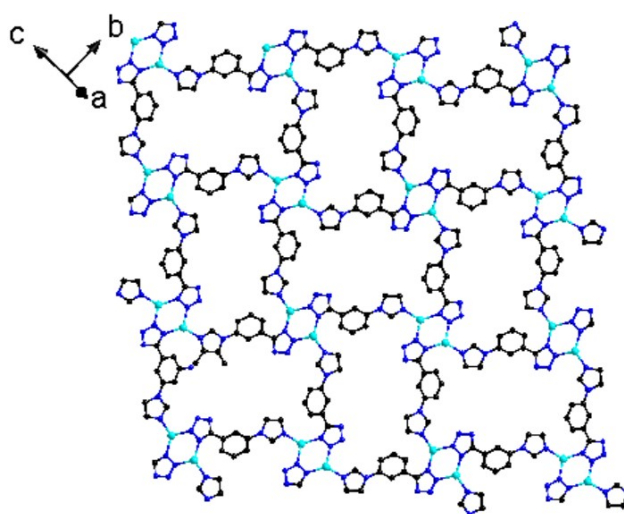
Complex	CP 1	CP 2
Empirical formula	C ₂₁ H ₁₅ CuN ₈ O	C ₂₄ H ₂₀ CuN ₉ O
Formula weight	458.95	514.03
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> 21/ <i>c</i>	<i>P</i> 21/ <i>n</i>
<i>a</i> / Å	15.5585(2)	10.6532(5)
<i>b</i> / Å	16.1849(3)	16.4768(8)
<i>c</i> / Å	7.67250(10)	13.9291(7)
α / °	90	90
β / °	100.337(2)	102.397(2)
γ / °	90	90
<i>V</i> / Å ³	1900.68(5)	2388.0(2)
<i>Z</i>	4	4
<i>D</i> / g cm ⁻³	1.604	1.430
μ / mm ⁻¹	1.906	0.951
<i>T</i> / K	293(2)	293(2)
<i>R</i> ^a / <i>wR</i> ^b	0.0333 / 0.0954	0.0627 / 0.1575
Total / unique	10388 / 3626	37660 / 5534

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$, ^b $wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$, where $w = 1 / [\sigma^2(F_o^2) + (aP)_2 + bP]$. $P = (F_o^2 + 2F_c^2) / 3$.

Table S2 Selected bond lengths [\AA] and angles [$^\circ$] for CPs 1-2

CP 1			
Cu(1)-N(1)	1.9644(17)	N(1)-Cu(1)-N(8)#1	133.67(7)
Cu(1)-N(8)#1	1.9757(17)	N(1)-Cu(1)-N(2)#2	113.22(7)
Cu(1)-N(2)#2	2.0204(18)	N(8)#1-Cu(1)-N(2)#2	112.34(7)
C(21)-N(8)-C(20)	105.20(16)	C(20)-N(8)-Cu(1)#3	131.78(13)
C(21)-N(8)-Cu(1)#3	122.85(14)		
CP 1			
Cu(1)-N(5)#1	1.983(3)	Cu(1)-N(3)	2.085(3)
Cu(1)-N(1)#2	2.021(3)	Cu(1)-N(6)#3	2.144(3)
N(5)#1-Cu(1)-N(1)#2	127.21(12)	N(5)#1-Cu(1)-N(6)#3	113.48(11)
N(5)#1-Cu(1)-N(3)	108.73(12)	N(1)#2-Cu(1)-N(6)#3	100.64(12)
N(1)#2-Cu(1)-N(3)	108.22(13)	N(3)-Cu(1)-N(6)#3	93.12(12)

Symmetry transformations used to generate equivalent atoms: #1 $x, -y + 1/2, z + 1/2$, #2 $-x + 1, -y + 1, -z + 2$, #3 $x, -y + 1/2, z - 1/2$ for CP 1; #1 $-x, -y, -z$, #2 $x + 1/2, -y + 1/2, z + 1/2$, #3 $x + 1, y, z$, #4 $x - 1, y, z$, #5 $x - 1/2, -y + 1/2, z - 1/2$ CP 2.

**Fig. S1** The hydrogen bond between DMF and the framework in CP 2.**Fig. S2** The 2D layer in CP 2.

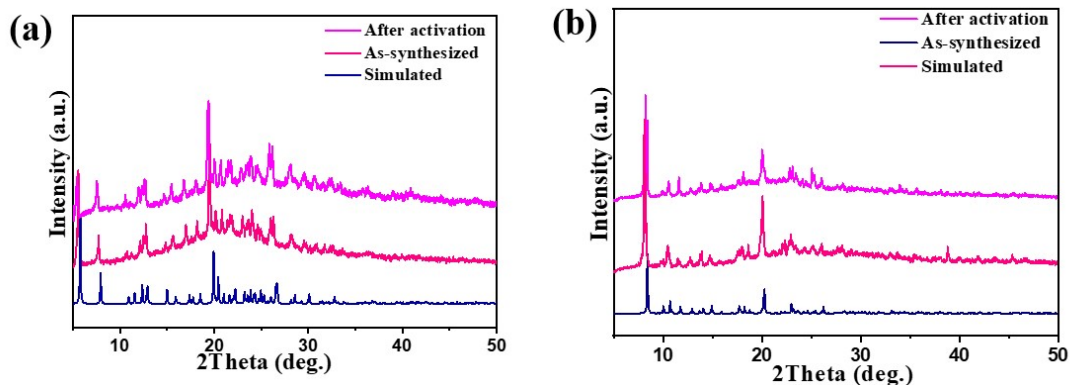


Fig. S3 PXRD of (a) Simulated and as-synthesized CP 1 and (b) Simulated and as-synthesized CP 2.

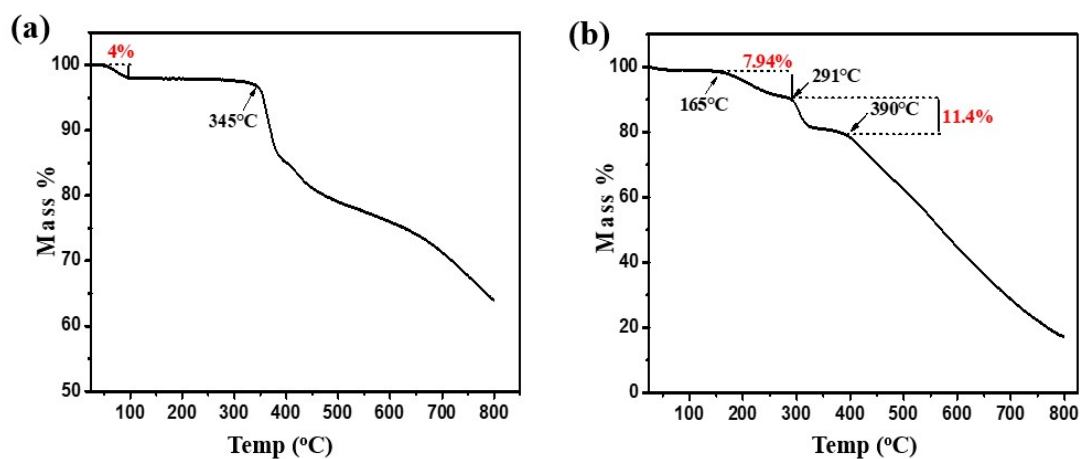


Fig. S4 TG curve of (a) CP 1 and (b) CP 2.

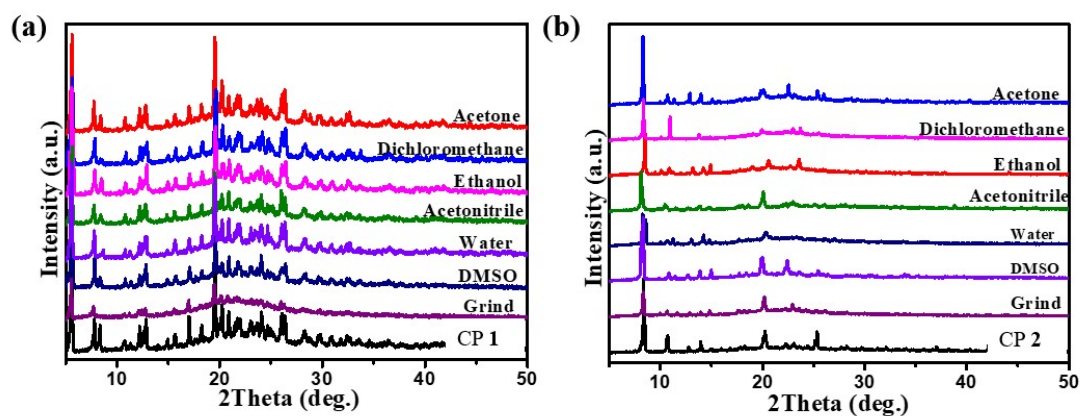


Fig. S5 PXRD of (a) CP 1 and (b) CP 2 after immersing in different organic solvents for 72 h.

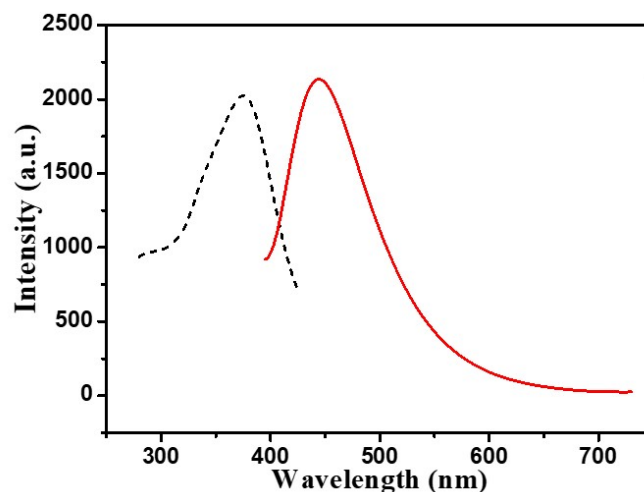


Fig. S6 Photoluminescence spectrum of HDBPT.

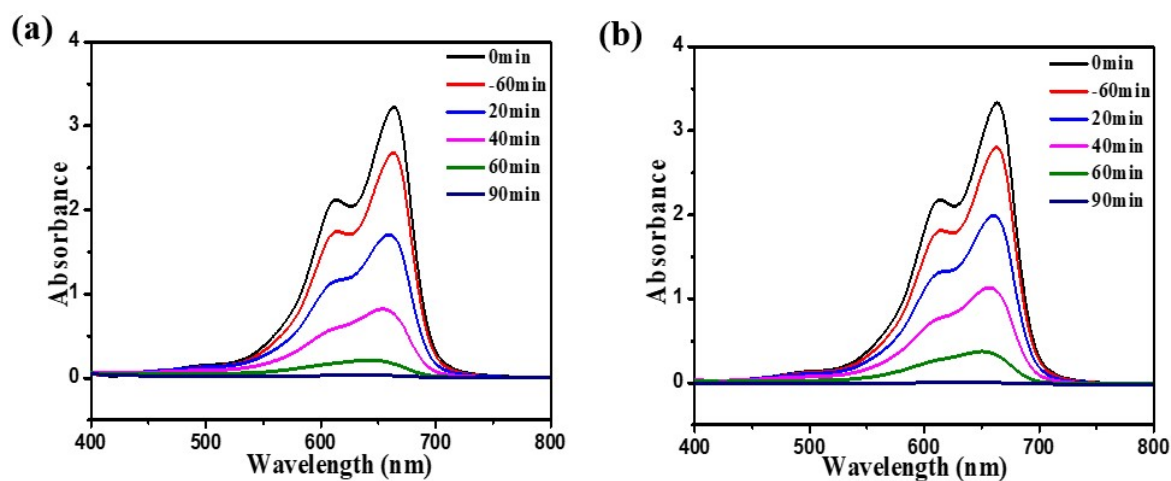


Fig. S7 UV-vis spectra of MLB solutions which were recorded after photocatalytic degradation had been performed for different lengths of time with (a) CP 1 and (b) CP 2 in the presence of H_2O_2 .

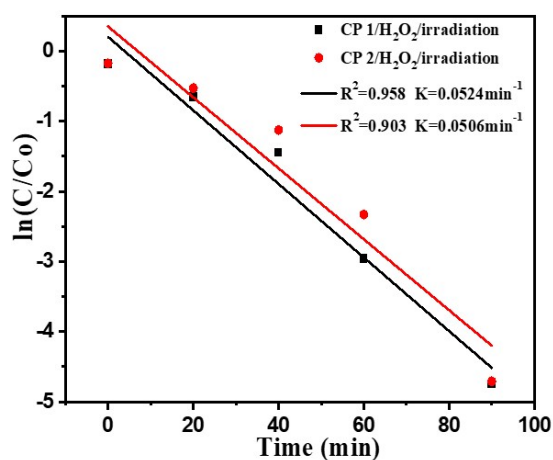


Fig. S8 Pseudo-first-order plot with respect to time for CP 1 and CP 2 in an aqueous MLB solution.

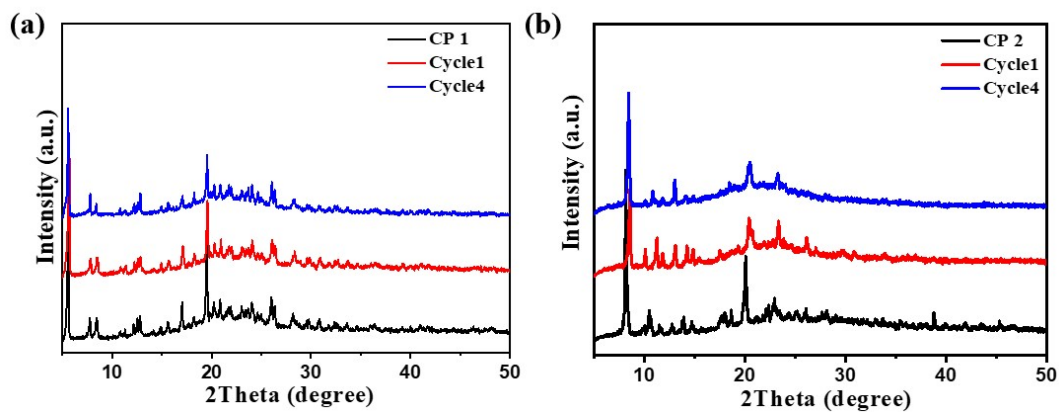


Fig. S9 The PXRD patterns of (a) CP 1 and (b) CP 2 before and after recycling test for MLB.

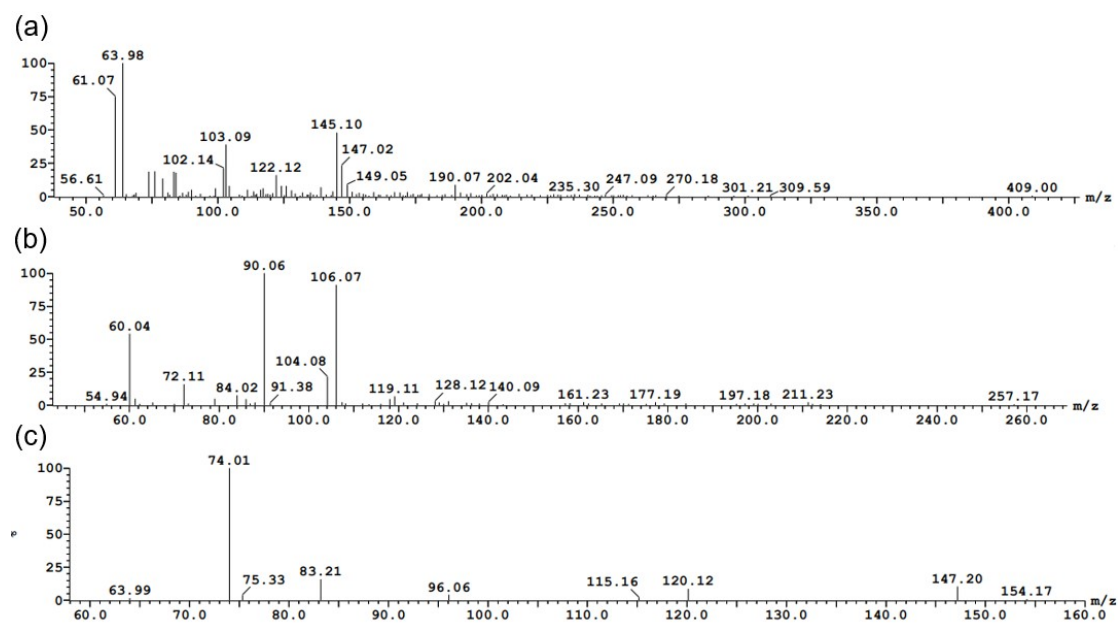


Fig. S10 The ESI positive mass spectra of MLB after reacting for 90 min by using CP 1 as catalyst.

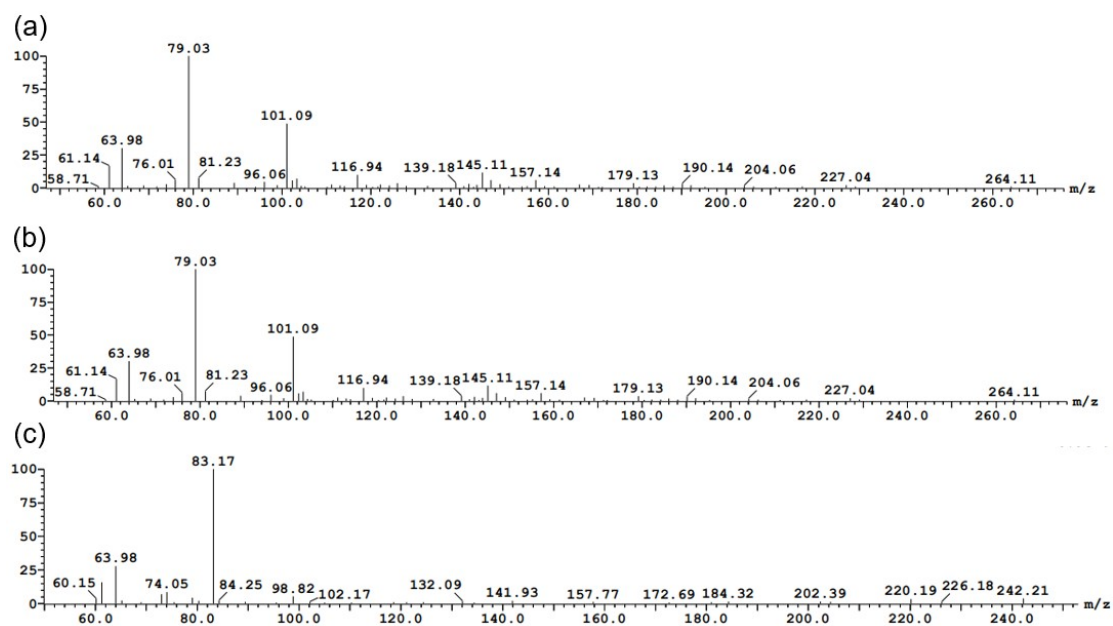


Fig. S11 The ESI positive mass spectra of MLB after reacting for 90 min by using CP 2 as catalyst.