

Supporting Information

A High Performance Asymmetric Supercapacitor Device Based on CoO@CoAl-LDH Hierarchical 3D Nanobouquet Arrays

Yuan Song^{1,2,3}, Qiang Shen^{1,*}, Guo-Xiang Pan^{1,*}, Chao Ye², Yi-Fan Zhang¹, Lin Song

³

¹ Department of Materials Chemistry, Huzhou University, Huzhou 313000, China

² Yangtze River Delta Research Institute, Northwestern Polytechnical University, Taicang 215400, China

³ School of Materials Science and Engineering, Northwestern Polytechnical University, Xi'an 710072, China

*Corresponding authors: Department of Materials Chemistry, Huzhou University, Huzhou 313000, China

E-mail addresses: qiangshen@zjhu.edu.cn (Q. Shen), pgxzjut@163.com (GX. Pan)

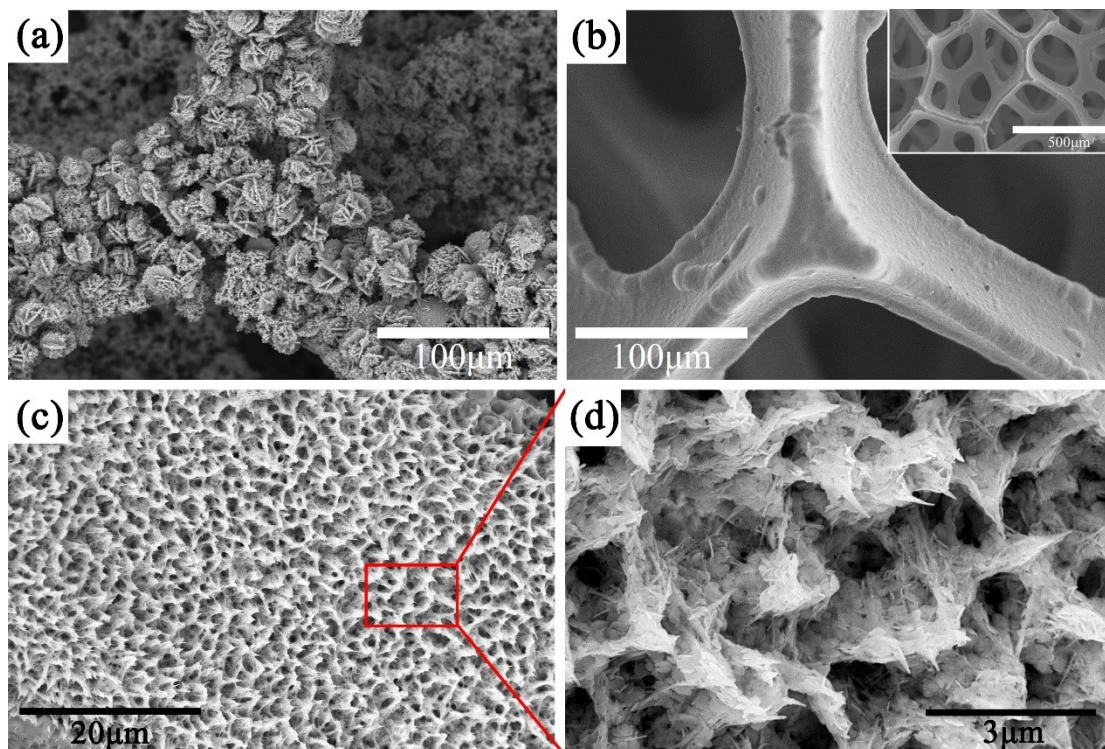


Fig. S1 SEM image of (a)CoO@CoAl-LDH/NF, (b) pure Nickel Foam with an inset of low magnification, CoAl-LDH nanoneedles with low magnification (c) and high magnification (d).

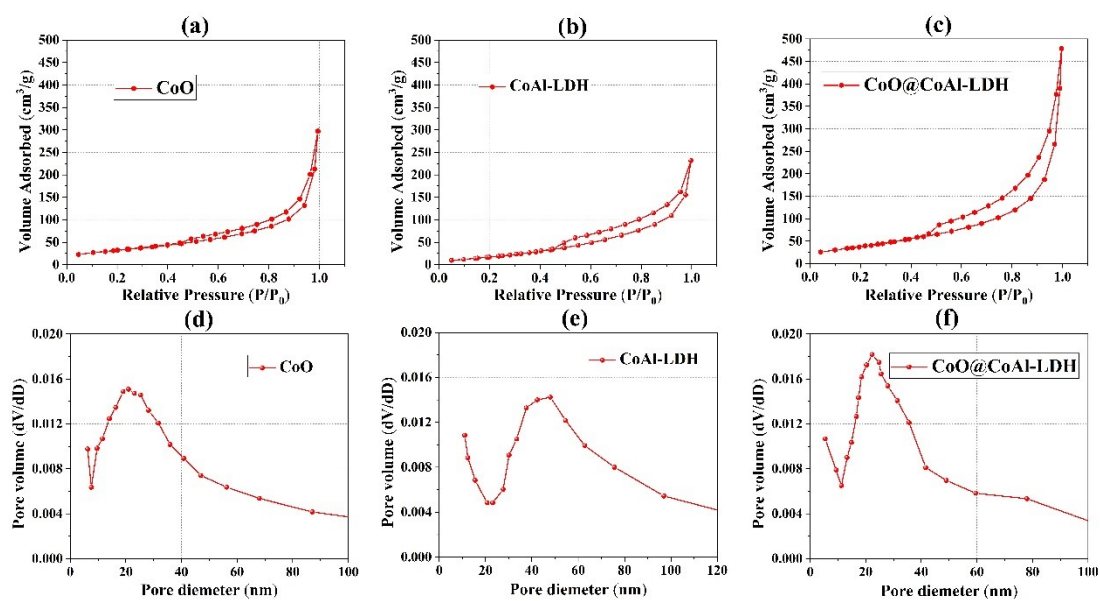
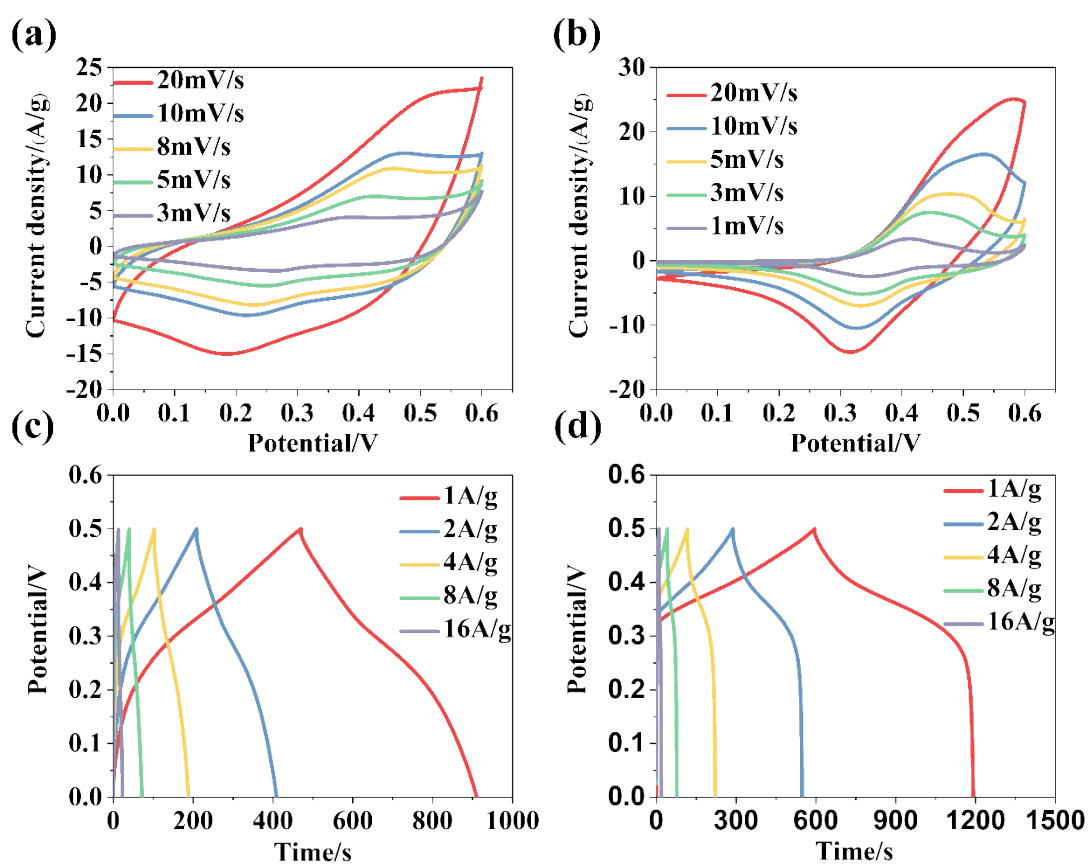


Fig. S2 Nitrogen adsorption-desorption isotherms of (a) CoO nanosheets, (b) CoAl-LDH nanoneedles, (c)CoO@ CoAl-LDH nanobouquets, pore size distribution curves of (d) CoO nanosheets, (e) CoAl-LDH nanoneedles, (f) CoO@ CoAl-LDH nanobouquets

Table. S1 Data analysis of nitrogen absorption and desorption

Materials	S _{BET} (m ² g ⁻¹)	Average Pore Size (nm)
CoO nanosheets	115.6	28.6
CoAl-LDH nanoneedles	87.3	46.2
CoO@ CoAl-LDH nanobouquets	152.8	32.4

**Fig. S3** CV curves of (a) the CoO/NF and (b) the CoAl-LDH/NF at various scan rates; GCD curves of (c) the CoO/NF and (d) the CoAl-LDH/NF at diverse current densities.

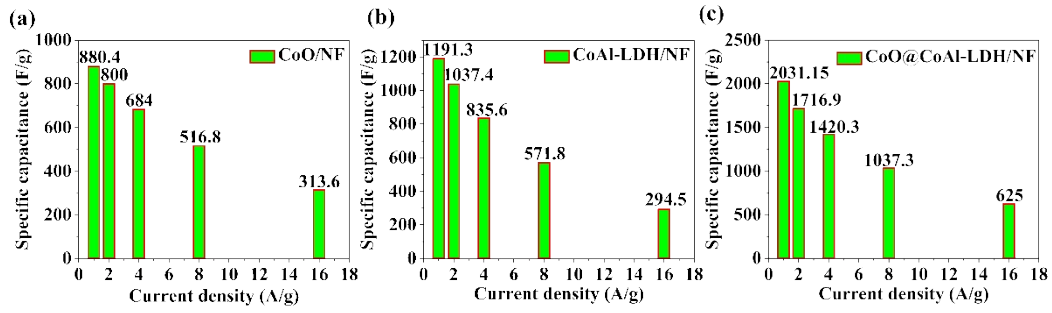


Fig. S4 Rate capability (histogram) at different scan rates of (a) CoO/NF, (b) CoAl-LDH/NF and (c) CoO@CoAl-LDH/NF

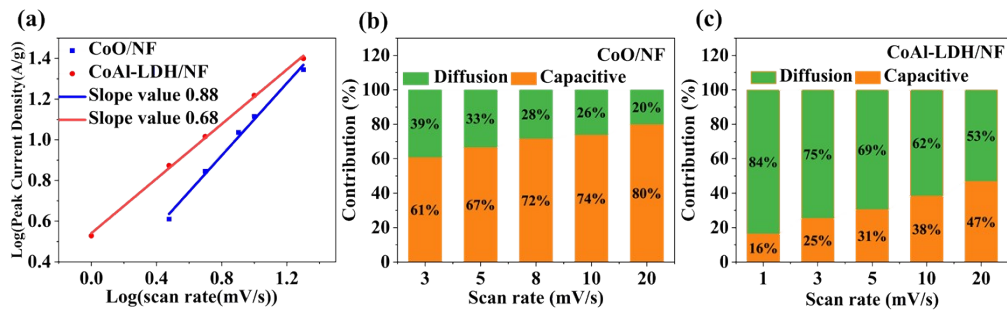


Fig. S5 (a) The value of the slope at different peak currents after linear fitting and pseudo-capacitance contribution rate (histogram) at different scan rates of (b) CoO/NF and (c) CoAl-LDH/NF

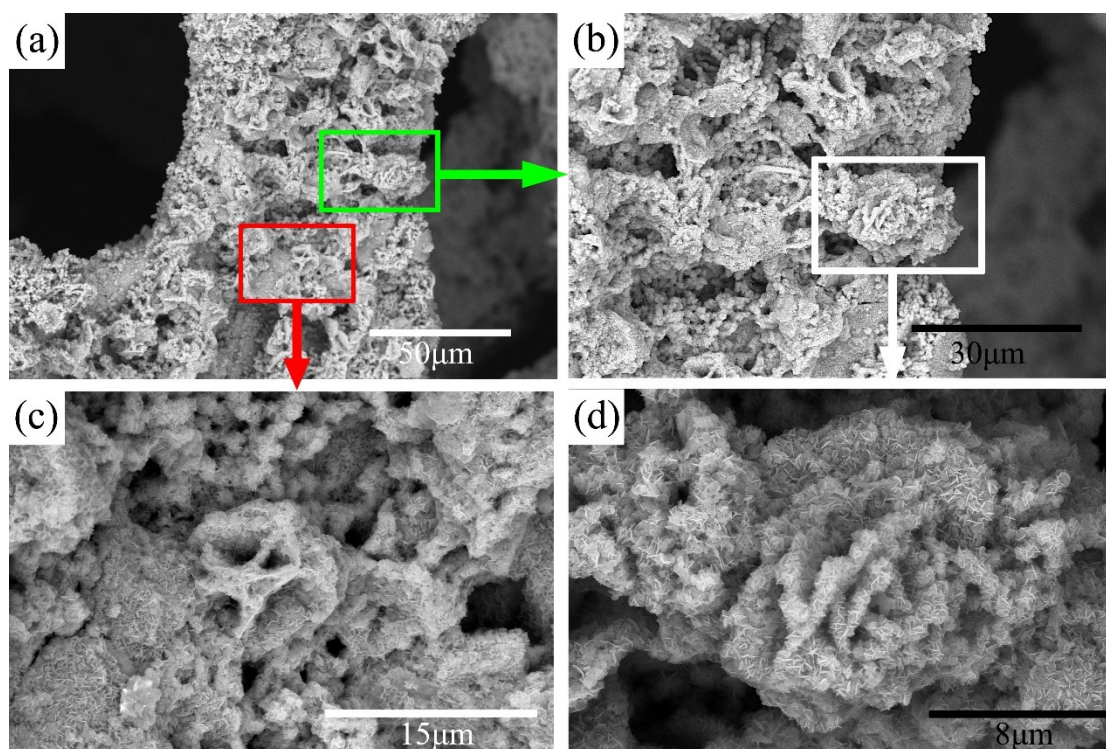


Fig. S6 SEM images of CoO@CoAl-LDH nanobouquets arrays after 5000 cycles

Table. S2 Comparison of electrochemical properties of similar substance

Electrode materials	Specific capacitance	Cycle stability	Ref
CoAl-LDH/rGO	1492F/g (at 1A/g)	94.3% (after 5000 cycles)	[1]
Co ₃ O ₄ @CoAl-LDH	1899.4F/g (at 1A/g)	96.8% (after 5000 cycles)	[2]
Ni ₃ S ₂ /CoAl-LDH/rGO	2457.5F/g (at 1A/g)	90.0% (after 5000 cycles)	[3]
NiCo ₂ O ₄ @NiCoAl-LDH	1814.2F/g (at 1A/g)	93.0% (after 2000 cycles)	[4]
O _v -NiCo-LDH	2577.8F/g (at 1A/g)	73.5% (after 5000 cycles)	[5]
CuCo ₂ O ₄ @MoO ₄	1153F/g (at 1A/g)	76.56% (after 5000 cycles)	[6]
α-phase NiCo-LDH	1120F/g (at 1A/g)	93.8% (after 1000 cycles)	[7]

MnO ₂ @NiCo-LDH	1547F/g (at 1A/g)	82.3% (after 2000 cycles)	[8]
(Ni ₃ Co)Se ₂ /CC	2.85F/cm ² (at 2mA/cm ²)	80.8% (after 2000 cycles)	[9]
ZIF-8-C@NiAl-LDH	1370F/g (at 1A/g)	77.0% (after 1000 cycles)	[10]
CoO@CoAl-LDH	2031.2F/g (at 1A/g)	88.2% (after 5000 cycles)	This work

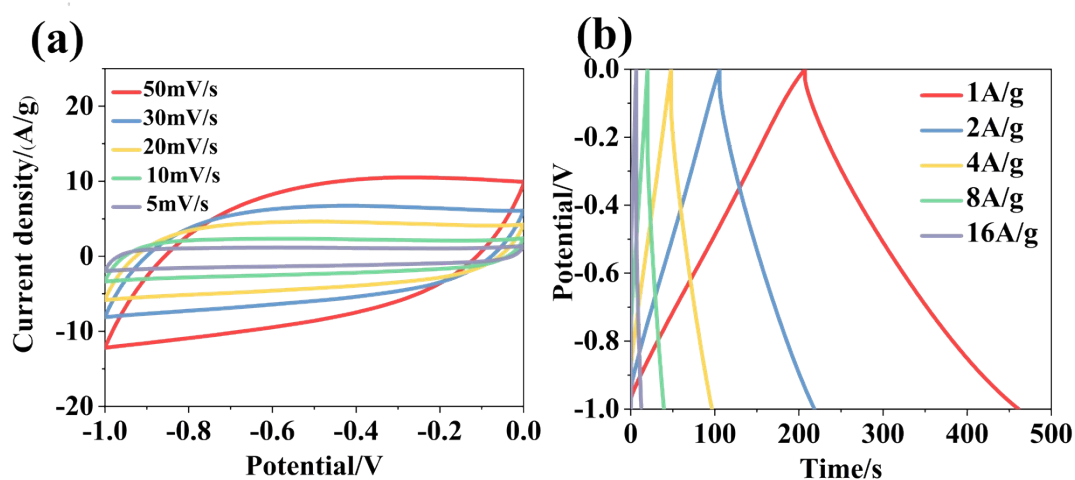


Fig. S7 (a) CV curves of the AC at diverse scan rates; (b) GCD curves of the AC at diverse current densities

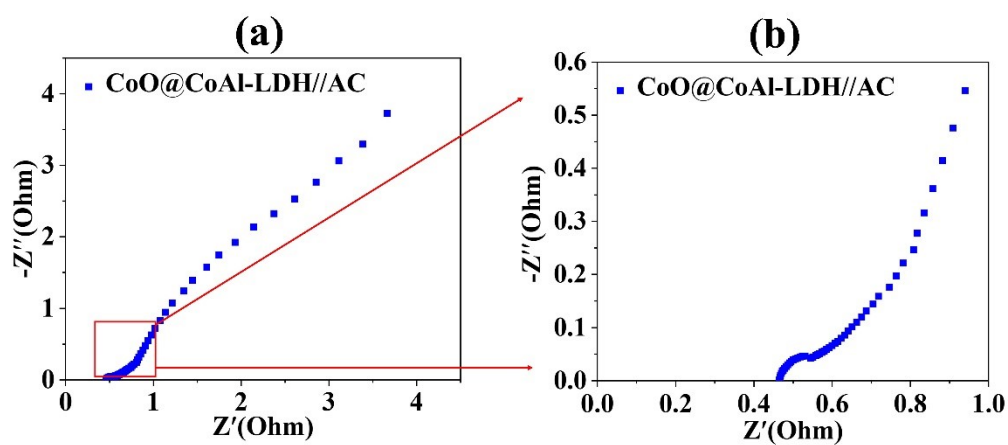


Fig. S8 (a) Nyquist spectra of CoO@CoAl-LDH/NF and (b) enlarged Nyquist spectra of CoO@CoAl-LDH/NF at Z' value range of 0-1.0 ohm.

Reference

1. Li, J., et al., Structure-controlled Co-Al layered double hydroxides/reduced graphene oxide nanomaterials based on solid-phase exfoliation technique for supercapacitors. *Journal of Colloid and Interface Science*, 2019. **549**: p. 236-245.
2. Long, Y.-W., et al., Hierarchical $\text{Co}_3\text{O}_4@\text{CoAl}$ hydrotalcite grown on Ni foam for high-performance asymmetric supercapacitors. *Applied Clay Science*, 2021. **210**.
3. Xuan, H., et al., Hierarchical design of core-shell structured $\text{Ni}_3\text{S}_2/\text{CoAl-LDH}$ composites on rGO/Ni foam with enhanced electrochemical properties for asymmetric supercapacitor. *Journal of Alloys and Compounds*, 2021. **873**.
4. He, X., et al., Hierarchical $\text{NiCo}_2\text{O}_4@\text{NiCoAl}$ -layered double hydroxide core/shell nanoforest arrays as advanced electrodes for high-performance asymmetric supercapacitors. *Journal of Alloys and Compounds*, 2017. **724**: p. 130-138.
5. Wang, T., et al., Oxygen vacancy-rich flower-like nickel cobalt layered double hydroxides for supercapacitors with ultrahigh capacity. *Ceramics International*, 2022.
6. Hao, C., et al., Fabrication of flower-shaped $\text{CuCo}_2\text{O}_4@\text{MgMoO}_4$ nanocomposite for high-performance supercapacitors. *Journal of Energy Storage*, 2021. **41**.
7. Li, J., et al., High-stable α -phase NiCo double hydroxide microspheres via microwave synthesis for supercapacitor electrode materials. *Chemical Engineering Journal*, 2017. **316**: p. 277-287.
8. Wang, X.H., et al., Unique MOF-derived hierarchical MnO_2 nanotubes@NiCo-LDH/ CoS_2 nanocage materials as high performance supercapacitors. *Journal of Materials Chemistry A*, 2019. **7**(19): p. 12018-12028.
9. Song, W., et al., Rational construction of self-supported triangle-like MOF-derived hollow (Ni,Co) Se_2 arrays for electrocatalysis and supercapacitors. *Nanoscale*, 2019. **11**(13): p. 6401-6409.

10. Han, B., et al., Three dimensional hierarchically porous ZIF-8 derived carbon/LDH core-shell composite for high performance supercapacitors. *Electrochimica Acta*, 2018. **263**: p. 391-399.