

Electronic Supplementary Material (ESI) for Physical Chemistry Chemical Physics.

^aInstitute of Theoretical and Computational Chemistry, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, China

^bKey Laboratory of Theoretical and Computational Photochemistry, Ministry of Education, College of Chemistry, Beijing Normal University, 19
Xin-Jie-Kou-Wai Street, Beijing 100875, China

*Corresponding author email address: Lianghui Gao (lhgao@bnu.edu.cn); Weihai Fang (fangwh@bnu.edu.cn)

Supplementary Material: A Top-down and Bottom-up Combined Strategy for Parameterization of Coarse-grained Force Fields for Phospholipids

Mingwei Wan,^{ab} Junjie Song,^b Ying Yang,^b Lianghui Gao,^{*b} and Weihai Fang^{*ab}

SI.1. OPT1.35 FF and its predictions.

Table SI-1. Bond length and non-bonded parameters.

| Pairs | L_b (nm) | ϵ (kJ/mol) | R_0 (nm) | α | β |
|---|------------|---------------------|------------|----------|---------|
| ^a C ₃ -C ₃ | 0.410 | 3.091 | 0.534 | 11.442 | 11.907 |
| ^a C ₄ -C ₄ | 0.450 | 4.357 | 0.578 | 12.798 | 14.254 |
| C ₅ -C ₅ | 0.490 | 4.820 | 0.613 | 14.965 | 12.754 |
| ^a C ₃ -C ₄ | 0.430 | 3.671 | 0.556 | 12.101 | 13.028 |
| C ₃ -C ₅ | 0.450 | 4.582 | 0.596 | 13.839 | 13.483 |
| C ₄ -C ₅ | 0.470 | 3.859 | 0.574 | 13.085 | 12.323 |
| ^b W _d -W _d | 0.5054 | 3.015 | 0.4969 | 6.249 | 9.253 |
| C ₃ -W _d | | 2.470 | 0.486 | 8.456 | 10.496 |
| ^b C ₄ -W _d | | 2.935 | 0.492 | 8.943 | 11.484 |
| C ₅ -W _d | | 3.017 | 0.503 | 9.670 | 10.863 |
| COC-COC | 0.364 | 4.033 | 0.512 | 11.392 | 15.557 |
| C ₃ -COC | 0.387 | 3.260 | 0.524 | 11.417 | 13.610 |
| C ₄ -COC | 0.407 | 3.844 | 0.546 | 12.075 | 14.891 |
| C ₅ -COC | 0.427 | 4.139 | 0.558 | 13.057 | 14.086 |
| W _d -COC | | 4.137 | 0.492 | 8.437 | 11.998 |
| NC ₃ -NC ₃ | 0.450 | 4.632 | 0.574 | 13.678 | 14.314 |
| C ₃ -NC ₃ | 0.430 | 3.671 | 0.540 | 12.510 | 13.055 |
| C ₄ -NC ₃ | 0.450 | 4.501 | 0.567 | 13.231 | 14.284 |
| C ₅ -NC ₃ | 0.470 | 4.696 | 0.581 | 14.307 | 13.512 |
| W _d -NC ₃ | | 4.444 | 0.478 | 9.245 | 11.509 |
| COC-NC ₃ | | 4.508 | 0.528 | 12.483 | 14.923 |
| COOC-COOC | 0.392 | 5.108 | 0.513 | 7.638 | 17.538 |
| C ₃ -COOC | 0.401 | 3.334 | 0.559 | 9.348 | 14.451 |
| C ₄ -COOC | 0.421 | 4.013 | 0.560 | 9.887 | 15.811 |
| C ₅ -COOC | 0.441 | 4.248 | 0.562 | 10.691 | 14.956 |
| W _d -COOC | | 4.045 | 0.500 | 6.909 | 12.739 |
| COC-COOC | | 4.446 | 0.514 | 9.328 | 16.518 |
| NC ₃ -COOC | | 4.890 | 0.532 | 10.221 | 15.844 |
| PO ₄ -PO ₄ | | 8.318 | 0.528 | 15.665 | 16.804 |
| C ₃ -PO ₄ | 0.370 | 3.995 | 0.522 | 13.388 | 14.145 |
| C ₄ -PO ₄ | | 4.533 | 0.526 | 14.159 | 15.477 |
| C ₅ -PO ₄ | | 4.721 | 0.526 | 15.311 | 14.640 |
| W _d -PO ₄ | | 6.197 | 0.512 | 9.894 | 12.469 |
| COC-PO ₄ | | 5.544 | 0.511 | 13.359 | 16.168 |
| NC ₃ -PO ₄ | | 6.183 | 0.488 | 14.638 | 15.509 |
| COOC-PO ₄ | | 7.012 | 0.529 | 10.938 | 17.167 |

^aParameters taken from ref. 1. ^bParameters taken from ref. 2.

Table SI-2. Thermodynamic properties of single-component systems.

| Molecule | ρ (kg/m ³) | ΔH_{vap} (kJ/mol) | Γ (mN/m) | ΔG_s (kJ/mol) |
|--|--------------------------------|-------------------------------------|--------------------|--------------------------|
| W _d -W _d | 999±4/997 | 45.9±0.2/44.0 | 69.4±2.0/72.0 | -26.4±1.0/-26.5 |
| C ₃ | 494±5/493 | 14.9±0.1/14.8 | 6.5±0.5/6.6 | -8.6±0.1/-8.3 |
| C ₄ | 573±4/573 | 21.2±0.2/21.0 | 12.1±0.7/11.9 | -11.8±0.1/-11.4 |
| C ₅ | 597±4/621 | 25.1±0.2/26.4 | 15.5±0.8/15.5 | -14.2±0.1/-14.2 |
| C ₃ -C ₄ | 664±4/682 | 36.4±0.3/36.6 | 19.5±0.9/19.8 | -20.0±0.1/-19.7 |
| C ₃ -C ₅ | 662±4/699 | 40.4±0.4/41.5 | 22.0±0.7/21.1 | -23.0±0.1/-22.2 |
| C ₄ -C ₅ | 680±4/715 | 45.5±0.4/46.6 | 22.4±0.8/22.4 | -25.5±0.2/-24.7 |
| COC | 652±4/655 | 18.7±0.1/18.5 | 11.3±0.7/11.4 | -10.1±0.1/-10.2 |
| COC-C ₃ | 704±3/739 | 32.6±0.3/32.4 | 19.3±0.9/19.5 | -17.8±0.1/-17.5 |
| COC-C ₄ | 723±3/756 | 38.0±0.3/36.9 | 21.5±0.8/21.5 | -20.5±0.1/-20.6 |
| C ₅ -COC-C ₅ | 754±3/790 | 66.9±0.5/63.5 | 26.9±0.9/25.4 | -36.9±0.2/-36.0 |
| NC ₃ | 599±4/629 | 22.8±0.2/24.0 | 14.6±0.7/13.4 | -12.4±0.1/-11.7 |
| NC ₃ -C ₃ | 683±3/716 | 37.0±0.3/34.4 | 20.0±0.8/19.2 | -20.1±0.1/-19.6 |
| NC ₃ -C ₄ | 702±3/732 | 39.2±0.3/39.1 | 20.3±0.8/20.8 | -23.6±0.1/-23.2 |
| NC ₃ -C ₅ | 713±3/744 | 43.2±0.4/43.8* | 21.9±0.8/21.5 | -25.8±0.2/-25.8 |
| ^a COOC-C ₃ | 825±3/890 | 38.2±0.4/40.2 | 23.7±0.8/24.6 | -22.0±0.1/-21.1 |
| ^a COOC-C ₄ | 840±3/883 | 42.4±0.4/44.1* | 24.1±0.9/25.7 | -24.3±0.2/-24.3 |
| ^a COOC-C ₅ | 845±3/878 | 45.4±0.4/48.1* | 24.7±0.9/26.6 | -26.6±0.2/-26.7 |
| ^b PO ₄ -C ₃ | 1231±4/1202 | 52.3±0.4/50.0 | 32.9±0.8/37.0 | -29.0±0.2/-30.4 |

Properties (left) in green and blue mean the deviation from experiment (right and black)³⁻⁵ is in the range of $\delta < 5.0\%$ and $\delta \geq 5.0\%$, respectively. *Properties experimentally unavailable are generated by extrapolating from the properties of the corresponding homologues.

^aReference properties are set as the average of the properties of COOC_x and C_xOOC with x equaling to 4, 5, or 6. ^bExperimental properties³⁻⁵ of trimethyl phosphate are set as reference.

Table SI-3. Properties of binary mixtures.

| Cross-term | ρ_{mix} (kg/m ³) | E_{int} (kJ/mol) | f_{pe} | ΔG_{hyd} or ΔG_{c} (kJ/mol) | Γ_{int} (mN/m) |
|--|---|------------------------------|-----------------|--|---------------------------------|
| C ₃ -W _d | 725±3/713 | -29.7±0.3/-29.6 | 1.5% | hyd: 7.7±0.3/8.2 | 43.6±0.6/48.5 |
| C ₄ -W _d | 752±3/738 | -36.6±0.3/-36.7 | 1.5% | hyd: 8.1±0.3/8.7 | 47.8±0.3/49.4 |
| C ₅ -W _d | 742±3/753 | -42.9±0.4/-43.7 | 3.6% | hyd: 9.7±0.4/9.8 | 49.6±0.5/50.0 |
| COC-C ₃ | 551±4/549 | -23.2±0.2/-23.1 | 0.4% | | |
| COC-C ₄ | 594±4/593 | -27.6±0.3/-27.1 | 2.0% | C ₁₆ : -4.4±0.2/-6.2 | |
| COC-C ₅ | 622±4/626 | -28.5±0.3/-29.1 | 3.9% | | |
| COC-W _d | 801±3/870 | -54.7±0.5/-57.4 | 8.8% | hyd: -7.9±0.4/-8.0 | |
| NC ₃ -C ₃ | 572±4/601 | -29.9±0.3/-32.1 | 5.3% | | |
| NC ₃ -C ₄ | 602±4/631 | -37.1±0.4/-37.0 | 5.8% | C ₁₆ : -9.7±0.5/-9.25 | |
| NC ₃ -C ₅ | 620±4/651 | -37.5±0.4/-39.8 | 3.3% | | |
| NC ₃ -W _d | 896±4/860 | -68.6±0.7/-69.8 | 8.0% | hyd: -12.1±0.6/-13.4 | |
| NC ₃ -COC | 658±4/691 | -40.9±0.4/-43.5 | 3.7% | | |
| COOC-C ₃ | 683±4/715 | | 3.5% | | |
| COOC-C ₄ | 727±3/738 | | 3.3% | C ₁₆ : -7.6±0.4/-8.3 | |
| COOC-C ₅ | 743±3/752 | | 4.4% | | |
| COOC-W _d | 973±5/996 | | 4.4% | hyd: -11.9±0.5/-11.6 | |
| COOC-COC | 844±4/847 | | 4.1% | | |
| COOC-NC ₃ | 801±3/839 | | 2.7% | | |
| ^a C ₃ PO ₄ -C ₃ | 794±3/807 | | 11.0% | C ₆ : -21.0±0.9/-24.4 | |
| ^a C ₃ PO ₄ -C ₄ | 812±3/825 | | 8.0% | C ₇ : -19.6±0.8/-23.4 | |
| ^a C ₃ PO ₄ -C ₅ | 804±3/832 | | 5.8% | C ₈ : -19.2±0.9/-22.4* | |
| ^a C ₃ PO ₄ -W _d | 1085±5/1133 | | 3.9% | hyd: -35.0±1.5/-36.4 | |
| ^a C ₃ PO ₄ -COC | 921±5/962 | | 5.1% | | |
| ^a C ₃ PO ₄ -NC ₃ | 907±5/933 | | 5.8% | | |
| ^a C ₃ PO ₄ -COOC | 1099±5/1132 | | 2.9% | | |

CG Properties (left) in green and blue mean the deviation from the reference data (right) is in the range of $\delta < 5.0\%$ and $\delta \geq 5.0\%$. Here, the reference ρ_{mix} , E_{int} , and f_{pe} are obtained from AA MD simulations by using the OPLS FF.⁶ Free energy and interfacial tension are taken from experimental materials.⁷⁻⁸ *Properties experimentally unavailable are generated by extrapolating from the properties of the corresponding homologues. ^aThe C₃PO₄ group is represented by C₃-PO₄ in CG MD simulations and trimethyl phosphate in the case of AAMD simulations and experiment.

Table SI-4. The bonded parameters of DMPC and DPPC lipids.

| Bond | L_b (nm) | k_b (kJ/mol/nm ²) | Angle | θ_a (degree) | k_a (kJ/mol) |
|-------|---------------|------------------------------------|----------|------------------------|-------------------|
| DMPC | | | | | |
| 1-2 | 0.470 | 1250 | 1-2-3 | 140 | 7.0 |
| 2-3 | 0.390 | 1250 | 2-3-4 | 135 | 7.0 |
| 3-4 | 0.340 | 1250 | 2-3-5 | 180 | 7.0 |
| 3-5 | 0.430 | 1250 | 3-5-6 | 180 | 11.0 |
| 5-6 | 0.520 | 1250 | 5-6-7 | 180 | 11.0 |
| 6-7 | 0.470 | 1250 | 4-8-9 | 180 | 11.0 |
| 4-8 | 0.430 | 1250 | 8-9-10 | 180 | 11.0 |
| 8-9 | 0.520 | 1250 | | | |
| 9-10 | 0.470 | 1250 | | | |
| DPPC | | | | | |
| 1-2 | 0.470 | 1250 | 1-2-3 | 140 | 7.0 |
| 2-3 | 0.390 | 1250 | 2-3-4 | 135 | 7.0 |
| 3-4 | 0.340 | 1250 | 2-3-5 | 180 | 7.0 |
| 3-5 | 0.430 | 1250 | 3-5-6 | 180 | 11.0 |
| 5-6 | 0.520 | 1250 | 5-6-7 | 180 | 11.0 |
| 6-7 | 0.420 | 1250 | 6-7-8 | 180 | 11.0 |
| 7-8 | 0.360 | 1250 | 4-9-10 | 180 | 11.0 |
| 4-9 | 0.430 | 1250 | 9-10-11 | 180 | 11.0 |
| 9-10 | 0.520 | 1250 | 10-11-12 | 180 | 11.0 |
| 10-11 | 0.420 | 1250 | | | |
| 11-12 | 0.360 | 1250 | | | |

SI.2. Meta-multilinear interpolation parameterization (meta-MIP)

The meta-MIP algorithm² was applied here to optimize parameters. It constructs local databases in an on-the-fly pattern, the connection of which leads to an optimal path, thus ensuring a high parameterization efficiency. The basic procedure is introduced as followings.

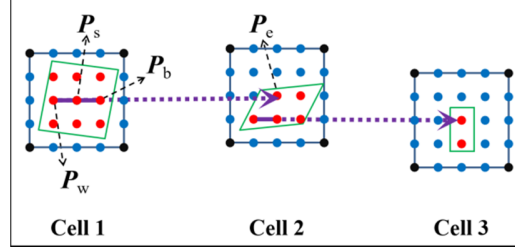


Fig. SI-1. Schematic diagram of meta-MIP.

Step 1. Starting from an initial parameter set P_s (Figure SI-1), Cell 1 encircling P_s is constructed. $\{\rho, \Delta H_{vap}, \Gamma\}$ at 2^d vertexes (black points) of Cell 1 are evaluated by performing simulations.

Step 2. The resolution inside Cell 1 is raised by multilinear interpolation via

$$f(P_x) = \sum_{n=1,2^d} \left(\left(\prod_{m=1,d} w_{n,m} \right) \times f(P_n) \right), w_{n,m} = \frac{p_{n+1,m} - p_{x,m}}{p_{n+1,m} - p_{n,m}}. \quad (SI-1)$$

Here, $f(P)$ is the property of a d -dimensional P ($d=2$ in Figure SI-1). The value of $f(P_x)$ is predicted by learning from the 2^d neighbors P_n . The weighting factor $w_{n,m}$ stands for the degree of deviation of P_x from P_{n+1} . $p_{x,m}$, $p_{n,m}$, and $p_{n+1,m}$ is the component of P_x , P_n , and P_{n+1} in the m th direction, respectively.

The high-resolution version of database Cell 1 is obtained (blue and red points). At each parameter point, the absolute deviations of $\{\rho, \Delta H_{vap}, \Gamma\}$ in relative to experimental values are calculated as $\{\delta_1, \delta_2, \delta_3\}$.

Step 3. The interpolated parameter sets are screened (red points) by defining a screening condition (SC) condition. For instance, $SC = \{\delta_1 < 200\%, \delta_2 < 200\%, \delta_3 < 200\%\}$ restricts all three properties loosely; $SC = \{\delta_1 < 5\%, \delta_2 < 5\%, \delta_3 < 200\%\}$ reserves accuracy for ρ and ΔH_{vap} ; $SC = \{\delta_1 < 5\%, \delta_2 < 5\%, \delta_3 < 5\%\}$ reserves accuracy for all three properties.

Step 4. The screened parameter sets are ranked by defining a ranking condition (RC). For instance, $RC = \{\text{decreasing } \delta_a = (\delta_1 + \delta_2 + \delta_3)/3\}$ outputs the parameter sets in the ascending order of δ_a ; $RC = \{\text{decreasing } \delta_3\}$ outputs the parameter sets in the ascending order of δ_3 ; $RC = \{\text{increasing } \varepsilon\}$ outputs the parameter sets in the descending order of ε . Afterwards, the worst/best point P_w/P_b is obtained as the last/first ranked parameter set. Then, the optimal path in Cell 1 (purple solid line) is roughly obtained as the vector from P_w to P_b .

Step 5. The optimization moves to the next cell (P_e) by extrapolating the optimal path of Cell 1 (purple dash line). This is reasonable, because the optimal path is roughly linear in local areas, and further screening and ranking could adjust the path direction. An extrapolation factor $EF = |P_e - P_b| / |P_b - P_w|$ (suggested value of 0 to 1) is set to control the speed and stability of optimization.

Exit condition. If the predicted properties of P_b from interpolation are satisfying, further MD simulations are performed to test this point. If the simulated properties of P_b (or one of its neighbouring points) are satisfying, the optimization program exits.

Steps 1 to 5 are iterated until an exit criterion is met in Step 5 or an unphysical error (such as crystallization) occurs in any step. A detailed illustration for the usage of meta-MIP could be found in our previous works.^{2,9}

SI.3. Properties of Membranes

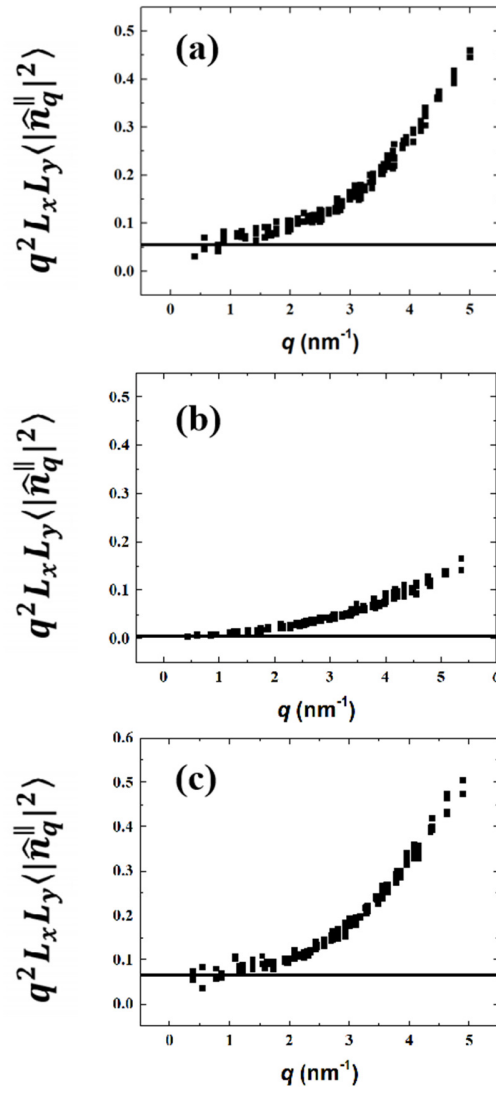


Fig. SI-2. Fluctuation spectra for (a) DMPC membrane at 303 K, (b) DPPC membrane at 303 K, and (c) DPPC membrane at 323 K.

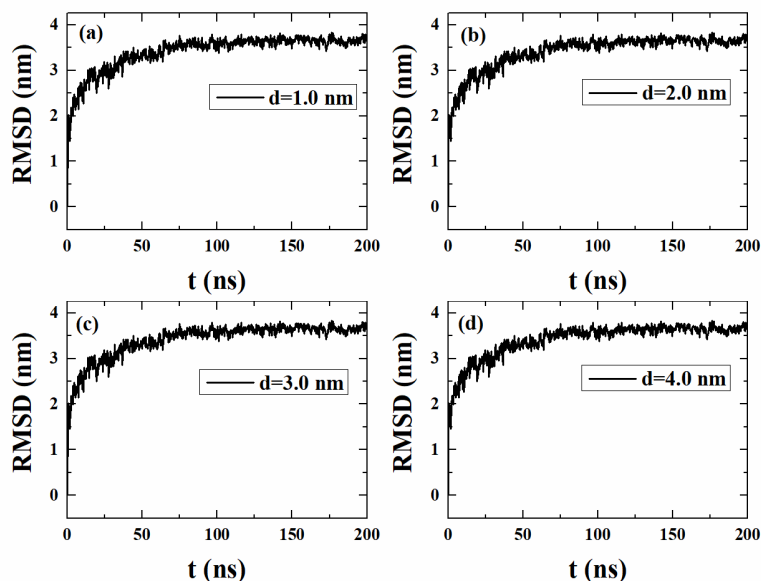


Fig. SI-3. Root mean-square deviation throughout a DMPC membrane at typical umbrella sampling windows where the normal position of a tagged lipid is constrained.

References

1. M. Wan, J. Song, W. Li, L. Gao, W. Fang, Development of Coarse-Grained Force Field by Combining Multilinear Interpolation Technique and Simplex Algorithm. *J. Comput. Chem.* **2020**, *41*, 814-829.
2. M. Wan, J. Song, Y. Yang, L. Gao, W. Fang, Development of coarse-grained force field for alcohols: an efficient meta-multilinear interpolation parameterization algorithm. *Phys. Chem. Chem. Phys.* **2021**, *23*, 1956-1966.
3. J. Rumble, *CRC Handbook of Chemistry and Physics*. 97 ed.; CRC press: Boca Raton, 2017.
4. C. L. Yaws, *Thermophysical properties of chemicals and hydrocarbons*. 2 ed.; William Andrew: Beaumont, 2014.
5. C. L. Yaws, *The Yaws handbook of vapor pressure: Antoine coefficients*. 2 ed.; Gulf Professional Publishing: Houston, 2015.
6. M. J. Robertson, J. Tirado-Rives, W. L. Jorgensen, Improved Peptide and Protein Torsional Energetics with the OPLS-AA Force Field. *J. Chem. Theory Comput.* **2015**, *11* (7), 3499-3509.
7. S. Cabani, P. Gianni, V. Mollica, L. Lepori Group Contributions to the Thermodynamic Properties of Non-Ionic Organic Solutes in Dilute Aqueous Solution. *J. Solution Chem.* **1981**, *10* (8), 563-595.
8. S. Zeppieri, J. Rodríguez, Interfacial Tension of Alkane + Water Systems. *J. Chem. Eng. Data* **2001**, *46* (5), 1086-1088.
9. J. Song, M. Wan, Y. Yang, L. Gao, W. Fang, Development of Accurate Coarse-Grained Force Fields for Weakly Polar Groups by an Indirect Parameterization Strategy. *Phys. Chem. Chem. Phys.* **2021**, *23*, 6763-6774.