Supporting information

Construction and Engineering of Interfacial Structure in Cu_x/FeMgO_y Catalyst for Photoreduction of CO₂ to Ethylene

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Figure S1. Production of (A) CO, (B) CH_4 on Cu_x/Fe_1MgO_y , $Cu/MgAlO_y$ and $Fe/MgAlO_y$ catalysts



Figure S2. Evolution rate of CO and CH_4 at 2h over Cu_x/Fe_1MgO_y , $Cu/MgAlO_y$ and $Fe/MgAlO_y$ catalysts



Figure S3. Evolution rate of CO and CH_4 at 3h over Cu_x/Fe_1MgO_y , $Cu/MgAlO_y$ and $Fe/MgAlO_y$ catalysts



Figure S4. Evolution rate of CH₄ at 4h over Cu_x/Fe₁MgO_y, Cu/MgAlO_y and Fe/MgAlO_y catalysts

| Catalysts | catalyst mass • (mg) | Product evolution rate (μ mol g ⁻¹ h ⁻¹) | | |
|---|-------------------------|--|-----------------|----------|
| | | СО | CH ₄ | C_2H_4 |
| Cu ₁ /Fe ₁ MgO _y | 1 | 40.1 | 0.9 | 9.9 |
| 6%CuO/WO ₃ ¹ | 50 | 1.6 | - | - |
| 3%Pt/p-C ₃ N ₄ ² | 100 | 4.7 | 4.0 | - |
| $Cu^{\delta +}$ /CeO ₂ -TiO ₂ ³ | 10 | 3.5 | 1.6 | 4.5 |
| Cu ₂ O@Cu@UiO- 66-NH ₂ ⁴ | 3 | 20.9 | 8.3 | - |
| Ag/TiO ₂ ⁵ | 10 | 5.2 | - | - |
| a-Fe ₂ O ₃ /BCN ⁶ | 4 | 11.0 | 0.7 | - |
| $\alpha\text{-}Fe_2O_3/Cu_2O^7$ | 100 | 3.1 | 0.6 | - |
| Cu@Cu ₂ O/N-GCs ⁸ | 50 | 4.0 | 5.6 | - |
| Ni-SA-x/ZrO ₂ ⁹ | 10 | 11.8 | - | - |
| Ru-Bi ₂ MoO ₆ ¹⁰ | 30 | 23.8 | - | - |
| ZnSe– CsSnCl ₃ ¹¹ | 5 | 54.0 | 3.0 | - |
| $TiO_2@Cu^{12}$ | 10 | 25.4 | 5.8 | - |
| $SnTa_2O_6^{13}$ | 20 | 31 | - | - |
| Mn:CsPb(Cl/Br)314 | 10 | 64 | - | - |
| Pt SA/ZrO ₂ ¹⁵ | 20 | 16.6 | 0.4 | - |
| BiOIO ₃ nanostrips ¹⁶ | 20 | 17.33 | - | - |
| $Au_{SA}Cd_{1-x}S^{17} \\$ | 30 | 32.2 | 11.3 | - |
| $CeO_2 @CdS-Cu_xS_y{}^{18}$ | 5 | - | 18.2 | - |
| Cu/Fe-CNC ¹⁹ | 20 | 6.66 | - | - |

 Table S1. Performance comparison of Cu₁/Fe₁MgO_y and some reported photocatalysts for CO₂

 reduction



Figure S5. Recycling experiments of Cu₁/Fe₁MgO_y photocatalyst for three cycles



Figure S6. In situ FTIR spectra of humid CO₂ adsorbed on (A) Cu₄/Fe₁MgO_y and (B) Fe/MgAlO_y

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