

## Supplementary information

### Depth resolved oxidational studies of Be/Al periodic multilayers investigated by X-ray photoelectron spectroscopy

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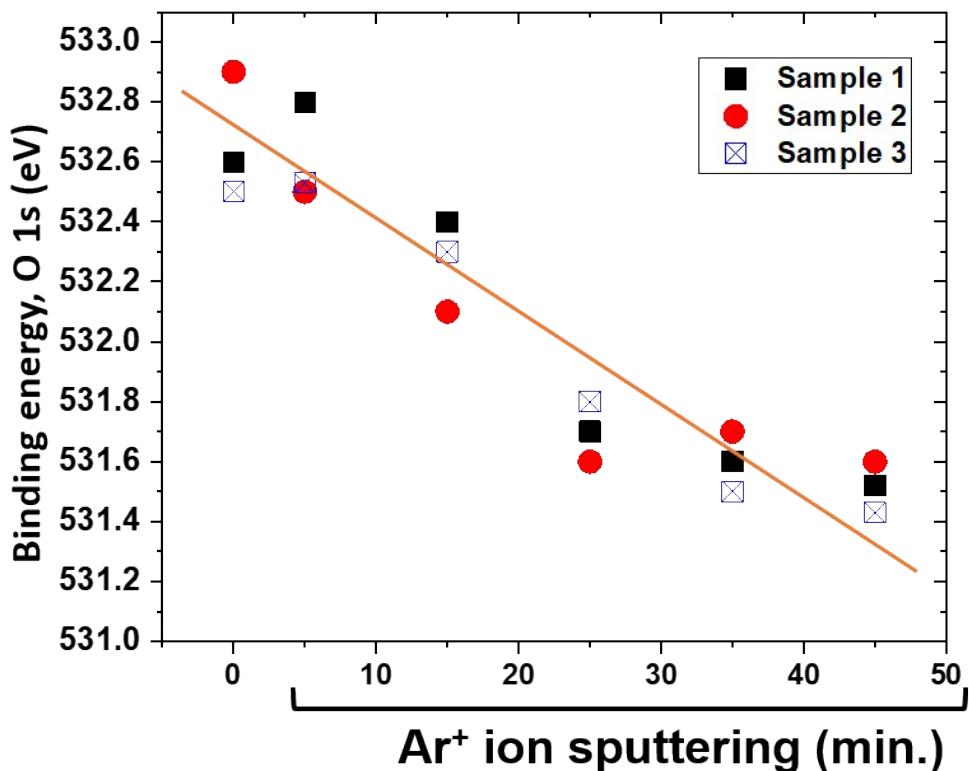
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**Fig. 1-SI.** O 1s peak shift for as – deposited (0), and Ar<sup>+</sup> ion sputtered samples for 5, 15, 25, 35 and 45 minutes.

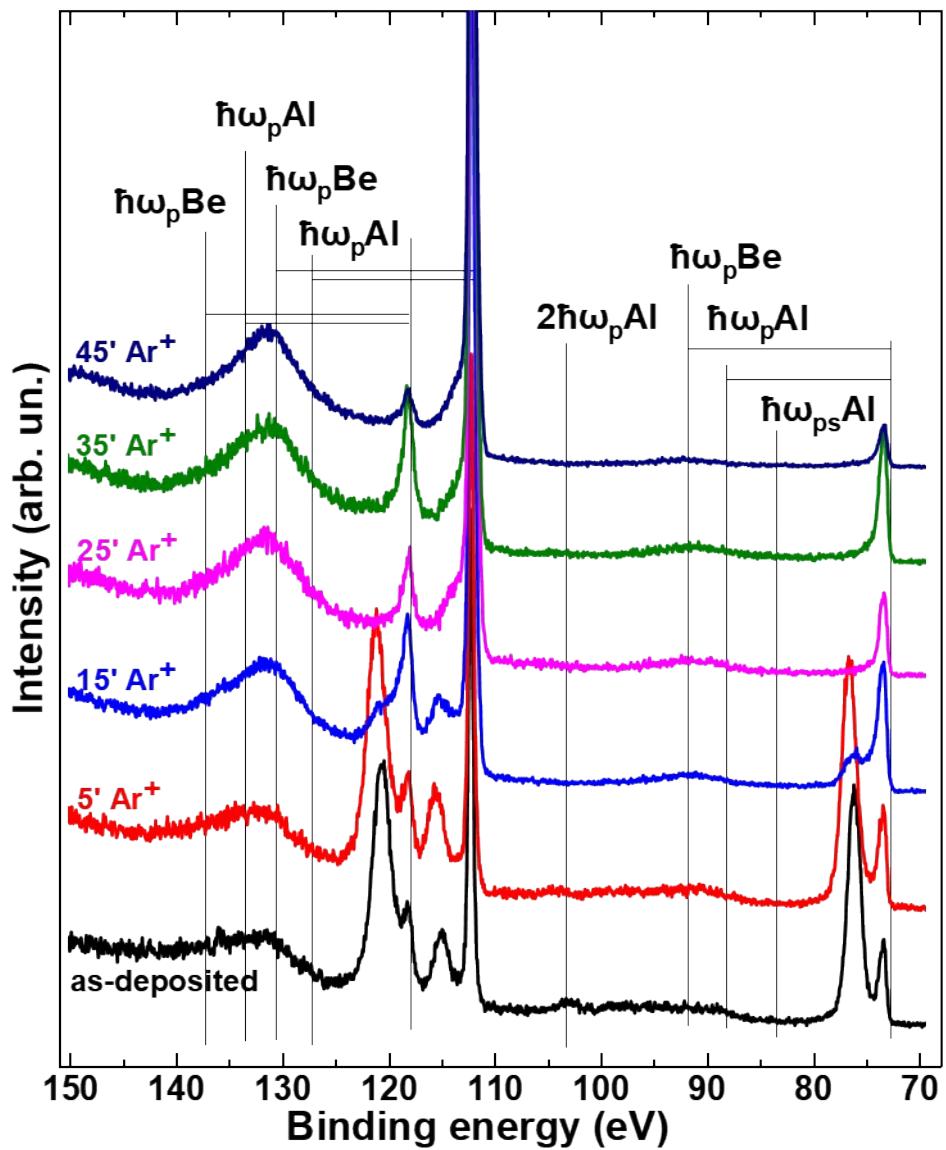
**Fig. 2-SI.** Plasmon shift in XPS spectra of Be<sub>7.86</sub>/Al<sub>0.62</sub> (sample 1):  $\hbar\omega_p$ Be and  $\hbar\omega_p$ Al – first order plasmon of Be and Al, respectively; 2 $\hbar\omega_p$ Be and 2 $\hbar\omega_p$ Al – second order plasmon of Be and Al, respectively;  $\hbar\omega_{ps}$ Al – surface plasmon of Al.

**Fig. 3-SI.** Plasmon shift in XPS spectra of Be<sub>4.44</sub>/Al<sub>4.10</sub> (sample 2):  $\hbar\omega_p$ Be and  $\hbar\omega_p$ Al – first order plasmon of Be and Al, respectively; 2 $\hbar\omega_p$ Be and 2 $\hbar\omega_p$ Al – second order plasmon of Be and Al, respectively;  $\hbar\omega_{ps}$ Al – surface plasmon of Al.

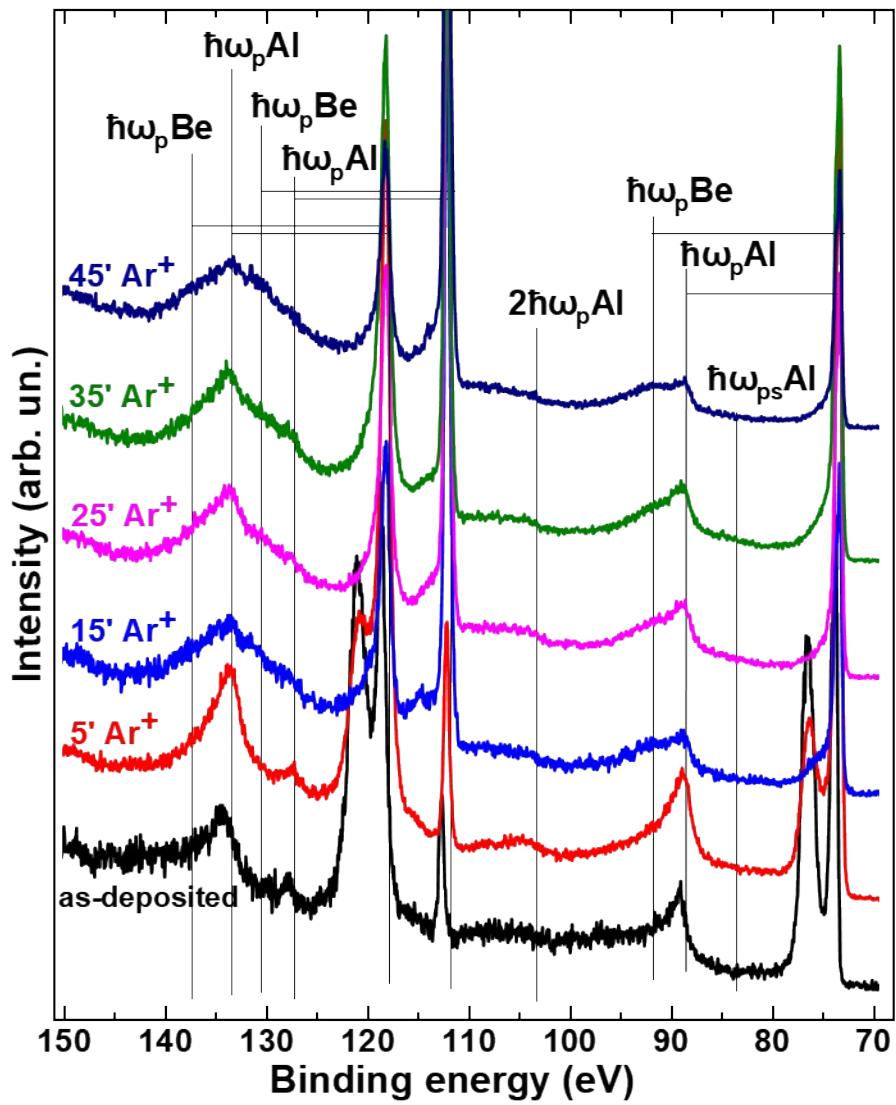
**Fig. 4-SI.** Plasmon shift in XPS spectra of Be<sub>1.62</sub>/Al<sub>6.90</sub> (sample 3):  $\hbar\omega_p$ Be and  $\hbar\omega_p$ Al – first order plasmon of Be and Al, respectively; 2 $\hbar\omega_p$ Be and 2 $\hbar\omega_p$ Al – second order plasmon of Be and Al, respectively;  $\hbar\omega_{ps}$ Al – surface plasmon of Al.



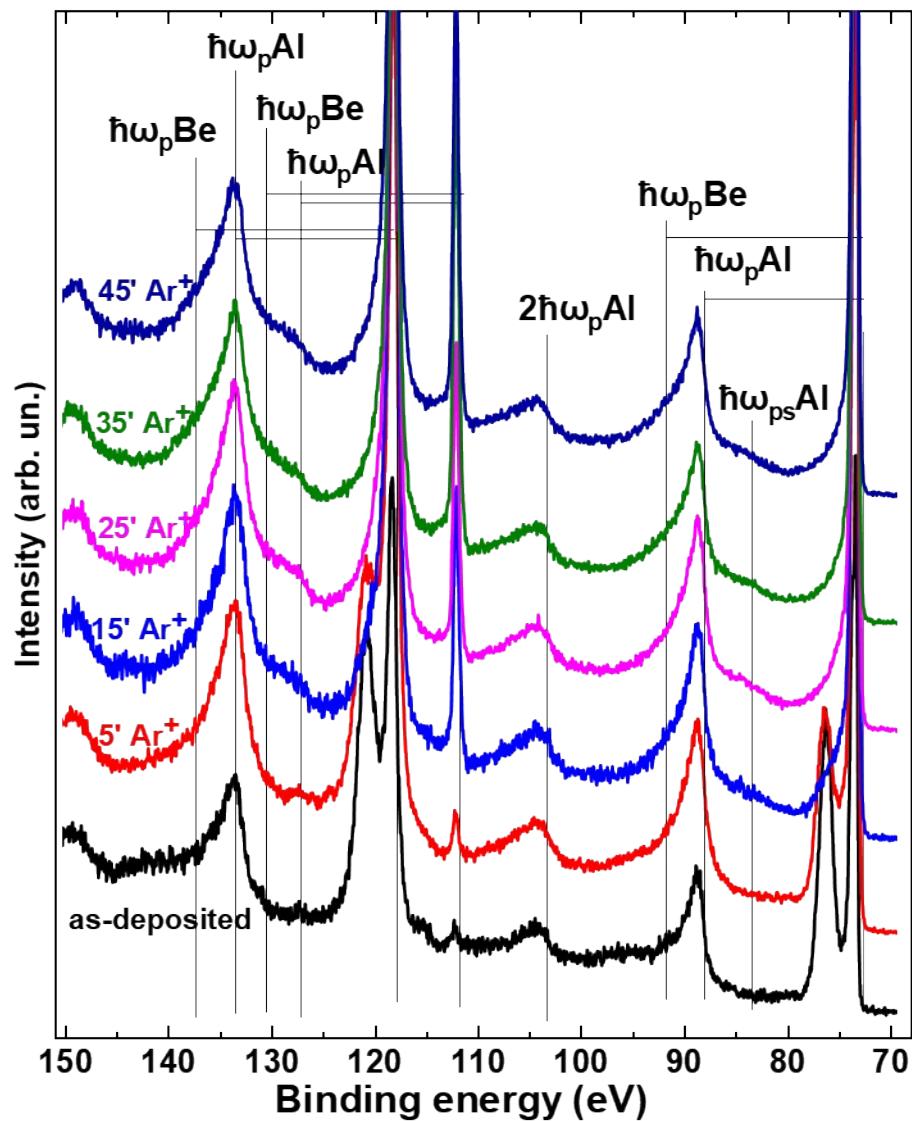
**Fig. 1-SI.** O 1s peak shift for as – deposited (0), and Ar<sup>+</sup> ion sputtered samples for 5, 15, 25, 35 and 45 minutes.



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**Fig. 3-SI.** Plasmon shift in XPS spectra of  $\text{Be}_{4.44}/\text{Al}_{4.10}$  (sample 2):  $\hbar\omega_p \text{Be}$  and  $\hbar\omega_p \text{Al}$  – first order plasmon of Be and Al, respectively;  $2\hbar\omega_p \text{Be}$  and  $2\hbar\omega_p \text{Al}$  – second order plasmon of Be and Al, respectively;  $\hbar\omega_{ps} \text{Al}$  – surface plasmon of Al.



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