Supporting Information for

Three Multi-responsive Luminescent Zn-CPs for Detection of

Antibiotics/Cations/Anions in Aqueous Media

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Materials and General Characteristics

The used materials of this work were of AR grade and purchased from commercial sources without further purification. The experimental powder X-ray diffraction (PXRD) of **YMUN 6–8** samples were collected on a Rigaku Miniflex 600 instrument. A Nicolet 170SX spectrometer was employed to record the infrared (IR) spectra with KBr pellets, and the measurements were in the range of 4000–400 cm⁻¹. We used the Perkin-Elmer TG-7 thermogravimetric analyzer to conduct thermogravimetric analysis (TGA) under nitrogen condition, and the heating rate was 10 °C min⁻¹ from 30 to 800 °C. A Shimadzu UV-2550 spectrophotometer was used to obtain the UV–vis spectra. Luminescence spectra were collected on a Perkin Elmer LS55 spectrophotometer.



Figure S1. View of the noncovalent supramolecular interactions in YMUN **6**. The weak interactions in the neighboring A and B segments by C-H…O hydrogen bonding interactions (the distance of C-H…O and the angle of \angle C-H…O are listed in Table S2).



Figure S2. View of the noncovalent supramolecular interactions in YMUN 7. The neighboring 2D three-fold interpenetrated structures are interlined by C-H···O hydrogen-bonding interactions (the distance of C-H···O and the angle of \angle C-H···O are listed in Table S2).



Figure S3. View of the noncovalent supramolecular interactions in YMUN 8. The neighboring 2D three-fold interpenetrated structures are interlined by C-H···O hydrogen-bonding interactions (the distance of C-H···O and the angle of \angle C-H···O are listed in Table S2).



Figure S4. PXRD patterns of YMUN 6, simulated and as-synthesized.



Figure S5. PXRD patterns of YMUN 7, simulated and as-synthesized.



Figure S6. PXRD patterns of YMUN 8, simulated and as-synthesized.



Figure S8. Photoluminescence of H₂L ligand and **YMUN 6–8** at room temperature in the solid state.



Figure S9. (A)-(C) Fluorescence spectra of YMUN 6 powder and YMUN 6 powder introduced into different antibiotics (0.5 mM), metal ions (1.0 mM), and cations (1.0 mM) in the aqueous solution at room temperature ($\lambda_{ex} = 275$ nm). (E)-(G) Fluorescence spectra of YMUN 7 powder and YMUN 7 powder introduced into different antibiotics (0.5 mM), metal ions (1.0 mM), and cations (1.0 mM) in the aqueous solution at room temperature ($\lambda_{ex} = 271$ nm). (G)-(I) Fluorescence spectra of YMUN 8 powder and YMUN 8 powder introduced into different antibiotics (0.5 mM), metal ions (1.0 mM), and cations (1.0 mM) in the aqueous solution at room temperature ($\lambda_{ex} = 275$ nm).



Figure S10. The fluorescence intensity trend spectra and the S-V linear relationship curve of **YMUN 6** after adding Ag⁺ solution.



Figure S11. The fluorescence intensity trend spectra and the S-V linear relationship curve of **YMUN 7** after adding Ag⁺ solution.



Figure S12. The fluorescence intensity trend spectra and the S-V linear relationship curve of YMUN 8 after adding Ag⁺ solution.



Figure S13. The fluorescence intensity trend spectra and the S-V linear relationship curve of YMUN 6 after adding $Cr_2O_7^{2-}$ solution.



Figure S14. The fluorescence intensity trend spectra and the S-V linear relationship curve of YMUN 7 after adding $Cr_2O_7^{2-}$ solution.



Figure S15. The fluorescence intensity trend spectra and the S-V linear relationship curve of YMUN 8 after adding $Cr_2O_7^{2-}$ solution.



Figure S16. The fluorescence spectra of **YMUN 6** in various antibiotics (0.5 mM) (Black lines), and adding equivalent **Chl** in various antibiotics solution.



Figure S17. The fluorescence spectra of YMUN 6 in various metal ions (1.0 mM) (Black lines), and adding equivalent Ag^+ in various metal ions solution.



Figure S18. The fluorescence spectra of **YMUN 6** in various anions (1.0 mM) (Black lines), and adding equivalent $Cr_2O_7^{2-}$ in various anions solution.



Figure S19. The recycling fluorescence intensity of YMUN 6 for Chl antibiotic, Ag^+ ion, and $Cr_2O_7^{2-}$ ion, respectively, after four times.



Figure S20. The comparison of PXRD patterns for YMUN 6 after Chl antibiotic, Ag⁺,

 $Cr_2O_7^{2-}$ ions sensing.



Figure S21. The UV-vis spectra of Chl antibiotic/ $Cr_2O_7^{2-}$ ion in aqueous solutions and the excitation spectrum of YMUN 6.

Table S1 Selected bond lengths (Å) and angles (°) for YMUN 6-8.

YMUN 6				
Zn1—O2	1.968 (2)	Zn2—O8 ⁱⁱⁱ	1.956 (2)	-
Zn1—O6 ⁱ	1.956 (2)	Zn2—O11	1.931 (2)	
Zn1—N4 ⁱⁱ	2.022 (3)	Zn2—N8 ^{iv}	2.033 (3)	
Zn1—N1	2.004 (3)	Zn2—N5	2.023 (3)	
O2—Zn1—N4 ⁱⁱ	91.51 (11)	O6 ⁱ —Zn1—N1	98.39 (11)	
O2—Zn1—N1	107.74 (11)	N1—Zn1—N4 ⁱⁱ	124.10 (12)	
O6 ⁱ —Zn1—O2	117.63 (10)	O8 ⁱⁱⁱ —Zn2—N8 ^{iv}	115.24 (11)	

$O6^{i}$ —Zn1—N4 ⁱⁱ	118.40 (11)	O8 ⁱⁱⁱ —Zn2—N5	91.15 (11)
O11—Zn2—O8 ⁱⁱⁱ	122.29 (11)	C1—O2—Zn1	120.1 (2)
O11—Zn2—N8 ^{iv}	95.87 (11)	C47—O8—Zn2 ⁱⁱⁱ	115.4 (2)
O11—Zn2—N5	114.58 (11)	C66—O11—Zn2	122.4 (2)
$N5$ — $Zn2$ — $N8^{iv}$	119.90 (12)	C20—O6—Zn1 ⁱ	113.9 (2)
C87—N8—Zn2 ⁱⁱ	135.7 (2)	C46—N4—Z $n1^{iv}$	133.8 (2)
C86—N8—Zn2 ⁱⁱ	117.6 (2)	C40—N4—Z $n1^{iv}$	118.9 (2)
C21—N1—Zn1	137.2 (2)	C67—N5—Zn2	117.6 (2)
C27—N1—Zn1	116.9 (2)	C68—N5—Zn2	130.2 (2)

Symmetry codes: (i) -*x*, -*y*, -*z*; (ii) *x*, *y*-1, *z*-1; (iii) -*x*+1, -*y*+1, -*z*+1; (iv) *x*, *y*+1, *z*+1.

YMUN 7			
Zn2—O5 ⁱ	1.977 (2)	Zn1—O1 ^{iv}	1.961 (2)
Zn2—O5	1.977 (2)	Zn1—O1	1.961 (2)
Zn2—N4 ⁱⁱ	2.026 (2)	$Zn1$ — $N1^{iv}$	2.012 (3)
Zn2—N4 ⁱⁱⁱ	2.026 (2)	Zn1—N1	2.012 (3)
O5 ⁱ —Zn2—O5	115.21 (14)	N4 ⁱⁱ —Zn2—N4 ⁱⁱⁱ	108.02 (14)
$O5^{i}$ —Zn2—N4 ⁱⁱⁱ	118.02 (10)	O1 ^{iv} —Zn1—O1	119.21 (14)
O5 ⁱ —Zn2—N4 ⁱⁱ	99.20 (9)	$O1^{iv}$ —Zn1—N1 ^{iv}	102.48 (10)
O5—Zn2—N4 ⁱⁱ	118.02 (10)	O1—Zn1—N1 ^{iv}	108.57 (10)
O5—Zn2—N4 ⁱⁱⁱ	99.20 (9)	O1 ^{iv} —Zn1—N1	108.57 (10)
O1—Zn1—N1	102.48 (10)	N1 ^{iv} —Zn1—N1	116.23 (15)
N1 ^{iv} —Zn1—N1	116.23 (15)	C21—O5—Zn2	110.49 (19)
C21—O5—Zn2	110.49 (19)	C1—O1—Zn1	112.8 (2)
O1—Zn1—N1	102.48 (10)	C35—N4—Zn2 ^v	129.1 (2)
C24—N1—Zn1	132.5 (2)	C22—N1—Zn1	121.6 (2)

Symmetry codes: (i) -*x*+1, *y*, -*z*+3/2; (ii) -*x*+1, *y*-3, -*z*+3/2; (iii) *x*, *y*-3, *z*; (iv) -*x*+2, *y*, -

z+3/2; (v) *x*, *y*+3, *z*.

YMUN 8			
Zn1—O3	1.952 (2)	Zn1—N1	2.030 (3)
Zn1—O7 ⁱ	1.974 (2)	Zn1—N4 ⁱⁱ	2.048 (3)
O3—Zn1—O7 ⁱ	113.32 (11)	O7 ⁱ —Zn1—N1	95.75 (11)
O3—Zn1—N1	137.25 (12)	O7 ⁱ —Zn1—N4 ⁱⁱ	93.57 (11)

O3—Zn1—N4 ⁱⁱ	106.48 (11)	N1—Zn1—N4 ⁱⁱ	101.99 (12)	
C19—O3—Zn1	104.9 (2)	C1—N1—Zn1	117.6 (3)	
C38—O7—Zn1 ⁱ	120.6 (2)	C2—N1—Zn1	136.4 (2)	
C18—N4—Zn1 ⁱⁱⁱ	122.2 (2)	C17—N4—Zn1 ⁱⁱⁱ	132.6 (3)	
Symmetry codes: (i) - <i>x</i> +2, - <i>y</i> +2, - <i>z</i> +1; (ii) <i>x</i> +1/2, - <i>y</i> +1/2, <i>z</i> +1/2; (iii) <i>x</i> -1/2, - <i>y</i> +1/2, <i>z</i> -1/2.				

Table S2 Hydrogen bond distances (Å) and angles (°) of YMUN 6-8					
D-H···A	d(D-H)	d(H···A)	$d(D \cdots A)$	∠DHA	
		YMUN 6			
C18-H18…O7	0.95	2.54	3.441(4)	159	
С25-Н25…О12	0.95	2.27	3.080(5)	143	
С27-Н27…О7	0.95	2.32	3.149(4)	145	
C42-H42…O10	0.95	2.33	3.141(5)	143	
C45-H45…O5	0.95	2.54	2.932(6)	105	
С70-Н70…О4	0.95	2.37	3.308(5)	171	
С82-Н82…Об	0.95	2.46	3.099(5)	124	
С86-Н86…О1	0.95	2.33	3.134(4)	142	
С86-Н86…О7	0.95	2.49	3.113(5)	123	
YMUN 7					
C25-H25A…O2	0.99	2.47	3.061(4)	117	
С32-Н32В…О6	0.99	2.53	3.009(4)	109	
YMUN 8					
C1-H1…O6	0.95	2.43	3.226(5)	141	
С13-Н13…Об	0.95	2.20	3.153(4)	178	