

## 1 Supporting Information

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### 3 Mechanistic study to reveal steric and electronic aspects involved 4 in the formation of microstructures during Pd catalyzed 5 olefin/divinyl formal copolymerization: Reactivity to catalyst 6 choice

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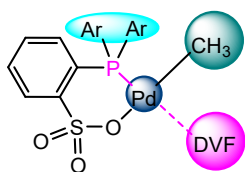
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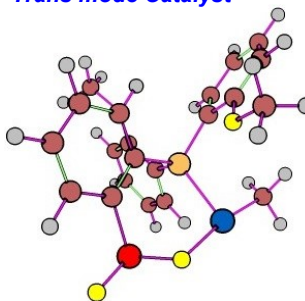
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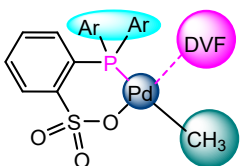
(a) *Trans mode insertion site  
for DVF in Catalyst (A)*



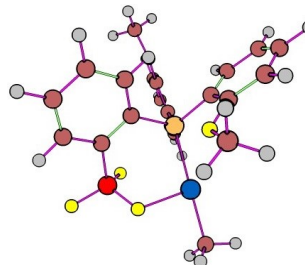
*Trans mode Catalyst*



(b) *Cis mode insertion site  
for DVF in Catalyst (A)*

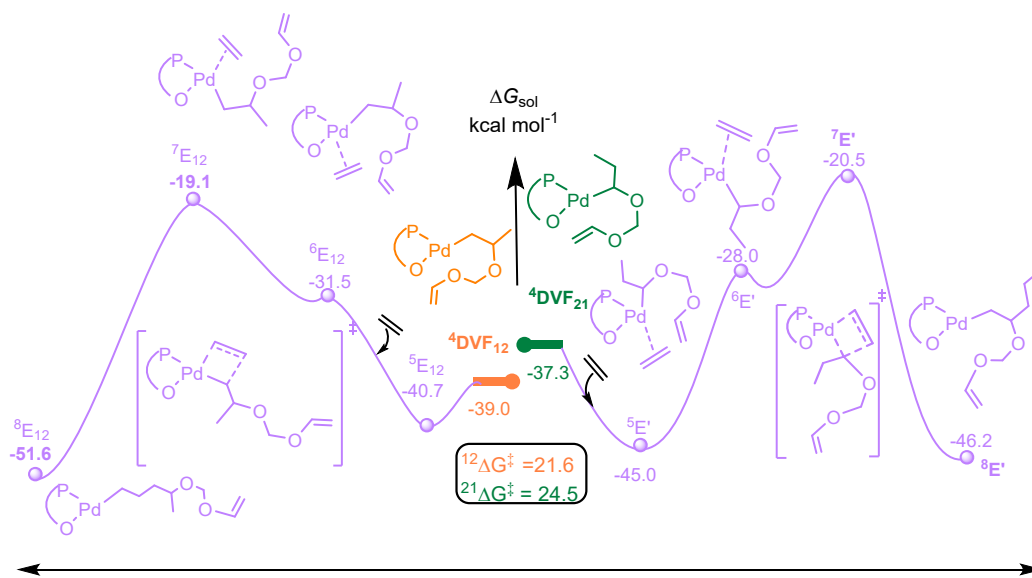


*Cis mode catalyst*



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17 **Fig S1.** Two coordination manners (trans and cis, Ar = 2-MeOC<sub>6</sub>H<sub>4</sub>) of the Ethylene and DVF correspond  
18 to the P-atom of phosphine-sulfonate-based Pd complexes.



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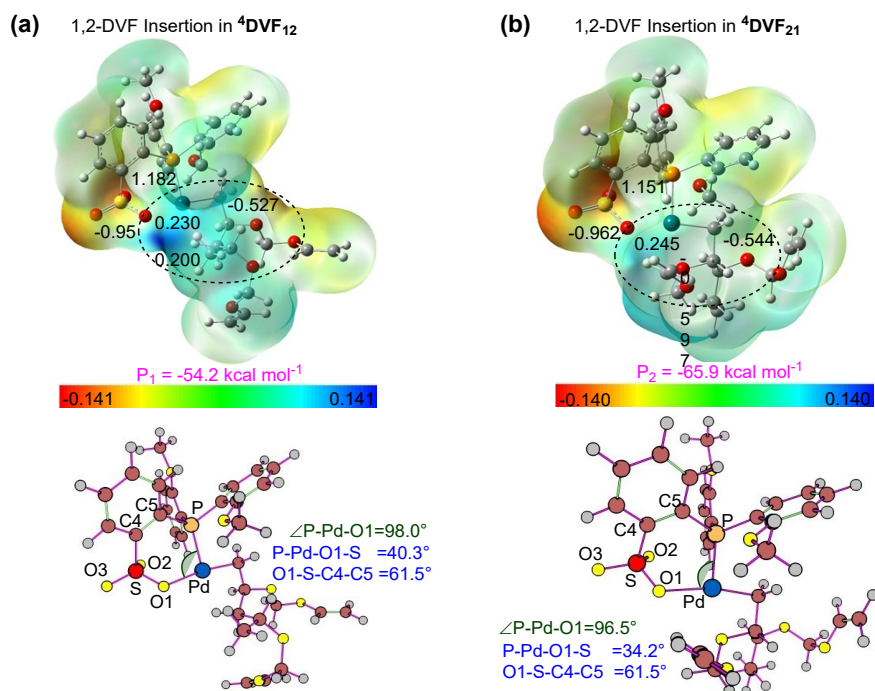
20 **Fig S2.** Energy profiles for chain propagation with E after 1,2/2,1-insertion of DVF in the 1,2/2,1-DVF pre-  
 21 insertion chains.

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23 **Table S1** Computed energies for copolymerization and homo-polymerization steps. All coordination  
 24 complexes (**C**<sup>1</sup>, **C**<sup>2</sup>, **C**<sup>3</sup>, and **C**<sup>4</sup>) and transition states (**TS**<sup>1</sup> and **TS**<sup>2</sup>) of DVF homo-polymers.

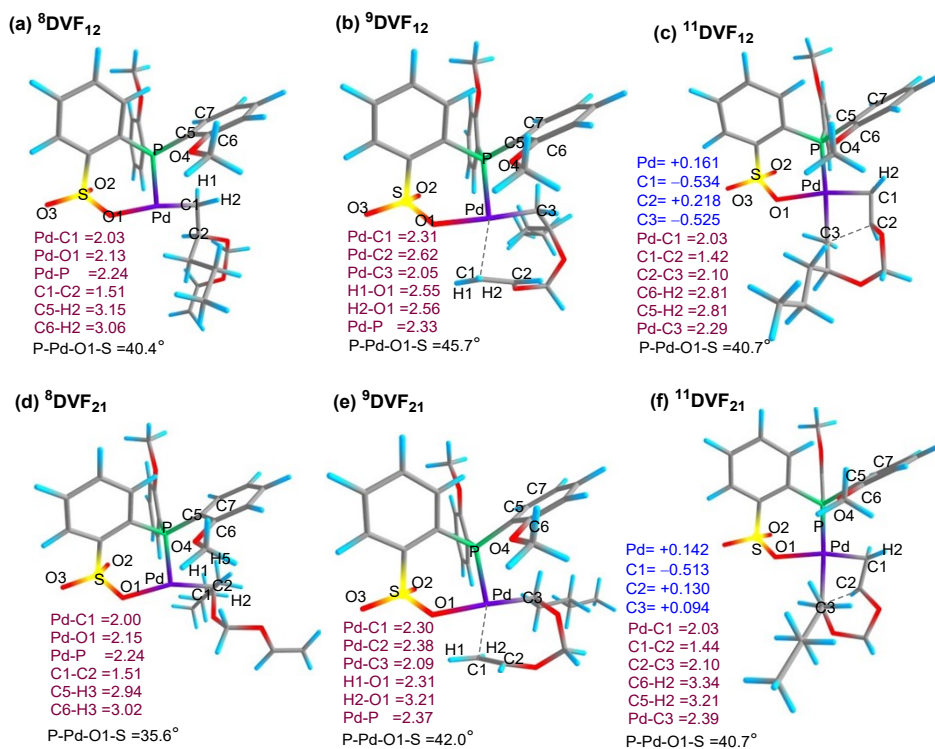
Monomer	E/1,2/2,1	E/1,2/2,1	E/1,2/2,	E/1,2/2,	E/1,2/2,1	E/1,2/2,	E/1,2/2,	E/1,2/2,	E/1,2/2,1	E/1,2/2,1
	<b>C</b> <sup>1</sup>	<b>C</b> <sup>2</sup>	<b>TS</b> <sup>1</sup>	<b>P</b> <sup>1</sup>	$\Delta G^{\ddagger}$	<b>C</b> <sup>3</sup>	<b>C</b> <sup>4</sup>	<b>TS</b> <sup>2</sup>	<b>P</b> <sup>2</sup>	$\Delta G^{2\ddagger}$
	<b>2<sup>nd</sup> insertion in <sup>4</sup>DVF<sub>12</sub></b>					<b>2<sup>nd</sup> insertion in <sup>4</sup>DVF<sub>21</sub></b>				
<b>1,2-DVF</b>	-54.4	-42.4	-32.2	-54.2	<b>22.2</b>	-55.7	-49.9	-32.6	-65.9	<b>23.1</b>
<b>2,1-DVF</b>	-61.9	-49.2	-30.4	-66.5	<b>31.5</b>	-59.5	-46.0	-26.4	-40.5	<b>33.1</b>

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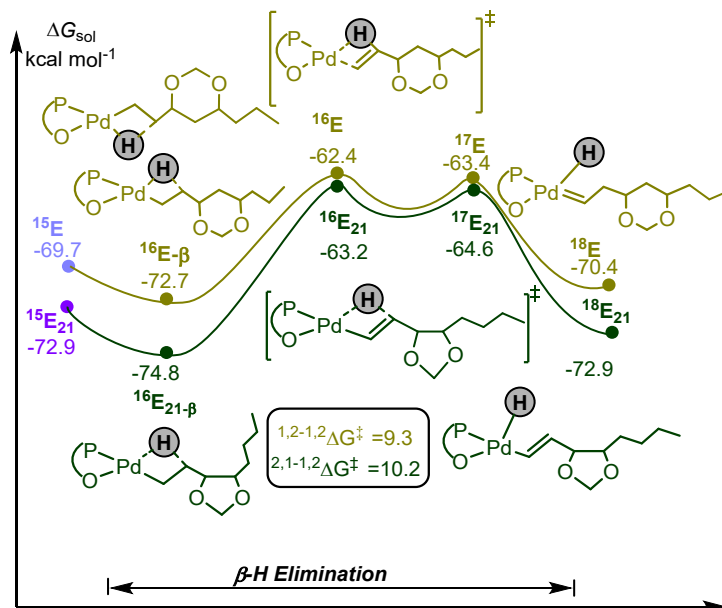
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27 **Fig S3.** Molecular electrostatic potential (MEP) formed by mapping of total density over the electrostatic  
 28 potential of products (a) 1,2-DVF Insertion in  ${}^4\text{DVF}_{12}$ , (b) 1,2-DVF Insertion in  ${}^4\text{DVF}_{21}$  geometric parameter  
 29 analysis. Energies in  $\text{kcal mol}^{-1}$ , and angles in degree ( $^\circ$ ). NBO charges are also mentioned on important  
 30 atoms.



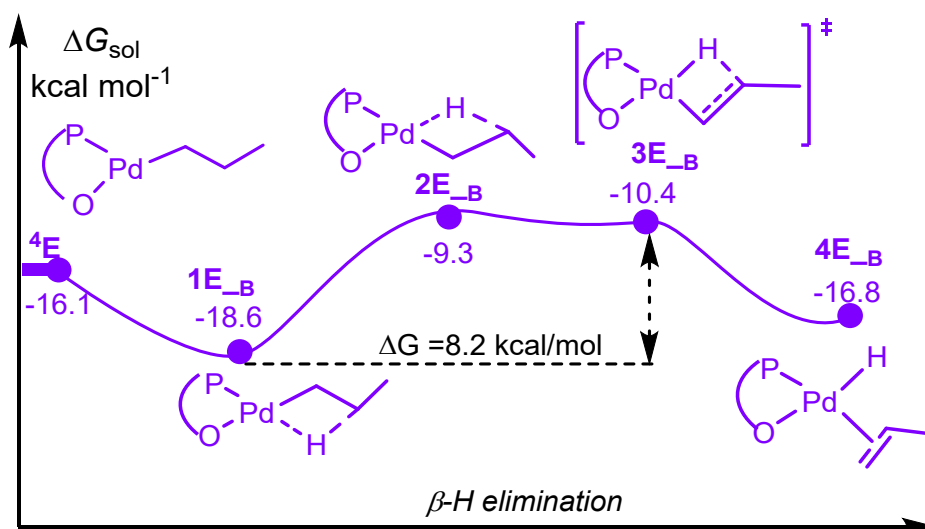
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32 **Fig S4.** Geometrical analysis of (a)  ${}^8\text{DVF}_{12}$ , (b)  ${}^9\text{DVF}_{12}$ , (c)  ${}^{11}\text{DVF}_{12}$ , (d)  ${}^8\text{DVF}_{21}$ , (e)  ${}^9\text{DVF}_{21}$ , and (f)  
 33  ${}^{11}\text{DVF}_{21}$ . Bond distances in Å and dihedral angles in degree ( $^\circ$ ). Hydrogen atoms of the catalyst's ligand  
 34 have been omitted for clarity.



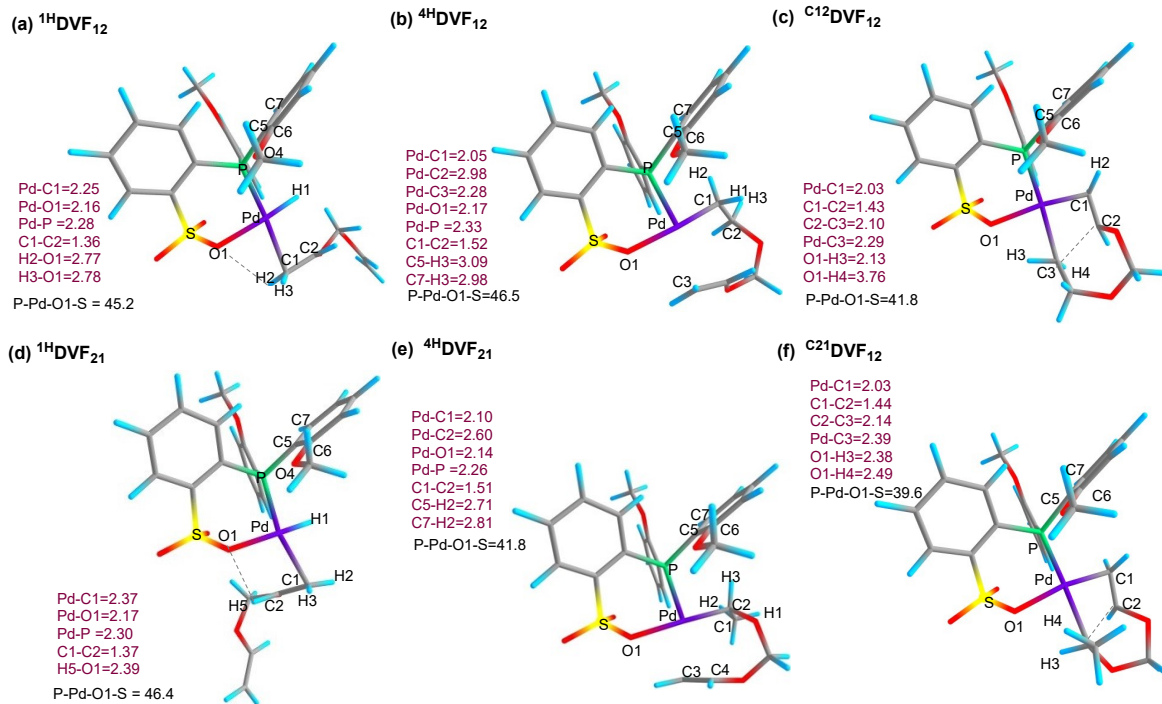
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**Fig S5.** Energy profiles for the  $\beta$ -hydride elimination after cyclization.



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**Fig S6.** Energy profiles for the  $\beta$ -hydride elimination after ethylene insertion.



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46 **Fig S7.** Geometrical analysis of (a)  $^1\text{H DVF}_{12}$ , (b)  $^4\text{H DVF}_{12}$ , (c)  $^{12}\text{C DVF}_{12}$ , (d)  $^1\text{H DVF}_{21}$ , (e)  $^4\text{H DVF}_{21}$ , and (f)  
47  $^{21}\text{C DVF}_{12}$ . Bond distances in Å and dihedral angles in degree ( $^\circ$ ). Hydrogen atoms of the catalyst's ligand  
48 have been omitted for clarity.