

Appendix A. Supplementary material

Metal Dimers Embedded Vertically in Defect-graphene as Gas Sensors: A First-Principles Study

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Table S1 The binding energy (E_b) per M atom of $M_2 \perp$ gra species, the cohesive energy (E_{coh}) of M metals (in eV).

| structure | E_b | E_{coh} | structure | E_b | E_{coh} | structure | E_b | E_{coh} |
|-----------|--------------|-----------|-----------|--------------|-----------|-----------|--------------|-----------|
| 2Sc-gra | -5.90 | -4.34 | 2Y-gra | -5.92 | -4.40 | | | |
| 2Ti-gra | -5.93 | -5.51 | 2Zr-gra | -6.64 | -6.45 | 2Hf-gra | -7.07 | -6.71 |
| 2V-gra | -4.96 | -5.34 | 2Nb-gra | -6.05 | -7.04 | 2Ta-gra | -6.81 | -8.37 |
| 2Cr-gra | -3.29 | -4.07 | 2Mo-gra | -5.00 | -6.33 | 2W-gra | -5.35 | -8.41 |
| 2Mn-gra | -4.28 | -3.79 | 2Tc-gra | -5.44 | -6.88 | 2Re-gra | -4.99 | -7.83 |
| 2Fe-gra | -4.90 | -4.61 | 2Ru-gra | -6.44 | -7.08 | 2Os-gra | -6.42 | -8.51 |
| 2Co-gra | -5.84 | -5.04 | 2Rh-gra | -6.52 | -5.93 | 2Ir-gra | -7.23 | -7.59 |
| 2Ni-gra | -6.17 | -5.04 | 2Pd-gra | -4.80 | -3.75 | 2Pt-gra | -6.31 | -5.52 |
| 2Cu-gra | -3.37 | -3.50 | 2Ag-gra | -2.03 | -2.49 | 2Au-gra | -2.38 | -2.99 |
| 2Zn-gra | -0.98 | -1.11 | 2Cd-gra | -0.57 | -0.75 | | | |

Table S2 The Bader charge (Q) on the M atoms, the band energy gap (E_g), the magnetic order (MO) and the magnetic moment on M atoms (M) of the five $M_2 \perp$ gra (M = Co, Ni, Rh, Ir and Pt) and $M_1@$ gra (M = Co, Ni, Rh and Pt) structures.

| system | Q ($ e $) | E_g (eV) | MO | M (μ_B) |
|-----------------------------|---------------|------------|------|-----------------|
| Co ₂ \perp gra | +0.67/+0.67 | 0.00 | FM | 0.66/0.66 |
| Ni ₂ \perp gra | +0.68/+0.68 | 0.02 | NM | 0.00/0.00 |
| Rh ₂ \perp gra | +0.35/+0.35 | 0.00 | FM | 0.24/0.24 |
| Ir ₂ \perp gra | +0.39/+0.39 | 0.00 | FM | 0.24/0.24 |
| Pt ₂ \perp gra | +0.30/+0.30 | 0.22 | NM | 0.00/0.00 |
| Co ₁ @gra | +0.75 | 0.00 | | 0.29 |
| Ni ₁ @gra | +0.74 | 0.25 | NM | 0.00 |
| Rh ₁ @gra | +0.43 | 0.00 | | 0.17 |
| Ir ₁ @gra | +0.82 | 0.00 | | 0.26 |
| Pt ₁ @gra | +0.50 | 0.18 | NM | 0.00 |

Table S3 The magnetic anisotropy energy (MAE, in meV/M atom) for the $M_2 \perp$ gra and M_2 -gra with two M atoms on two sides and one side of graphene, respectively.

| system | E(100)–E(001) | E(011)–E(001) | E(010)–E(001) | E(101)–E(001) | E(110)–E(001) | E(111)–E(001) |
|-----------------------------|---------------|---------------|---------------|---------------|---------------|---------------|
| Co ₂ \perp gra | 2.81 | 1.28 | 2.79 | 1.28 | 2.80 | 1.75 |
| Co ₂ -gra | 0.58 | 0.34 | 0.60 | 0.14 | 0.59 | 0.32 |
| Rh ₂ -gra | 5.39 | 2.72 | 5.38 | 2.52 | 5.37 | 3.50 |
| Ir ₂ -gra | 16.71 | 13.78 | 21.95 | 13.15 | 21.47 | 19.84 |

| system | E(100)–E(010) | E(011)–E(010) | E(001)–E(010) | E(101)–E(010) | E(110)–E(010) | E(111)–E(010) |
|-----------------------------|---------------|---------------|---------------|---------------|---------------|---------------|
| Rh ₂ \perp gra | 0.01 | 0.11 | 0.20 | 0.11 | 0.01 | 0.08 |
| Ir ₂ \perp gra | 0.01 | 4.86 | 13.63 | 4.86 | 0.01 | 2.89 |

Table S4 The magnetic moment on M atoms of the three $M_2 \perp$ gra (M = Co, Rh and Ir) and three M_2 -gra (M = Co, Rh and Ir) structures.

| system | Co | Rh | Ir |
|-----------------|-----------|-----------|-----------|
| $M_2 \perp$ gra | 0.66/0.66 | 0.24/0.24 | 0.24/0.24 |
| M_2 -gra | 0.03/1.91 | 0.04/1.33 | 0.08/1.50 |

Table S5 The binding energy (E_b) per M atom of $M_2 \perp$ gra, M_2 -gra and $M_1@$ gra species.

| E_b | Co | Ni | Rh | Ir | Pt |
|-----------------|-------|-------|-------|--------|-------|
| $M_2 \perp$ gra | -5.84 | -6.17 | -6.52 | -7.23 | -6.31 |
| M_2 -gra | -5.13 | | -5.81 | -7.12 | |
| $M_1@$ gra | -8.25 | -7.52 | -9.05 | -10.28 | -7.97 |

Table S6 The adsorption energy (E_{ad}), the charge transferred from the monolayer to molecule (ΔQ), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g), the recovery time (τ), the magnetic order (MO) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the $\text{Co}_2 \perp \text{gra}$.

| system | E_{ad} (eV) | ΔQ ($ e $) | D (\AA) | E_g (eV) | τ (s) 300 K | τ (s) 500 K | MO | M (μ_B) |
|----------------------|---------------|----------------------|----------------------|------------|-----------------------|-----------------------|------|-----------------|
| O_2 | -2.04 | -0.69 | 1.86 | 0.00 | 3.72×10^{21} | 1.41×10^9 | FM | 0.73/0.39 |
| N_2 | -1.49 | -0.27 | 1.80 | 0.30 | 1.78×10^{12} | 1.50×10^3 | NM | |
| CO | -2.23 | -0.27 | 1.79 | 0.30 | 6.16×10^{24} | 1.63×10^{11} | NM | |
| CO_2 | -1.02 | -0.51 | 1.90 | 0.19 | 1.93×10^4 | 1.19×10^{-2} | NM | |
| NO | -3.05 | -0.36 | 1.66 | 0.00 | 4.85×10^{38} | 1.30×10^{20} | FM | 0.02/0.27 |
| NO_2 | -2.87 | -0.59 | 1.86 | 0.00 | 4.32×10^{35} | 1.45×10^{18} | FM | 0.05/0.06 |
| NH_3 | -1.66 | +0.11 | 2.00 | 0.34 | 1.35×10^{15} | 1.05×10^5 | NM | |
| H_2O | -1.07 | +0.05 | 1.98 | 0.31 | 1.35×10^5 | 4.12×10^{-2} | NM | |
| H_2S | -1.41 | -0.00 | 2.17 | 0.29 | 7.83×10^{10} | 2.04×10^2 | NM | |
| SO_2 | -1.57 | -2.34 | 2.06 | 0.26 | 4.03×10^{13} | 1.11×10^4 | NM | |

Table S7 The adsorption energy (E_{ad}), the charge transferred from the monolayer to molecule (ΔQ), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g), the recovery time (τ), the magnetic order (MO) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the $\text{Ni}_2 \perp \text{gra}$.

| system | E_{ad} (eV) | ΔQ ($ e $) | D (\AA) | E_g (eV) | τ (s) 300 K | τ (s) 500 K | MO | M (μ_B) |
|----------------------|---------------|----------------------|----------------------|------------|------------------------|------------------------|------|-----------------|
| O_2 | -1.30 | -0.62 | 1.93 | 0.00 | 1.07×10^9 | 13.0 | FM | +0.14/+0.06 |
| N_2 | -0.73 | -0.27 | 1.85 | 0.09 | 2.35×10^{-1} | 8.43×10^{-6} | NM | |
| CO | -1.46 | -0.27 | 1.81 | 0.10 | 5.51×10^{11} | 7.11×10^2 | NM | |
| CO_2 | -0.22 | -0.51 | 2.02 | 0.01 | 5.35×10^{-10} | 2.45×10^{-11} | NM | |
| NO | -2.08 | -0.40 | 1.74 | 0.08 | 1.77×10^{22} | 3.83×10^9 | AFM | -0.05/+0.13 |
| NO_2 | -2.09 | -0.58 | 1.94 | 0.00 | 2.61×10^{22} | 4.92×10^9 | FM | +0.21/+0.14 |
| NH_3 | -1.20 | +0.11 | 2.05 | 0.02 | 2.16×10^7 | 1.07 | NM | |
| H_2O | -0.68 | +0.05 | 2.04 | 0.00 | 3.31×10^{-2} | 2.40×10^{-6} | NM | |
| H_2S | -0.94 | +0.00 | 2.25 | 0.03 | 8.50×10^2 | 1.61×10^{-3} | NM | |
| SO_2 | -1.03 | -2.34 | 2.20 | 0.06 | 2.85×10^4 | 1.52×10^{-2} | NM | |

Table S8 The adsorption energy (E_{ad}), the charge transferred from the monolayer to molecule (ΔQ), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g), the recovery time (τ), the magnetic order (MO) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the Rh₂ \perp gra.

| system | E_{ad} (eV) | ΔQ ($ e $) | D (\AA) | E_g (eV) | τ (s) 300 K | MO | M (μ_B) |
|------------------|---------------|----------------------|----------------------|------------|-----------------------|------|-----------------|
| O ₂ | -1.48 | -0.47 | 1.95 | 0.00 | 1.20×10^{12} | FM | +0.01/+0.01 |
| N ₂ | -0.92 | -0.19 | 2.00 | 0.23 | 3.89×10^2 | NM | |
| CO | -1.78 | -0.19 | 1.79 | 0.24 | 1.46×10^{17} | NM | |
| CO ₂ | -0.38 | -0.39 | 2.02 | 0.00 | 2.75×10^{-7} | NM | |
| NO | -2.65 | -0.29 | 1.74 | 0.00 | 8.08×10^{31} | NM | |
| NO ₂ | -2.57 | -0.50 | 1.94 | 0.00 | 3.56×10^{30} | NM | |
| NH ₃ | -1.30 | +0.16 | 2.05 | 0.24 | 1.07×10^9 | NM | |
| H ₂ O | -0.70 | +0.06 | 2.04 | 0.23 | 7.23×10^{-2} | NM | |
| H ₂ S | -1.13 | +0.10 | 2.25 | 0.23 | 1.41×10^6 | NM | |
| SO ₂ | -1.23 | -2.29 | 2.20 | 0.18 | 6.98×10^7 | NM | |

Table S9 The adsorption energy (E_{ad}), the charge transferred from the monolayer to molecule (ΔQ), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g), the recovery time (τ), the magnetic order (MO) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the Ir₂ \perp gra.

| system | E_{ad} (eV) | ΔQ ($ e $) | D (\AA) | E_g (eV) | τ (s) 300 K | MO | M (μ_B) |
|------------------|---------------|----------------------|----------------------|------------|-----------------------|------|-----------------|
| O ₂ | -1.88 | -0.82 | 1.91 | 0.00 | 7.22×10^{18} | FM | +0.15/+0.37 |
| N ₂ | -1.33 | -0.36 | 1.91 | 0.00 | 3.45×10^9 | FM | +0.07/+0.32 |
| CO | -2.78 | -0.39 | 1.84 | 0.16 | 1.29×10^{34} | NM | |
| CO ₂ | -0.63 | -0.52 | 2.07 | 0.00 | 4.74×10^{-3} | NM | |
| NO | -3.19 | -0.47 | 1.81 | 0.00 | 1.14×10^{41} | NM | |
| NO ₂ | -2.87 | -0.60 | 2.00 | 0.00 | 4.32×10^{35} | NM | |
| NH ₃ | -1.47 | +0.15 | 2.17 | 0.00 | 8.14×10^{11} | FM | +0.22/+0.52 |
| H ₂ O | -0.76 | +0.07 | 2.20 | 0.00 | 7.51×10^{-1} | FM | +0.28/+0.54 |
| H ₂ S | -1.43 | +0.06 | 2.28 | 0.00 | 1.71×10^{11} | FM | +0.15/+0.48 |
| SO ₂ | -1.50 | -2.30 | 2.20 | 0.00 | 2.63×10^{12} | FM | +0.16/+0.43 |

Table S10 The adsorption energy (E_{ad}), the charge transferred from the monolayer to molecule (ΔQ), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g), the recovery time (τ), the magnetic order (MO) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the $\text{Pt}_2 \perp \text{gra}$.

| system | E_{ad} (eV) | ΔQ ($ e $) | D (\AA) | E_g (eV) | τ (s) 300 K | MO | M (μ_B) |
|------------------|---------------|----------------------|----------------------|------------|------------------------|------|-----------------|
| O ₂ | -0.91 | -0.54 | 2.17 | 0.00 | 2.64×10^2 | FM | +0.04/+0.03 |
| N ₂ | -0.53 | -0.17 | 2.02 | 0.07 | 9.58×10^{-5} | NM | |
| CO | -1.62 | -0.20 | 1.91 | 0.04 | 2.84×10^{14} | NM | |
| CO ₂ | -0.06 | -0.25 | 2.20 | 0.00 | 1.04×10^{-12} | NM | |
| NO | -1.91 | -0.37 | 1.85 | 0.06 | 2.33×10^{19} | AFM | -0.03/+0.09 |
| NO ₂ | -1.88 | -0.58 | 2.14 | 0.00 | 7.22×10^{18} | FM | +0.02/+0.03 |
| NH ₃ | -1.15 | +0.17 | 2.24 | 0.21 | 3.08×10^6 | NM | |
| H ₂ O | -0.54 | +0.07 | 2.40 | 0.22 | 1.40×10^{-4} | NM | |
| H ₂ S | -0.60 | +0.05 | 2.33 | 0.16 | 1.47×10^{-3} | NM | |
| SO ₂ | -0.97 | -2.26 | 2.29 | 0.01 | 2.74×10^3 | NM | |

Table S11 The adsorption energy (E_{ad}), the shortest distance between the molecule and monolayer (D), the band-gap widths (E_g) and the magnetic moment on M atoms (M) for the structures of the O₂/CO₂ (gas) and O₂+H₂O/CO₂+H₂O (mol+H₂O) adsorbed on the Ni₂ \perp gra/Pt₂ \perp gra.

| system | E_{ad} (eV) | | D (\AA) | | E_g (eV) | | M (μ_B) | |
|-----------------------|---------------|----------------------|----------------------|----------------------|------------|----------------------|-----------------|----------------------|
| | gas | gas+H ₂ O | gas | gas+H ₂ O | gas | gas+H ₂ O | gas | gas+H ₂ O |
| Ni (O ₂) | -1.30 | -1.30 | 1.93 | 1.97 | 0.00 | 0.00 | +0.14/+0.06 | +0.10/+0.02 |
| Pt (CO ₂) | -0.06 | -0.04 | 2.20 | 2.43 | 0.00 | 0.16 | 0.00/0.00 | 0.00/0.00 |

Table S12 The adsorption energy (E_{ad}) and the recovery time (τ) for the structures of the O₂ adsorbed on the Ni₂ \perp gra and the CO₂ adsorbed on the Pt₂ \perp gra.

| system | E_{ad} (eV) | | D (\AA) | | τ (s) 300K | |
|-----------------------|---------------|-------|----------------------|-------|------------------------|------------------------|
| | DFT | DFT+U | DFT | DFT+U | DFT | DFT+U |
| Ni (O ₂) | -1.30 | -1.16 | 1.93 | 1.93 | 1.07×10^9 | 4.54×10^9 |
| Pt (CO ₂) | -0.06 | -0.04 | 2.20 | 2.24 | 1.04×10^{-12} | 4.76×10^{-13} |

Table S13 The adsorption energy (E_{ad}), the shortest distance between the molecule and monolayer (D), the charge on the metal atom M (Q) and the magnetic moment on M atoms (M) for the structures of the molecules adsorbed on the $Rh_2 \perp gra$ and $Rh_1 @ gra$.

| system | E_{ad} (eV) | | D (Å) | | Q ($ e $) | | M (μ_B) | |
|------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| | Rh ₂ | Rh ₁ | Rh ₂ | Rh ₁ | Rh ₂ | Rh ₁ | Rh ₂ | Rh ₁ |
| O ₂ | -1.48 | -1.30 | 1.95 | 2.13 | +0.63/+0.37 | +0.80 | 0.01/0.01 | 0.05 |
| N ₂ | -0.92 | -0.37 | 2.00 | 2.20 | +0.49/+0.42 | +0.57 | 0 | 0.09 |
| CO | -1.78 | -0.99 | 1.79 | 2.04 | +0.43/+0.40 | +0.57 | 0 | 0.05 |
| CO ₂ | -0.38 | -0.13 | 2.02 | 2.20 | +0.47/+0.35 | +0.56 | 0 | 0 |
| NO | -2.65 | -1.77 | 1.74 | 1.98 | +0.58/+0.37 | +0.67 | 0 | 0 |
| NO ₂ | -2.57 | -2.30 | 1.94 | 2.13 | +0.61/+0.38 | +0.69 | 0 | 0 |
| NH ₃ | -1.30 | -0.80 | 2.05 | 2.32 | +0.42/+0.35 | +0.52 | 0 | 0.18 |
| H ₂ O | -0.70 | -0.42 | 2.04 | 2.37 | +0.46/+0.35 | +0.56 | 0 | 0.20 |
| H ₂ S | -1.13 | -0.59 | 2.25 | 2.46 | +0.34/+0.36 | +0.47 | 0 | 0.16 |
| SO ₂ | -1.23 | -0.86 | 2.20 | 2.55 | +0.37/+0.39 | +0.54 | 0 | 0.02 |

Table S14 The d-band centers (C_{d-band} , in eV) of $M_2 \perp gra$ and $M_1 @ gra$

| system | C_{d-band} | system | C_{d-band} |
|-----------------------------|--------------|-----------------------|--------------|
| Co ₂ \perp gra | -0.93 | Co ₁ @ gra | -1.78 |
| Ni ₂ \perp gra | -2.09 | Ni ₁ @ gra | -3.25 |
| Rh ₂ \perp gra | -1.33 | Rh ₁ @ gra | -2.48 |
| Ir ₂ \perp gra | -1.27 | Ir ₁ @ gra | -2.94 |
| Pt ₂ \perp gra | -3.20 | Pt ₁ @ gra | -4.20 |

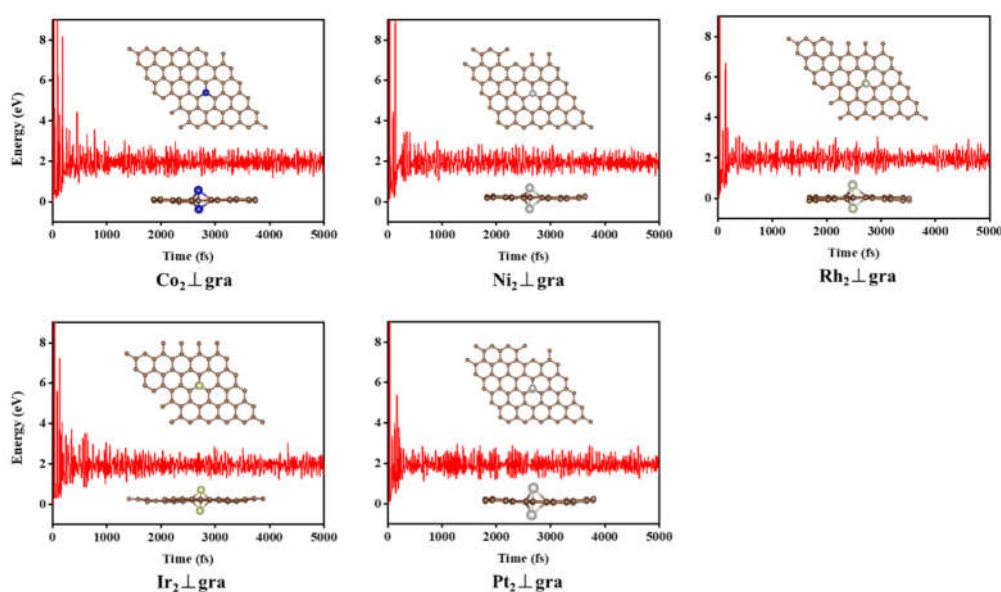


Fig. S1 The energy evolution of five $M_2 \perp gra$ ($M = Co, Ni, Rh, Ir$ and Pt) monolayers during 5 ps FPMD simulation at 300 K, the insets are the final annealed structures.

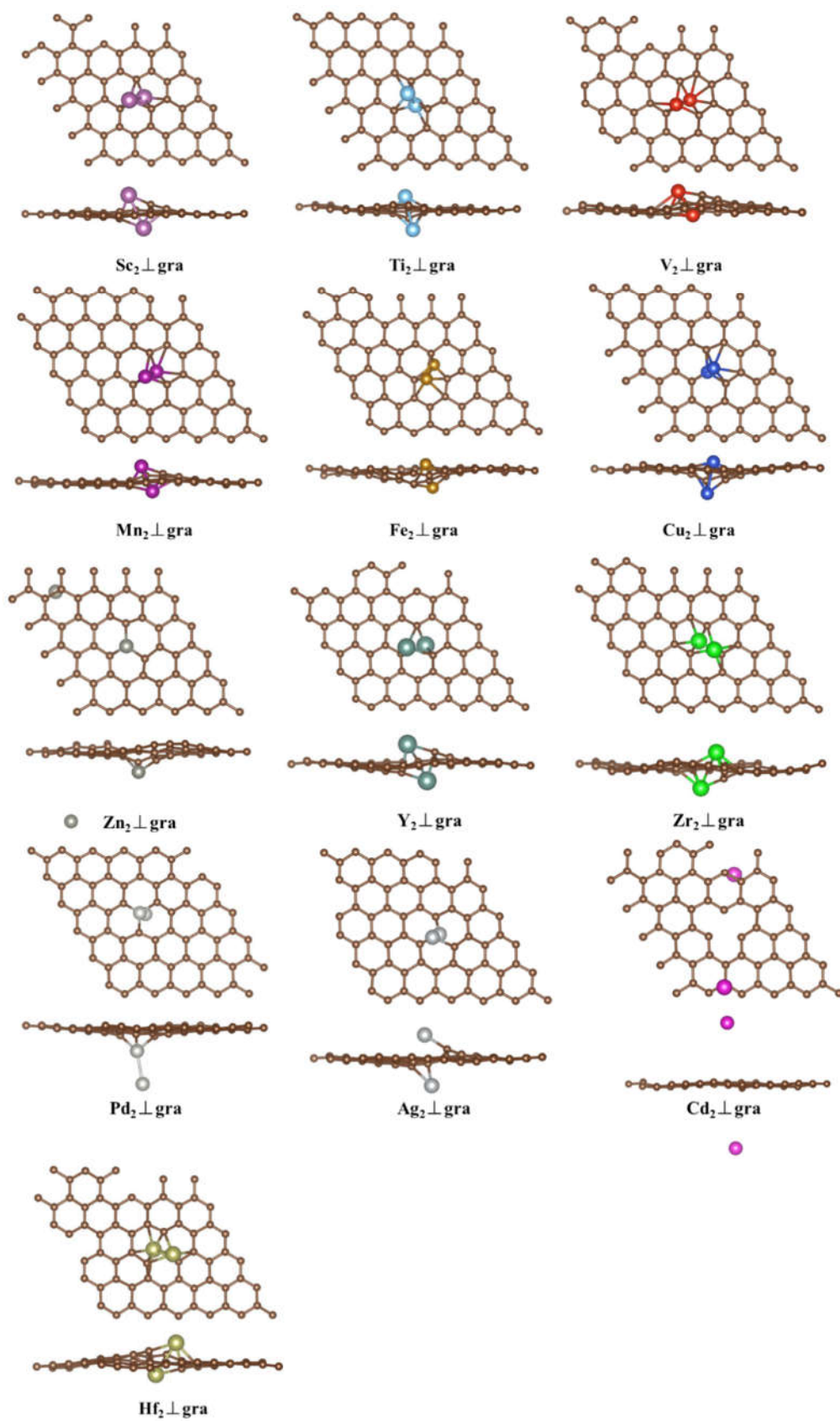


Fig. S2 The final structures of thirteen $\text{M}_2 \perp \text{gra}$ ($\text{M} = \text{Sc}, \text{Ti}, \text{V}, \text{Mn}, \text{Fe}, \text{Cu}, \text{Zn}, \text{Y}, \text{Zr}, \text{Pd}, \text{Ag}, \text{Cd}$ and Hf) monolayers annealed for 5 ps at 300 K.

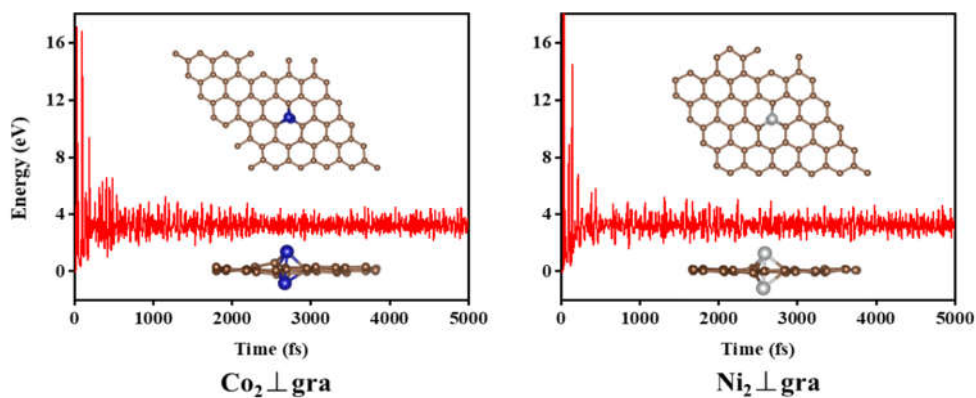


Fig. S3 The energy evolution of $\text{Co}_2 \perp \text{gra}$ (left) and $\text{Ni}_2 \perp \text{gra}$ (right) monolayers during 5 ps FPMD simulation at 500 K, the insets are the final annealed structures.

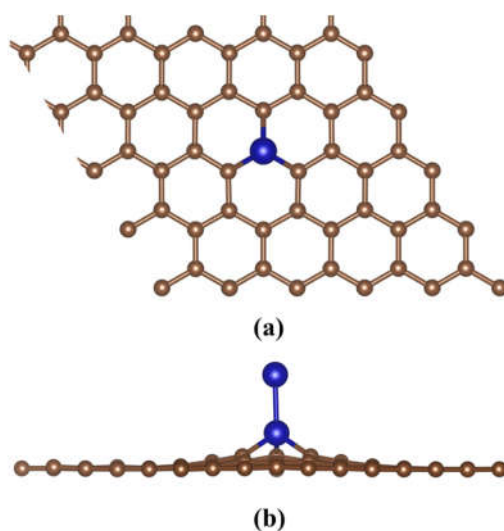


Fig. S4 Structural diagram of the $\text{M}_2\text{-gra}$ model, top view (a) and side view (b). Color scheme: C, brown; M, blue.

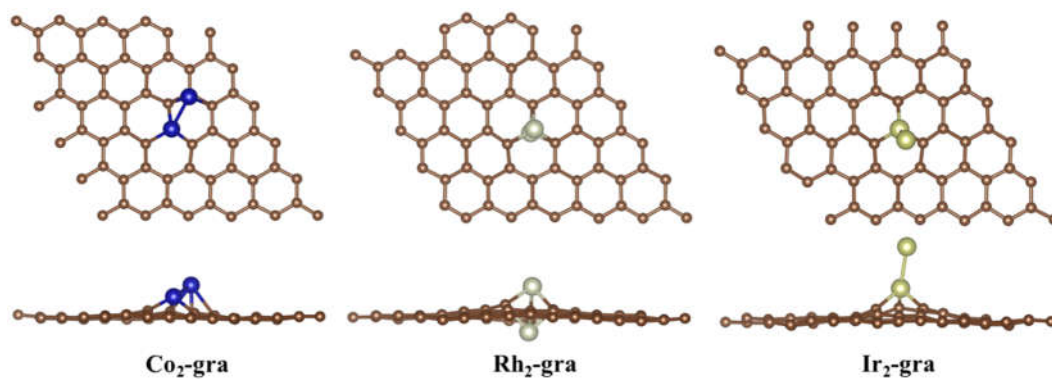


Fig. S5 The final structures of $\text{Co}_2\text{-gra}$, $\text{Rh}_2\text{-gra}$ and $\text{Ir}_2\text{-gra}$ monolayers annealed for 5 ps at 300 K.

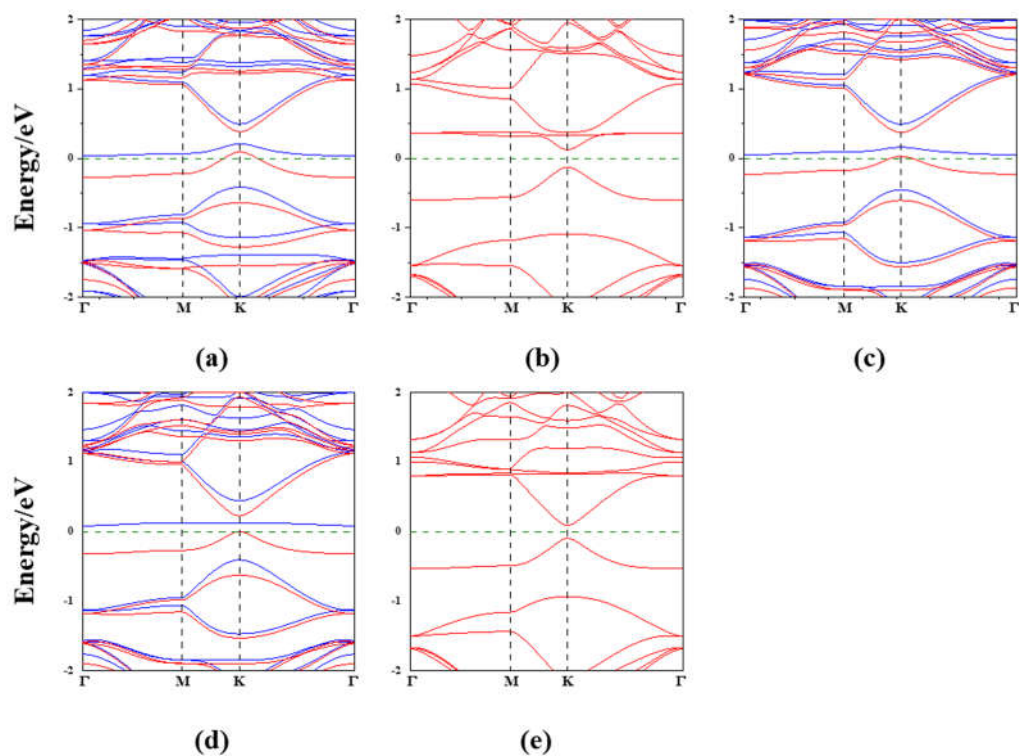


Fig. S6 The band structures of the five $M_1@gra$ structures: $Co_1@gra$ (a), $Ni_1@gra$ (b), $Rh_1@gra$ (c), $Ir_1@gra$ (d) and $Pt_1@gra$ (e).

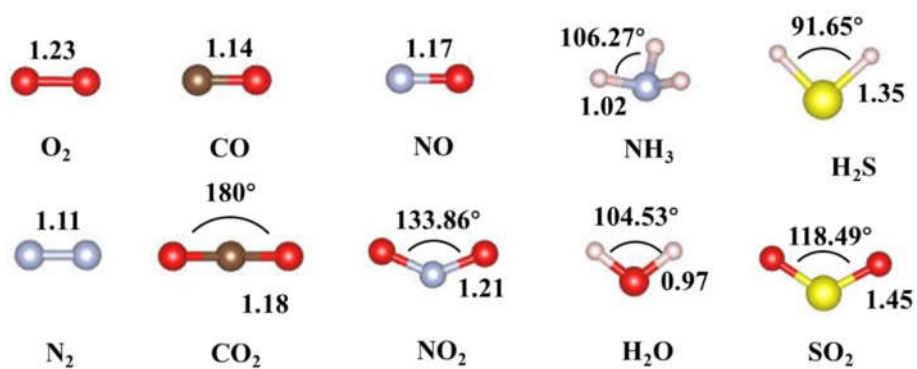


Fig. S7 The optimized structures of free gas molecules.

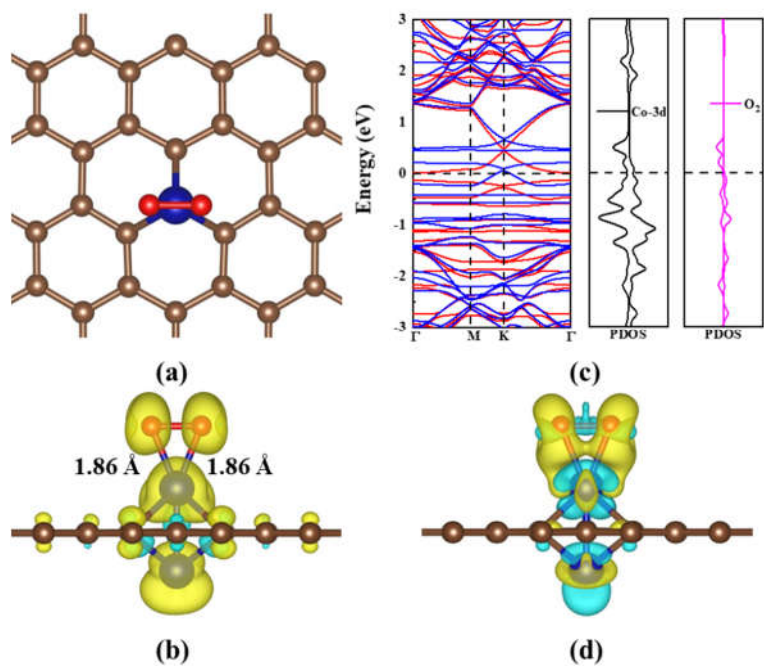


Fig. S8 The most favorable configuration of O₂ adsorption energy on the Co₂⊥ gra structure (top view (a)), spatial spin density distribution (b), energy band structure and PDOS (c), and CDD diagram (d) of the adsorbed system. The isosurface value was set 0.003 e/Å⁻³.

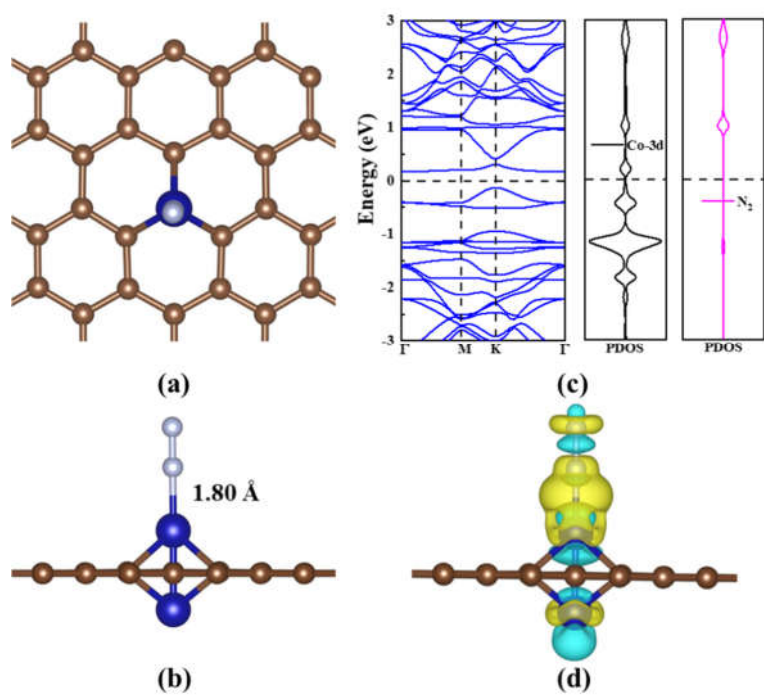


Fig. S9 The most favorable configuration of N₂ adsorption energy on the Co₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

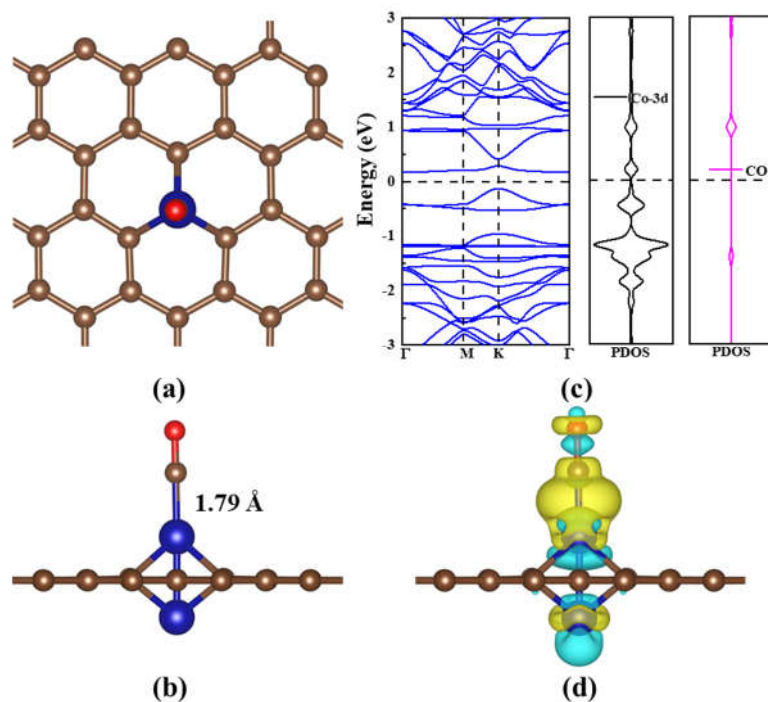


Fig. S10 The most favorable configuration of CO adsorption energy on the $\text{Co}_2\perp$ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

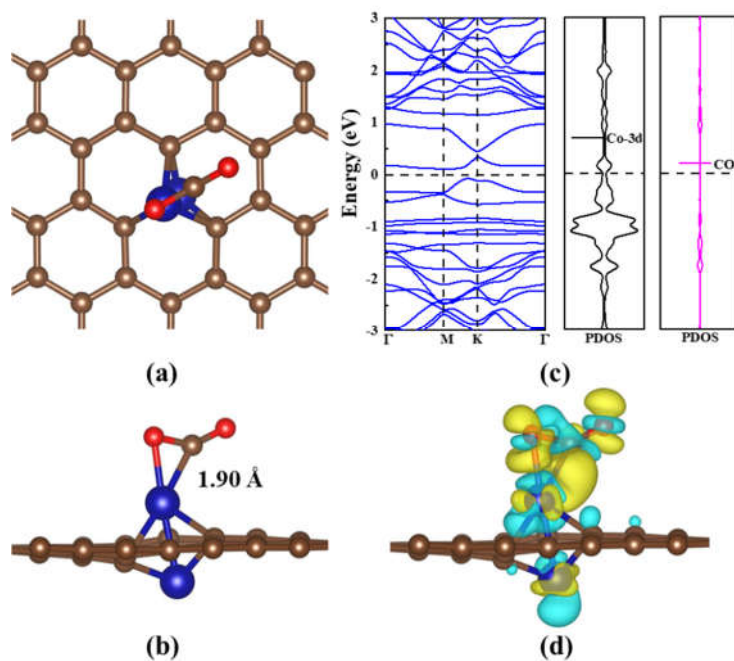


Fig. S11 The most favorable configuration of CO_2 adsorption energy on the $\text{Co}_2\perp$ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

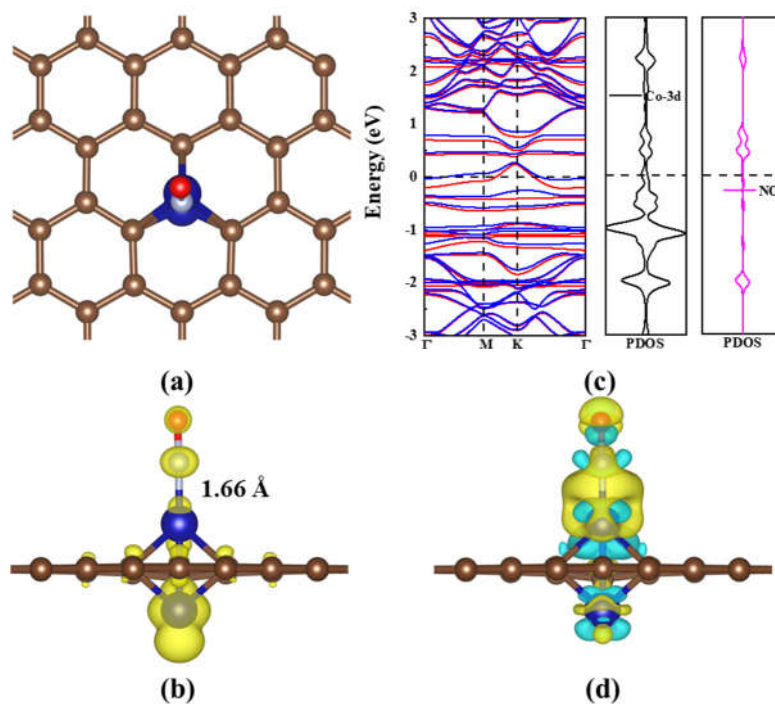


Fig. S12 The most favorable configuration of NO adsorption energy on the Co₂⊥ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

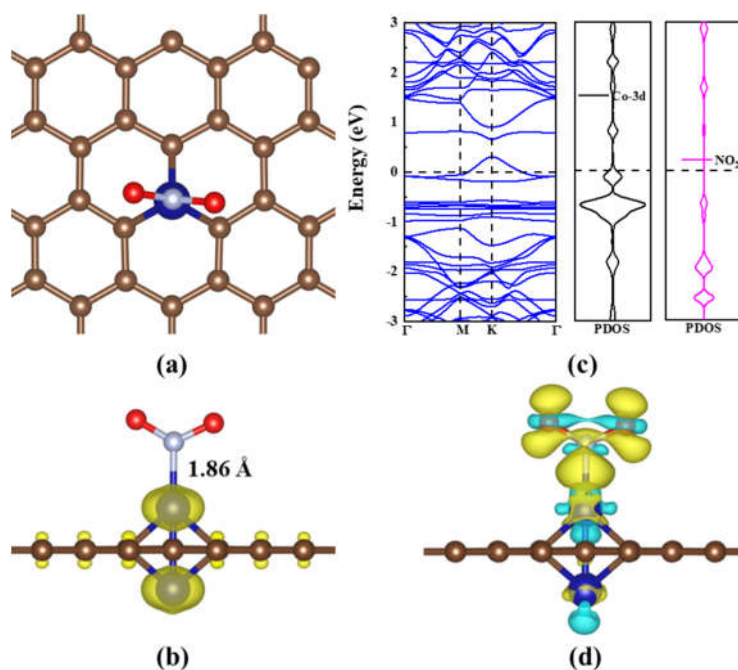


Fig. S13 The most favorable configuration of NO₂ adsorption energy on the Co₂⊥ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.0008 \text{ e}/\text{\AA}^{-3}$ for spatial spin density distributions and $0.003 \text{ e}/\text{\AA}^{-3}$ for CDD, respectively.

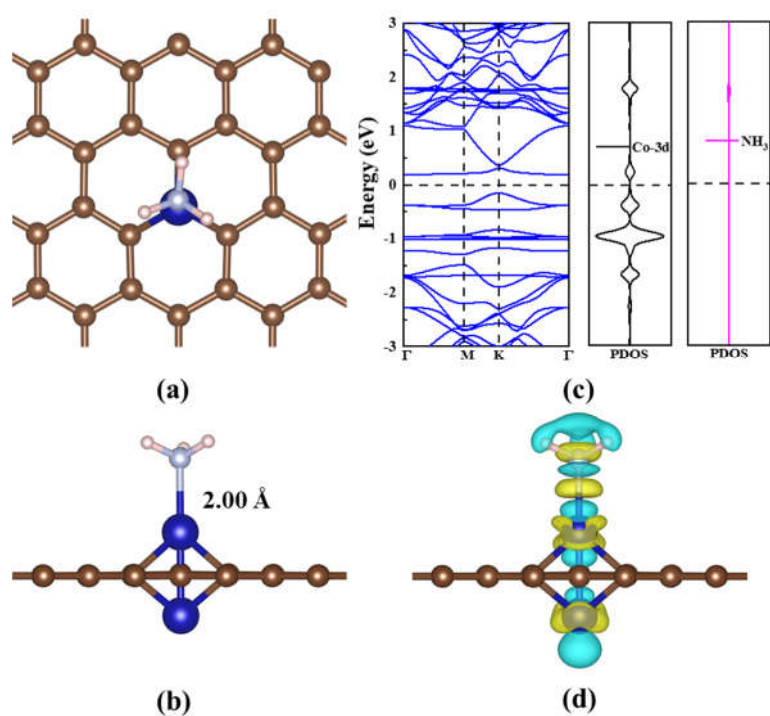


Fig. S14 The most favorable configuration of NH₃ adsorption energy on the Co₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

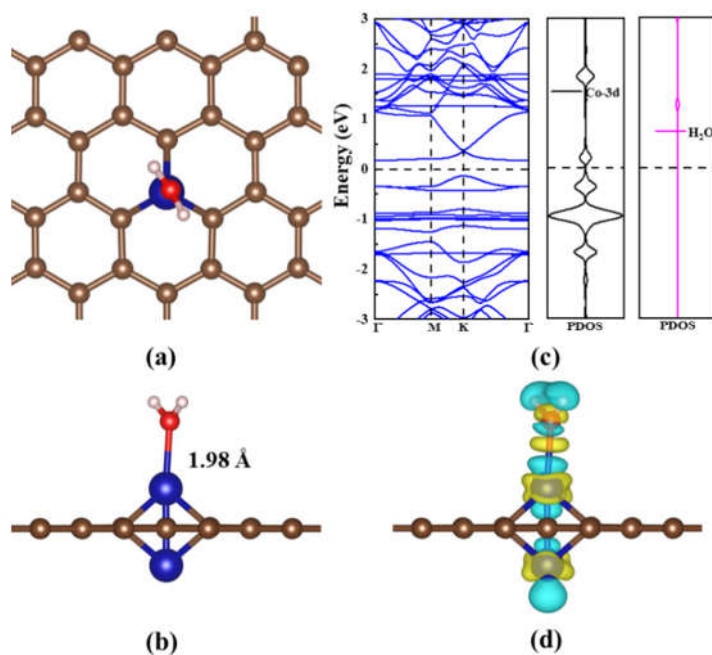


Fig. S15 The most favorable configuration of H₂O adsorption energy on the Co₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

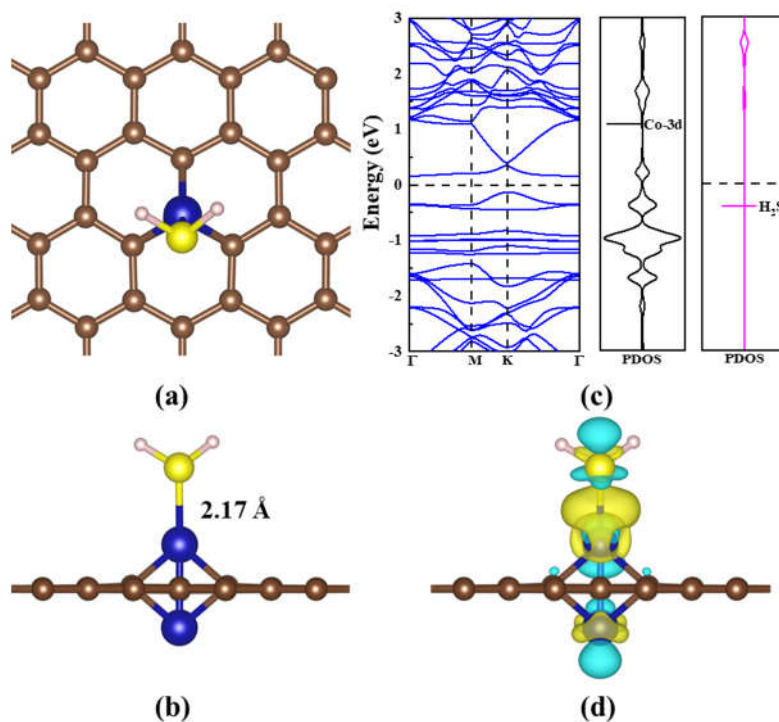


Fig. S16 The most favorable configuration of H₂S adsorption energy on the Co₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

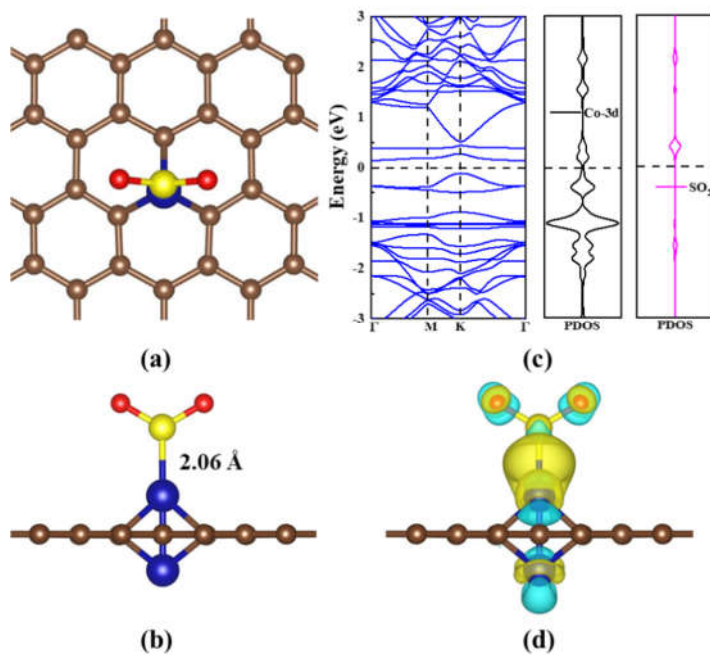


Fig. S17 The most favorable configuration of SO₂ adsorption energy on the Co₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

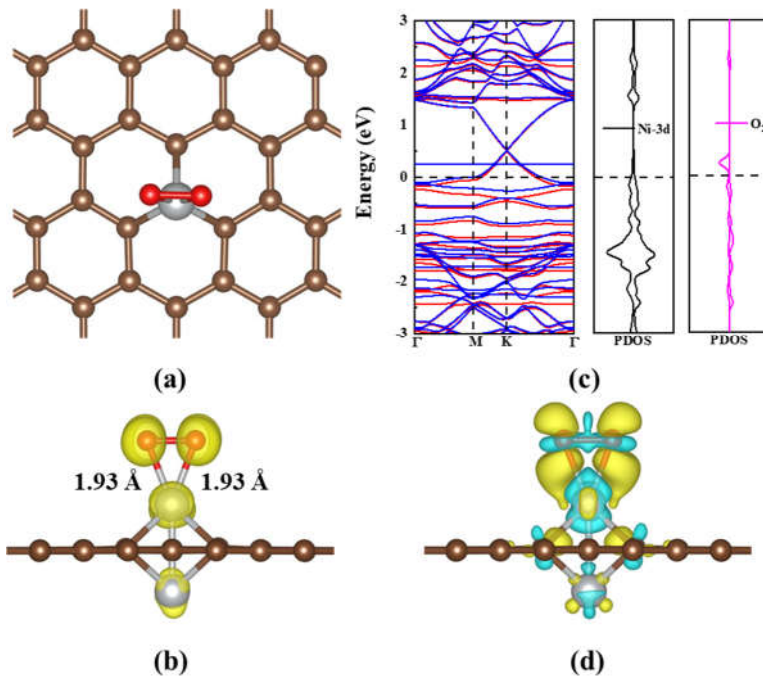


Fig. S18 The most favorable configuration of O₂ adsorption energy on the Ni₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.005 \text{ e}/\text{\AA}^{-3}$ for spatial spin density distributions and $0.003 \text{ e}/\text{\AA}^{-3}$ for CDD, respectively.

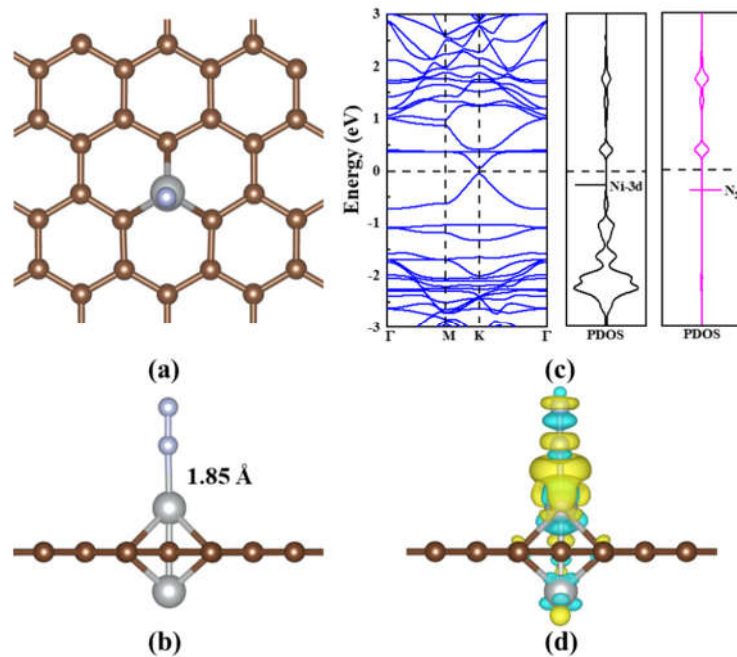


Fig. S19 The most favorable configuration of N₂ adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

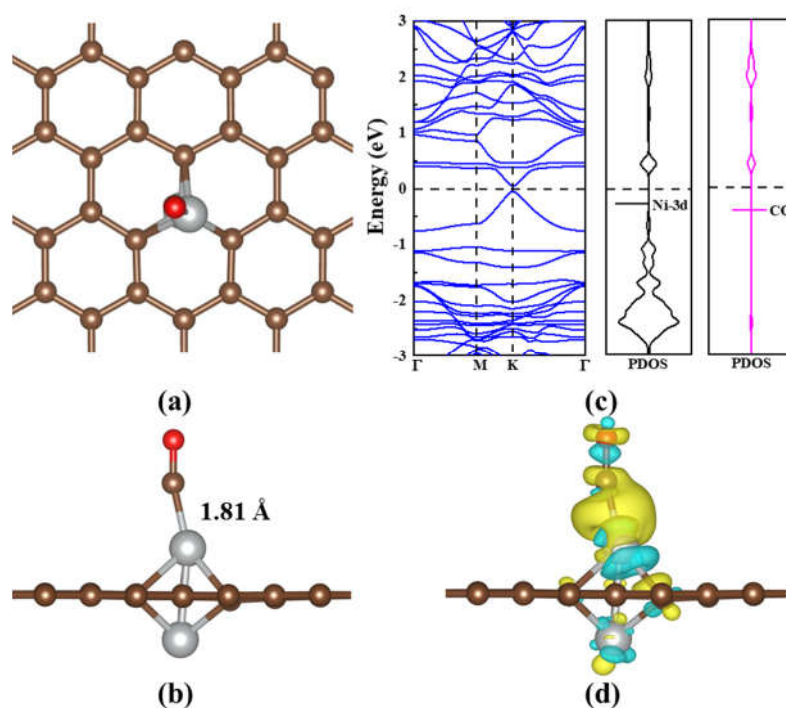


Fig. S20 The most favorable configuration of CO adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

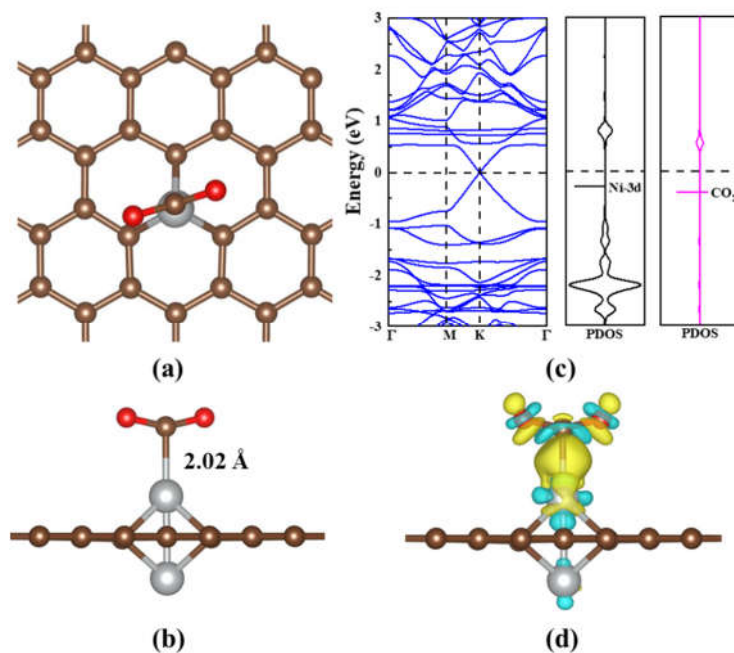


Fig. S21 The most favorable configuration of CO₂ adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

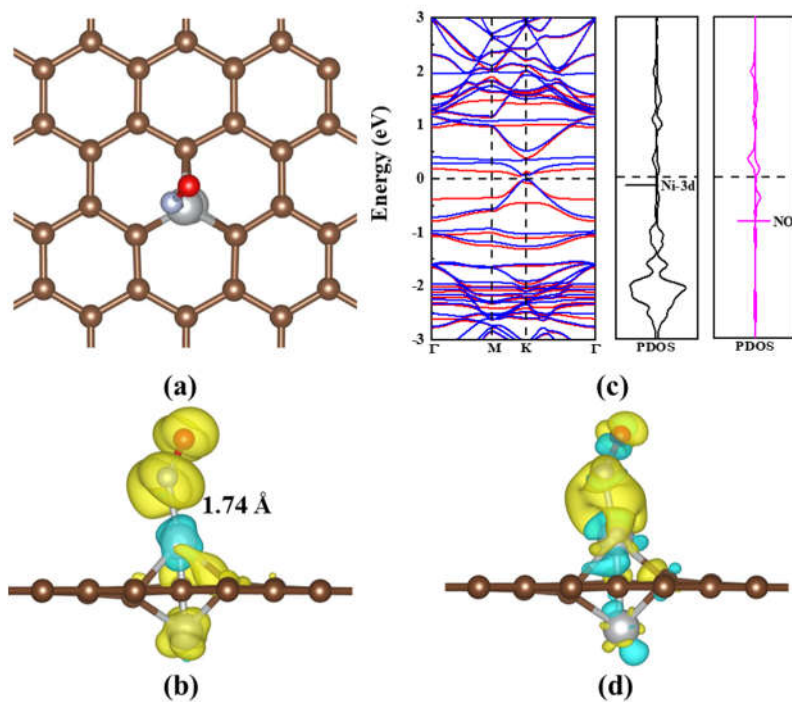


Fig. S22 The most favorable configuration of NO adsorption energy on the $\text{Ni}_2\perp$ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

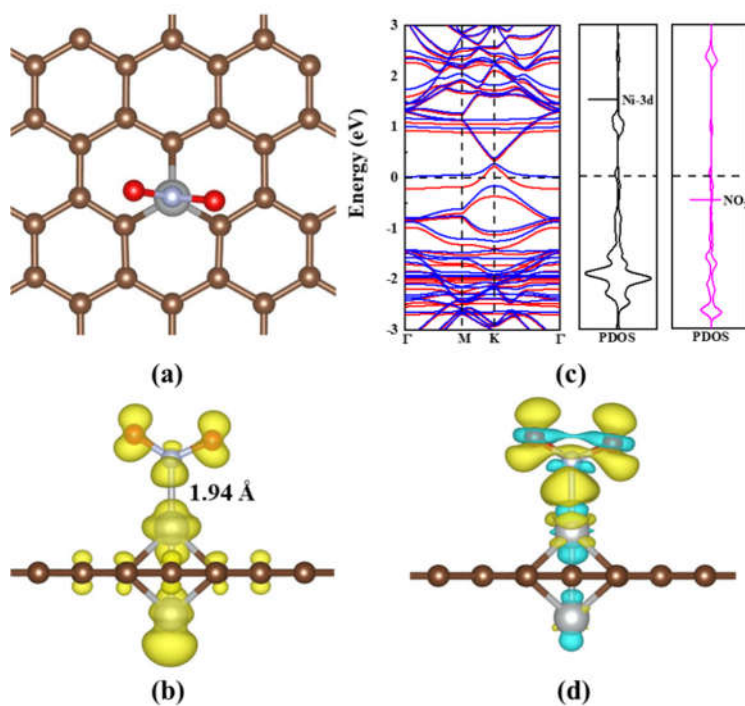


Fig. S23 The most favorable configuration of NO_2 adsorption energy on the $\text{Ni}_2\perp$ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

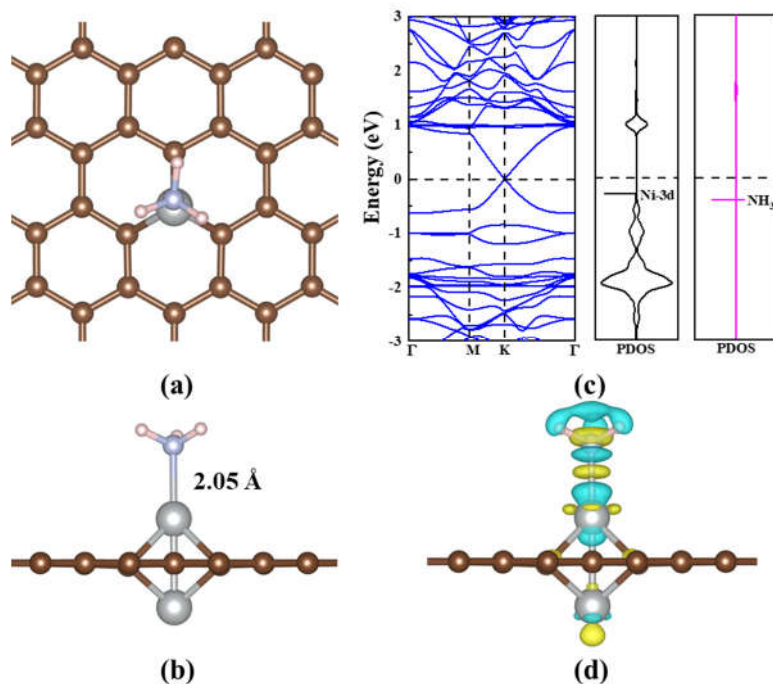


Fig. S24 The most favorable configuration of NH₃ adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

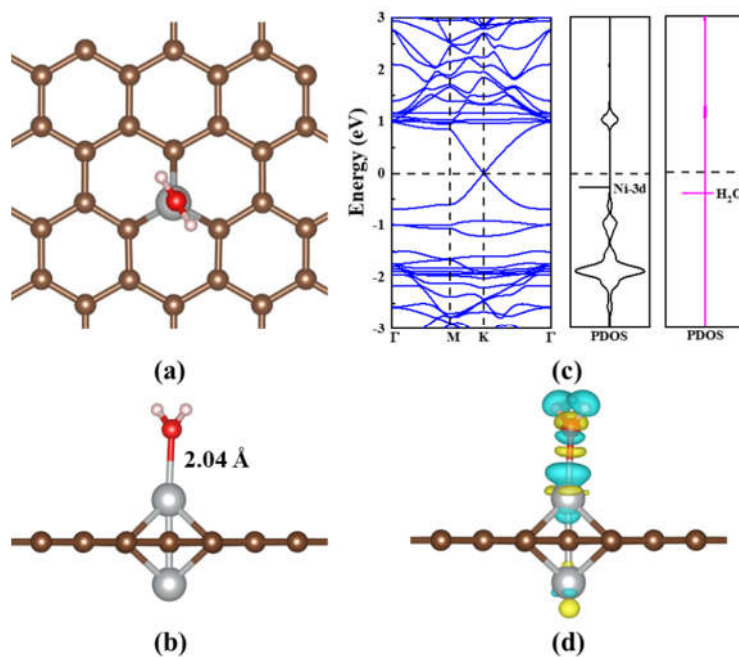


Fig. S25 The most favorable configuration of H₂O adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

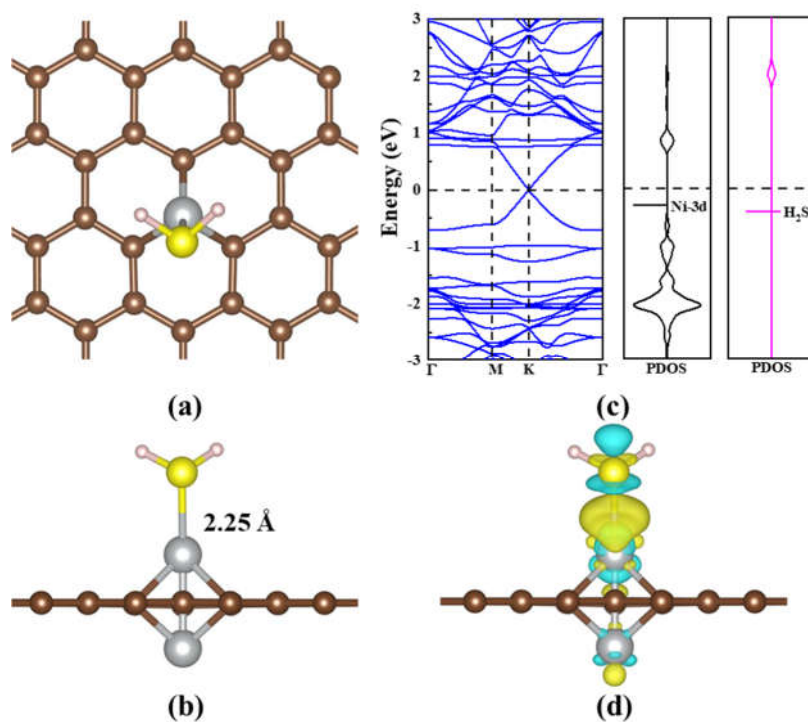


Fig. S26 The most favorable configuration of H₂S adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

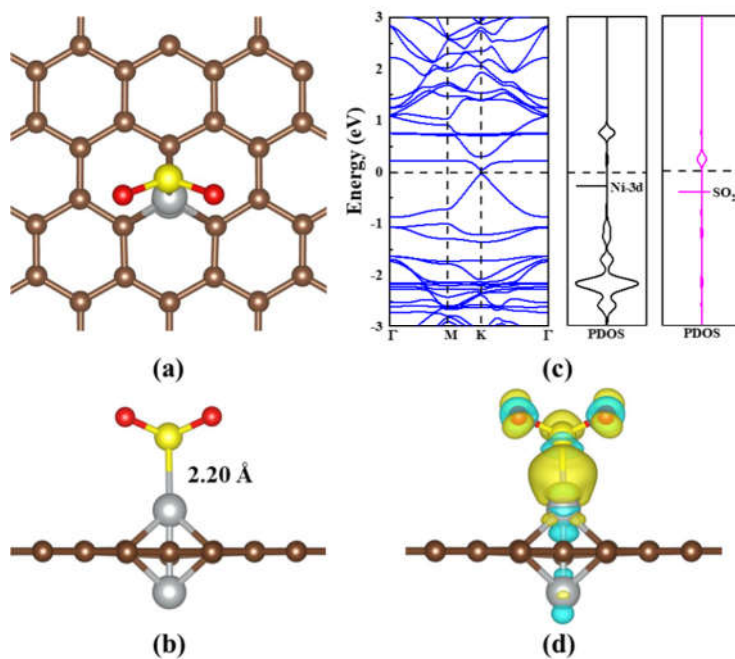


Fig. S27 The most favorable configuration of SO₂ adsorption energy on the Ni₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

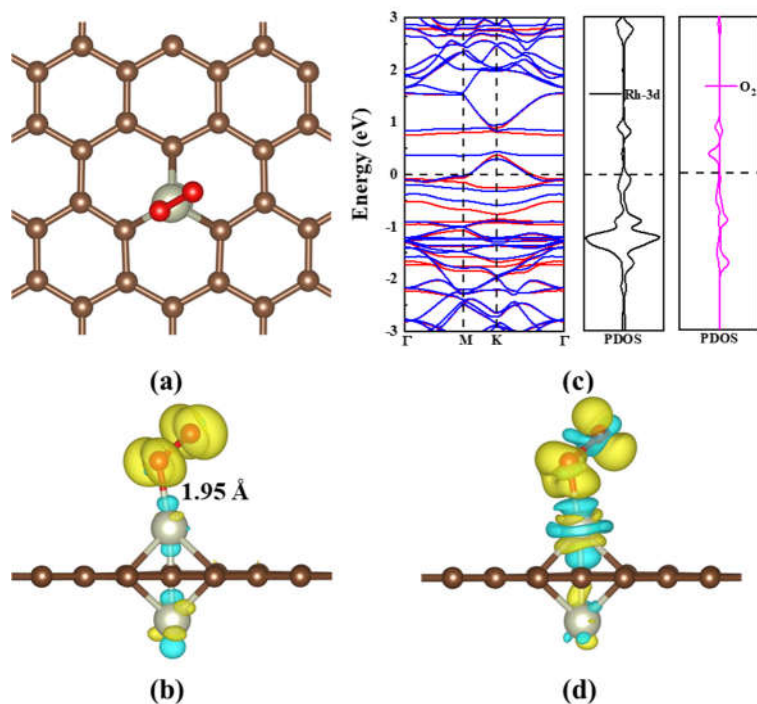


Fig. S28 The most favorable configuration of O₂ adsorption energy on the Rh₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

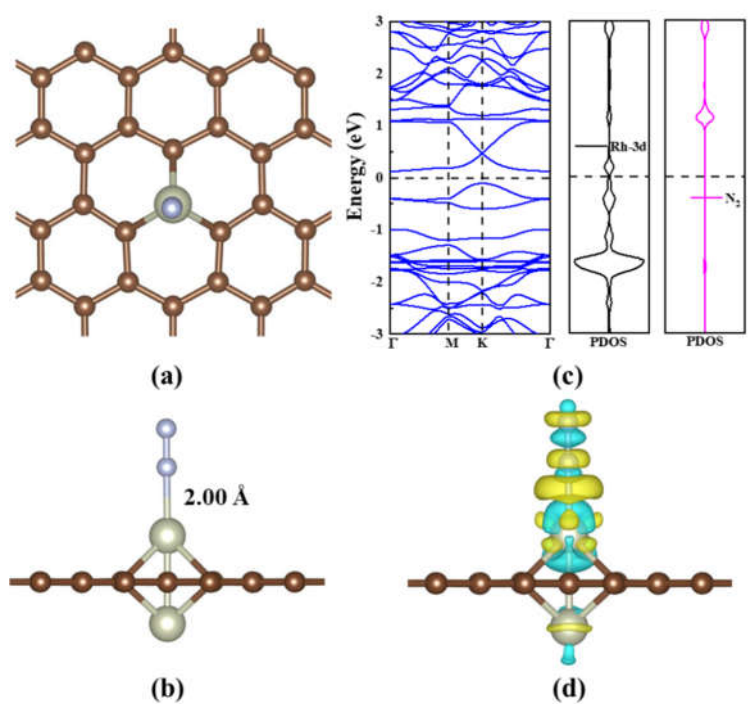


Fig. S29 The most favorable configuration of N₂ adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

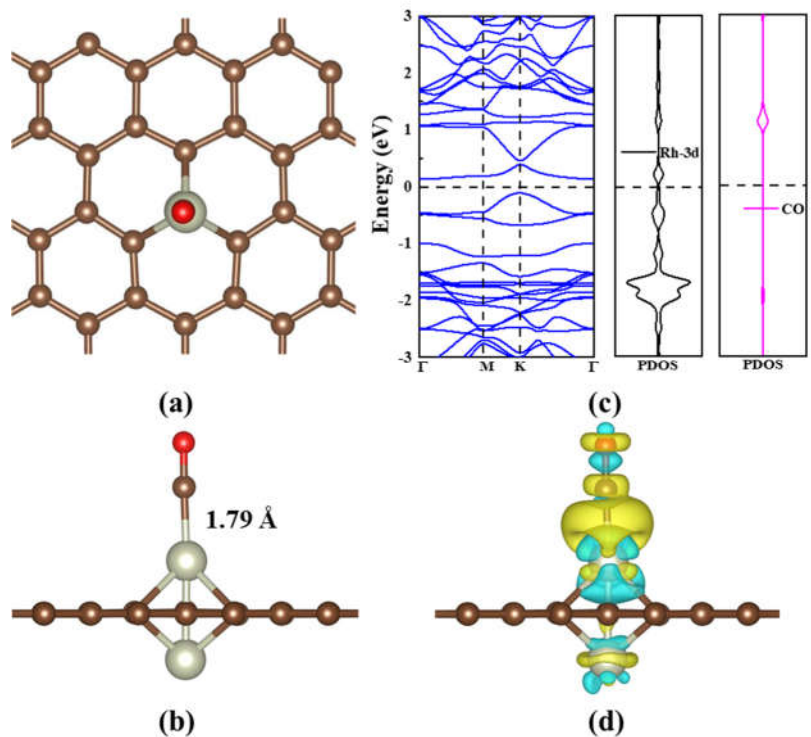


Fig. S30 The most favorable configuration of CO adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

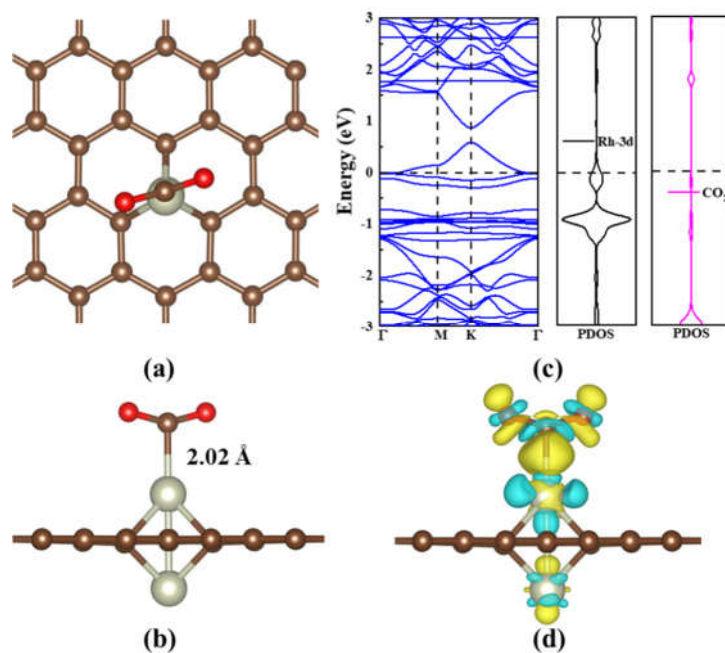


Fig. S31 The most favorable configuration of CO₂ adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

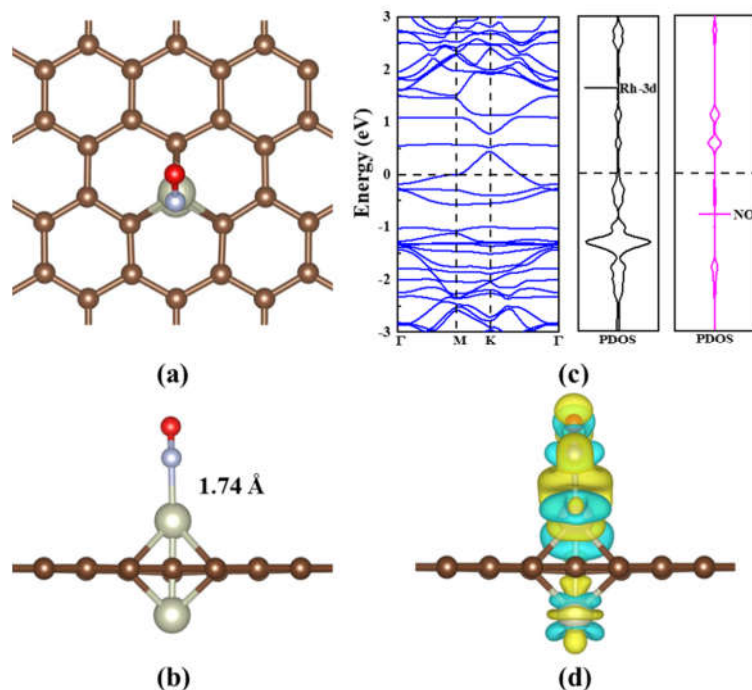


Fig. S32 The most favorable configuration of NO adsorption energy on the Rh₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

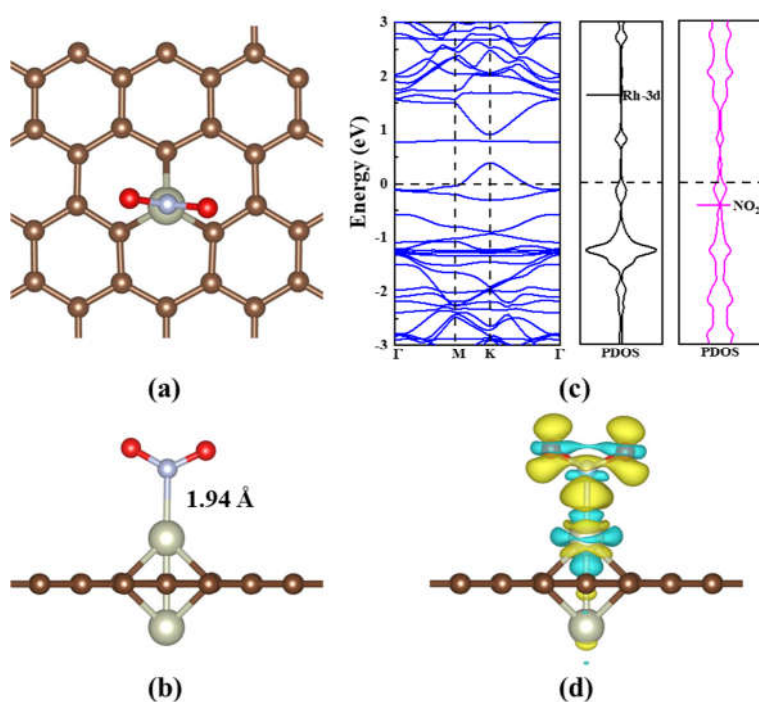


Fig. S33 The most favorable configuration of NO₂ adsorption energy on the Rh₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

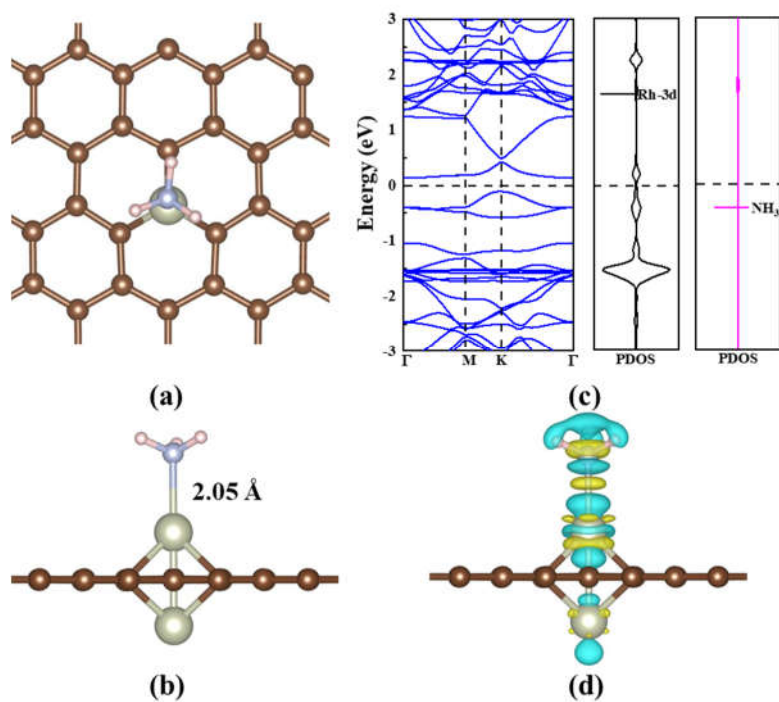


Fig. S34 The most favorable configuration of NH₃ adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

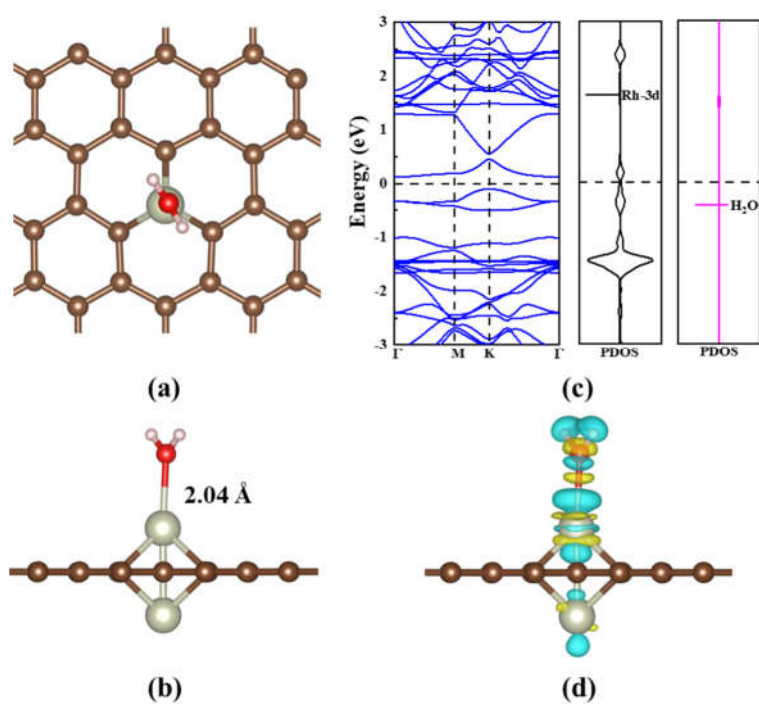


Fig. S35 The most favorable configuration of H₂O adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

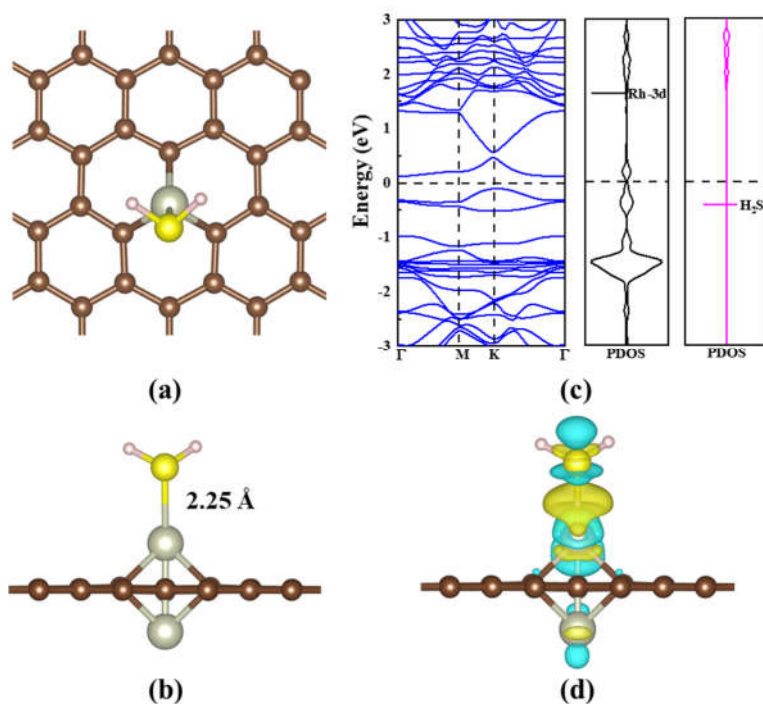


Fig. S36 The most favorable configuration of H₂S adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

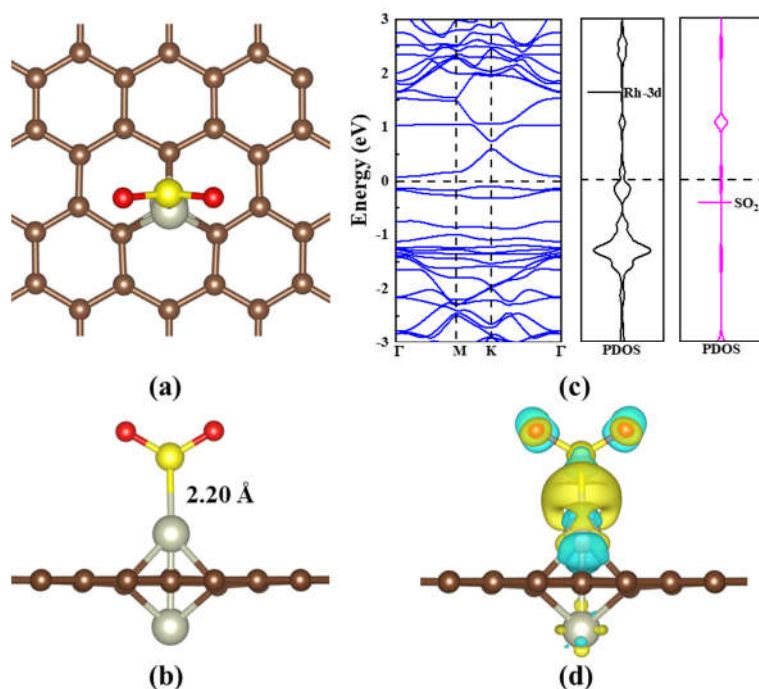


Fig. S37 The most favorable configuration of SO₂ adsorption energy on the Rh₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

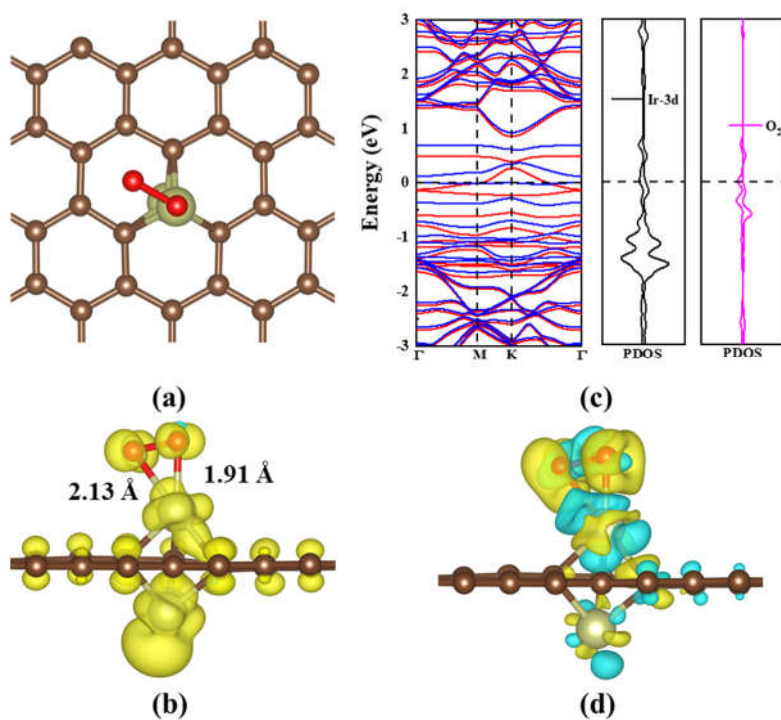


Fig. S38 The most favorable configuration of O₂ adsorption energy on the Ir₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

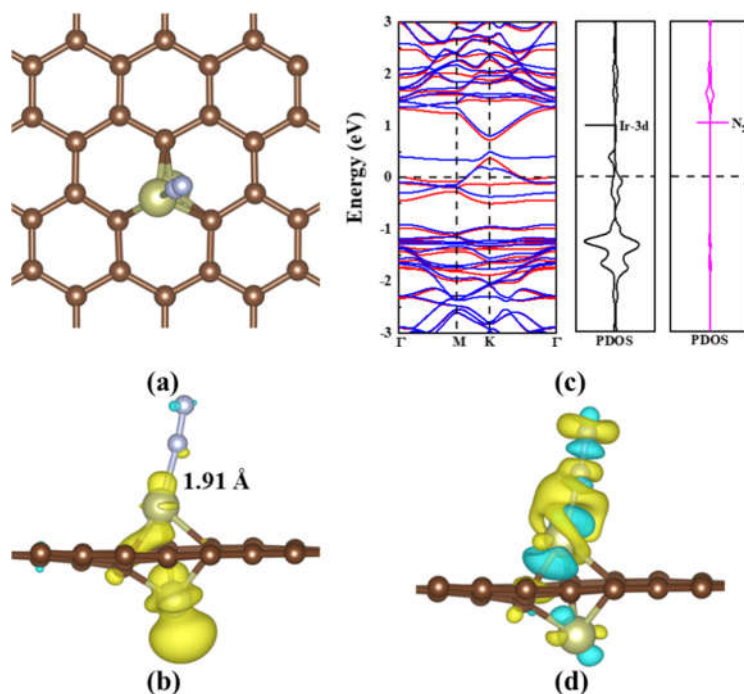


Fig. S39 The most favorable configuration of N₂ adsorption energy on the Ir₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

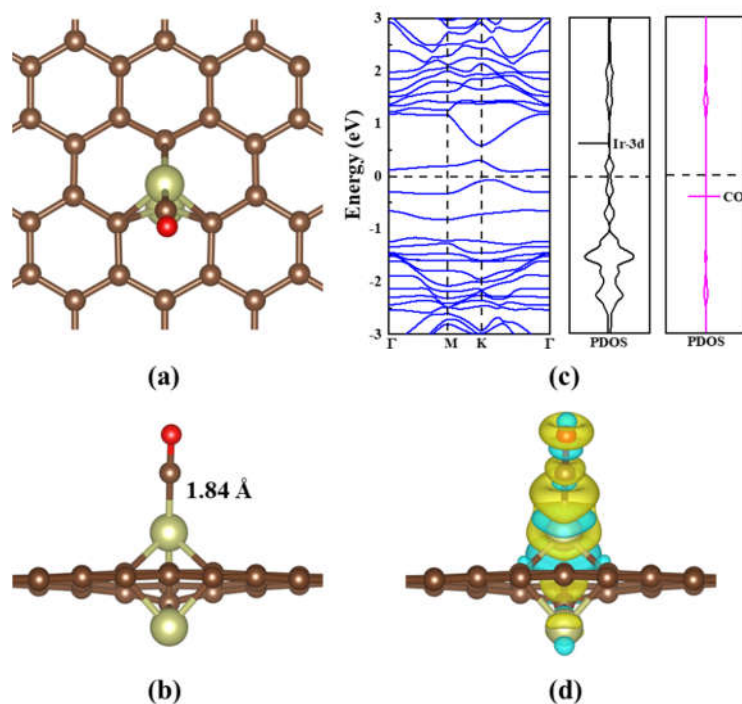


Fig. S40 The most favorable configuration of CO adsorption energy on the Ir₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

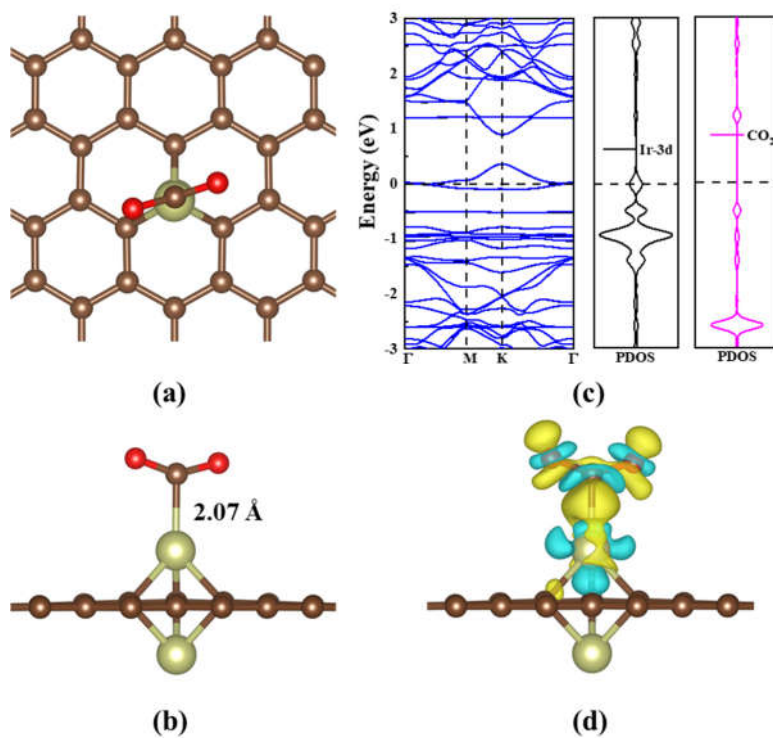


Fig. S41 The most favorable configuration of CO₂ adsorption energy on the Ir₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

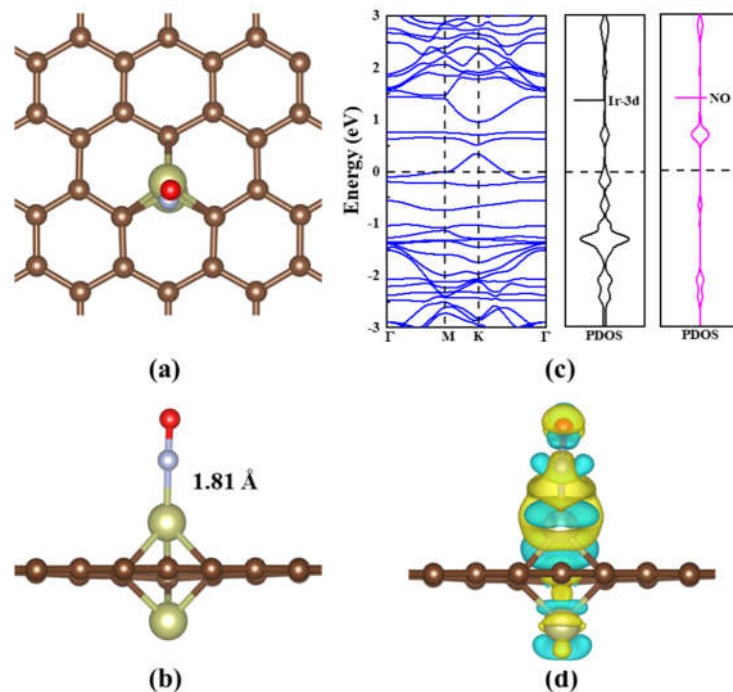


Fig. S42 The most favorable configuration of NO adsorption energy on the Ir₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

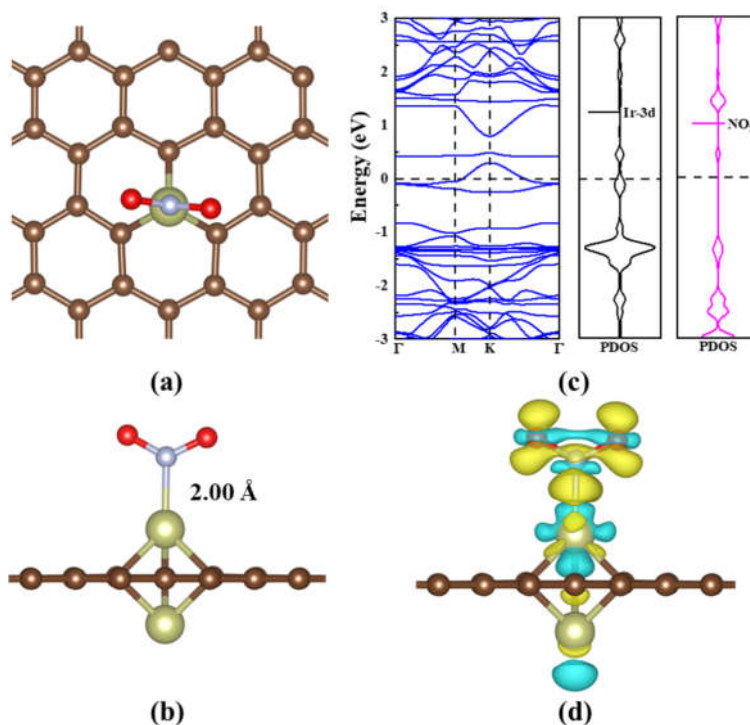


Fig. S43 The most favorable configuration of NO₂ adsorption energy on the Ir₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

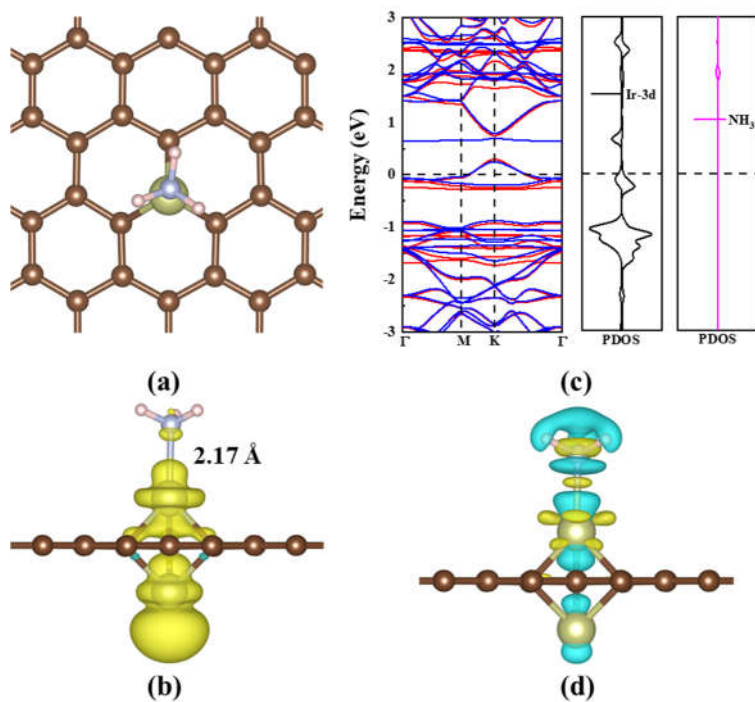


Fig. S44 The most favorable configuration of NH₃ adsorption energy on the Ir₂⊥ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

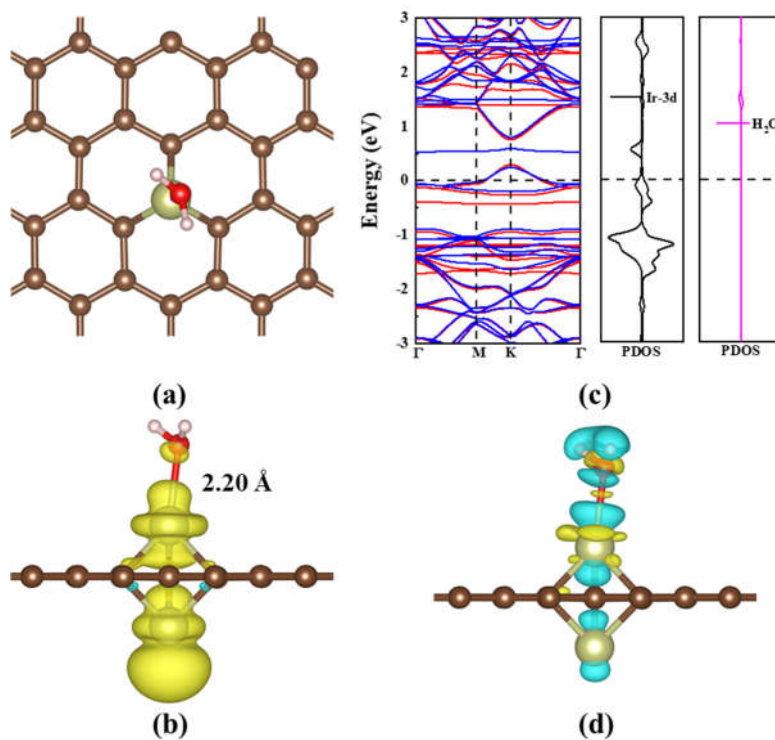


Fig. S45 The most favorable configuration of H₂O adsorption energy on the Ir₂⊥ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

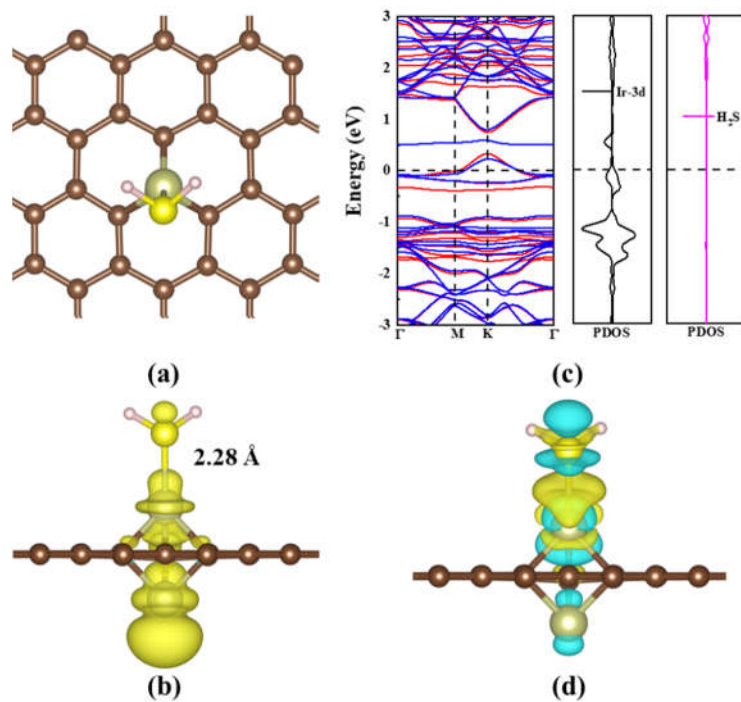


Fig. S46 The most favorable configuration of H₂S adsorption energy on the Ir₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

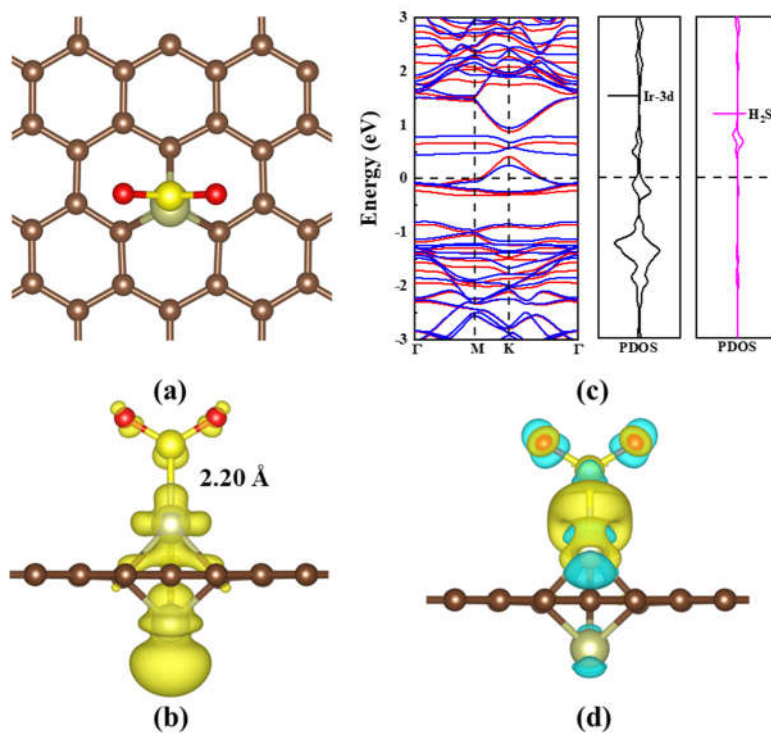


Fig. S47 The most favorable configuration of SO₂ adsorption energy on the Ir₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

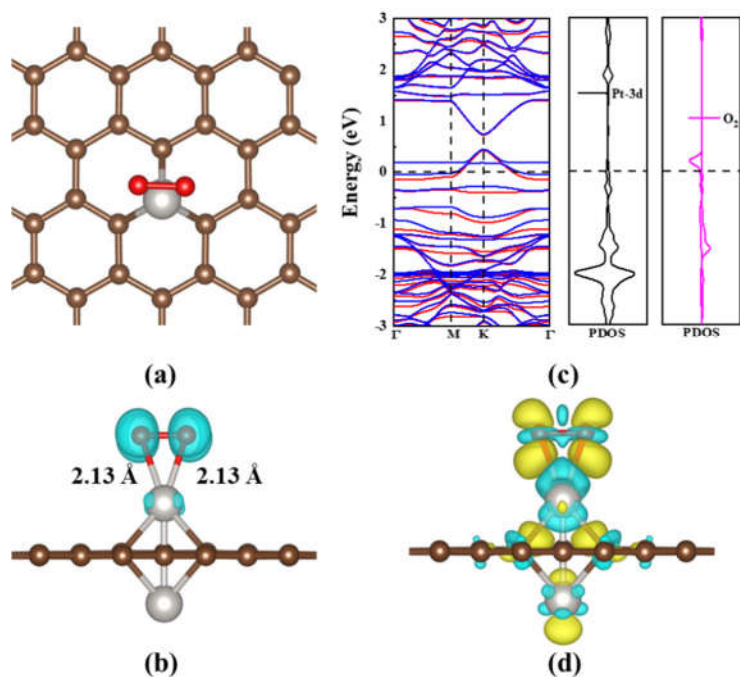


Fig. S48 The most favorable configuration of O₂ adsorption energy on the Pt₂⊥ gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

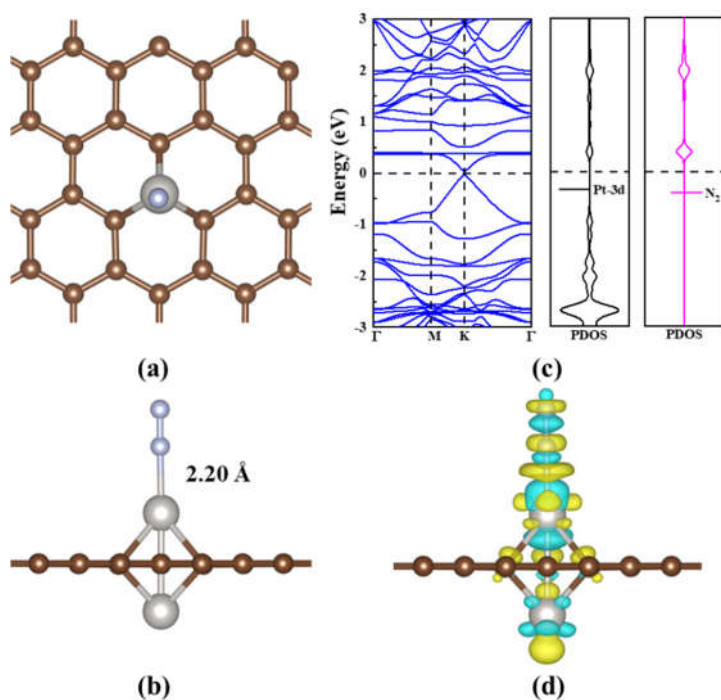


Fig. S49 The most favorable configuration of N₂ adsorption energy on the Pt₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

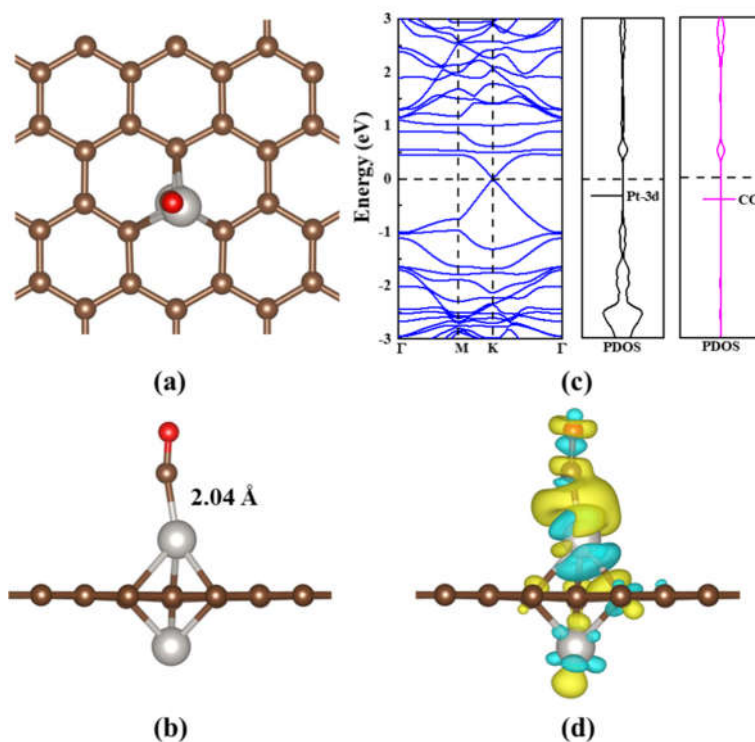


Fig. S50 The most favorable configuration of CO adsorption energy on the Pt₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

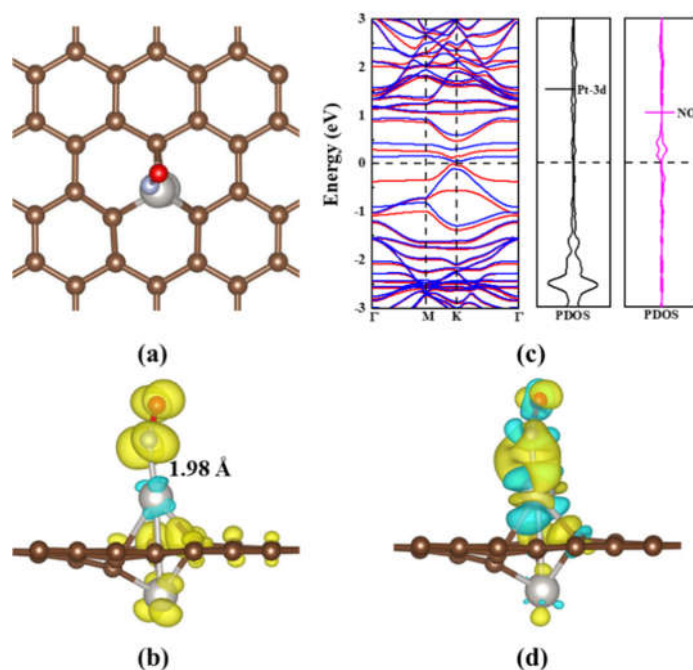


Fig. S51 The most favorable configuration of NO adsorption energy on the Pt₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

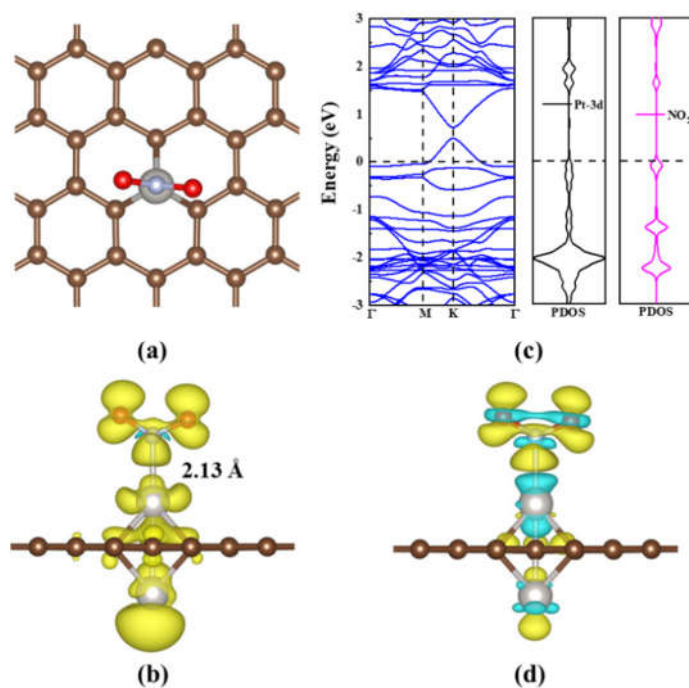


Fig. S52 The most favorable configuration of NO₂ adsorption energy on the Pt₂L gra structure (top view (a)), spatial spin density distributions (b), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.0003 \text{ e}/\text{\AA}^{-3}$ for spatial spin density distributions and $0.003 \text{ e}/\text{\AA}^{-3}$ for CDD, respectively.

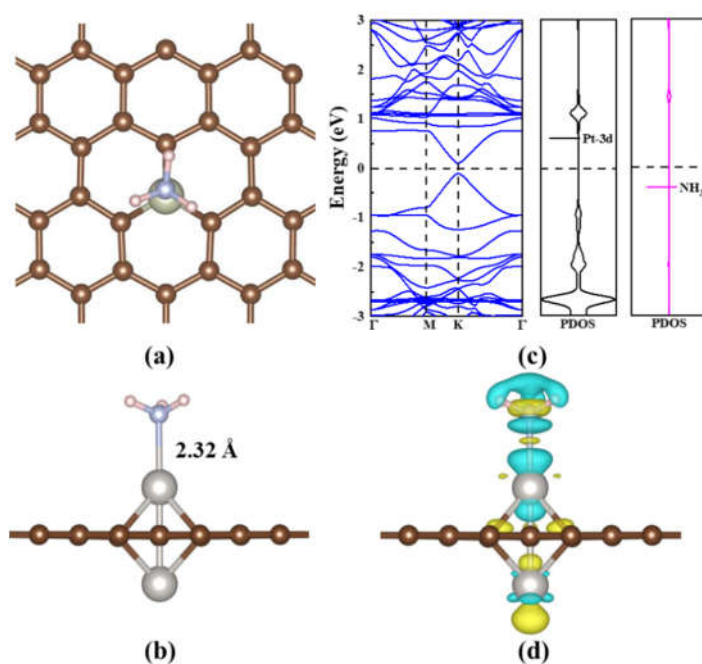


Fig. S53 The most favorable configuration of NH₃ adsorption energy on the Pt₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

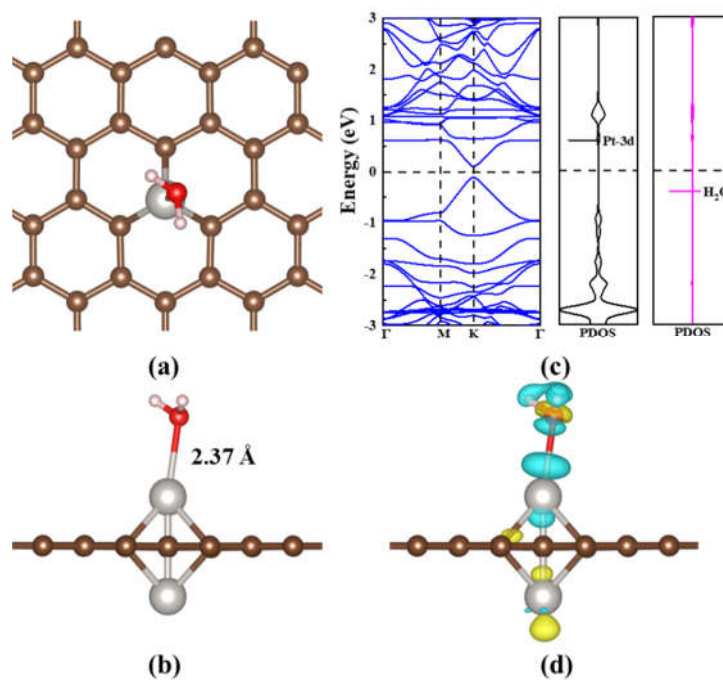


Fig. S54 The most favorable configuration of H₂O adsorption energy on the Pt₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

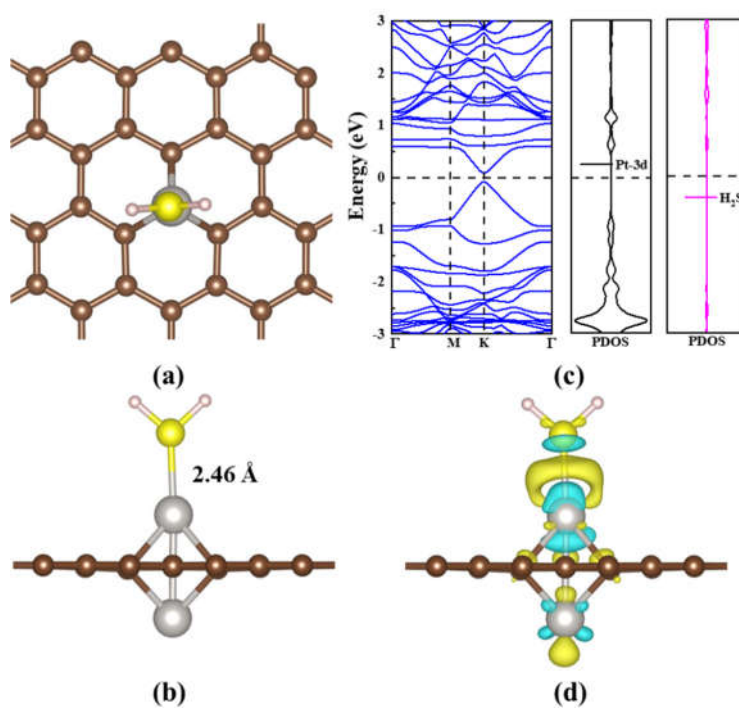


Fig. S55 The most favorable configuration of H₂S adsorption energy on the Pt₂L gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set $0.003 \text{ e}/\text{\AA}^{-3}$.

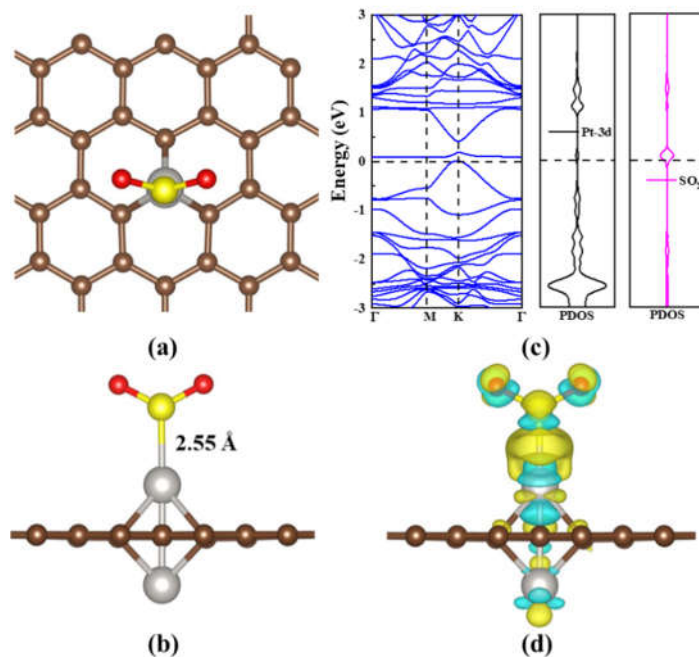


Fig. S56 The most favorable configuration of SO₂ adsorption energy on the Pt₂⊥ gra structure (top view (a) and side view (b)), energy band structure and PDOS (c), and CDD diagram (d). The isosurface value was set 0.003 e/Å⁻³.

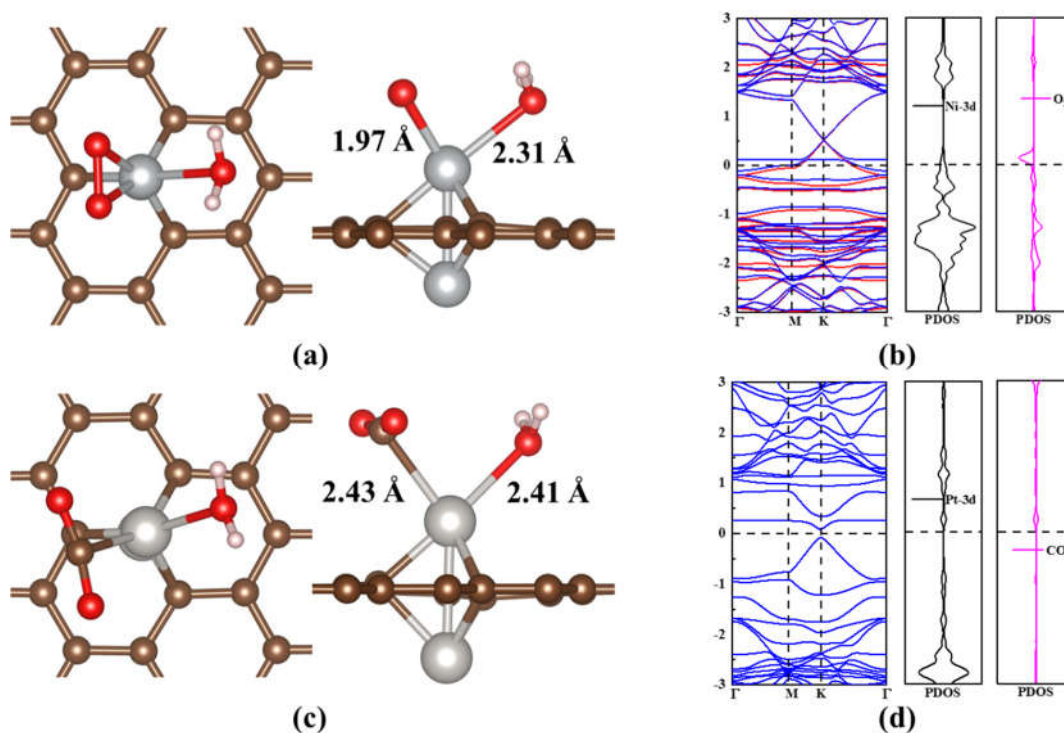


Fig. S57 The most favorable configuration of O₂/CO₂ and H₂O co-adsorption on the Ni₂⊥ gra/Pt₂⊥ gra structure (a/c), and their corresponding energy band structure and PDOS (b/d).