

Electronic Supplementary Information (ESI)

Grape bunches of novel conjugated chain bonded fullerene oligomers: Design of a potential electron trap carbonaceous molecular material

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Content of the Electronic Supplementary Information (ESI):

Fig. S1 Key bond length parameters (in Å) of the five **CBFOs** structures at B3LYP/6-31G(d), B3LYP/6-31+G(d) (in brackets) and B3LYP-GD3BJ/def2-tzvp (in italic)

Fig. S2 Kinetic study of geometries in Path 2 and the change in the reaction Gibbs free energies for **TDD-3-carbene (L-P3)**.

Fig. S3 Kinetic study of geometries in Path 3 and the change in the reaction Gibbs free energies for **TDD-3-carbene (LS-P3)**.

Fig. S4 Kinetic study of geometries in Path 4 and the change in the reaction Gibbs free energies for **TDD-3-carbene (SL-P3)**.

Fig. S5 Relative electrophilic index (*I*) of **D** and **CBFOs** for EA_{a-2s} (singlet) and EA_{v-2s} (singlet).

Fig. S6 Highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) and LUMO+1 of C₆₀, **D** and **T2c**

Fig. S7 Density of states (DOS) of **D**, **T1t**, **T2t** and **T3**. The brown lines represent the Fermi levels, defined as the midpoint between the highest occupied and lowest unoccupied

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states.

Fig. S8 Energetical band structure of **2D-T3** along high symmetry points, and the density of states, the partial density states of *p* orbitals of carbon atoms

Fig. S9 Electron localization function (ELF) for (a) three dimension and (b) two-dimension plot of **2D-T3**.

Fig. S10 UV-vis spectrum of the five **CBFOs** under PBE0/6-31G(d) theoretical level

Fig. S11 Relationship between absorption index and radiation energy for **D**, **CBFOs** and **2D-T3**.

Table S1 The energies of *cis* and *trans* configurations of **TDD-1** and **TDD-2**

Table S2 The energies of *cis* and *trans* configurations of **T1c**, **T1t**, **T2c** and **T2t**

Table S3 The first vertical electron affinity (EA_{v1} , eV) of C_{60} , dimer (**D**) and five **CBFOs** at B3LYP/6-31G(d) and B3LYP/6-31+G(d) levels of theory

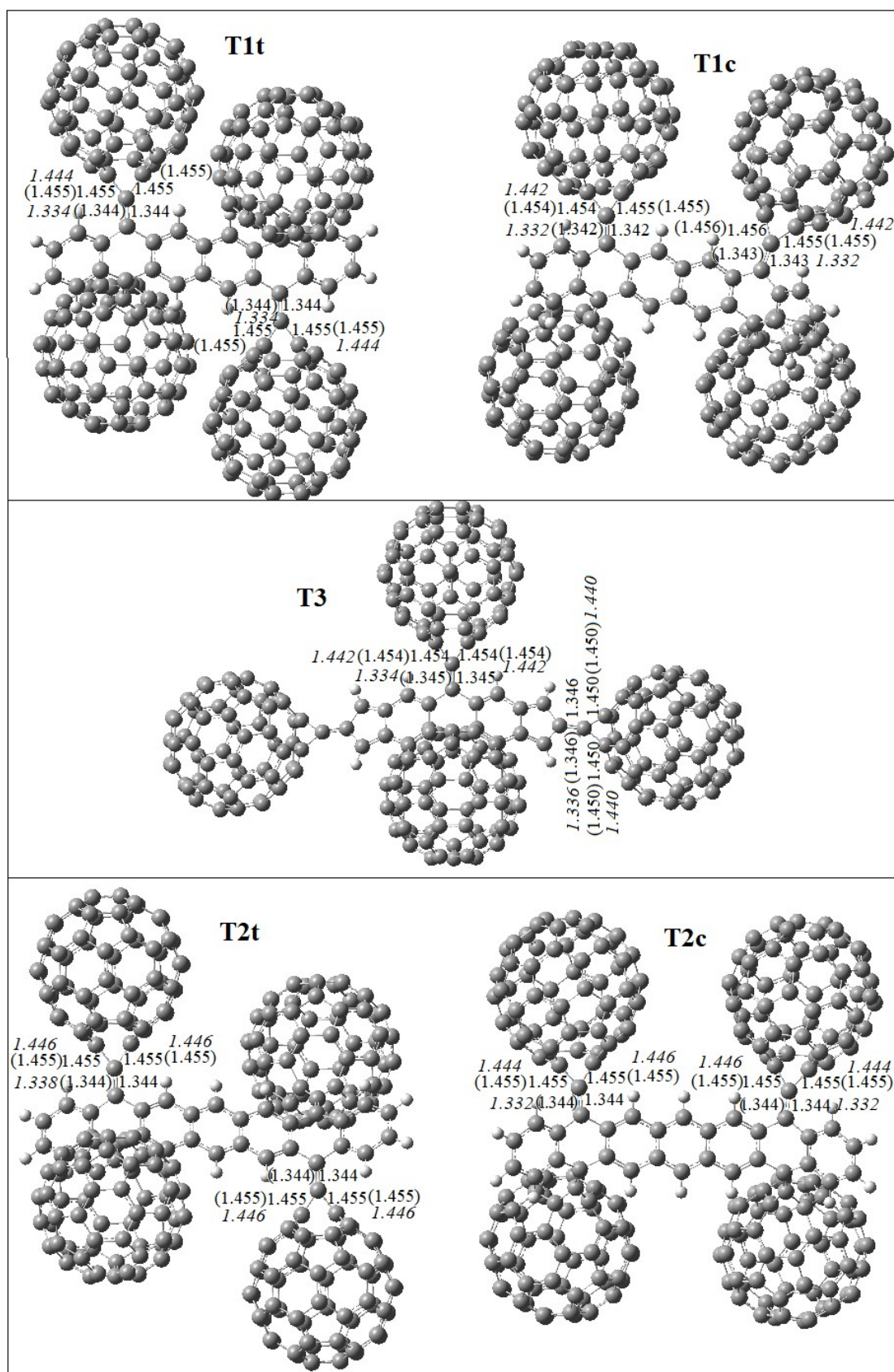


Fig. S1 Key bond length parameters (in Å) of the five **CBFOs** structures at B3LYP/6-31G(d), B3LYP/6-31+G(d) (in brackets) and B3LYP-GD3BJ/def2-tzvp (in italic)

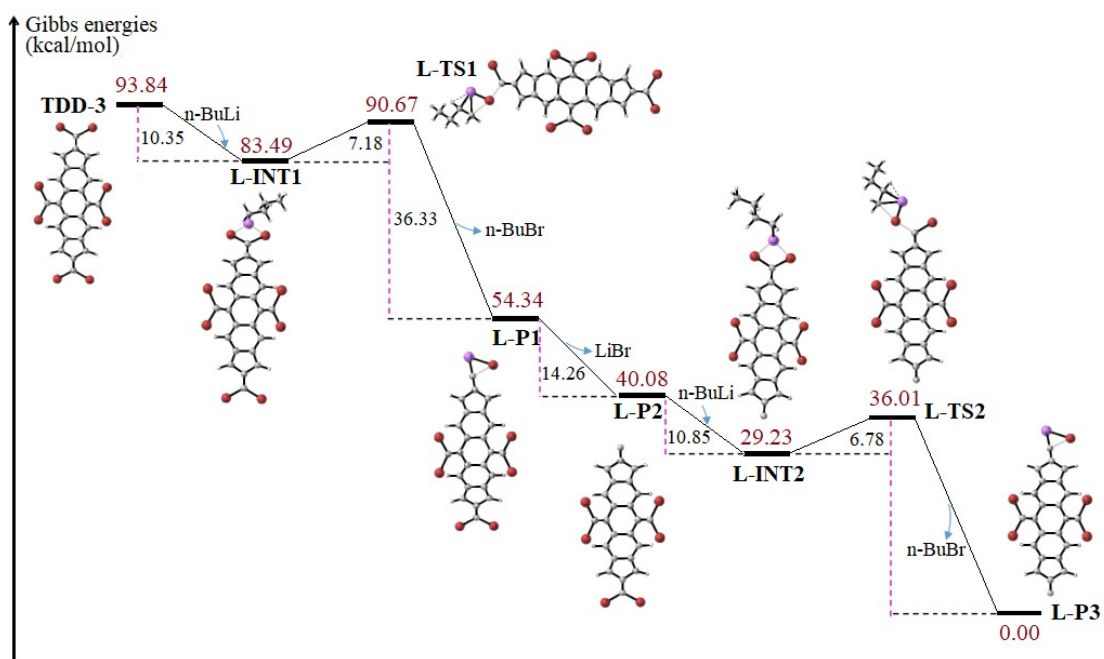


Fig. S2 Kinetic study of geometries in Path 2 and the change in the reaction Gibbs free energies for TDD-3-carbene (L-P3).

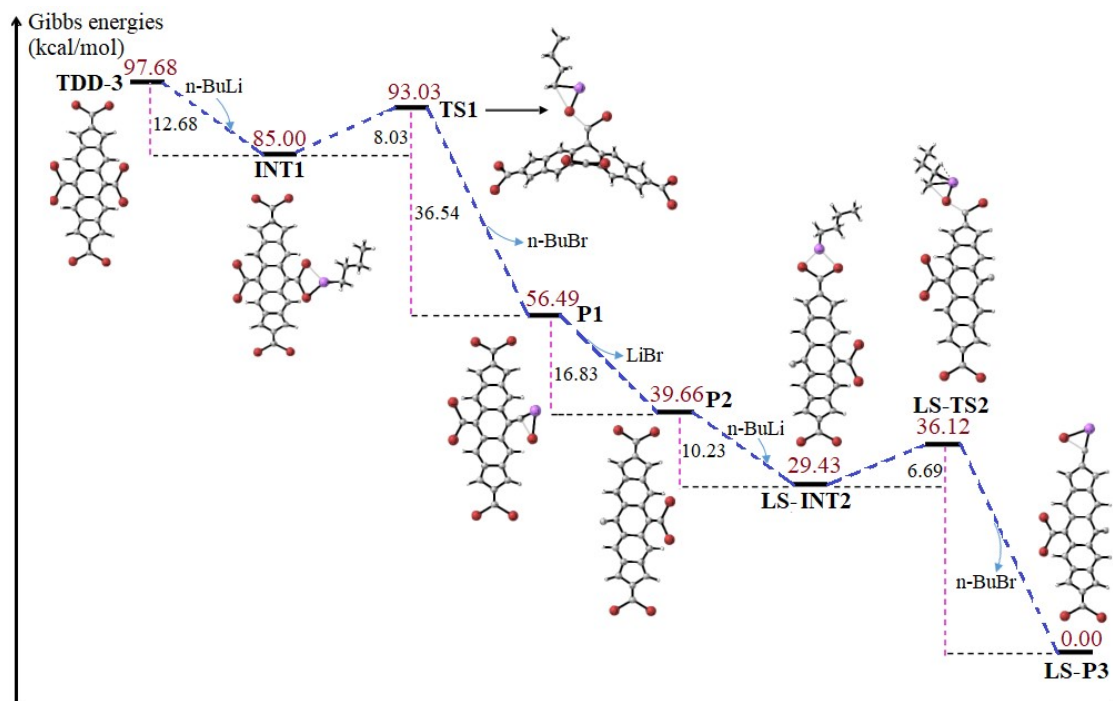


Fig. S3 Kinetic study of geometries in Path 3 and the change in the reaction Gibbs free energies for TDD-3-carbene (LS-P3).

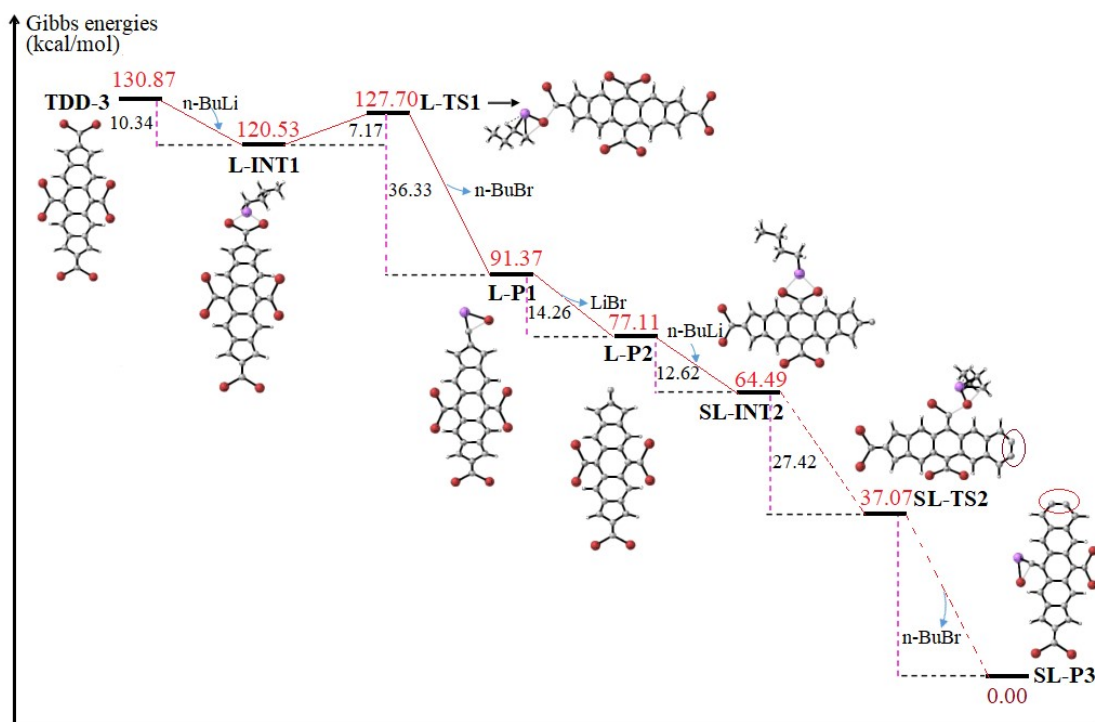


Fig. S4 Kinetic study of geometries in Path 4 and the change in the reaction Gibbs free energies for phenylalkyne intermediate (SL-P3).

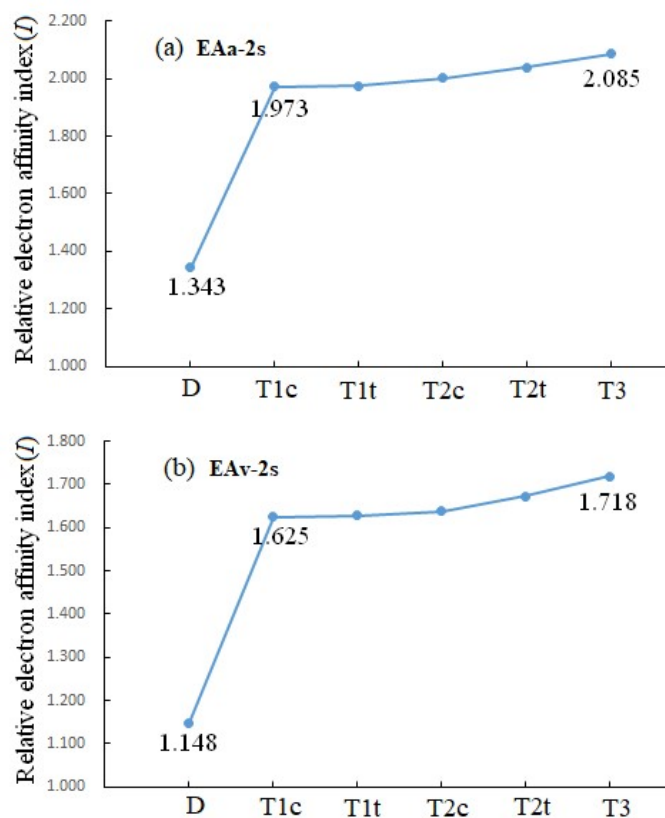


Fig. S5 Relative electrophilic index (*I*) of **D** and **CBFOs** for EA_{a-2s} (singlet) and EA_{v-2s} (singlet).

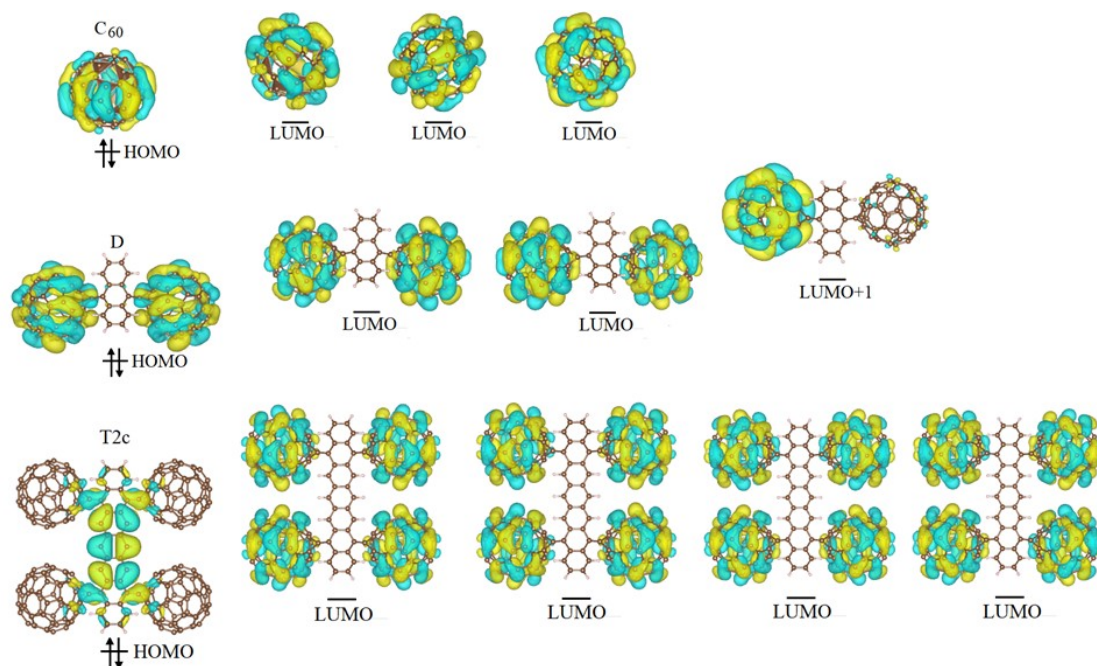


Fig. S6 Highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) and LUMO+1 of C₆₀, **D** and **T2c**.

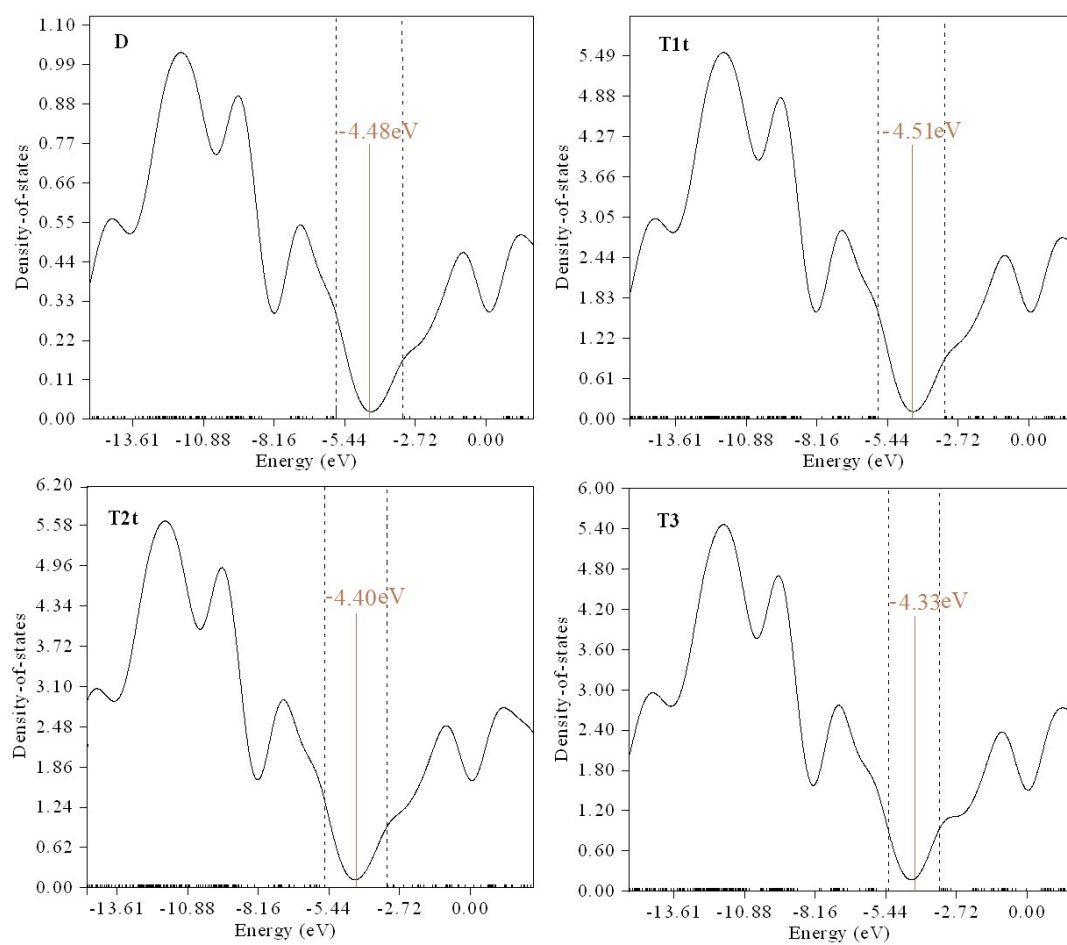


Fig. S7 Density of states (DOS) of **D**, **T1t**, **T2t** and **T3**. The brown lines represent the Fermi levels, defined as the midpoint between the highest occupied and lowest unoccupied states.

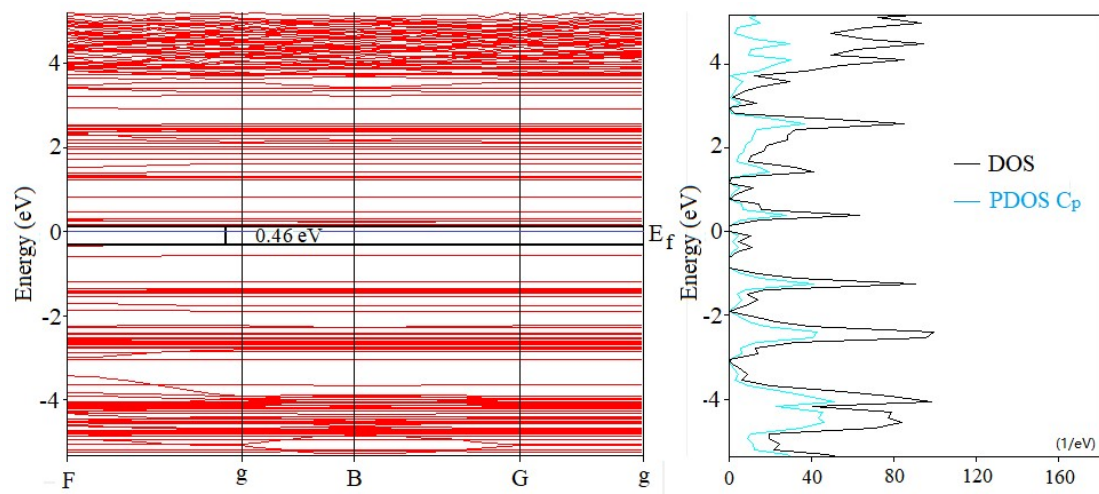


Fig. S8 Energetical band structure of **2D-T3** along high symmetry points, and the density of states, the partial density states of p orbitals of carbon atoms.

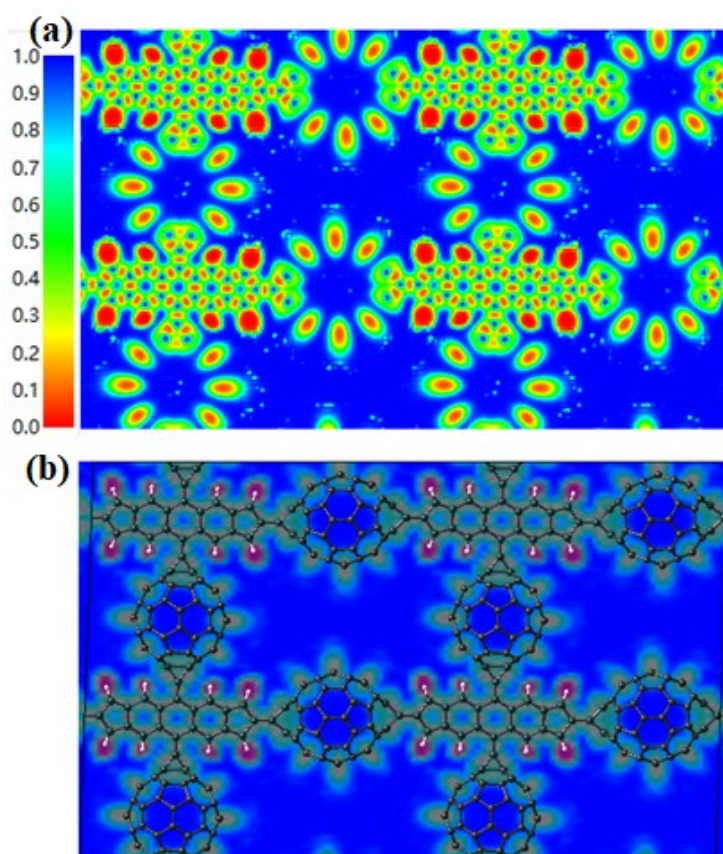


Fig. S9 Electron localization function (ELF) of (a) three-dimension and (b) two-dimension plot for **2D-T3**.

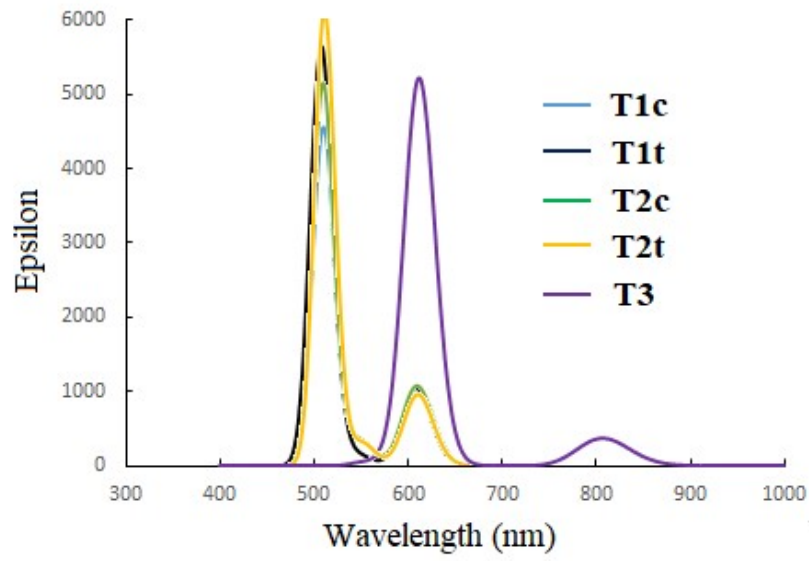


Fig. S10 UV-vis spectrum of the five CBFOs under PBE0/6-31G(d) theoretical level.

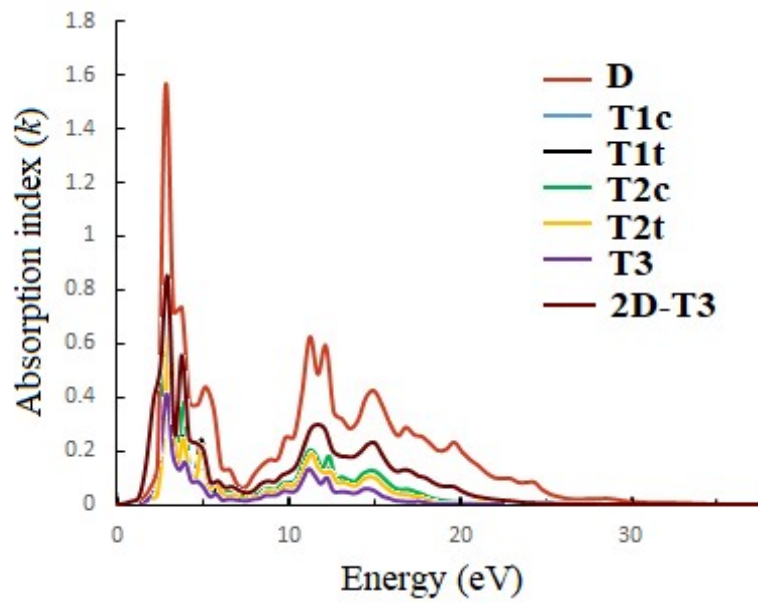


Fig S11 Relationship between absorption index and radiation energy for D, CBFOs and 2D-T3.

Table S1 The energies of *cis* and *trans* configurations of **TDD-1** and **TDD-2**

B3LYP/6-31G(d)					
Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹	Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹
<i>cis</i> - TDD-1	-21743.76962	0.07	<i>cis</i> - TDD-2	-21897.44201	0.01
<i>trans</i> - TDD-1	-21743.76974	0	<i>trans</i> - TDD-2	-21897.44199	0
B3LYP-GD3BJ/def2-tzvp					
Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹	Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹
<i>cis</i> - TDD-1	-21744.38516	0	<i>cis</i> - TDD-2	-21898.09332	0.031
<i>trans</i> - TDD-1	-21744.38512	0.025	<i>trans</i> - TDD-2	-21898.09337	0

Table S2 The energies of *cis* and *trans* configurations of **T1c**, **T1t**, **T2c** and **T2t**

B3LYP/6-31G(d)					
Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹	Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹
T1c	-10295.012129	1.87	T2c	-10448.652896	0
T1t	-10295.015137	0	T2t	-10448.652851	0.03
B3LYP/6-31+G(d)					
Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹	Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹
T1c	-10295.231754	2.06	T2c	-10448.876825	0
T1t	-10295.235037	0	T2t	-10448.876049	0.48
B3LYP-D3BJ/def2-tzvp					
Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹	Configurations	Energy/a.u.	Relative energy kcal·mol ⁻¹
T1c	-10299.690314	0.13	T2c	-10453.395411	0
T1t	-10299.690520	0	T2t	-10453.391247	2.61

Table S3 The first vertical electron affinity (EA_{v1}, eV) of C₆₀, dimer (**D**) and five **CBFOs** at B3LYP/6-31G(d) and B3LYP/6-31+G(d) levels of theory

	C ₆₀	D	T1c	T1t	T2c	T2t	T3
EA _{v1} (6-31+g(d))	-2.52	-2.86	-3.07	-3.08	-3.07	-3.1	-3.16
EA _{v1} (6-31g(d))	-1.99	-2.38	-2.61	-2.61	-2.61	-2.63	-2.71
ΔEA _{v1}	0.53	0.48	0.46	0.47	0.46	0.47	0.45

ΔEA_{v1} is the difference between two levels of theory for the corresponding structure