Supporting Information

The structural evolution of Mo₂C and Mo₂C/SiO₂ under dry reforming of methane conditions: morphology and support effects

Alexey Kurlov,[†]a Dragos Stoian,^b Ali Baghizadeh,^c Evgenia Kountoupi,^a Evgeniya B. Deeva,^a Marc Willinger,^{††c} Paula M. Abdala,^{*a} Alexey Fedorov^{*a} and Christoph R. Müller^{*a}

^a ETH Zürich, Department of Mechanical and Process Engineering, Leonhardstrasse 21, CH 8092 Zürich, Switzerland

^b Swiss-Norwegian Beamlines at the European Synchrotron Radiation Facility, 71 Avenue des Martyrs, Grenoble, France

^c ETH Zürich, Scientific Center for Optical and Electron Microscopy, Auguste-Piccard-Hof 1, CH 8093 Zürich, Switzerland

[†] Present address: Paul Scherrer Institute, Laboratory for Bioenergy and Catalysis, 5232 Villigen PSI, Switzerland

^{††} Present address: Department of Chemistry, Technical University Munich, Lichtenbergstrasse 4, 85748 Garching, Munich

E-mail:

- * abdalap@ethz.ch
- * fedoroal@ethz.ch
- * muelchri@ethz.ch

List of Figures

Figure S1. TEM images of exfoliated <i>d</i> -MoO ₃
Figure S2. XRD pattern of carburized Mo2C-nb and simulated Mo2C reference pattern (ICSD #246146)
Figure S3. Catalytic performance of Mo ₂ C-nb in DRM (laboratory scale reactor, 800 °C, 8 bar, $W/F = 75 - 300 \text{ ms } g_{Mo} \text{ mL}^{-1}$)
Figure S4. XRD pattern of Mo ₂ C-nb after the DRM reaction (as presented in Figure S3) and simulated Mo ₂ C (ICSD #246146) and MoO ₂ (ICSD #24322) reference patterns
Figure S5. Catalytic performance of Mo_2C/SiO_2 in DRM (laboratory scale reactor, 800 °C, 8 bar, $W/F = 5.25 \text{ ms } g_{Mo} \text{ mL}^{-1}$, $CH_4 : CO_2 : N_2 = 4 : 3 : 3$)
Figure S6. Energy edge position of the Mo XANES spectra vs. Mo oxidation state in reference materials
Figure S7. In situ carburization of MoO_3 -nb (left) and MoO_3/SiO_2 (right) under H_2 : $CH_4 = 4 : 1$ gas mixture followed by Mo K-edge XANES
Figure S8. Mo K-edge XANES spectra (top), k^3 -weighted EXAFS (middle) and (uncorrected for the phase shift) Fourier-transformed EXAFS functions (bottom) of reference materials ($\Delta k = 3 - 15 \text{ Å}^{-1}$) 8
Figure S9. In situ k^3 -weighted EXAFS (left) and the corresponding Fourier-transform (right) for MoO ₃ -nb during carburization. ($\Delta k = 3 - 10 \text{ Å}^{-1}$)
Figure S10. Comparison of Fourier-transformed EXAFS functions and EXAFS fittings of α -MoO ₃ , α -MoO ₃ -nb and MoO ₃ /SiO ₂ . ($\Delta k = 3 - 10 \text{ Å}^{-1}$)
Figure S11. In situ k ³ -weighted EXAFS (left) and the corresponding Fourier-transform (right) for MoO_3/SiO_2 during its carburization. ($\Delta k = 3 - 10 \text{ Å}^{-1}$)
Figure S12. Comparison of individual Mo K-edge spectra of Mo ₂ C-nb (left) and Mo ₂ C/SiO ₂ (right) at different stages in the DRM reaction. 12
Figure S13. Comparison of individual Mo K-edge spectra of Mo_2C/SiO_2 after TOS = 0 and 120 min, with a subsequent exposure to a CO_2 flow, and then to a CH_4 flow
Figure S14. PCA analysis of MoO ₃ -nb
Figure S15. Theoretical XANES spectra of the components obtained by MCR-ALS analysis during the carburization of α -MoO ₃ -nb
Figure S16. PCA analysis of MoO ₃ /SiO ₂
Figure S17. Theoretical XANES spectra of the components obtained by MCR-ALS analysis during the carburization of MoO ₃ /SiO ₂
Figure S18. SEM images of Mo ₂ C-nb after the DRM reaction
Figure S19. TEM images of a Mo ₂ C-nb cross-section after the DRM reaction revealing a heterogeneous porous structure
Figure S20. TEM images of a Mo ₂ C-nb cross-section after the DRM reaction revealing graphitic and/or carbon covering the surface of the nanoparticles. The arrows point at the graphitic layers around the used Mo ₂ C-nb



Figure S1. TEM images of exfoliated *d*-MoO₃.

We note that TEM images of exfoliated *d*-MoO₃ have been reported by us previously, see Kurlov et al., *Nanoscale*, **2020**, *12*, 13086–13094, and are reproduced here for completeness and comparison to TEM characterization of MoO₃/SiO₂.



Figure S2. XRD pattern of carburized Mo₂C-nb and simulated Mo₂C reference pattern (ICSD #246146).

Average crystallite size of Mo₂C_nb (determined using the full width at half maximum, FWHM, of the reflection centered at 34.4 in 2Θ) = 26 nm.



Figure S3. Catalytic performance of Mo₂C-nb in DRM (laboratory scale reactor, 800 °C, 8 bar, $W/F = 75 - 300 \text{ ms } g_{Mo} \text{ mL}^{-1}$).



Figure S4. XRD pattern of Mo₂C-nb after the DRM reaction (as presented in Figure S3) and simulated Mo₂C (ICSD #246146) and MoO₂ (ICSD #24322) reference patterns.



Figure S5. Catalytic performance of Mo_2C/SiO_2 in DRM (laboratory scale reactor, 800 °C, 8 bar, $W/F = 5.25 \text{ ms } g_{Mo} \text{ mL}^{-1}$, $CH_4 : CO_2 : N_2 = 4 : 3 : 3$).



Figure S6. Energy edge position of the Mo XANES spectra vs. Mo oxidation state in reference materials.



Figure S7. In situ carburization of MoO₃-nb (left) and MoO₃/SiO₂ (right) under $H_2 : CH_4 = 4 : 1$ gas mixture followed by Mo K-edge XANES.



Figure S8. Mo K-edge XANES spectra (top), k^3 -weighted EXAFS (middle) and (uncorrected for the phase shift) Fourier-transformed EXAFS functions (bottom) of reference materials ($\Delta k = 3 - 15 \text{ Å}^{-1}$).



Figure S9. *In situ* k³ -weighted EXAFS (left) and the corresponding Fourier-transform (right) for MoO₃-nb during carburization. ($\Delta k = 3 - 10 \text{ Å}^{-1}$)

At ca. 600 °C Mo-C and Mo-Mo peaks at 1-2 Å and 2-3 Å appear in the (non-phase corrected) EXAFS Fouriertransform data, in line with the formation of a Mo_2C phase. Due to the high thermal disorder in the EXAFS signal no quantitative information could be extracted, yet it can be observed that the changes in the EXAFS data qualitatively agree with the XANES analysis.



Figure S10. Comparison of Fourier-transformed EXAFS functions and EXAFS fittings of α -MoO₃, α -MoO₃-nb and MoO₃/SiO₂. ($\Delta k = 3 - 10 \text{ Å}^{-1}$).

Table S1. EXAFS fitting of reference MoO₃, α-MoO₃-nb and MoO₃/SiO₂.

Description of the fitting procedure. EXAFS fittings were performed in R-space with $\Delta k = 3 - 10 \text{ Å}^{-1}$ and $\Delta R = 1 - 4 \text{ Å}$. The amplitude reduction factor S₀ was determined to be 1 from the fitting of the α -MoO₃ reference.

The fittings for the α -MoO₃ reference and α -MoO₃-nb were performed using a local structure model based on the crystallographic structure of α -MoO₃, with fixed coordination numbers (CN), free interatomic distances R, while the Debye–Waller factors i.e., the variance of the distance distribution (σ^2), were determined and restrained to be the same for all the Mo-O and Mo-Mo paths. This model fitting used 9 variable of 13 independent points.

For MoO₃/SiO₂, the α -MoO₃ model was not applicable, and the best fit was obtained when using two Mo-O spheres, one Mo-Si and one Mo-M sphere, with varying CN. We also tested a local structure model for MoO₃/SiO₂ without including the Mo-Si path, yet the agreement factor (R-factor) was above 0.2, demonstrating that a Mo-Si path was needed to account for the local structure of this material. This model fitting used 9 variable of 13 independent points.

Reference α-MoO3 (R-factor 0.005)				
Shell	CN	R (Å)	σ^2 (Å ²)	ΔE (eV)
$Mo - O_1$	2	1.71 (1)	0.003 (1)	
$Mo - O_2$	2	1.97 (1)	0.003 (1)	5 (1)
$Mo - O_3$	1	2.31 (1)	0.003 (1)	
Mo — Mo ₁	2	3.45 (1)	0.003 (1)	
$Mo - Mo_2$	2	3.83 (3)	0.003 (1)	
Mo — Mo ₃	2	3.98(1)	0.003 (1)	

α-MoO3-nb (R-factor 0.017)				
Shell	CN	R (Å)	σ^2 (Å ²)	ΔE (eV)
$Mo - O_1$	2	1.71 (1)	0.004 (1)	
$Mo - O_2$	2	1.96 (1)	0.004 (1)	
$Mo - O_3$	1	2.29 (2)	0.004 (1)	2(1)
$Mo - Mo_1$	2	3.44 (1)	0.004 (1)	5(1)
$Mo - Mo_2$	2	3.70 (3)	0.004 (1)	
Mo — Mo ₃	2	3.96(1)	0.004 (1)	

MoO ₃ /SiO ₂ (R-factor 0.015)				
Shell	CN	R (Å)	σ^2 (Å ²)	ΔE (eV)
$Mo - O_1$	2(1)	1.75 (2)	0.012 (5)	
$Mo - O_2$	3 (1)	1.98 (2)	0.012 (5)	7 (1)
$Mo - Si_1$	0.6 (4)	2.7 (5)	0.010 (4)	/ (1)
$Mo - Mo_1$	3 (2)	3.41 (1)	0.010 (4)	

MoO ₃ /SiO ₂ without Mo-Si (R-factor 0.027)				
Shell	CN	R (Å)	σ^2 (Å ²)	ΔE (eV)
$Mo - O_1$	2(1)	1.75 (2)	0.010 (5)	
$Mo - O_2$	2(1)	1.98 (2)	0.010 (5)	6(1)
$Mo - Mo_1$	3 (2)	3.41 (1)	0.011(4)	



Figure S11. In situ k³ -weighted EXAFS (left) and the corresponding Fourier-transform (right) for MoO₃/SiO₂ during its carburization. ($\Delta k = 3 - 10 \text{ Å}^{-1}$)



Figure S12. Comparison of individual Mo K-edge spectra of Mo₂C-nb (left) and Mo₂C/SiO₂ (right) at different stages in the DRM reaction.



Figure S13. Comparison of individual Mo K-edge spectra of Mo_2C/SiO_2 after TOS = 0 and 120 min, with a subsequent exposure to a CO_2 flow, and then to a CH_4 flow.



Figure S14. PCA analysis of MoO₃-nb.



Figure S15. Theoretical XANES spectra of the components obtained by MCR-ALS analysis during the carburization of α -MoO₃-nb.



Figure S16. PCA analysis of MoO₃/SiO₂.



Figure S17. Theoretical XANES spectra of the components obtained by MCR-ALS analysis during the carburization of MoO₃/SiO₂.



Figure S18. SEM images of Mo₂C-nb after the DRM reaction.



Figure S19. TEM images of a Mo₂C-nb cross-section after the DRM reaction revealing a heterogeneous porous structure.



Figure S20. TEM images of a Mo₂C-nb cross-section after the DRM reaction revealing graphitic and/or carbon covering the surface of the nanoparticles. The arrows point at the graphitic layers around the used Mo₂C-nb.