

## ELECTRONIC SUPPORTING INFORMATION

### Tailoring the physical properties of hybrid magnetic quinuclidine-based plastic compounds via weak interactions

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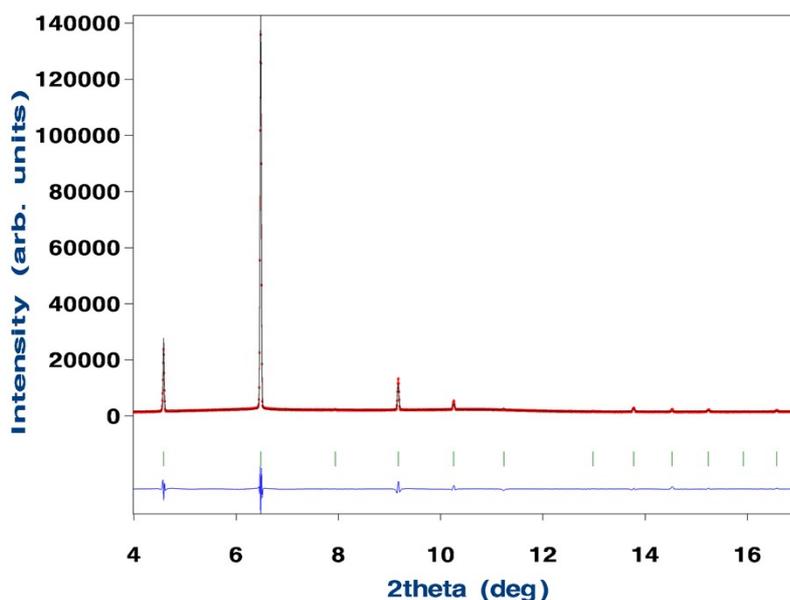
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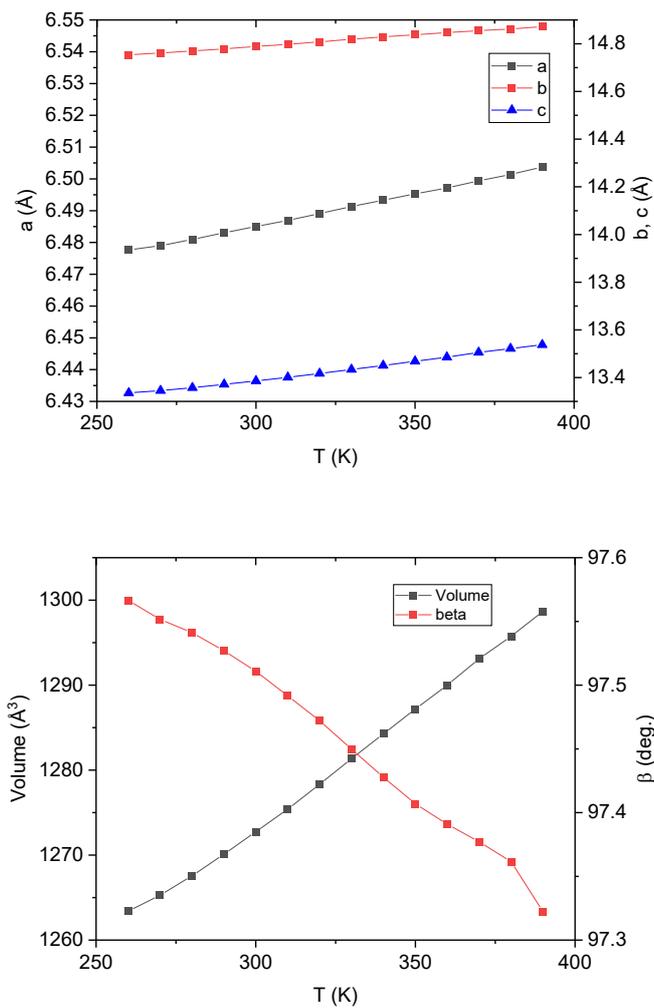
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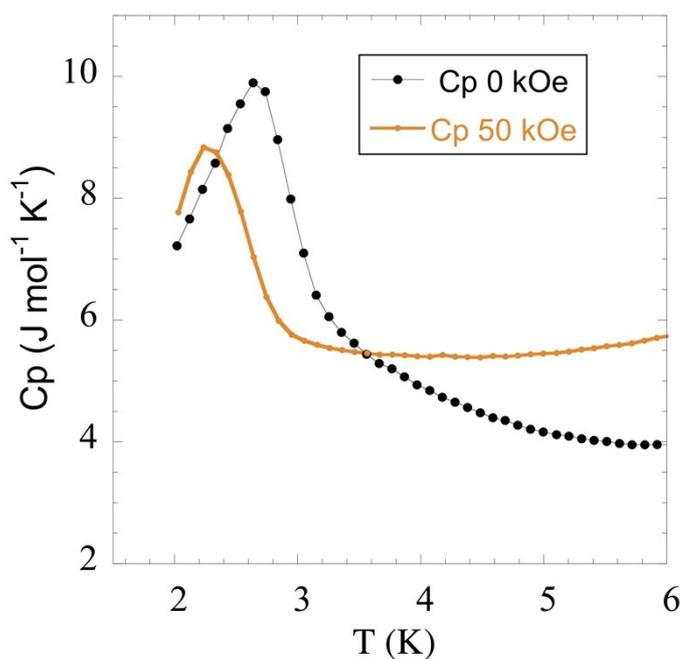
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**Figure S1:** Rietveld refinement to the SXPd data at 420 K of compound **A** using the spherical shell model. Observed (red points) and calculated (black solid line) powder diffraction patterns for positions of the Bragg reflections are represented by green vertical bars. The observed-calculated difference patterns are depicted as a blue line.  $R_{\text{Bragg}} = 4.76$ .



**Figure S2:** Evolution of the volume and cell parameters of compound **A** from 260 to 390 K obtained from pattern matching to the SXPD data.



**Figure S3:** The solid lines are guide for the eye.

**Table S1:** Relevant distances of compound **A** at 300 K obtained from single-crystal X-Ray data.

<b>(3-oxoquinuclidinium)</b>	<b>Length (Å)</b>	<b>FeCl4</b>	<b>Length (Å)</b>
C1-N1	1.472(7)	Fe1-Cl1	2.1736(16)
C1-C4	1.504(7)	Fe1-Cl3	2.189(2)
C2-N1	1.490(7)	Fe1-Cl2	2.207(2)
C2-C5	1.520(8)	Fe2-Cl6	2.1858(15)
C6-N1	1.492(6)	Fe2-Cl4	2.1947(17)
C3-C6	1.503(6)	Fe2-Cl5	2.1972(15)
C4-C7	1.534(8)		
C5-C7	1.510(7)		
C7-C3	1.470(7)		
C3-O1	1.208(5)		

**Table S2:** Relevant distances of compound **B** at 100 K obtained from single-crystal X-Ray data.

<b>(3-oxoquinuclidinium)</b>	<b>Length (Å)</b>	<b>FeCl4</b>	<b>Length (Å)</b>
C2-N1	1.4926(19)	Fe1-Cl3	2.1871(5)
C2-C3	1.512(2)	Fe1-Cl1	2.1990(6)
C8-C7	1.530(3)	Fe1-Cl2	2.2007(7)
C8-C4	1.542(3)	Fe2-Cl4	2.1868(7)
C4-C3	1.498(2)	Fe2-Cl5	2.1977(5)
C4-C5	1.537(3)	Fe2-Cl6	2.2182(6)
C7-N1	1.505(2)		
C6-N1	1.507(2)		
C6-C5	1.530(3)		
C3-O2	1.209(2)		

**Table S3:** Crystallographic data and single-crystal refinement details of compound **A** at 13 K from the refinement of the neutron single-crystal data.

<b>A<sub>II</sub> @ 13 K</b>	
Empirical formula	C <sub>28</sub> H <sub>48</sub> Cl <sub>16</sub> Fe <sub>4</sub> N <sub>4</sub> O <sub>4</sub>
Formula weight	1295.34
Crystal system	monoclinic
Space group	<i>P</i> 1 2 <sub>1</sub> / <i>m</i> 1
<i>a</i> (Å)	6.445
<i>b</i> (Å)	14.509
<i>c</i> (Å)	13.037
$\beta$ (°)	97.96
<i>V</i> (Å <sup>3</sup> )	1207.3
<i>Z</i>	4
$\rho$ (g·cm <sup>-3</sup> )	1.782
wavelength (Å <sup>-1</sup> )	1.45567
Reflections [ <i>I</i> >2 $\sigma$ ( <i>I</i> )]	1909
Reflections [all data]	1911
parameters	244
Goodness of fit ( <i>S</i> ) <sup>a</sup>	1.293
R <sub>1</sub> <sup>b</sup> /wR <sub>2</sub> <sup>c</sup> [ <i>I</i> >2 $\sigma$ ( <i>I</i> )]	0.0422 / 0.1011
R <sub>1</sub> <sup>b</sup> /wR <sub>2</sub> <sup>c</sup> [all data]	0.0422 / 0.1011

**Table S4:** Relevant distances of compound **A** at 13 K obtained from single-crystal neutron diffraction measured at D19.

<b>(3-oxoquinuclidinium)</b>	<b>Length (Å)</b>	<b>FeCl4</b>	<b>Length (Å)</b>
C1-N1	1.5045(23)	Fe1-Cl1	2.1780(15)
C1-C4	1.5375(26)	Fe1-Cl3	2.1971(18)
C2-N1	1.5071(21)	Fe1-Cl2	2.2104(24)

C2-C5	1.5367(25)	Fe2-Cl6	2.1899(18)
C6-N1	1.4961(23)	Fe2-Cl4	2.1984(23)
C3-C6	1.5201(23)	Fe2-Cl5	2.2009(15)
C4-C7	1.5375(27)		
C5-C7	1.5406(24)		
C7-C3	1.5065(24)		
C3-O1	1.2035(25)		
<b>X-H...Cl<sup>a</sup></b>	<b>Length (Å)</b>	<b>Angle (°)</b>	<b>a (symm. op.)</b>
N1-H1...Cl4	2.712(4)	128.3(3)	-1+x, y, z
N1-H1...Cl5	2.733(4)	120.1(3)	-1+x, 0.5-y, z
C2-H2A...Cl5	2.835(4)	131.6(3)	1-x, 0.5+y, -z
C2-H2B...Cl5	2.936(4)	118.3(3)	-1+x, 0.5-y, z
C3-H6A...Cl5	2.793(4)	124.9(3)	-1+x, 0.5-y, z
C4-H4A...Cl1	2.647(4)	152.3(3)	1+x, y, z
C4-H4B...Cl3	2.777(4)	175.4(3)	-x, 0.5+y, 1-z
C5-H5A...Cl5	2.698(4)	145.7(3)	2-x, 0.5+y, -z
C5-H5B...Cl2	2.765(5)	142.9(3)	-x, 0.5+y, 1-z
C7-H7...Cl2	2.818(4)	150.8(3)	1-x, 0.5+y, 1-z
C7-H7...Cl1	2.867(4)	127.7(3)	1-x, 1-y, 1-z
<b>Fe...Fe<sup>b</sup></b>	<b>Length (Å)</b>		<b>b (symm. op.)</b>
Fe1...Fe2	6.340(2)		-1+x, y, z
Fe1...Fe1	6.4450(19)	Intra-plane	-1+x, y, z
Fe1...Fe2	6.879(2)		-1+x, y, 1+z
Fe1...Fe2	7.527(2)		-2+x, y, z
Fe1...Fe1	7.5402(6)	Inter-plane	-x, 0.5+y, 1-z
Fe2...Fe2	7.4048(5)		2-x, -0.5+y, -z