

## Supporting Information for

# Oxygen Vacancies Regulated Selective Hydrogenation of $\alpha,\beta$ -unsaturated Aldehydes over LDH Surface Group Coordinated Transition Metal Photocatalysts

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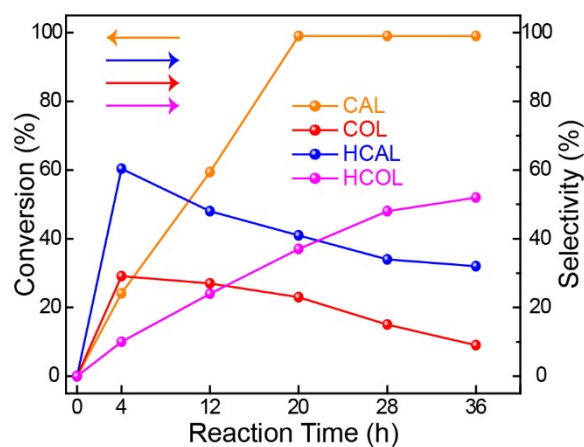
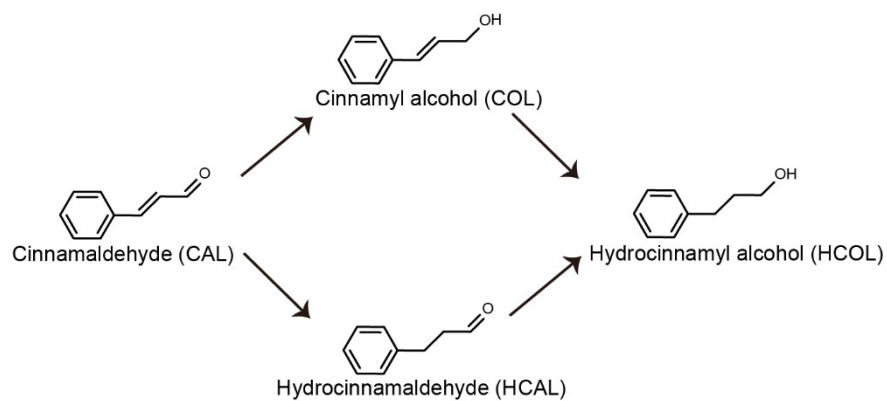
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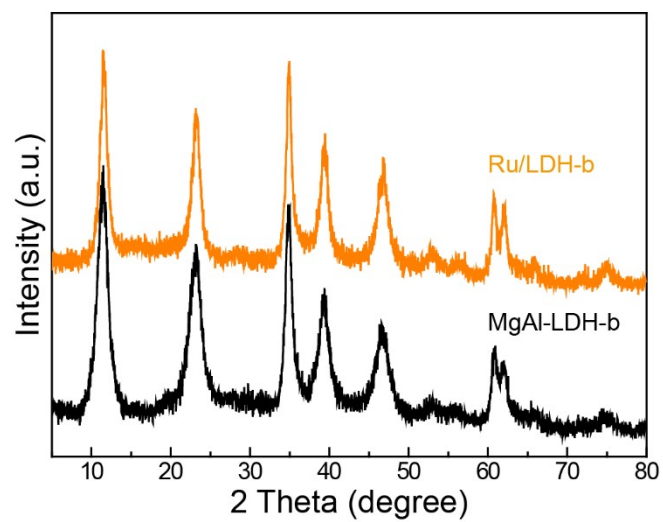
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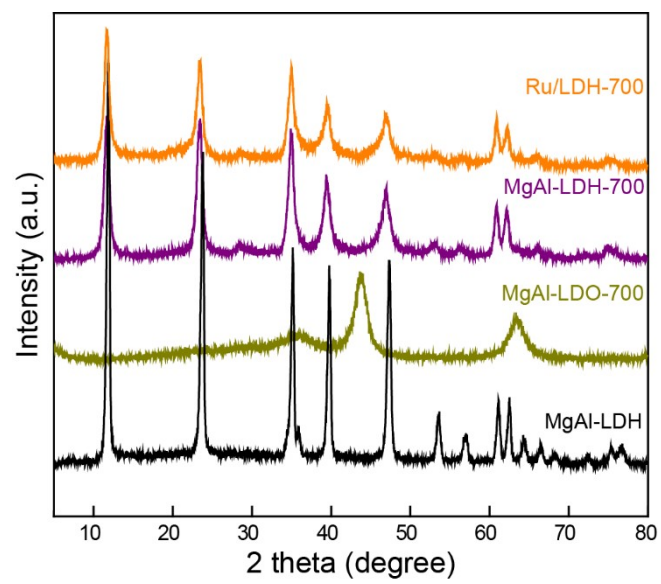
‡ *These authors contributed equally to this work.*



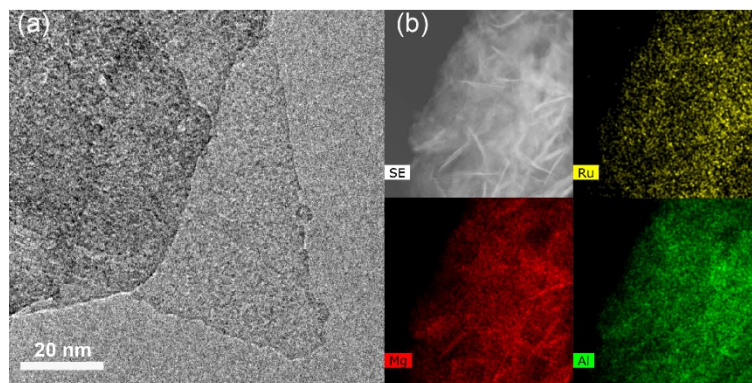
**Fig. S1** Time-conversion plots for cinnamaldehyde reduction under visible light irradiation catalyzed by Ru/LDH catalyst.



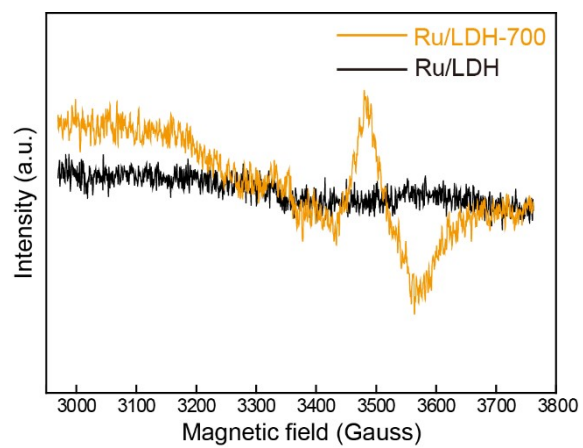
**Fig. S2** XRD patterns of LDH-b and Ru/LDH-b.



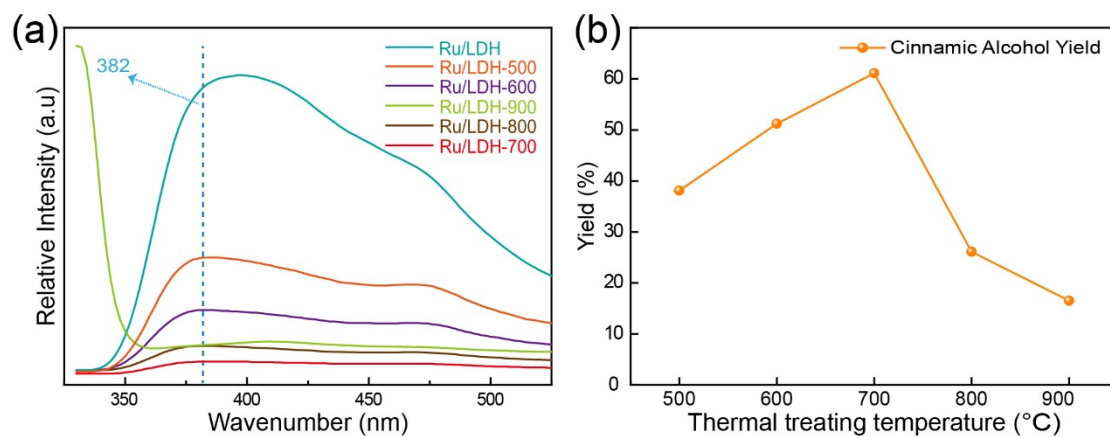
**Fig. S3** The XRD patterns of various samples.



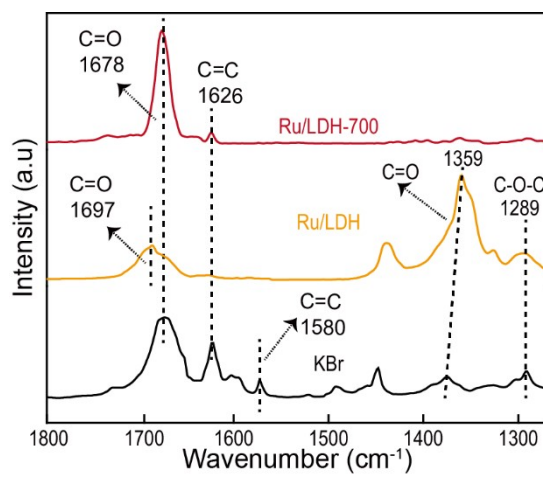
**Fig. S4** (a) TEM image and (b) EDX mappings of Ru/LDH-700 catalyst.



**Fig. S5** EPR patterns of Ru/LDH and Ru/LDH-700 catalysts.

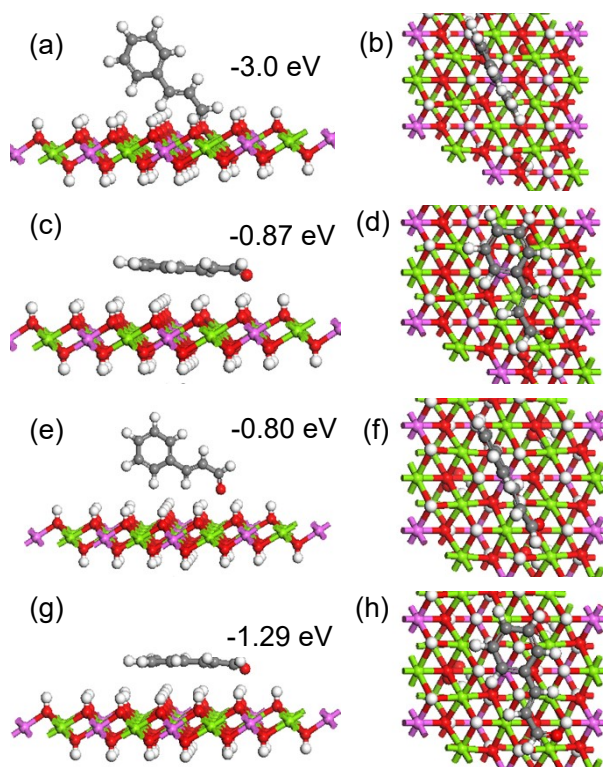


**Fig. S6** (a) The PL spectra and (b) Effect of calcination temperatures on the yield of cinnamyl alcohol.

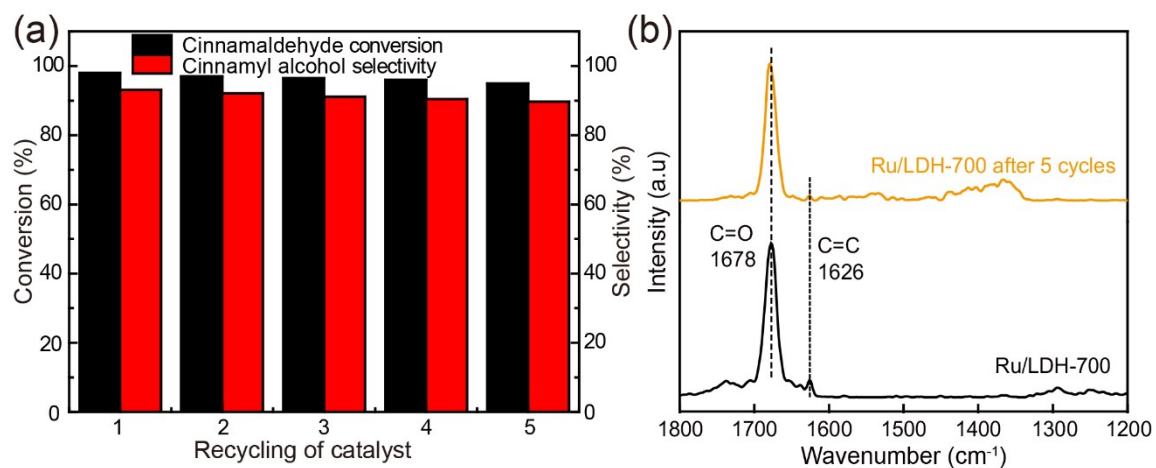


**Fig. S7** DRIFTS of cinnamaldehyde adsorbed on KBr, Ru/LDH and Ru/LDH-700.

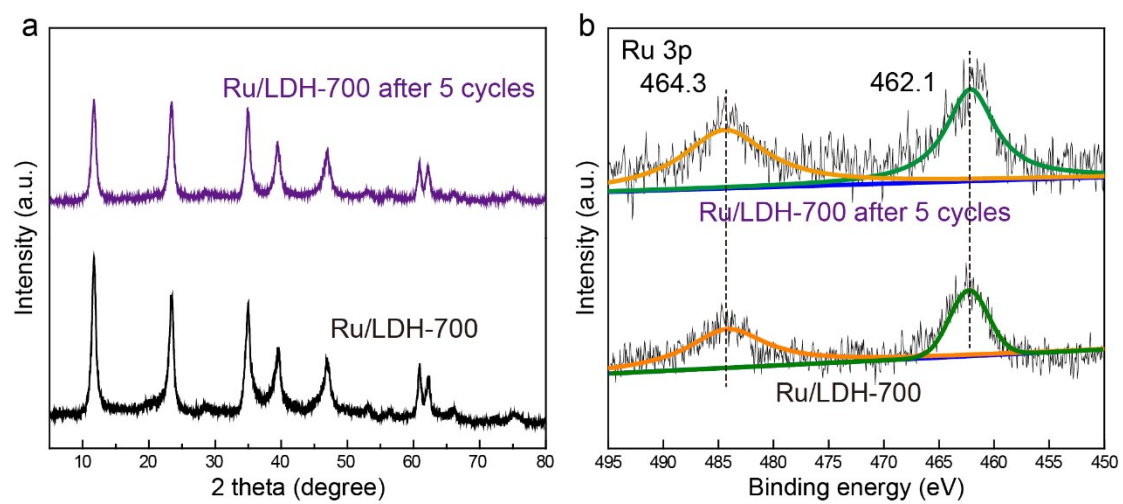




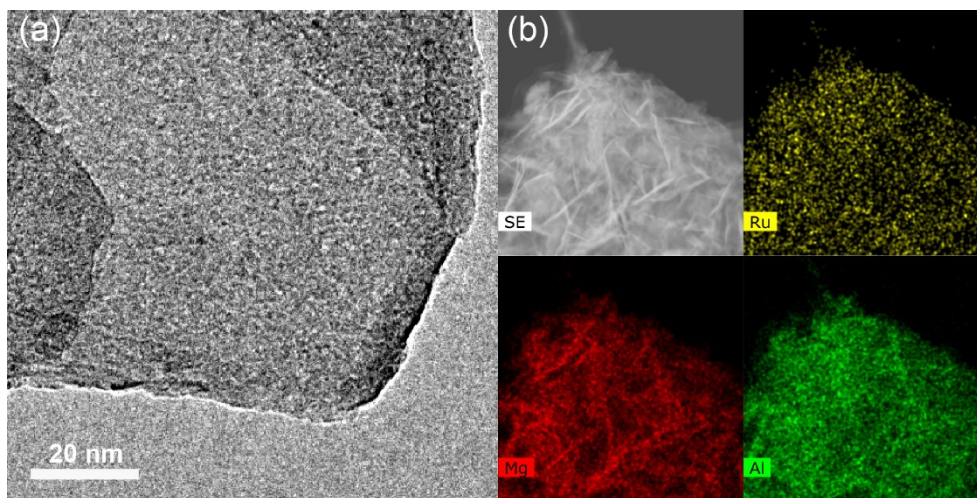
**Fig. S8** The optimized structure for adsorption of cinnamaldehyde molecule using  $\text{C}=\text{O}$  (a, b) or  $\text{-C}=\text{C}$  group (c, d) on MgAl-LDH-700. The optimized structure for adsorption of cinnamaldehyde molecule using  $\text{C}=\text{O}$  (e, f) or  $\text{-C}=\text{C}$  group (g, h) on MgAl-LDH catalyst. Atom coloring: Mg, purple; Al, green; O, red; H, white; C, black.



**Fig. S9** (a) Recyclability in selective hydrogenation of cinnamaldehyde over Ru/LDH-700 catalyst and (b) DRIFTS of cinnamaldehyde adsorbed on fresh and used Ru/LDH-700 photocatalyst.



**Fig. S10** (a) XRD patterns and (b) XPS spectra of the fresh and used Ru/LDH-700 photocatalyst.



**Fig. S11** TEM image (a) and EDX mappings (b) of Ru/LDH-700 photocatalyst used for five cycles.

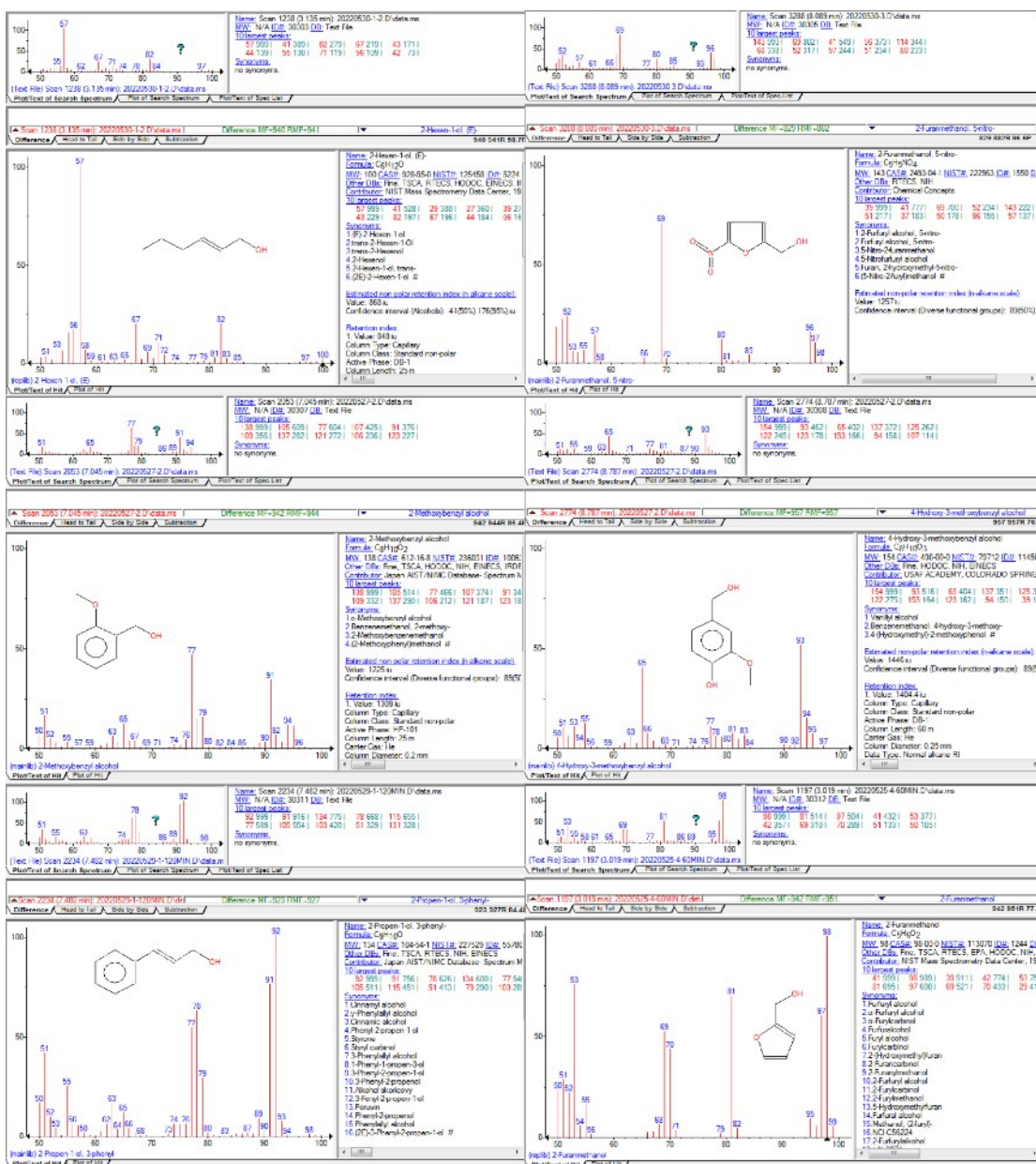
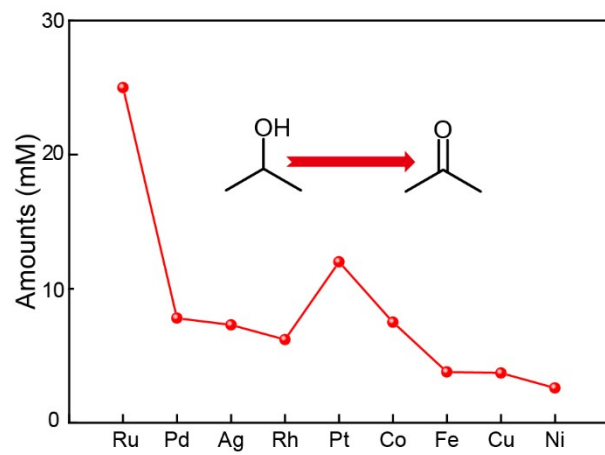
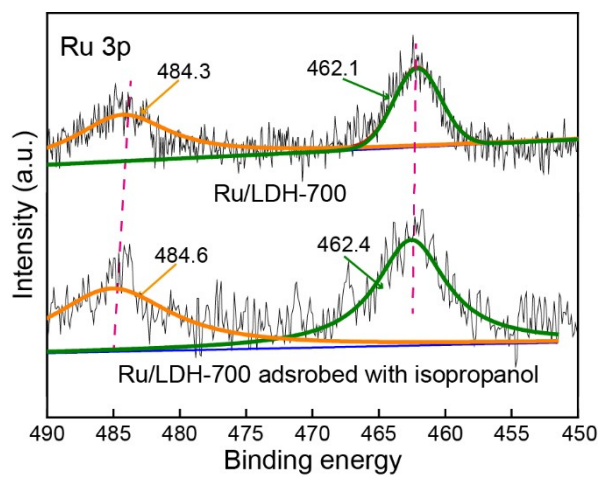


Fig. S12 Mass spectrometry data of various reaction products.



**Fig. S13** Comparison of isopropanol activation ability of various catalysts.



**Fig. S14** Ru 3p XPS spectra of Ru/LDH-700 and Ru/LDH-700 after isopropyl alcohol adsorption.

**Table S1** Effect of Ru loadings on the cinnamaldehyde hydrogenation<sup>a</sup>

Entry	Sample	Ru loading (wt%) <sup>b</sup>	Incident light	Conv. (%)	Sel. (%)
1	LDH-700	0	Visible	2.4	91.6
2	LDH-700	0	Dark	1.2	92.8
3	1.9 wt% Ru/LDH-700	1.9	Visible	4.7	91.4
4	1.9 wt% Ru/LDH-700	1.9	Dark	1.6	92.1
5	3.2 wt% Ru/LDH-700	3.2	Visible	43.7	91.9
6	3.2 wt% Ru/LDH-700	3.2	Dark	17.6	92.7
7	4.3 wt% Ru/LDH-700	4.3	Visible	67.1	91.6
8	4.3 wt% Ru/LDH-700	4.3	Dark	24.3	93.4
9	6.2 wt% Ru/LDH-700	6.2	Visible	71.5	90.7
10	6.2 wt% Ru/LDH-700	6.2	Dark	36.4	92.1

<sup>a</sup> Reaction conditions: 0.1 mmol reactant, 2 mL isopropyl alcohol, 80 °C, 20 mg catalyst, 4 h, Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>), 1 atm Ar.

<sup>b</sup> Determined by inductively coupled plasma (ICP) analysis.



**Table S2** Effect of catalyst usage on the cinnamaldehyde hydrogenation

Entry	catalyst amount (mg)	Incident light	Conversion (%)	Selectivity (%)
1	10	Visible	4.5	91.5
2	10	Dark	1.3	92.4
3	15	Visible	39.7	93.4
4	15	Dark	12.6	92.7
5	20	Visible	67.1	91.6
6	20	Dark	24.3	93.4
7	30	Visible	88.4	91.4
8	30	Dark	42.7	93.1

Reaction conditions: catalyst 4.3 wt% Ru/LDH-700, cinnamaldehyde 0.1 mmol, isopropyl alcohol 2 mL, 80 °C, 4 h, Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>), 1 atm Ar.

**Table S3** Effect of cinnamaldehyde concentration

Entry	Cinnamaldehyde concentration (mol L <sup>-1</sup> )	Incident light	Conversion (%)	Selectivity (%)
1	0.025	Visible	99.9	93.1
2	0.025	Dark	66.1	93.9
3	0.05	Visible	67.1	91.6
4	0.05	Dark	24.3	93.4
5	0.75	Visible	52.5	91.1
6	0.75	Dark	20.8	92.5
7	0.1	Visible	39.1	90.7
8	0.1	Dark	17.4	91.4

Reaction conditions: 20 mg 4.3 wt% Ru/LDH-700, 2 mL isopropyl alcohol, 80 °C, 4 h,

Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>), 1 atm Ar.

**Table S4** Influence of reaction temperature

Entry	T (°C)	Incident light	Conv. (%)	Sel. (%)
1	60	Visible	7.2	94.2
2	60	Dark	1.8	93.5
3	80	Visible	38.4	91.9
4	80	Dark	13.1	93.6
5	100	Visible	97.5	91.1
6	100	Dark	51.8	92.5

Reaction conditions: 0.1 mmol reactant, 2 mL isopropyl alcohol, 20 mg 4.3 wt%

Ru/LDH-700, 2 h, Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>), 1 atm Ar.

**Table S5** Textural properties of various samples.

Entry	Catalyst	BET surface area (m <sup>2</sup> g <sup>-1</sup> )	Pore volume (cm <sup>3</sup> g <sup>-1</sup> )	Pore size (nm)
1	MgAl-LDH	23.86	0.071	12.56
2	Ru/LDH	21.37	0.073	10.67
3	MgAl-LDH-700	91.51	0.50	20.77
4	Ru/LDH-700	84.77	0.43	19.17
5	MgAl-LDH-b	106.31	0.58	25.34
6	Ru/LDH-b	97.46	0.51	23.97

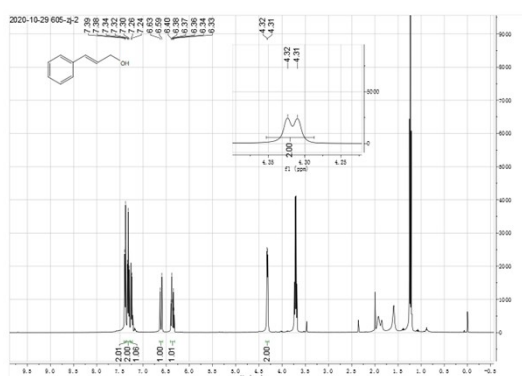
**Table S6** Catalytic performance of Ru/LDH-b for cinnamaldehyde reduction

Entry	Catalyst	Incident light	Conv. (%)	Sel. (%)
1	4.3 wt% Ru/LDH-b	Visible	17.2	28.7
2	4.0 wt% Ru/LDH-b	Dark	8.1	31.4

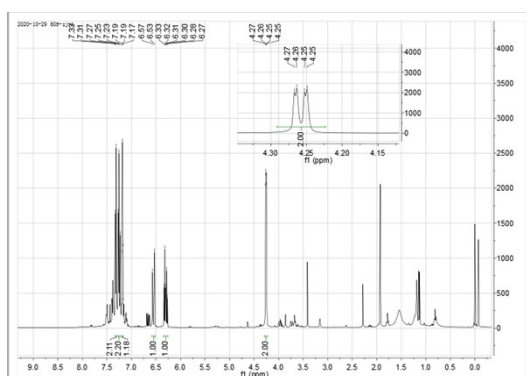
Reaction conditions: 0.1 mmol reactant, 2 mL isopropyl alcohol, 80 °C, 20 mg catalyst, 4 h, Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>), 1 atm Ar.

## Isotope labeling experiments

Reaction conditions: cinnamaldehyde 0.1 mmol, Ru/LDH-700 20 mg, 2-propanol-OD 2 mL, Ar 1 atm, 80 °C, 6 h, Xe lamp irradiation (420–800 nm, 500 mW cm<sup>-2</sup>). The product was determined by <sup>1</sup>H NMR and <sup>13</sup>C NMR. The position of deuterium was confirmed by comparing the <sup>1</sup>H NMR, and <sup>13</sup>C NMR of cinnamic alcohol standard sample and the product.



**Product of isotope labeling experiments:** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, TMS): δ 7.38 (d, J = 4.0 Hz, 2H), 7.32 (t, J = 16.0 Hz, 2H), 7.25 (d, J = 24.0 Hz, 1H), 6.61 (d, J = 16.0 Hz, 1H), 6.36 (dt, J = 16.0 & 8 Hz, 1H), 4.31 (d, J = 4.0 Hz, 2H).



**Cinnamic alcohol standard sample:** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, TMS): δ 7.32 (d, J = 8.0 Hz, 2H), 7.25 (t, J = 8.0 Hz, 2H), 7.18 (d, J = 8.0 Hz, 1H), 6.55 (d, J = 16.0 Hz, 1H), 6.36 (dt, J = 16.0 & 4 Hz, 1H), 4.26 (dd, J = 8.0 & 4.0 Hz, 2H).