

## Solvent induced conformational polymorphism

Konrad Dyk<sup>1</sup>, Vasyl Kinzhybalo<sup>2</sup>, Grzegorz Czernel<sup>3</sup>, Wojciech Grudziński<sup>4</sup>, Yuriy Horak<sup>5</sup>, Serhii Butenko<sup>5</sup> and Daniel M. Kamiński<sup>1\*</sup>

<sup>1</sup>Institute of Chemical Sciences, Maria Curie-Skłodowska University, pl. Marii Curie-Skłodowskiej 3, 20-031 Lublin, Poland

<sup>2</sup>Institute of Low Temperature and Structure Research, Polish Academy of Sciences, Okólna 2 str., 50-422 Wrocław, Poland

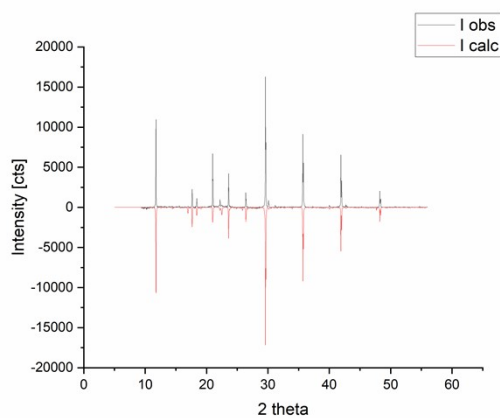
<sup>3</sup>Department of Biophysics, University of Life Sciences in Lublin, Akademicka 13, 20-950 Lublin, Poland

<sup>4</sup>Institute of Physics, Maria Curie-Skłodowska University, pl. Marii Curie-Skłodowskiej 1, 20-031 Lublin, Poland

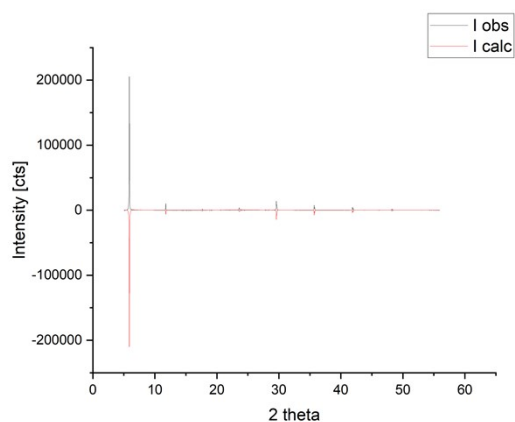
<sup>5</sup>Ivan Franko National University of Lviv, Kyryla I Mefodiya St. 6, 79005 Lviv, Ukraine

## Supplementary

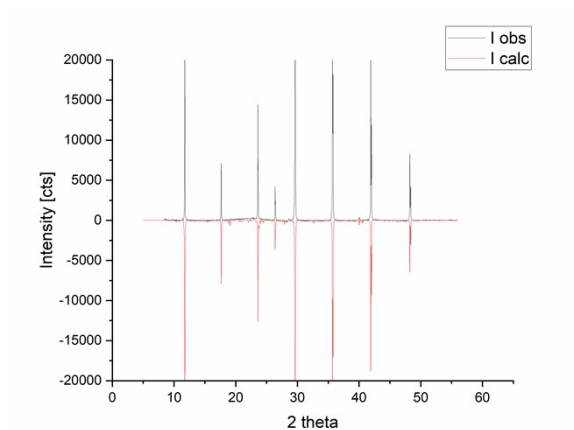
**Synthesis of 2,5-Bis(3-bromophenyl)furan.** A solution of sodium nitrite (6.38 g, 92.5 mmol) in H<sub>2</sub>O (14 ml) was added dropwise to the stirred mixture of 3-bromoaniline 14.4 g (83.7 mmol), concentrated HCl (36%, 0.3 mol, 30 ml), and H<sub>2</sub>O (15 ml) at a temperature below +5°C. The cold solution was filtrated and the diazonium salt was slowly added to a vigorously stirred solution of 2-furoic acid (4.6 g, 41.0 mmol) and CuCl<sub>2</sub>·2H<sub>2</sub>O (1.0 g, 5.9 mmol) in acetone (70 ml) at the room temperature. The rate of addition was controlled by the nitrogen and carbon dioxide evolution (3–4 bubbles/sec, 0.5–1 h). At the end of the reaction, after the evolution of nitrogen and carbon dioxide ceased, water (200 ml) was added, extracted with CH<sub>2</sub>Cl<sub>2</sub>, washed with 10% K<sub>2</sub>CO<sub>3</sub> (2×200 ml), and H<sub>2</sub>O (3×200 ml), and the organic layers were dried with anhyd. Na<sub>2</sub>SO<sub>4</sub> concentrated and purified by column chromatography (eluent hexane). Yield 23% (3.56 g).



A) EtOH Form  $\alpha$ ;  $\beta$  43%



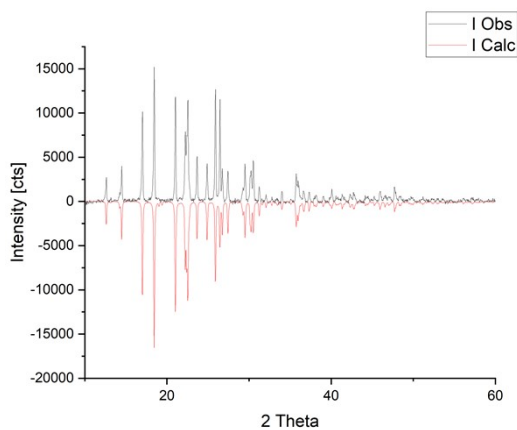
B) ACN form  $\beta$



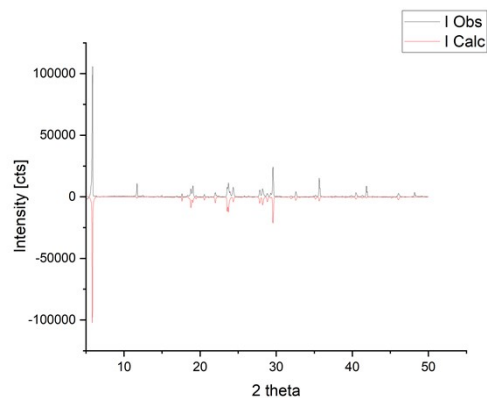
C) EtOH Form  $\alpha$  7%;  $\beta$  93%

S. Figure 1. PXRDs of fast crystallization on glass from A) & C) ethanol; B) acetonitrile. Grey-intensity observed, red-intensity calculated from the structural data.

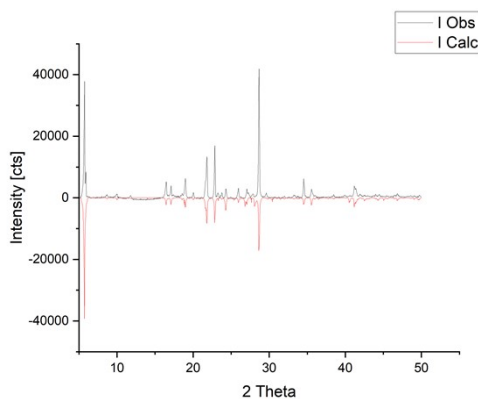
$\alpha$



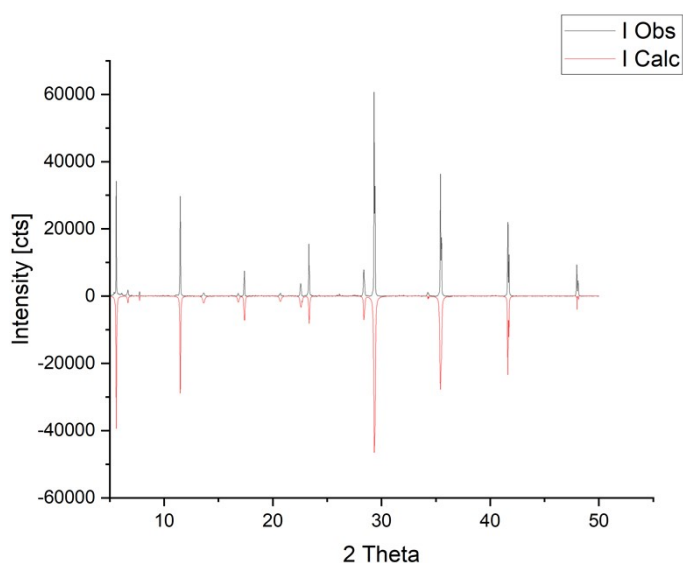
$\beta$



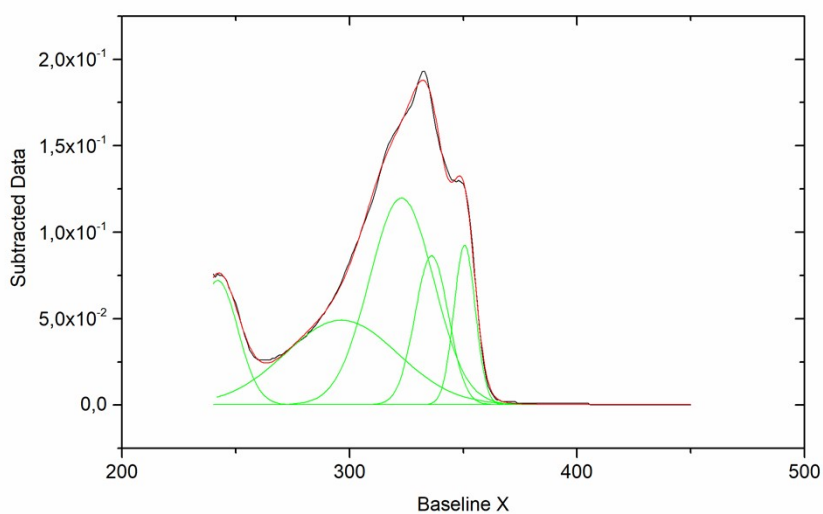
$\gamma$



S. Figure 2. The PXR of the polymorphs structures of the MBPF. Grey-intensity observed, red-intensity calculated from the structural data.



**S. Figure 3.** PXRD of the sample after crystallization from the vapor. The diffractograms indicates presents of forms  $\beta$  and  $\gamma$  with dominant phase  $\gamma$ .



**Fitting Results**

Peak Index	Peak Type	Area Intg	FWHM	Max Height	Center Crvty	Area IntgP
1	Gaussian	0,94668	20,97637	0,07173	242,0516	8,73709
2	Gaussian	2,99134	58,27469	0,04877	296,43641	27,6077
3	Gaussian	4,23335	33,31587	0,11937	322,82795	39,07049
4	Gaussian	1,57976	17,25641	0,086	336,1115	14,57996
5	Gaussian	1,08403	11,05644	0,09211	350,59085	10,00475

**S. Figure 4.** UV-Vis spectra of the **MBPF** from chloroform and its decomposition. The component at 322 and 336 nm correspond to monomers. The component at 350 nm corresponds to dimers.

**S. Table 1.** Crystal17 calculations of the crystal net energies for studied polymorphs using different basis sets and structure optimization.

$E_{total}$ [kJ]	$E_{mol D}$ [kJ]	$E_{mol noD}$ [kJ]	$E_{mol cpc}$ [kJ]	BSSE [kJ]	Form
------------------	------------------	--------------------	--------------------	-----------	------

**DZVP**

-61302025,59	-15325134,36	-15324947,36	-15325136,74	189,3832338	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -372,0361295</math></b>					$\alpha$
<b><math>E_{\text{coh+BSSE}} = -182,6528956</math></b>					
-61302905,8	-15325283,45	-15325097	-15325360,46	263,4596354	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -442,9998668</math></b>					$\beta$
<b><math>E_{\text{coh+BSSE}} = -179,5402314</math></b>					
-61302027,08	-15325132,4	-15324945,71	-15325139,42	193,711696	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -374,367502</math></b>					$\gamma$
<b><math>E_{\text{coh+BSSE}} = -180,6558059</math></b>					

**TZVP**

-61317490,65	-15329049,25	-15328862,24	-15329010,12	147,878497	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -323,4132546</math></b>					$\alpha$
<b><math>E_{\text{coh+BSSE}} = -175,5347576</math></b>					
-61318075,1	-15329194,42	-15329007,97	-15329161,45	153,4765996	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -324,3599133</math></b>					$\beta$
<b><math>E_{\text{coh+BSSE}} = -170,8833137</math></b>					
-61317491,03	-15329047,6	-15328860,91	-15329011,07	150,1674073	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -325,1592254</math></b>					$\gamma$
<b><math>E_{\text{coh+BSSE}} = -174,9918181</math></b>					

**Optimization - TZVP**

-61321672,73	-15329761,33	-15329576,05	-15328988,26	-587,7957436	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -656,8497117</math></b>					$\alpha$
<b><math>E_{\text{coh+BSSE}} = -1244,645455</math></b>					
-61321725,09	-15329761,78	-15329576,72	-1,53E+07	-584,321125	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -669,4925044</math></b>					$\beta$
<b><math>E_{\text{coh+BSSE}} = -1253,813629</math></b>					
-61321729,37	-15329761,77	-15329576,77	-15329011,08	-565,695114	
Z	4				
<b><math>E_{\text{coh Bartolomeo}} = -670,5760767</math></b>					$\gamma$
<b><math>E_{\text{coh+BSSE}} = -1236,27119</math></b>					

