## Electronic Supplementary Information for

## Stretching-mode specificity in the Cl + CH<sub>3</sub>D( $v_1$ -I, $v_1$ -II, and $v_4$

## = 1; $|jK\rangle$ ) reactions: Dependency on the initial $|jK\rangle$ selectivity

Sohidul Mondal, Huilin Pan, 1,2 and Kopin Liu 1,3,4\*

This pdf contains

Fig. S1 (raw images) and Table S1 (rotational energy disposal)

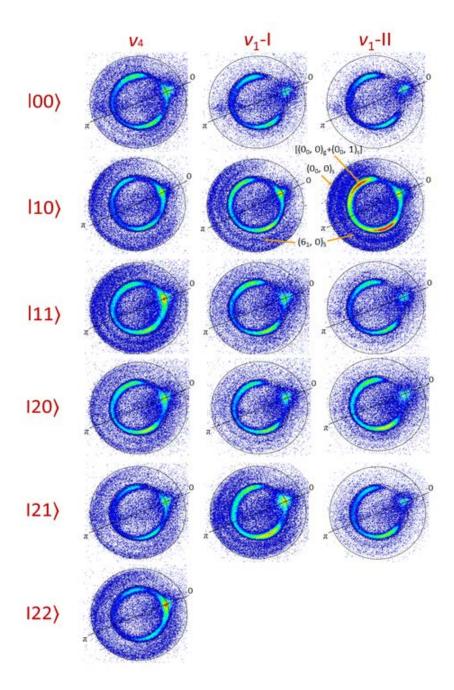
<sup>&</sup>lt;sup>1</sup> Institute of Atomic and Molecular Sciences (IAMS), Academia Sinica, P. O. Box 23-166, Taipei 10699, Taiwan

<sup>&</sup>lt;sup>2</sup> Department of Chemistry, Southern University of Science and Technology, Shenzhen 518055, P. R. China

<sup>&</sup>lt;sup>3</sup> Aerosol Science Research Center, National Sun Yat-sen University, Kaohsiung 80424, Taiwan

<sup>&</sup>lt;sup>4</sup> State Key Laboratory of Molecular Reaction Dynamics, Dalian Institute of Chemical Physics, CAS, Dalian 116023, P. R. China

<sup>\*</sup> Email: kliu@po.iams.sinica.edu.tw



**Fig. S1** Raw difference-images, obtained by [(IR-on) –  $(1-n_s/n_0)$ ·(IR-off)], of the probed CH<sub>2</sub>D( $0_0/6_1$ ) products in 16 ro-vibrationally selected reactions of Cl + CH<sub>3</sub>D( $v_4$ ,  $v_1$ -I,  $v_1$ -II;  $|jK\rangle$ ) at  $E_c = 5.4$  kcal mol<sup>-1</sup>.

**Table S1** Comparison of the averaged rotational energy,  $\langle E_R \rangle$  in kcal mol<sup>-1</sup>, of the  $(0_0, 0)$  product pairs in both the stretch-excited and the ground-state reactions of Cl + CH<sub>3</sub>D. The  $\langle E_R \rangle$  values and the quoted uncertainties ( $\pm$  two standard deviations) were deduced from the measured TKER distributions of the corresponding  $(0_0, 0)_{s/g}$  pairs acquired in three to five independent experiments (see text for details).

<i>jK</i>  >	$(0_0, 0)_g$	$(0_0,0)_s$		
		$v_4$	$v_1$ -I	v <sub>1</sub> -II
00>	$0.52 \pm 0.20$	$3.01 \pm 0.32$	$2.86 \pm 0.32$	$2.79 \pm 1.06$
10>		$2.52 \pm 0.21$	$2.54 \pm 0.25$	$2.40 \pm 0.14$
11>		$3.54 \pm 0.20$	$3.16 \pm 1.16$	$3.05\pm0.30$
20>		$3.22\pm0.20$	$3.46 \pm 0.16$	$3.02\pm0.28$
21)		$3.24 \pm 0.36$	$3.15 \pm 0.98$	$3.02 \pm 0.68$
22}		$2.98 \pm 0.04$		