Electronic Supplementary Information

Contribution of Lamellar Morphology on Photocatalytic Activity of Alkaline-Hydrothermally Treated Titania in Rhodamine B Photodegradation.

Fry V. Steky^{a,b}, Didi P. Benu^{a,b,c}, Kemal L. H. Putra^b, Muhamad N. Siddik^b, Damar R. Adhika^{d,e}, Rino R. Mukti^{b,d}, Brian Yuliarto^{d,e}, Irma Mulyani^b, Veinardi Suendo^{*,b,d}

^aDoctoral Program of Chemistry, Faculty of Mathematics and Natural Sciences, of Department of Chemistry, Institut Teknologi Bandung, Bandung 40132, Indonesia.

^bDivision of Inorganic and Physical Chemistry, Faculty of Mathematics and Natural Sciences, Institut Teknologi Bandung, Jl. Ganesha No. 10, Bandung 40132 Indonesia.

^eDepartment of Chemistry, Universitas Timor, Jl. Eltari, Kefamenanu 85613, Indonesia.

^dResearch Center for Nanosciences and Nanotechnology, Institut Teknologi Bandung, Jl. Ganesha No 10, Bandung 40132, Indonesia.

^eEngineering Physics Department, Faculty of Industrial Technology, Institut Teknologi Bandung, Jl. Ganesha No. 10, Bandung 40132 Indonesia.

S1. Particle Size Distribution



Figure S1. Particles Size Distribution of (a) S-AS, (b) S-HT, and (c) S-HT-500.

S2. Particle Size Distribution



Figure S2. XRD pattern of (a) S-500 and (b) S-500-HT. The Blue area represents the anatase phase, while the green area represents the amorphous phase.

S.3 Determining of {001} facet percentage

The interplanar spacing or *d*-spacing of the anatase crystal can be determined using Bragg's equation:

$$2d_{hkl}\sin\theta = n\lambda\tag{S1}$$

$$\sin\theta = \frac{n\lambda}{2\,d_{hkl}}\tag{S2}$$

The first-order reflection from (100) planes occurs at an angle given by:

$$\sin\theta \left(1^{st} \text{ order } 001\right) = \frac{1\lambda}{2\,d_{001}} \tag{S3}$$

The second-order reflection from the same set of planes then occurs at an angle:

$$\sin\theta \left(2^{nd} \operatorname{order} 001\right) = \frac{2\lambda}{2\,d_{001}} \tag{S4}$$

It is always referred to as the first-order reflection from (200) planes, i.e.

$$\sin\theta \left(1^{st} \text{ order } 002\right) = \frac{1\lambda}{2\,d_{002}} \tag{S5}$$

Similarly, the third and fourth-order reflection from (001) planes is at an angle:

$$\sin\theta (3^{rd} \text{ order } 001) = \frac{3\lambda}{2\,d_{001}} \equiv \sin\theta (1^{st} \text{ order } 003) = \frac{1\lambda}{2\,d_{003}}$$
 (S6)

$$\sin\theta \left(4^{rd} \text{ order } 001\right) = \frac{4\lambda}{2\,d_{001}} \equiv \sin\theta \left(1^{st} \text{ order } 004\right) = \frac{1\lambda}{2\,d_{004}} \tag{S7}$$

Note that the $d_{001} = 2d_{002} = 3d_{003} = 4d_{004}$, thus we can obtain the d_{001} value by knowing the d_{004} value. We have tabulated the calculation result of d_{hkl} spacing of anatase based on the XRD pattern in Table S3-1.

 d_{001} Samples d_{004} d_{200} d_{100} θ S-HT-500 2.3783 1.8921 9.5132 3.7842 68.3° S-500-HT 2.3764 1.8919 9.5056 3.7838 68.3° A-HT-500 2.3758 1.8934 9.5032 3.7868 68.3° A-500-HT 2.3799 1.8899 9.5196 3.7798 68.3° P25 Degussa 2.3810 1.8950 9.5240 3.7900 68.3°

Table S3-1. The unit cell parameters of samples calculated using Bragg's equation



Figure S3-1. The schematic of the anatase unit cell.

The crystal system of anatase is tetragonal (I41/amd) with unit cell $a = b \neq c$; $\alpha = \beta = \gamma = 90^{\circ}$, where $a = b = d_{100}$ and $c = d_{001}$. We know θ is the angle between the (101) and (001) planes, as shown in Figure S3-1, where:

$$\theta = \tan^{-1} \frac{d_{001}}{d_{100}} \tag{S8}$$

thus, using equation S8, we can calculate θ for each sample as tabulated in Table S3-1.

Since the crystallite consists of several unit cells, thus the ratio of d_{100} to d_{001} is equal to the ratio of the average crystallite size along [100] direction (D_{100}) to the average crystallite size along [001] direction (D_{001}), which can be written as:

$$\theta = \tan^{-1} \frac{d_{001}}{d_{100}} = \tan^{-1} \frac{D_{001}}{D_{100}} \cong 68.3^{\circ}$$
(S9)



Figure S3-2. (a) HRTEM images of the calcined sample show the square–bipyramidal shape of crystallites. (b) Projected crystallites to {010} plane are depicted as blue hexagons and at higher magnification, (c) a hexagonal shaped crystallite shows vertical lattice fringes representing the {004} interplanar distance. (d) The geometric scheme of a truncated square–bipyramidal shaped anatase crystallite.

a<mark>₊t</mark>b

{200}

The percentage of {001} facet can be calculated by knowing the crystallite truncated length (*a*). We use the geometry approach to calculate the *a* value since it has a relation with D_{100} and D_{001} . The average crystallite size (D_{hkl}) calculated using the Scherrer equation:

$$D_{hkl} = \frac{K \times \lambda}{B_{hkl} \times \cos \theta_{hkl}}$$
(S10)

Where D_{hkl} is the mean crystallite size in [hkl] direction, K is a dimensionless shape factor (in this case, we use K = 0.94), λ is the source X-ray wavelength (in this case, we use Cu $K_{\alpha} = 1.540$ Å), B_{hkl} is the line broadening at half the maximum intensity (FWHM) of (*hkl*) peak, θ_{hkl} is the Bragg angle.

As shown in Figure S3-2, we can calculate the values of *a* and *l* using the following equations:

$$a = D_{100} - \frac{D_{001}}{\tan \theta}$$
(S11)

$$l = \frac{D_{001}}{2\sin\theta} \tag{S12}$$

The percentage of {001} facet can be calculated by dividing the area of (001) plane to the crystallite total surface area:

$$A_{\{001\}} = a \times a = \left(D_{100} - \frac{D_{001}}{\tan\theta}\right)^2$$
(S13)

$$A_{\{101\}} = \left(\frac{D_{100} + a}{2}\right) \times l = \left(2D_{100} - \frac{D_{001}}{\tan\theta}\right) \times \frac{D_{001}}{2\sin\theta}$$
(S14)

%{001} facet =
$$\frac{\left(D_{100} - \frac{D_{001}}{\tan\theta}\right)^2}{\left(D_{100} - \frac{D_{001}}{\tan\theta}\right)^2 + \frac{2D_{001}}{\sin\theta}\left(2D_{100} - \frac{D_{001}}{\tan\theta}\right)} \times 100\%$$
(S15)



S4. The Rietveld refinement and the corresponding Williamson-Hall plot

Figure S4. The Rietveld refinement results and the corresponding Williamson-Hall plot of (a) S-HT-500, (b) S-500-HT, (c) A-HT-500, and (d) A-500-HT.

S5. N₂ physisorption results



Figure S5. N₂ physisorption isotherm of (a) S-500 vs. S-500-HT and (b) A-500 vs. A-500-HT.

S6. HRTEM images of each synthesized step



Figure S6. (a-c) TEM and (d-f) HRTEM images of samples: (a,d) as-synthesized, (b,e) hydrothermally-treated, and (c,f) calcined samples. The inset of (d-f) is the corresponding SAED pattern.

S7. Photodegradation of Rhodamine B



Figure S7. Evolution of PL spectra of rhodamine B with respect to the irradiation time and the corresponding first-order kinetic plots.