

Supporting information

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An Ab-initio Study of the Graphyne (GY) for Detection of Rare Bases in DNA

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Figure S2: Optimized geometries and interplanar distances of the complexes of rare bases (a:Cyt, b:5-meCyt, c:5-hmCyt, d:5-fCyt, e:5-caCyt) with N-doped γ -GY, The top views as well as the side views of the complexes are shown. (gray ball is C atom; white ball is H atom; red ball is O atom; blue ball is N atom.)

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Table S1: The cohesive energies of graphene and B/N-doped grapheme, pristine, B- and N-doped γ -GY.

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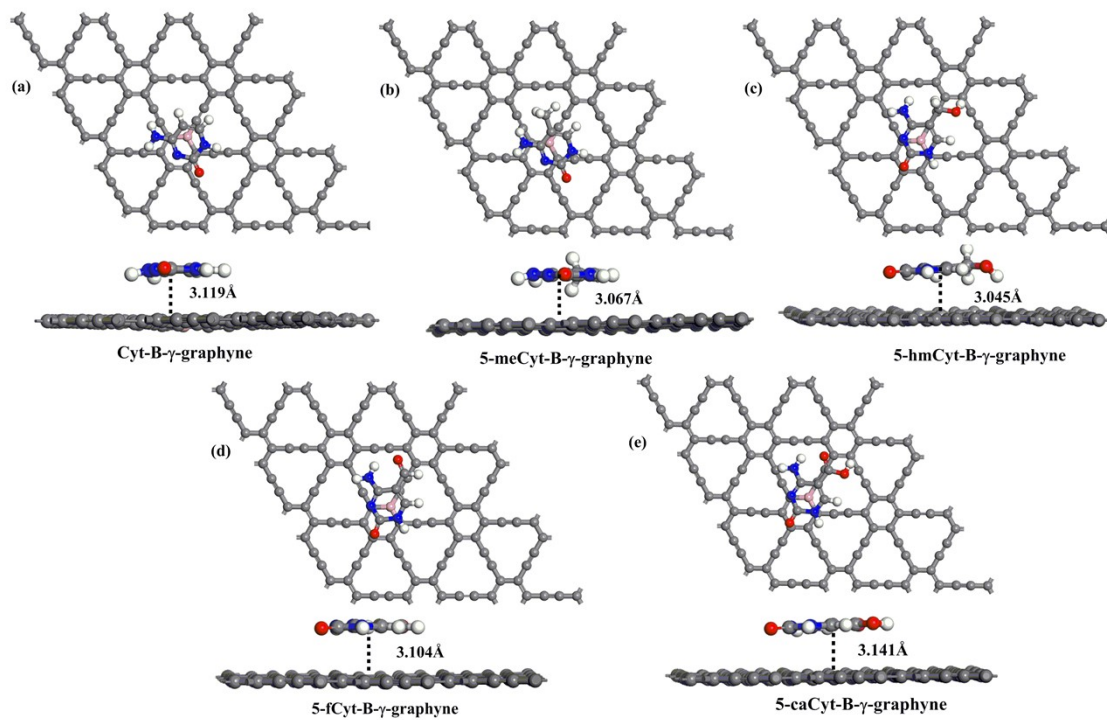


Fig. S1

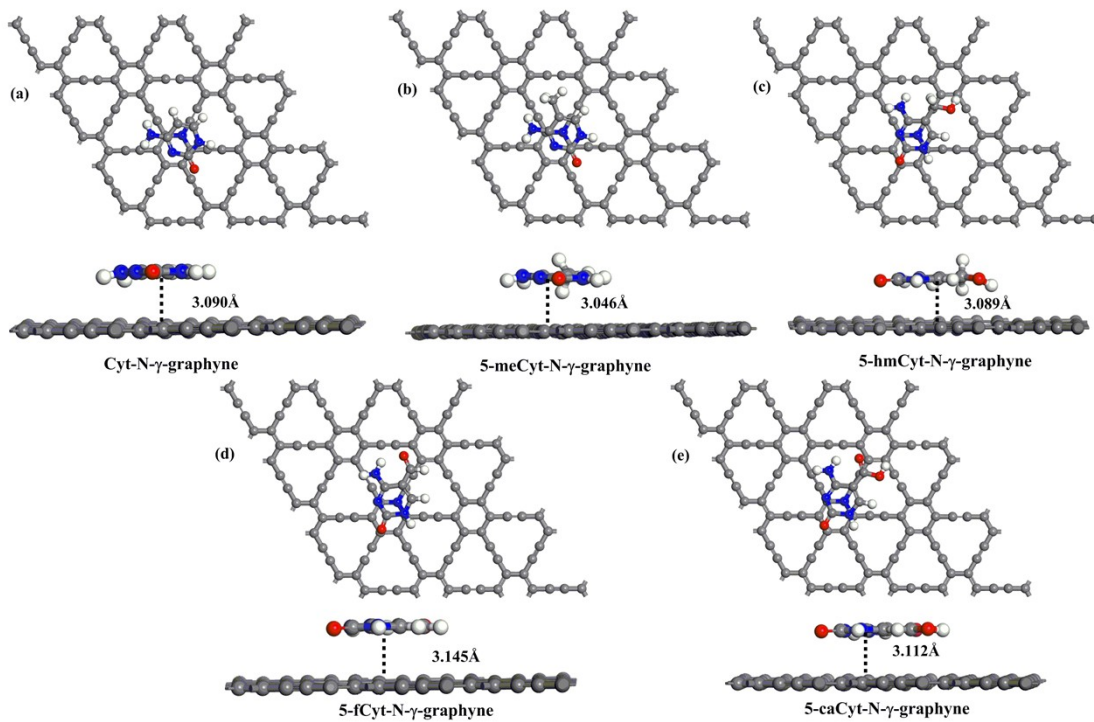


Fig. S2

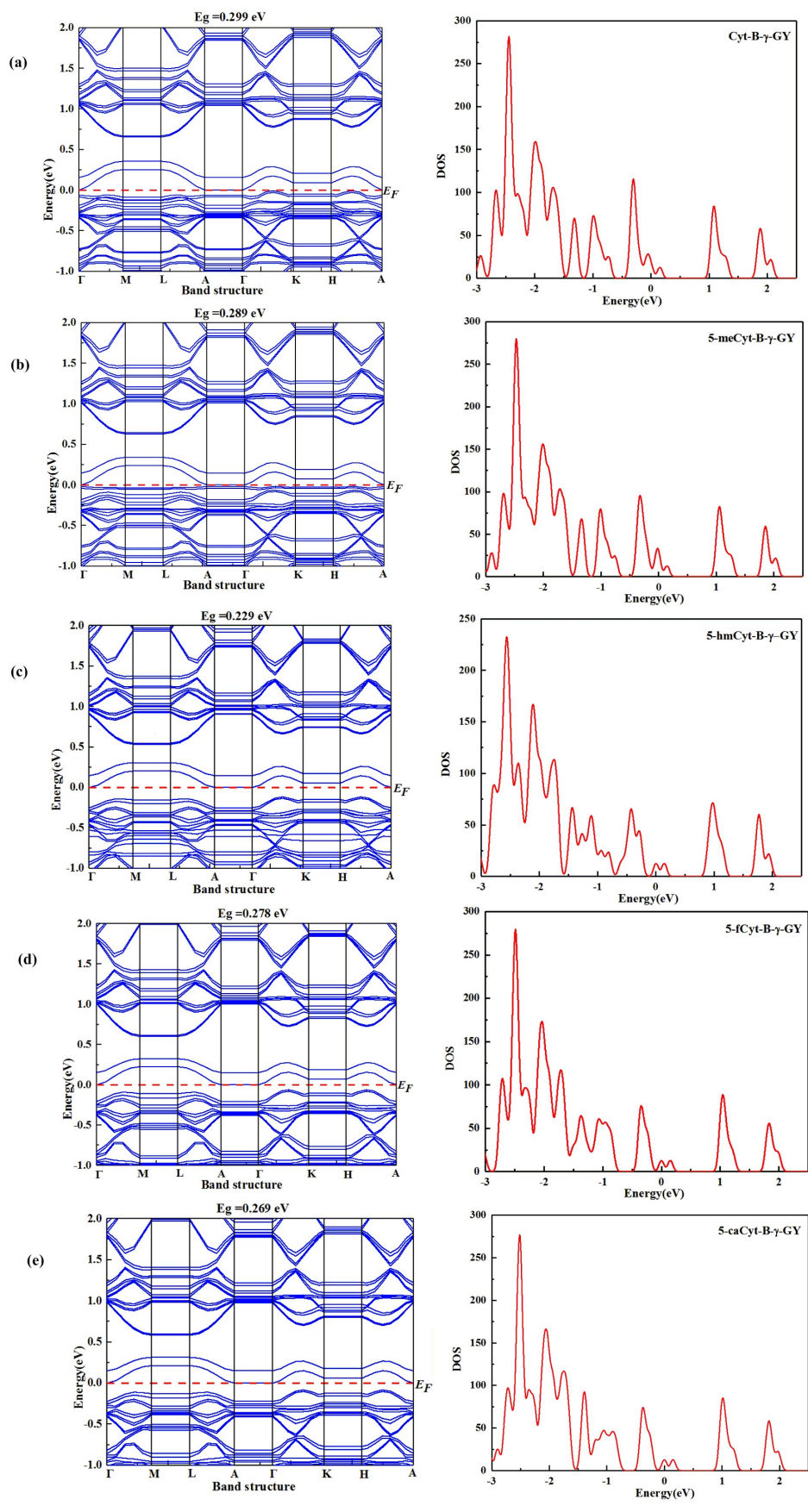


Fig. S3

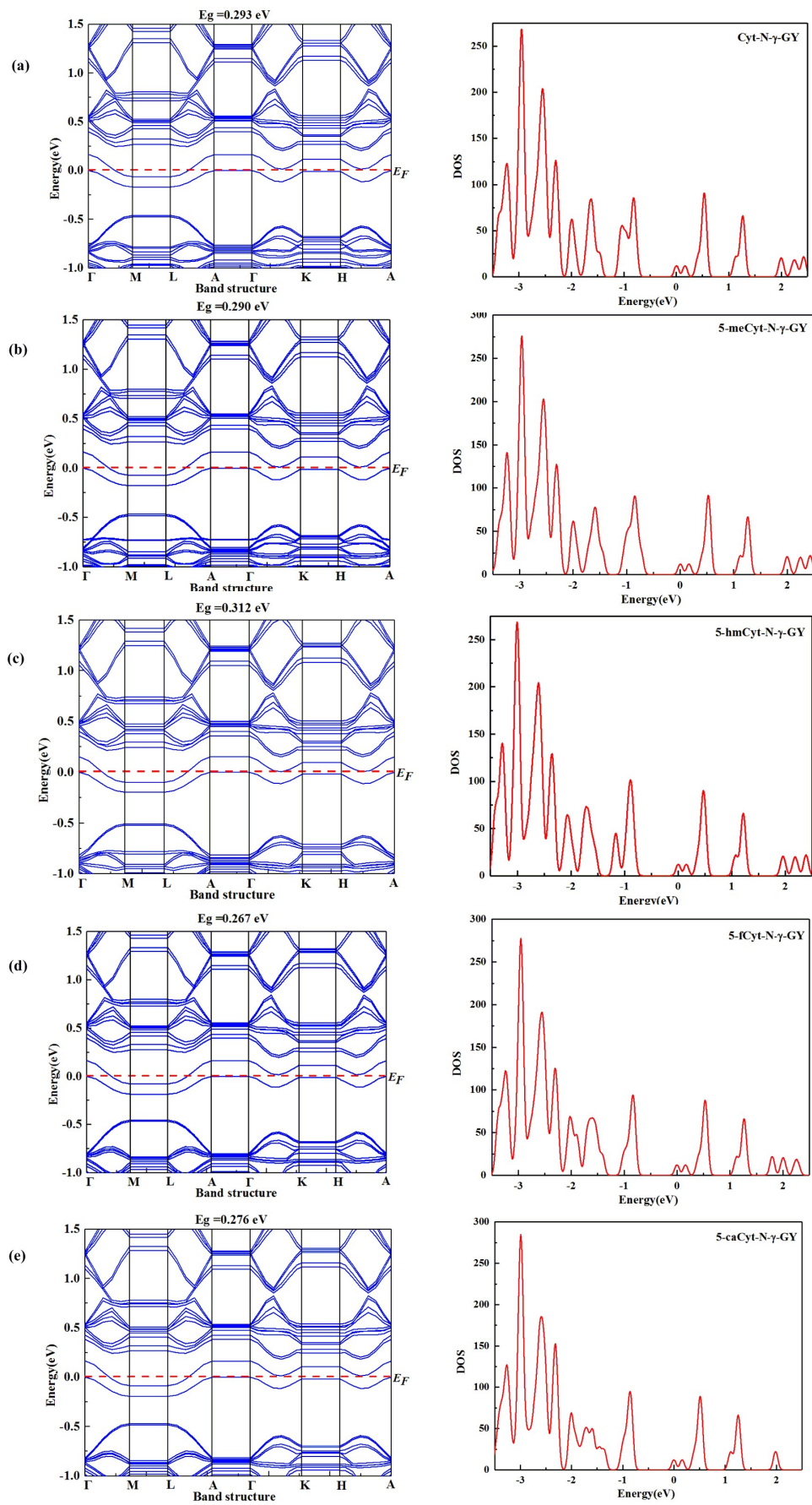


Fig. S4

Table S1: The cohesive energies (eV) of graphene and B/N-doped graphene, pristine, B- and N-doped γ -GY

Nanostructure	$E_{\text{coh}}/ \text{eV}$
γ -GY	-8.79
B- γ -GY	-8.78
N- γ -GY	-8.74
graphene	-9.35
B-graphene	-9.28

Table S2 The adsorption energy (E_{ad}/eV) upon adsorption of rare base on different nanostructures

Rare base and Nanostructure	γ -GY	B- γ -GY	N- γ -GY	graphene	B-graphene	N-graphene
Cyt	-0.781	-0.986	-0.853	-1.156	-1.340	-1.298
5-meCyt	-0.814	-1.033	-0.905	-1.462	-1.633	-1.588
5-hmCyt	-0.957	-1.122	-0.933	-1.188	-1.470	-1.399
5-fCyt	-0.689	-0.903	-0.697	-1.553	-1.666	-1.551
5-caCyt	-0.791	-0.979	-0.778	-1.166	-1.322	-1.177