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Table S1: Fluorescence decay parameters for NCS1 and 1,8-ANS: NCS1 complex in the presence/absence of								
NCC1	- (f(0/)	- ($\int (0/)$	- ($\int (0/)$		2
NCSI	τ_1 (ns)	$J_{1}(\%)$	τ_2 (ns)	$J_{2}(\%)$	τ_{3} (ns)	$J_{3}(\%)$	$<\tau>(ns)$	χ²
Аро	0.66	9	3.20	56	6.30	36	4.11	0.9
Ca ²⁺	0.64	16	2.99	53	5.65	41	3.63	1.5
Tb ³⁺	0.29	5	2.12	36	4.84	59	3.60	2.0
Eu ³⁺	0.47	8	2.45	44	4.99	48	3.54	1.5
NCS1:ANS								
Аро	0.28	8	4.25	15	16.60	77	13.50	1.8
Ca ²⁺	0.28	7	4.47	17	17.20	76	14.05	1.5
Tb ³⁺	0.28	7	5.08	14	17.45	79	14.80	1.1
Eu ³⁺	0.28	7	5.12	14	17.20	79	14.25	1.1

Table S2: CCS values determined for Tb ³⁺ and	Eu ³⁺ ion adducts at 9+ charge state using
TIMS-MS.	
Th ³⁺ ion	Eu ³⁺ ion

Tb ³⁻	⁺ ion	Eu ³⁺ ion		
$^{\text{TIMS}}\text{CCS}_{N2}$ (Å ²)	Ion specie	$^{\text{TIMS}}\text{CCS}_{N2}(\text{\AA}^2)$	Ion specie	
2182	[M-3Ca+3H] ⁹⁺	2192	[M-3Ca+3H] ⁹⁺	
2204	[M-Tb+6H] ⁹⁺	2201	[M-3Ca-Na+2H]9+	
2189	[M-Ca-Tb+4H]9+	2204	[M-4Ca+H] ⁹⁺	
2201	[M-2Ca-Tb+2H] ⁹⁺	2198	[M-2Na-Eu+4H] ⁹⁺	
2179	[M-3Ca-Tb] ⁹⁺	2179	[M-2Ca-Eu+2H]9+	
2195	[M-2Tb+3H] ⁹⁺	2163	[M-3Ca-Eu] ⁹⁺	



Figure S1: Determination of the stoichiometry of Tb^{3+}/Eu^{3+} bound NCS1.



Figure S2: Far-UV circular dichroism spectra of NCS1 in absence and presence of metal ions. Conditions as in the figure 1.



Figure S3: Frequency domain intensity decay of apo- and metal bound NCS1. Solid and open symbols indicate modulation ratio and phase delay data points, respectively. The solid line corresponds to the fit of the experimental data using a sum of three discreet exponential decay model. Conditions: 10 μ M NCS1, 20 mM Tris buffer, pH 7.40, and 1 mM EDTA or 1 mM Ca²⁺, 40 μ M Tb³⁺, 40 μ M Eu³⁺.



Figure S4: Titration curves of Tb³⁺/Eu³⁺ binding to Ca²⁺NCS1:1,8-ANS. Conditions: 10 μ M NCS1, 20 μ M 1,8-ANS, 1 mM Ca²⁺, 40 μ M Tb³⁺, 32 μ M Eu³⁺ and λ_{exc} =350 nm.



Figure S5: Frequency domain intensity decay of apo- and metal bound NCS1:1,8-ANS complex. Open and solid symbols indicate modulation ratio and phase delay data points, respectively. The solid line corresponds to the fit of the experimental data using a sum of three discreet exponential decay model. Conditions: $10 \mu M$ NCS1, $20 \mu M$ 1,8-ANS, 20 mM Tris buffer, pH 7.40, and 1 mM EDTA or 1 mM Ca²⁺, $40 \mu M$ Tb³⁺, $40 \mu M$ Eu³⁺.



Figure S6: Luminescence decay of Eu^{3+} itself and in presence of NCS1. Conditions: 40 μM NCS1 and 20 μM $Eu^{3+}.$



Figure S7: Steady state anisotropy of isolated D2R peptide and NCS1:D2R complex in the absence or presence of metal ions. Conditions: 10 μ M NCS1, 500 nM D2R, 1 mM EDTA, 1 mM Ca²⁺, 40 μ M Tb³⁺, 40 μ M Eu³⁺ and λ_{exc} =490 nm.



Figure S8: Expanded view of the 9+ charge state with annotated adduct species and the isotopic pattern of the ion $[M\bullet2Na\bulletEu+4H]^{9+}$ shown in the inset (A). Typical broadband nESI-FT-ICR MS spectrum of NCS1 in the presence of $Eu^{3+}(B)$. Deconvoluted spectrum showing the neutral species. The isotopic pattern shown in the inset confirms the theoretical isotopic average mass of the NCS1 protein 22.976 kDa (C).



Figure S9: Comparison of experimental isotopic distributions of NCS1:Tb (red) and NCS1:Eu (blue) adduct species obtained from nESI-FT-ICR MS with their theoretical isotope patterns (black). Note the good agreement between profiles indicated by the dashed lines.



Figure S10: ${}^{\text{TIMS}}_{\text{N2}}$ profiles for the charge state 9+ of NCS1 in the presence of Tb³⁺ (left) and Eu³⁺ (right) obtained from nESI-TIMS-TOF MS.